## C-210

09.2-28 CRYSTAL STRUCTURE OF ( $\pm$ )-3a-METHYL-I, 2,3,3a,6,7-HEXAHYDROPENTALENO 1, 6a-a NAPHTHALEN -4-ONE, Cl7H180. B.N.Lahiri, Dept of Physics, University of Burdwan, Golapbag, Burdwan 713104, India, Birendra Nath Das,Dept of Physics, Vivekananda College, Calcutta 700063, and S.Chaudhuri, RSIC, Bose Institutedalcutta 700009.

The title compound has recently been synthesized at IACS,Calcutta by Prof U.R. Ghatak et al. (J.C.S.,Perkin I, in press)through a novel carbenium ion rearrangement arising from an acid-catelyzed intramolecular alkylation of a rigid $\beta, \gamma$-unsaturated diazomethyl ketone. The complete stereostructure of this highy strained tetracyclic unsaturated ketone having an angular methyl group has been established by the present x-ray crystal structure analysis. It may be mentioned that the usual spectroscopic methods, viz. ${ }^{2} \mathrm{H}$ N $N \mathbb{R},{ }^{12} \mathrm{C}$ MMR, $\operatorname{IR}$, UV, mass spectrometry etc. failed to give unambiguously the structure and stereochemistry of this ketone.
The compound erystallizes, from a solution in petroleum ether, in the orthorhombic space group Pbea with 8 molecules in the unit cell of dimensions $a=11.312(1), b=12.834(3), c=18.158(4)$ A. The structure has been solved by direct methods and refined by full matrix least squares with anisotropic thermal parameters for nonhydrogen atoms to a final conventional $R$ index of 0.054 for 657 independent reflexions observed above $2.5 \sigma\left(F_{0}^{2}\right)$ level out of 2235 reflexions in the range of $2 \theta=4^{\circ}-130^{\circ}$ measured on a four-circle diffractometer with graphite monochromatized CuK radiation. All the hydrogen atoms have been located as dominant peaks on a difference eleotron density map and refined isotropically. A perspective view of the molecule is shown.


Benzenoid ring $A, C_{s p^{2}} \mathrm{C}_{\mathrm{sp}} \mathrm{p}^{2}$ bond lengths 1.3601.439 , mean $1.386(9)$ A ; endocyclic angles $117.8_{-}$ 123.0, mean 120.0(8). Cyclopentane ring D, $\mathrm{Cssp}^{3}-\mathrm{Csp}^{3}$ bond lengths 1.526-1.561, mean 1.543(10) A; endocyclic angles 101.3-106.3,mean 103.7(7). Cyclohexene ring $\mathrm{B}_{2} \mathrm{C}_{\mathrm{Sp}}{ }^{2}-\mathrm{C}_{s p^{3}}$ bond lengths 1.463-1.514, mean I. 497(9) A, Cspo Csp $^{3} 1501$ (12) A; endocyclic angles at the sp carbons 117.0-123.0, mean 120.8(7) and at the so carbons 108.7-111.5, mean 110.3(7). Cyclopentenome ring $\mathrm{C}, \mathrm{C} \mathrm{sp}^{2}-\mathrm{Csp}^{2}$ single bond 1. 445 ( 10 A double bond $1.325(10) \mathrm{A}, \mathrm{Csp}^{2}-\mathrm{C} \mathrm{sp}^{3} 1542(8) \AA$, $\mathrm{C}_{\mathrm{Sj} \mathrm{J}^{3} \mathrm{CH} 3} 1.523(9) \mathrm{A}$ and $\mathrm{C}=01.197(8)$ A; endocyclic angles at the $\mathrm{sp}^{2}$ carbons 106.8-112.2,mean $110.3(7)^{\circ}$ and at the $\mathrm{sp}^{3}$ carbons $103.5(6)^{\circ}$ and 103.6(6).

The molecule contains planar A ring. The $B$ ring has unexpected nearly planar conformation. This is probably due to the influence of A ring leading to partial pi electron delocalisation in $B$ ring. It is confirmed by the shortage of $\mathrm{Csp}^{2}-\mathrm{C}_{\mathrm{sp}}{ }^{3}$ 1.463(10) $\AA$ in the ring. The shortest intermolecular contact of 3.494(9)A indicates that in the solid state the molecules are held together only by van der Waals forces.
09.2-29 THE CRYSTAL STRUCTURE OF DRIMENOL. By C. Escobar ańd 0. Wittk.e. חpto. de Fisica, Fac. de Ciencias Fisicas y Matemáticas. Universidad de Chile. Casilla 5487, Santiago, Chile.

This natural product was isolated more than twenty years ago.Its constitution and stereochemistry was determined by chemical and spectroscopical methods (H.H. Appel, C.J.W. Brooks and K.H. Overton, J. Chem. Soc. (1959)3322), but no X-Ray Structure investigation has been published yet. orimenol, $\mathrm{C}_{15} \mathrm{H}_{26} 0$, is monoclinic with $a=12.384(1), b=22.615(2), \quad c=7.462(1) \AA$, $B=93.23(1)^{\circ}, Z=6$, space group $P 2_{1}$.

The structure was solved by direct methods (MULTAN76 program) and refined to an $R=0.058$ for 3177 reflexions with $I>\sigma$ (I) measured with an automatic diffractometer. The least squared refinement included all non-hydrogen atoms with anisotropic thermal parameters and the $H$ atoms of the radical $\mathrm{CH}_{2} \mathrm{OH}$. All other $H$ atoms were theoretically 2 positioned.

The three molecules of the asymmetric unit have the same configuration and, making due allowance for the estimated errors, the same dimensions and angles. The cyclohexane ring (chair) and the cyclohexene ring (half chair) are fused trans. H bonds through the $O H$ groups originate chains of trimers running parallel to the $c$ axis. Between these chains there are only weak van der Naals forces.
09.2-30 CRYSTAL STRUCTURE OF A BRIDGED HYDROXY CYCLOPENTANONE, $\mathrm{C}_{1} 7 \mathrm{H}_{2} \mathrm{O}_{2}$. BY S.C. Kunda Das and B.S. Basak, X-ray Laboratory, Presidency College, Calcutta-700073, India. A facet of diazocarboxyl chemistry that generates a great deal of interest is the acid - catalysed cyclisation of diazomethyl ketones through m-participation of suitably situated olefinic bonds. In this study the title compound was obtained by cyclisation of a rigid polycyclic unsaturated diazomethyl ketone under certain reaction conditions. The fact that a highly rearranged compound was obtained by interesting rearrangement of an initially formed carbo-cation was established and confirmed by X-ray crystal structure determination of the compound. The compound (Fig. 1) has been synthesized by Prof. Uj.R. Ghatak et al. (U.R. Ghatak et al., J. Chem. Soc., Perkin Trans. 1, 1981, in press). The crystals belong to the tetragonal system having space group If with 8 molecules in the unit-cell of dimensions $a=b=16.30$ A.
$c=10.31$ A. Photographic data were collected by Weissenberg camera using equi-inclination technique. Intensities were measured visually for 1392 reflections of which $72.4 \%$ were above the threshold value of observation within the Cuk - sphere. The structure has been solved by direct methods.
Full-matrix least-squares refinement with isotropic thermal parameters brought the R value to 0.149 .

