02-Methods for Structure Determination and Analysis, Computing and Graphics

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02.08 - Crystallographic Computing

OCM-02.08.01 HYPERTEXT FOR CRYSTALLOGRAPHY by R.Diamond*, Medical Research Council, Laboratory for Molecular Biology, Hills Road, Cambridge, CB2 2QE, England and A Mummery, Oxford University Press, Walton Street, Oxford, OX2 6DP, UK

"Hypertext" designates a system of text and related material, prepared in machine readable form so that its presentation on a computer screen includes within it 'links' which enable the reader to invoke supporting material, references, figures etc. as he wishes, using a mouse or similar device to control the process. One such system, 'Molecular Structures in Biology', will be described both from the users' point of view and in terms of its internal construction, which is based on a 'web' which is a file defining many 'nodes' which may themselves invoke software to provide a pop-up reference, or another text, or graphics software or a stored bit map image etc. The web itself is traversed by software known as the 'spider' which puts into execution actions specified by nodes of the web. Such techniques, coupled with the high storage capacity of CD-ROM (600 Mbytes), provide an efficient means of publishing high-volume numerically intensive data, especially data with a high degree of permanence, as in data banks, where its crystallographic application seems most likely to be fruitful. MSB itself is both a textbook and a reference work, containing over 500 coordinate sets from the Protein Data Bank, plus twelve chapters of text exploiting hypertext techniques, plus many references, over 1000 illustrations (as bit maps), and the capability for the user to create an unlimited number of others.

OCM-02.08.02 IF WE CAN DO IT, YOU CAN DO IT: WRITING A GUI INTERFACE FOR A CRYSTALLOGRAPHIC PROGRAM by Paul N. Swepston* and Beverly R. Vincent, Molecular Structure Corporation, 3200 Research Forest Drive Woodlands, TX 77381 USA

Many crystallographic programs are still based on card-image input. The user is required to use a text editor prepare are instruction file for program execution. While this can be a simple effective means of program control, it fails to take advantage of modern software developments.

Examples will be given of graphical user interfaces (GUI) that have been developed for crystallographic programs using the MOTIF tool kit. Actual examples of source code will be shown in order to demonstrate easy it is to develop a modern graphical interface.

OCM-02.08.03 PC - A CRYSTALLOGRAPHIC COMPUTING TOOL FOR EXPERTS AND NOVICES. By V.K. Pecharsky, Dept of Inorganic Chemistry, L'viv State University, L'viv, Ukraine.

The choice of type of computer(s) for daily use in the laboratory for the tasks of diffraction-data processing, crystal-structure solution, refinement and final representation faces every crystallographer. For work on inorganic crystal structures and on small organic molecular structures (up to 100-150 independent atoms in the unit cell, with ca.1000 free least-squares parameters), personal computers (PCs) can be used very effectively. PCs have

many advantages, viz: they are open and friendly systems, they can easily be used by students, by regular scientific staff and by experts, and their capital cost is low. With proper software and suitable user interface, success is assured for everyone. PCs are excellent tools for crystallographic computing today and probably in the future.

We will illustrate our experience with PCs by describing the software package CSD (Crystal Structure Determination) for crystallographic computing on single-crystal and powder, X-ray and neutron diffraction data developed over the past few years. It will be shown that the average time spent by a crystallographer in the solution and refinement of a crystal structure is significantly less than the time necessary to collect the diffraction data. It does not usually exceed 1-4 hours, depending on the number of free least-squares parameters. Over two thirds of this time is spent on the final stages of refinement.

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OCM-02.08.04

THE PROFILE FITTING IN SINGLE-CRYSTAL X-RAY DIFFRACTOMETRY. By Ewa Gałdecka, Institute of Low Temperature and Structure Research, Polish Academy of Sciences, ul.Okólna 2, 50-950 Wrocław, Poland.

The whole-profile fitting that playes a fundamental role in crystal-structure determination from powder samples (the Rietveld method) is rather occasionally used for processing the single-crystal intensities [Clegg (1981). Acta Cryst. A37, 22-28 and 437; Oatley & French (1982). Acta Cryst. A38, 537-549]. Usually, the traditional background-peak-background procedure is considered to be sufficient in the latter case. The purpose of the paper is to work out a suitable method for approximation the single-crystal diffraction profiles and to test the effect of the careful data processing on the precision and reliability of the crystal-structure determination. The subject; of detailed considerations &r2 such problems as criteria of the goodness of fit, the choice of the basic approximating function (shape function), the proper number of independent adjustable parameters, dependence of parameters of the profiles on the Bragg angle and direction cosines of the diffraction vectors, and - recently discussed by Schwarzenbach & Flack [Acta Cryst. (1991), A47, 134-137] and Lenstra, Geise & Vanhouteghem [Acta Cryst. (1991), A47, 597-604] - treatment of the background and 'negative' reflections. The basis for the considerations is the papers mentioned above and two papers by the author [Gałdecka (1993). Acta Cryst. A49, 106-115 and 116-126]. Results of the present work are currently being incorporated into a computer procedure which approximates the diffraction profile and calculates the integrated net intensities. The newest results of the crystal-structure determination that includes the profile fitting, as compared with those obtained using. the common approach, will be presented.

OCM-02.08.05THE USE OF ORTHONORMAL FUNCTIONS FOR THE REPRESENTATION OF 3-DIMENSIONAL INFORMATION

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Many scientific fields deal with the calculation or observation of 3-dimensional data, for example electron density, Van der Waals surface and electrostatic potential calculations. Conventionally, this data is stored in grid form where a large number of data points are needed to fully describe it, which leads to problems with both storage and analysis.

An alternative approach is to approximate the grid by fitting a set of orthonormal functions to it. Once this has been carried out only the function coefficients need to be stored - i.e. a few hundred numbers instead of several hundred thousand. The 3-d information can then be regained when required by reversing the calculation.

The functions we have chosen are very similar to those used to describe atomic orbitals, they differ only in that they have been scaled to represent a whole molecule and not just the volume associated with a single atom. This technique has been applied previously for electron density display and fitting in the field of Protein Crystallography but has not been used extensively for smaller molecules.

We are particularly interested in the following applications of this technique:-

Crystallography:

The representation of molecular shape and charge for use in crystal packing studies.

The method can be used to represent voids within crystals, something that is difficult to achieve by other means.

Drug Design:

Molecular shapes are important for explaining and/or predicting the interaction of a drug with a putative binding site.

QSAR studies on the coefficients can be easily carried out. The number of parameters involved is relatively small and the nature of the functions allow the use of fast rotation algorithms to aid molecular comparisons.

The applications of the technique are limited only by the types of information that such functions can be expected to represent. If required another set of functions can be used for other cases as the choice is only limited by the need for orthonormality. Also, as the functions are able to represent a complex 3-d grid, two independent sets of data for the same molecule can be dealt with simultaneously. One is entered as though it was the real part of the grid and the other as the imaginary part; the two sets of numbers are kept completely separate during the calculations but may be displayed and analysed together if required.

OCM-02.08.06MOLECULAR SCENE ANALYSIS: A TOPOLOGICAL APPROACH FOR THE AUTOMATED INTERPRETATION OF PROTEIN ELECTRON DENSITY MAPS L. Leherte*, S. Fortier and J. Glasgow, Dept. of Computing and Information Science, Dept. of Chemistry, Queen's University, Kingston, Canada K7L 3N6.

As part of our project in Molecular Scene Analysis (Fortier, S., Castleden, I., Glasgow, J. I., Conklin, D., Walmsley, C., Leherte, L. & Allen, F. H. (1993). Acta Cryst. D 49, 168-178), we have been investigating methods to assist in the spatial and visual analysis of electron density maps at varying resolution. In particular, we have assessed the usefulness of the topological approach for the segmentation of medium (3 Å) resolution maps of proteins and their interpretation in terms of structural motifs. We have followed the approach implemented by Johnson (Johnson, C. K. (1977). ORCRIT. The Oak Ridge Critical Point Network Program. Chemistry Division, Oak Ridge National Laboratory, USA) in the program ORCRIT, which provides a global representation of the electron density distribution through the location, identification and linkage of its critical points (points where the density gradient vanishes). In the first part of our study, the topological approach was applied to ideal (calculated) maps of three proteins of small to medium size so as to develop a methodology - rules, heuristics or templates - that could then be used for analyzing maps of medium resolution. The methodology was then applied to both calculated and experimental maps of penicillopepsin at 3 Å resolution. The study shows that the networks of critical points provide a useful segmentation of the maps, tracing the protein main chains and capturing their conformation. In addition, these networks can be parsed in terms of secondary structure motifs, through a geometrical analysis of the critical points. The procedure adopted for secondary structure recognition was phrased in terms of geometry-based rules. It provides a basis for an automated implementation through the use of artificial intelligence techniques.

PS-02.08.07 THE USE OF A MODIFIED DOUBLE PATTERSON FUNCTION IN DIRECT PHASE DETERMINATION.

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Nowadays in X-ray crystallography most single crystal structures are solved by direct methods, which are based on probability theory. From this theory the values of structure invariants and seminvariants are estimated, one of them being the triplet phase sum ψ_1 . If interatomic triangles $(\mathbf{r}_{\kappa} \cdot \mathbf{r}_{\nu}, \mathbf{r}_{\lambda} \cdot \mathbf{r}_{\nu})$ can be obtained, $<\cos(\psi_1)>$ and $<\sin(\psi_2)>$ can be expressed in more accurate formulas. (Kronenburg, Thesis, Univ. of Amsterdam, 1992, 67-75). It is commonly known that the Patterson function consists of interatomic vectors. Two arbitrary Patterson vectors, however, do not necessary share a common atom, so the construction of triangles $(\mathbf{r}_{\kappa} \cdot \mathbf{r}_{\nu}, \mathbf{r}_{\lambda} \cdot \mathbf{r}_{\nu})$ from Patterson vectors is a complicated task. In this respect the double Patterson function (Vaughan, Acta Cryst., 1958, 11, 111-115):

$$P(\mathbf{u},\mathbf{v}) = V^2 \int_V \rho(\mathbf{r}) \rho(\mathbf{r}+\mathbf{u}) \rho(\mathbf{r}+\mathbf{v}) d\mathbf{r}$$

is more interesting for two related reasons;

- 1) P(u,v) is the Fourier transform of the triplet phase sum;
- non-zero P(u,v) are possible only if u and v form an interatomic triangle.

A modification of the double Patterson function is proposed. Calculations for both model and real structures are presented which show that interatomic triangles can be constructed more safely with the double Patterson function than with the normal Patterson function.

PS-02.08.08 A PACKAGE OF FAST FOURIER TRANSFORM ROUTINES FOR MACROMOLECULAR CRYSTALLOGRAPHY.

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