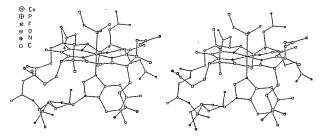
09. STRUCTURES OF ORGANIC, ORGANOMETALLIC AND COORDINATION COMPOUNDS C-235

09.4-36 THE CRYSTAL AND MOLECULAR STRUCTURE OF FLUOROMETHYLPHOSPHITO-P'-COBALAMIN.

By R. Bieganowski, Institute für Physiol. Chemie, Universität Hamburg, Grindallee 117 and <u>J. Kopf</u>, W. Hinrichs, K. von Deuten, Institut für Anorg. und Angew. Chemie, Universität Hamburg, Martin-Luther-King-Platz 6, D-2000 Hamburg 13, West Germany.

Recently we have reported the crystal and molecular structure of dimethylphosphito-P'-cobalamin [R. Bieganowski, W. Friedrich, J. Kopf, K. von Deuten, Symposium ECM-6, Barcelona 1980]. This compound was isolated from the reaction of cyanocobalamin with tri-fluorophosphine in methanol. From the same reaction we isolated a second product, the title compound, which was identified and characterized by means of electrophoresis and NMR [R. Bieganowski, W. Friedrich, Z. Naturforsch. and molecular structure of this compound. Crystals of fluoromethylphosphito-P'-cobalamin were ob-

Crystals of fluoromecnyiphosphilo-r -cobatamin were set tained by slow evaporation of methanol/water. The space group is $P2_12_12_1$ with lattice parameters <u>a</u> = 2570.0(14), <u>b</u> = 2241.4(13) and <u>c</u> = 1583.0(9) pm. The structure was refined anisotropically to R = 0.152 and R = 0.116 for 5078 reflections.



Stereo-plot of fluoromethylphosphito-P'-cobalamin

The title compound is much less soluble in water than cyanocobalamin. In view of the good binding of fluoro-methylphosphito-P'-cobalamin to the intrinsic factor, this compound may gain some biological importance as a depot-B12-preparation [T. Kamikubo, R. Bieganowski, B. Senkpiel, W. Friedrich; in preparation].

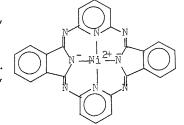
09.4-37 MOLECULAR STRUCTURE OF A FOUR-COORDI-NATED MACROCYCLIC Ni-COMPLEX [Ni(C₂₆H₁₄N₈)].

By S. Peng, <u>Y. Wang</u> and T. Ho, Department of Chemistry, National Taiwan University, Taipei, Taiwan, ROC.

[Ni($C_{26}H_{14}N_8$)] crystallizes in space group P1 with cell parameters: a=3.7637(4), b=15.8254(9), c=16.5865(6) R, α =85.67(1), β =83.50(1), γ =83.16 $(1)^{\circ}$, Z=2. The Ni-atom is coordinated to four nitrogen atoms of the porphyrin-like ligand in a square planar fashion. Unlike the planar metal-prophyrin complexes, the macrocyclic ligand shows a saddle like shape with the Ni atom at the saddle point. However, the atoms of the inner 16-membered ring are more or less in a plane. The dihedral angles pyridine/Ni-4N plane, benzene/Ni-4N plane are about 27 22 respectively. The two Ni-N(pyridine and respectively. The two Ni-N(pyridine) dis-

tances, 2.00 Å are slightly longer than the distances Ni-N(amine), 1.86 Å. The results of the structure determination indicate that the molecular structure is as in the figure with 4 C=N double bonds. The molecular structure

can be compared with a 14-membered macrocyclic iron complex, [Fe($C_{22}H_{22}N_4$)], and a 16-membered TAAB (C₂₈H₂₀N₄) Ni com-plex. (Hawkinson et. al., Inorg. Chem. <u>8</u>, 2402 (1969)).



09.4-38 THE STRUCTURE OF TRIS(S-METHYLDI -THIOCARBAZATE)NICKEL(II) COMPLEXES. By T.J. Malinowsky, V.A. Neverov, C.F. Belyaeva and V.N. Biyushkin, Institute of Applied Physics, Academy of Sciences of Moldavian S.S.R., Kishinev, U.S.S.R.

I. [NiC₆H_{17.5}N₆S₆]Cl_{1.5}·3H₂O, trigonal, R3, a = b₂ = 9.875(4), c = 39.228(5) Å, D_m = 1.64 Mg m⁻³, Z = 6 (Belyaeva et al., Coordinatzi-onnaya Chimya, U.S.S.R., (1981) 7, 433). The geometry of the coordination around the Ni atom is a distorted octahedron. The Ni atom, which lies on the threefold axis, is coordina-ting to the S and N atoms of three bidentate S-methyldithiocarbazate molecules with the formation of five-membered chelate rings. The formation of five-membered chelate rings. The ligand molecule is planar and has the cis-cis conformation. The Ni-S(2.409(3)Å) and Ni-N (2.14(1)Å) distances are typical for Ni octa-hedral S and N complexes. The sulphur and nitrogen atoms are on opposite planes with respect to the nickel atom. The interatomic distances, C-S(1.66(1)Å), C-N(1.34(2)Å) and N-N(1.42(1)Å) are very close to those in the free ligand (Manotti Lanfredi et al., J.Chem. Soc.Dalton Trans.(1977) 5, 417). The molecu-les form layers which are separated by c/3. One of the two Cl atoms is statistically dis-tributed between two equivalent positions on tributed between two equivalent positions on the threefold axis. It was concluded that two Ni complexes contained one deprotonated S methyldithiocarbazate and the structural for-mula might be written down as $[Ni(H_2NNHCSSCH_3)_2(H_2NNCSSCH_3)]$.

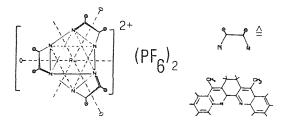
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. [Ni(H₂NNHCSSCH₃)₃] 3Cl \cdot 3H₂O \cdot A similar situation was found in the following compound II. [NiC₆H_{17.5}N₆S₆](ClO₄)_{1.5}, trigonal, P³, a = b = 9.670(4), c = 13.045(5)Å, Z = 2, with three [ClO₄] ions for the two Ni complexes. One of the two [ClO₄]⁻ ions is statistically disordered across a center of inversion at 001/2. Some bond parameters are: Ni-S= 2.386(2), Ni-N= 2.130(7), C-S= 1.672(7), C-N= 1.319(7), N-N= 1.401(9)Å. The two crystals I and II have almost the same structure. III. [NiC₆H₁₈N₆S₆](ClO₄)₂, monoclinic, P2₁/b,

a = 11.361(2), b = 22.323(5), c = 9.045(2)A, %= 101°10(2)', Z = 4. Some lengths are: Ni-S= 2.370-2.393(2), Ni-N= 2.132-2.135(6), C-S= 1.650-1.667(7), C-N= 1.305-1.317(8), N-N= 1.402-1.409(8)A.

The Ni(H₂NNHCSSCH₃) $_{3}^{2+}$ cation in this complex retains the pseudooctahedral NS coordination geometry. The difference between the complexes II and III exists in the molecular packing.

09.4-40 CRYSTAL STRUCTURE OF RU-TRIS (4,7-DIMETHYL-2,3,8,9-DIBENZ-5,6-DIHYDRO-PHENANTHROLINE) DI-HEXAFLUOROPHOSPHATE. By Margareta Zehnder, Institut für Anorg. Chemie der Universität Basel, Switzerland Peter Belser and Alex von Zelewsky, Institut de chimie inorganique de l'Université de Fribourg, Switzerland. The title compound C₆₆^H54^F12^N6^P2^{Ru} crystallizes in the trigonal space group P31c with a=b=14.180, c=17.160 and Z=2. The cation itself having exact D3 symmetry occupies a 32 special position of the space group. The structure was solved by direct methods by first eliminating the large structure factors of the metal ion. Spectroscopic data of the compound are given.



Octahedral complexes containing two 2,2'-bipyridine (bpy) ligands generally are found to have the cis geometry, consistent with the expected instability of the trans configuration due to steric interaction between opposing hydrogen atoms of the bpy ligands. The title cations represent the second and third structures reported of octahedral complexes with trans bpy groups and these, along with four square-planar di-bpy complexes, allow generalizations on the modes of accommodating the steric strain. Three of the complex ions have bpy units which are in a "twist" arrangement whereby the two \preccurlyeq hydrogen atoms of each bpy move to opposite sides of the idealized MN4 plane. The other

complexes display a "bowed" configuration in which the \nsim hydrogen atoms of each bpy are displaced on the same side of the plane, opposite the \nsim hydrogens of the trans bpy. These distortions are readily analyzed and contrasted by use of the bend and twist coordinates developed for the discussion of non-planar peptide linkages. The two title compounds are found to be in the bowed configuration, in contrast to the complex

 $Ru(bpy)_{2}(H_{2}O)(OH)^{+2}$ which has a twist conformation.

09.4-41 TRIBUTYLPHOSPHINE SUBSTITUTED POLYNUCLEAR Ru-CARBONYL CARBOXYLATES. By M.Bianchi & P.Frediani, Istituto di Chimica Organica dell'Università di Firenze, Via G.Capponi 6, Firenze, and <u>M.Nardelli</u> & G.Pelizzi, Istituto di Chimica Generale e Inorganica dell'Università di Parma, Centro di Studio per la Strutturistica Diffrattometrica del C.N.R., Via M.D'Azeglio 85, Parma (Italy).

Two new polynuclear Ru-carbonyl tributylphosphine substituted carboxylates have been synthetized by reaction of $\operatorname{Ru}_{2}(\operatorname{CO})_{12}$ with acetic and glutaric acids respectively and subsequent treatment with tributylphosphine. Their crystal data are:

(I) $[\operatorname{Ru}_2(\operatorname{CO})_1 \{\operatorname{OOC}(\operatorname{CH}_2)_2 \operatorname{COO}\}(\operatorname{Pbut}_3)_2]_2$, monoclinic $P_2 / n, a = 16.544(10), b = 23.819(15), c = 11.046(8) \text{ Å},$ $\beta = 103.72(6)^\circ, V = 4229(5) \text{ Å}^3, Z = 2, D_c = 1.333 \text{ g cm}^3,$ $\mu = 8.14 \text{ cm}^{-1}$ (Mo Ka)

(II) $[\operatorname{Ru}_{2}(\operatorname{CO})_{4}(\operatorname{CH}_{2}\operatorname{COO})_{2}(\operatorname{Pbut}_{3})]_{2}$, monoclinic $P2_{1}/n$, a = 14.411(8), b = 19.198(10), $\ddot{c} = 9.665(8)$ Å, $\beta = 93.45$ (6)°, V = 2669(3) Å³, Z = 2, $D_{c} = 1.579$ g cm⁻³, $\mu = 12.06$ cm⁻¹ (Mo K α).

Both structures were solved by single crystal diffractometer data (Mo K α), and refined by full-matrix least--squares to R = 5.18% for (I) and R = 3.83% for (II).

The structure of (I) consists of couples of dimers $(Pbut_3)(CO)_2Ru-Ru(CO)_2(Pbut_3)$ joined together by glutarate bridges coordinating to the metal atoms through their oxygens, each carboxylate group bridging two Ru atoms. Coordination around each Ru is distorted octahedral and involves two *cis* carbonyl ligands and two oxygen atoms from two carboxylate groups in a coordination plane and one phosphine and a Ru-Ru bond, 2.734(3) A long, *trans* with respect to that plane.