07.9-2 SYNTHESIS AND CRYSTALLOGRAPHIC DATA ON A NEW SILICA ZEOLITE, SILICA-ZSM-48. By R.P. Gunawardane, Department of Chemistry, University of Peradeniya, Sri Lanka & H.Gies, Mimeralogy Institute, Kiel University, West Germany.

A high silica zeolite ZSH-48 has been synthesized (Schlenker, J.L. et al., Zeolites, 1985, 5, 355-8) from silica solutions containing trace amounts of aluminium. In the present study the aluminium-free end member of the present stand the administrate enhances of the zeolite ZSM-48 ie. silica-ZSM-48, has been crystall-ized hydrothermally at 160° - 200°C from a pure silisic acid solution in the presence of polyamines as templates. Triethylenetetramine, tetraethylenepentamine, 1,5,8,12tetraazadodecane etc act as efficient templates in this synthesis. Microprobe analysis of the product confirmed the absence of aluminium in the framework. The X-ray powder diffraction pattern was indexed on the basis of an orthorhombic cell and the least square refinement using X-ray powder data gave the unit cell parameters <u>a</u> =14.21(4), <u>b</u> =20.09(5) and <u>c</u> =8.40(3) \Re . Powder reflections indicate C or I centering suggesting Immm. Imma. Comcon and Common symmetries. HRTEM studies on silica-ZSM-48 confirmed the lattice parameters obtained from X-ray powder data. All specimens investigated by TEM show weak reflections indicating the space groups Pomm or P21nm. Disregarding the weak reflections the space group Immm is found to be the most plausible. Twinning of silica-ZSM-48 is a common feature and the twin plane is perpendicular to [001]. It is apparent from these data that a structure based on ferrierite sheets linking via oxygen atoms located on mirror planes and consisting of 1-D channel systems may be visualized for silica-ZSH-48. The 10-T ring channels are running parallel to the c axis. Its framework topology appears to be closely related to that of ZSM-23 and ZSM-22. Further work is in progress to grow single crystals in order to confirm the structure by single crystal structure refinement.

HYPOTHETICAL FRAMEWORK STRUCTURES RELATED TO 07.9-3 MOBIL ZEOLITE ZSM-48. By J. L. Schlenker and W. J. Rohrbaugh, Mobil Research and Development Corporation, Research Department, Paulsboro, NJ 08066, USA. ZSM-48 is a high-silica zeolite whose structure has been proposed to be a disordered linking of ferrierite sheets via bridging oxygen atoms located on mirror planes. The proposed framework topology was based on agreement between observed and calculated x-ray powder diffraction patterns. The Smith plot in best agreement was obtained from a disordered intergrowth of two ideal framework structures with Cmcm and Imma symmetries. Because x-ray and electron diffraction data seem to indicate an orthorhombic lattice with pseudo-I or pseudo-Ccentering, attention was initially restricted to the formulation of hypothetical centered structures. With lattice parameters of a=14.24 A and b=20.14 A, and a Cor I-centered orthorhombic lattice, the ferrierite sheet has only four independent T-atoms, each of which may point up (U) or down (D) in order to link with other sheets. Alternating the orientation of these independent T-atoms yields twenty-eight closely related hypothetical framework structures, e.g. UUDD-Cmcm, UDUD-Imma, etc.

07.9-4 A SINGLE CRYSTAL STRUCTURE OF ZEOLITE ZSM-39. By <u>J. Macicek</u>, V.P. Vulchev, G.N. Kirov, Institute of Applied Mineralogy, BAS, Sofia, Bulgaria.

Octahedrally shaped crystals of the zeolite up to 0.7mm in size were grown hydrothermally from a gel with a composition 33Na_0:16(TMA)_0: $16(\text{TrMA})_2\text{O:}21_2\text{O}_3:110\text{SiO}_2:7300\text{H}_2\text{O:}22\text{H}_2\text{SO}_4$ for 7 days at 473 K. A spherically ground transparent crystal with r = 0.19mm was investigated on a CAD4SDP diffractometric system. Crystal data: (Na,TMA)_4.4^Al_4.4^Si_331.5^Oz7z, Mr = 8337, cubic, Fd3m, a = 19.396(1) Å, V = 7297(1) Å^3, Z = 1, D_z = 1.897 g.cm^{-3}, $\lambda(\text{MoK}_\alpha)$ = 0.71073 Å, μ = 6.8 cm^{-1}, F(000) = 4180, T = 291 K. From 8998 total (8<35°) and 316 unique (R_int = 0.033) 298 reflections with I>3o(I) were used in calculations. Final R(F) = 0.035 by full-matrix least-squares. Residual max. $\Delta(\rho)$ = 0.09 e.Å^{-3}.

The ZSM-39 topology, known from the powder data study (Schlenker J.L. et al., Nature, 1981, v. 294, 340-2) was confirmed. Besides, disordered Al, Na and TMA were localized in the 16-hedral cage. The Al-atom occupies a tetrahedron which shares face with another, currently vacant SiO4-tetrahedron from the framework. The apical oxygen atom of AlO4 is pointed to the centre of the cage, where both the Na and TMA are disposed. The electron density peak near the centre of the 12-hedral cage is presumably a water-oxygen.

07.9-5
ABOUT CRYSTAL STRUCTURES OF ZSM-11 AND ZSM-39. By A. Hardy, K. Ahmadi, M. Benjdir, P. Gravereau, A. Ouali and J.L. Guth*, Laboratoire de Cristallochimie Minérale, Université de Poitiers, 40 av. Recteur Pineau, F - 86022 Poitiers cédex, France, *Laboratoire de Chemie Minérale Générale, E.N.S.C.M., 3 rue Alfred Werner, F - 68093 Mulhouse cédex, France,

Single crystals of zeolites were synthesized in size large enough to study their crystal structure especially ZSM-11 (MEL) and ZSM-39 (MTN) known up to now only by their powder X-ray diffraction diagrams. One MEL single crystal measuring $9^4x5^4x5^4\mu\text{m}^3$ with $\{100\}$ and $\{001\}$ faces allowed us to assign it to space group $P4_22_12$ in place of I4m2 suggested by Kokotailo et al. (Nature, 1978, 275, 119-20). Cell parameters are a-b-19.93(1)Å, c-13.35(1)Å, V-5304(9)ų. Independent reflections were collected using CAD.4 diffractometer allowing us to index ZSM-11 powder diagram without any ambiguity and confirming a non-centered space group. The Kokotailo framework hypothesis must be called into question.

A quasi-spherical single crystal of MTN, $544\mu m$ in diameter, was prepared and studied with MoK α_1 . Faces are $\{100\}$, $\{111\}$ and $\{211\}$, space group Fd_3m and unit cell parameter a=19.43(1)Å. Data and refinements confirm the framework hypothesis given by J.L. Schlenker et al. (Nature, 1981, 294, 340-2) from powder X-Ray diffraction data.