05.1-33 LOW-TEMPERATURE PHASE TRANSITION IN LICSSO,.

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The alkali metal lithium sulphates, MeLiSO,, con-

stitute an important group of materials with interesting ferroelectric and ferroelastic properties. Orthorhombic ${\rm LiCsSO}_4$ has been observed to undergo a structural pure

ferroelastic, typical second-order phase transition at 203 K. The ferroelastic behaviour similar to that observed in ${\rm LiNH}_{\rm L}{\rm SO}_{\rm L}$ and ${\rm LiKSO}_{\rm L}$ has been investigated in

same detail. The monoclinic angle in the low-temperature phase has been interpreted as an order parameter. Its temperature variation in the range of two degrees below T_c agrees very well with predictions of Landau theory.

Lattice parameters have been determined as a function of the temperature in the range from 160 \mbox{up} to 550 K by using the Bond single-crystal diffractometer and copper radiation. The monoclinic strain has been measured by applying an attachement consisting of an additional silicon crystal which has converted our diffractometer into a two-crystal spectrometer. On basis of obtained results the thermal expansion and strain tensor have been calculated.

Optical inspection by means of a polarizing microscope with a special attachement applying a cold nitrogen flow indicates that in the monoclinic phase below 203 K the optical indicatrix rotates around the c-axis with decreasing temperature. The ferroelastic domains are clearly observable and their pattern changes with a lowering temperature in a very characteristic way contrary to that observed in ${\rm LiNH}_4{\rm SO}_4$ and ${\rm LiKSO}_4.$

05.1-34 THE PHASE TRANSITION AND THE DOMAIN STRUCTU-RE FOR MIXED CRYSTALS Cs_Rb1-xLiS04. By A. Pietraszko, Institute for Low Temperature and Structure Research, 50-950 Wroclaw, Poland.

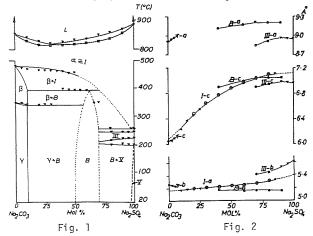
X-ray studies of ${\rm CsLiSO}_L$ - ${\rm RbLiSO}_L$ phase system have revealed successive structural phase transformations with ferroelastic and incommensurate phases. In our previous paper we presented the X-ray structure analysis of the RbLiSO $_4$, Cs $_{0.5}$ Rb $_{0.5}$ LiSO $_4$ and CsLiSO $_4$ crystals. New results has been obtained using X-ray analysis and optical method at temperature range from 140 K to 600 K. It was found that for concentration of Cs ions above x=0.4, the intermediate monoclinic phase B existed at low temperature. For the phase B the ferroelastic structure along b axis was investigated. The observation of the monoclinic phase II under polarizing microscope showed ferroelastic domain structure along c axis below the transition point which at low temperature gradualy disappeared. For small Cs concentration an additional intermediate ferroelectric - ferroelastic phases A and $B^{\,a}$ did appear. The domain structure of these phases is of typical mosaic pattern. Basing on the crystal structure analysis and the temperature dependence of domain structure the phase transitions for $Cs_x Rb_{1-x} LiSO_4$ mixed crystal have been discussed. The detected structural transformations are summarized in table 1.

	4	77 K	475 K		45	58 K	438 K	
Rbliso ₄	l Pcmn abc	inc	i comm. icβ	B P2 ₁ /c ab2cβ			Α ວmm. cβγ	ΙΙ P2 ₁ / abcγ
Cs _{.25} Rb.75 I LiSO ₄ Pcm ab				1/c		B ^I 377 K ? P 52cβ a		11 2 ₁ /n σcγ
Cs.5 ^{Rb} .5 Liso ₄			3	43 K B P2 ₁ / ab2c	/c cβ	2	35 Κ , Β ^ι , ? ab2cβ	
^{Cs} .75 ^{Rb} .2 LIS0 ₄	25	l Pcmn abc		255 K		5 K	B P2 ₁ /c ab2cβ	
CsLiSO ₄		l Pcmn abc		203 K			II P2 ₁ /n abcγ	

Table 1

05.1-35 UNLIMITED SOLUBILITY BETWEEN a-Na2CO3 AND Na, SO, I STRUCTURES. By B.N. Mehrotra*, Institut für Kristallographie, Technische Hochschule Aachen, W. Germany. The phase diagram of sodium carbonate and sodium sulphate has been investigated using X-ray and thermal methods, figure 1. The high temperature modifications α -Na₂CO₃, and Na₂SO₄I show complete solubility. Among low-temperature modifications Na2SO4V, Fddd, does not form any solid solution, but y-Na₂CO₃, C 2/m, form limited solid solution. A new compound Burkeite, B, is also formed at low temperature.

Unlimited solubility of high modifications is also con-firmed by fusing the two compounds, and quenching at room temperature. The shortening of the c-axis is the evidence that CO_3 groups are oriented perpendicular to III and Burkeite, B, after some time, figure 2.



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Single crystals of Na₂SO₄III are grown by metastable nucleation from aqueous solutions at 50° C. This phase is orthorhombic with space group Cmcm and cell constants a = $5.607A^{\circ}$, b = $8.96A^{\circ}$, c = $6.967A^{\circ}$, Z = 4. The structure is refined to R = 4.5%. The structure consists of tetrahedral and octahedral framework. The thermal vibrations of most of the atoms are large along c-axis (Zeit. Krist. - in Press). Single crystals of this phase III (fig. 3a: Weissenberg Photograph O-layer) on heating upto phase I and cooling, show oriented domains of phase III and phase II in the same crystals (fig. 3b: Weissenberg photograph O-layer).

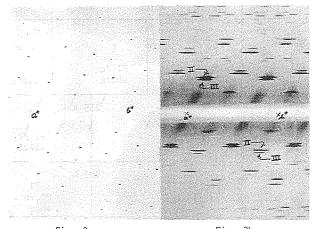


Fig. 3a Fig. 3b This work was supported by Deutsche Forschungsgemeinschaft.

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05.1-36 ON THE LOW TEMPERATURE PHASE TRANSITION IN (NH4)_2SO4. By Y.A. Badr, G. Said, M. Mikhail and S. Awad, Faculty of Science, Cairo University, Cairo, Egypt.

An ultracryostat of type KT"90S" and multidecameter (NF-Dekameter type WTW) provided with a special cell were used to determine the temperature dependence of the dielectric constant and dielectric loss over a wide range of frequencies of polycrystalline ammonium sulphate in the region of the low temperature phase transition. A sharp increase was observed in the values of ε' & ε'' at about -50°C. Besides, a dielectric dispersion was detected and it was found to be more pronounced in the higher temperature phase of (NH₄)₂SO₄. This dispersion was attributed to piezoelectric resonance (Badr, et al, Physica status solidi (a)53,K51(1979)).

The observed sudden increase in the values of the dielectric constant and dielectric loss below -50° C was inferred to the ferroelectric behaviour of phase II of $(NH_4)_2SO_4$. DTA studies showed that also a sharp peak at -50° C which indicated that the phase transition occurring in ammonium sulphate at -50° C was of the first order type. TMA thermograms showed that this transition was associated with a rapid increase in the expansion coefficient. Such increase in the lattice parameter might be attributed to the enhanced rotation of electric dipoles associated with the deformed $SO_{\overline{4}}^{\pm}$ ions. The deformation of the $SO_{\overline{4}}^{\pm}$ ions, in addition to their orientational motion, are thought to be responsible for the ferroelectric behaviour of $(NH_4)_2SO_4$ below -50° C. 05.1-37 CRYSTAL STRUCTURE CHARGES AND PRIN-CIFAL DEFECTS IN IRRADIATED CRYSTALS.By <u>E.V.</u> <u>Kolontsova</u>, A.E.Korneyev, V.F.Lutsenko.Moscow State University, Karpov Institute of Physics and Chemistry, Moscow, USSR.

Under the influence of radiation the crystals of Ca_2SiO_4 , $CaAlO_3$, $BaTiO_3$, $PbZrO_3$, ZrO_2 ,

 KIDO_3 , CsMO_3 and SiO_2 (quartz, cristobalite)

transform into the structural states, which are close to the high-temperature phases of unirradiated crystals. The investigation of radiation defects and comparison of crystal structure changes under the influence of irradiation and heating are important for the determination of such transformation mechanism.

By using X-ray diffuse scattering and X-ray crystal structure analysis it was shown that in neutron-irradiated quartz (E≥0.5 Mev;

$$= 6 \times 10^{19} \text{ncm}^{-2}; T_{\text{irrad}} 100^{\circ} \text{C}) \text{ structural}$$

states intermediate between \measuredangle and β -quartz are formed (E.V.Kolontsova,A.E.Korneyev, V.P. Lutsenko.Dokl.Akad.Kauk SSSR (1979),<u>247</u>,80). The dose dependences of these changes are qualitatively analogous to those in unirradiated crystals at about the \measuredangle - β phase transition temperature.The investigation of diffuse Xray scattering intensity (I_d) dependence of q (q-distance between reciprocal space point and the next nearest reciprocal lattice node)

showed,that $I_{\rm d}$ falls off like $q^{-2}.~I_{\rm d} \sim q^{-2}$

means (P.H.Dederichs.J.Phys. $(1973), \underline{F3}, 471$), that isolated point defects are principal defects in neutron-irradiated quartz and therefore they are responsible for the new structural states.

The changes of diffraction pattern from single crystal and powder ${\tt CsNO}_3$ during heating

and X and γ -irradiation (0.1 - 3.0×10⁹R; E= 1.3 Mev; T_{irrad} ≤ 40°C) were also studied.

Analysis of these experimental results allows to make a conclusion, that radiation-induced transformation of crystal structure is realized without forming of displacement spikes.

According to the analysis of our and literature data one can conclude that it is possible to consider isolated point defects concentration as an instability parameter under the influence of radiation for some compounds (such as $\measuredangle -SiO_2$, BaTiO₃, CsNO₃, and maybe

KNb03, PbZr03).