

07-Crystallography of Organometallic and Coordination Compounds

215

Turpeinen, U., Hämäläinen, R. & Reedijk, J. (1987). *Inorg. Chim. Acta*, **134**, 87–93; Turpeinen, U., Hämäläinen, R. & Reedijk, J. (1988). *Inorg. Chim. Acta*, **154**, 201–207. The title compound was obtained by slow evaporation of methanol solution containing Cu(II) 4-chlorobenzoate and 2-dimethylamino-ethanol. Blue crystals of $[\text{Cu}_4(\text{C}_7\text{H}_4\text{ClO}_2)_4(\text{C}_4\text{H}_{10}\text{NO})_4] \cdot 2\text{CH}_3\text{OH}$ are triclinic, space group $P\bar{1}$, $a=14.056(4)$, $b=14.704(4)$, $c=15.335(3)$ Å, $\alpha=106.27(2)$, $\beta=105.76(2)$, $\gamma=102.24(2)^\circ$, $Z=2$, $V=2782(1)$ Å³, $D_x=1.495$ Mg m⁻³, $M_r=1293.1$, final R value 0.039 for 5563 significant reflections.

The structure consists of discrete molecules. The Cu and bridging ethanolato O atoms form a cubane-type Cu_4O_4 core in which the short Cu–O bonds form an eight-membered ring folded in a boat-like conformation.

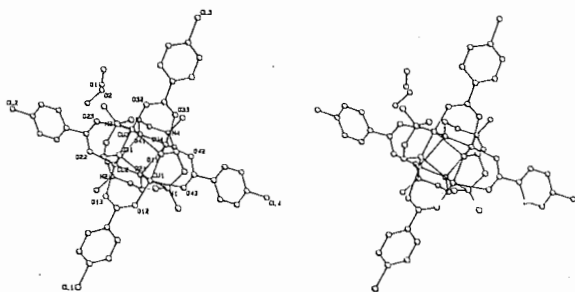


Fig. 1. Stereoview of the molecule.

Within the Cu_4O_4 core the long Cu–Cu distances are 3.664 and 3.796 Å and the four short Cu–Cu distances range from 3.096 to 3.191 Å. Each Cu atom is surrounded by two ethanolato oxygen atoms, a carboxyl oxygen atom and an amino nitrogen atom in a nearly square-planar arrangement with average Cu–O and Cu–N bonds of 1.936 and 2.055 Å. The axial sites of each Cu atom are occupied by an ethanolato oxygen and an oxygen atom of the carboxylate group with Cu–O distances of 2.409–2.815 Å.

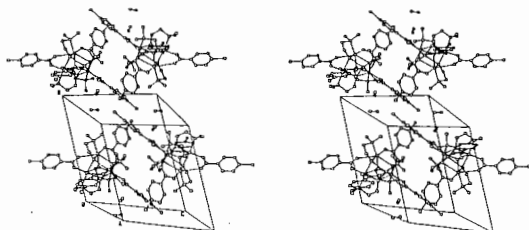


Fig. 2. Stereoscopic view of the packing.

PS-07.03.13 CRYSTAL STRUCTURE OF A TRINUCLEAR SULFIDOTUNGSTATE ANION. By Masood Parvez*, P. Michael Boorman and Meiping Wang, Department of Chemistry, University of Calgary, Calgary, Alberta, Canada T2N 1N4.

We have been exploring the controlled polymerization of WS_4^{2-} anions by alkylation reactions. The $\text{W}_3\text{S}_9^{2-}$ anion has been previously reported as a product of the reaction between WS_4^{2-} and alkyl halides. We have found that the complex $\text{Cl}_3\text{W}(\mu\text{-SEt}_2)_3\text{WCl}_3$ (**1**) is an effective ethylating agent, and will cause the trimerization of WS_4^{2-} to $\text{W}_3\text{S}_9^{2-}$. From ¹H NMR data we have evidence that the $\text{W}_3\text{S}_9^{2-}$ anion can itself be alkylated, hence we wished to see if clusters of higher nuclearity could be prepared this way. One product of the reaction between **1** and $[\text{PPh}_4]_2[\text{W}_3\text{S}_9]$ in a mixture of MeCN and N,N-dimethylformamide (DMF) was isolated and subjected to an X-ray crystal structure determination. It has been shown to be $[\text{PPh}_4]_2[\text{S}_2\text{W}(\mu\text{-S})_2\text{W}(\text{DMF}) (\text{S}) (\mu\text{-S})_2\text{WS}_2]$, rather than a cluster of higher nuclearity as was

anticipated. The structure has some interesting features, however, in that the sulfide and DMF ligands on the central tungsten are disordered over the two axial sites.

Crystal data. $\text{S}_9\text{W}_3(\text{C}_3\text{H}_6\text{NO}) \cdot 2(\text{C}_{24}\text{H}_{20}\text{P})$, monoclinic, space group $P2_1/n$, $a = 9.863(2)$, $b = 14.618(3)$, $c = 18.795(3)$ Å, $\beta = 97.98(1)^\circ$, $V = 2683.6(8)$ Å³, $Z = 2$, $R = 0.072$ for 3813 observed data ($I > 3\sigma(I)$) collected on a Rigaku AFC6S at 150(1) K.

PS-07.03.14 CRYSTAL STRUCTURE OF DIAMINE COMPLEXONATES OF COBALT WITH ALKALINE-EARTH CATIONS by L.A.Zassourskaya*, T.N.Polynova, V.B.Rybakov, M.A.Porai-Koshits. Department of Chemistry, Moscow State University, 119899, Moscow, Russia.

X-ray structure analysis and crystal chemistry investigation of $\text{M}'[\text{ML}]_2 \cdot n\text{H}_2\text{O}$ complexonates (M' -alkaline-earth cation, $\text{M}=\text{Co(II)}$, Co(III) , $\text{L}=(\text{edta})^{4-}$, $(\text{Hedta})^{3-}$, where edta-ethylenedia-minetetraacetic acid) were carried out in order to study the role of M' in the formation of their crystal structure. The active interaction between M' and the ligand leads to change the structural function of the complexon carboxylate groups. The interaction of M' with a single carbonyl oxygen atom leads change the structural function of the complexon carboxylate groups. The interaction of M' with a single carbonyl oxygen atom leads to the formation of an island structure. So, $[\text{Mg}(\text{CoHedta}(\text{NO}_2))_2 \cdot 9\text{H}_2\text{O}]$ consists of heterometallic binuclear complexes $[\text{Mg}(\text{H}_2\text{O})_5\text{CoHedta}(\text{NO}_2)]^+$ and $[\text{CoHedta}(\text{NO}_2)]^-$ anions. If M' is connected with several carbonyl atoms chain ($\text{Ba}(\text{Coedta})_2 \cdot 8\text{H}_2\text{O}$) or layer ($\text{Ca}(\text{CoHedta}(\text{CN})_2 \cdot 6\text{H}_2\text{O})$) structures are formed. Even more interesting are isostructural layer compounds $\text{M}'[\text{CoHedta}]_2 \cdot 6\text{H}_2\text{O}$, where $\text{M}'=\text{Ca}^{2+}$, Sr^{2+} , Ba^{2+} . Besides water molecules M' coordinates both carbonyl and carboxyl oxygen atoms forming four-membered cycles. The structure of $\text{Ca}(\text{Coedta})_2 \cdot 7\text{H}_2\text{O}$ is the most peculiar among calcium complexonates, because in it Ca^{2+} cation is completely hydrated and therefore plays the role of "outer sphere" cation. The wide diversity in the nature of the interaction between M' and carboxylate groups makes the investigation of the compounds of such type important.

PS-07.03.15 SYNTHESIS AND CRYSTAL STRUCTURE OF TETRACOBALT CLUSTER COMPLEX $[\text{Co}_4(\mu_3\text{-S})(\text{SCH}_2\text{CH}_2\text{S})_3(\text{PPh}_3)_3\text{Cl}]$. By F. L. Jiang*, Z. Y. Huang, B. S. Kang, M. C. Hong and H. Q. Liu, State Key Lab. of Struct. Chem. and Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, China

A new tetracobalt compound $[\text{Co}_4(\mu_3\text{-S})(\text{SCH}_2\text{CH}_2\text{S})_3(\text{PPh}_3)_3\text{Cl}]$ (**1**) was prepared by reacting $\text{CoCl}(\text{PPh}_3)_2$ and Na_2edt ($\text{H}_2\text{edt}=1,2$ ethanedithiol) in acetonitrile, and structurally characterized by X-ray diffraction methods. Compound **1** crystallizes in the trigonal space group $P\bar{3}$ with $a=16.505(8)$, $c=13.401(6)$ Å, $V=3162(3)$ Å³, $Z=2$ and $R(\text{w})=0.046$ (0.064) for 2636 reflections ($I > 3\sigma(I)$). The cluster contains four cobalt atoms in the form of a trigonal pyramid. The basal Co_3 triangle is capped by $\mu_3\text{-S}(3)$ below, and also di-bridged by $\text{S}(2)$ of edt^{2-} in each edge of the triangle, while the triangle is bridged to the apical $\text{Co}(2)$ above by another