## 242 07-Crystallography of Organometallic and Coordination Compounds

Metal alkoxides might represent an alternative, especially as they are generally more attractive for the design of mixed metal species. Barium alkoxides involving fluorinated groups, namely perfluoroisopropoxide OCH(CF<sub>3</sub>)<sub>2</sub> (ORf) have been obtained either by direct reaction between the metal and the alcohol or by alcoolysis reactions. Crystal structure and chemical formula of one of them, crystallized from hexane/tetrahydrofuran (THF), have been determined by single crystal X-ray diffraction at 220 K.

Crystal data are as follows: monoclinic symmetry; space group P21/C; cell parameters at 220 K: a = 24.667(7) Å, b = 13.925(3) Å, c = 25.856(9) Å,  $\beta$  = 113.15(3)°; asymmetric unit: one Ba5(OH)(OCH(CF3)2)9 (THF)4 H2O molecule + one not bonded THF molecule; Z = 4. The molecular structure consists of a classical pentanuclear cluster of barium atoms, arranged in a square based pyramid. A  $\mu_5$ -oxo ligand lies at the center of the pyramid squared basis; 5  $\mu_3$ -ORf ligands, 4  $\mu_2$ -ORf ligands, 4 THF and 1 H2O ligands are surrounding the cluster. Thus, each barium atom has 7 to 11 oxygen or fluor atoms as neighbours. Such a high coordination number for barium atoms explains the relative stability of this alkoxide

**PS-07.05.12** THE MODULATED CRYSTAL STRUCTURE OF A METALLOPORPHYRIN  $\pi$ -CATION RADICAL. Kenneth J. Haller\* and A. David Rae, School of Chemistry, University of New South Wales, P.O. Box 1, Kensington N.S.W. 2033, Australia.

The hexachloroantimonate salt of chloroiron tetramesityl porphyrin radical monocation crystallizes as a modulated structure with a tetragonal I4/m parent space group; a=18.008, c=9.028 Å, Z=2. A commensurate modulation exists corresponding to h+k+l = 2n+1 data being weakly present. The absence condition h+k = 2n+1 is maintained for the l = 0 data. There are also weak satellite reflections corresponding to a modulation wave vector 0.46 c\*.

A hierarchical approach to structure elucidation was used, starting with a disordered average structure in I4/m and progressing to structure refinement attempts in  $P4_2/n$ , and P4/n to try to identify the nature of the modulation.

The cation was located on a 4/m site in I4/m with the mesityl groups on four separate cations defining a channel about the 42 axis which contains the SbCl6 counterions. These ions appear streaked along c and individual atoms are not resolved. Reduction of symmetry to  $P4_2/n$  or P4/nallows the cation to adopt either a  $\frac{7}{4}$  or  $\frac{7}{4}$  symmetry respectively. Packing requires the SbCl<sub>6</sub> to have a 2 axis between adjacent Cl coincident with a 2 fold crystal axis parallel to c. In either spacegroup symmetry creates a second anion 90° rotated translated along the same axis. Both of these cannot fit in the axial repeat and one is not required by chemistry. Ordering to lower the spacegroup to  $P^{\frac{1}{4}}$  causes the n glide absence condition to be violated. The anions appear to contribute very little to the h+k+l =2n+1 data. The h+k+l=2n+1 data has intensity distributed in h and k rather than l. This appears to correspond to a correlated rotation of the mesityl groups which provides a mechanism for creating the necessary correlation between columns to see the incommensurate reflections. The c repeat distance would appear to be determined by contacts between the mesityl groups of different molecules but there appears to be a three atom contact between adjacent cations along c. This is consistent with having an asymmetric Cl<sub>3</sub> ion linking Fe atoms rather than a Cl

Dimensions are compatible. This would further enhance the necessary communication for the incommensurate modulation to be observed. Resolution of the structure options requires an ability to model the anions. The anions were initially included in the model as a series of spherically averaged SbCl<sub>6</sub> spaced at 1/8c with large U<sub>33</sub> thermal parameters. Subsequent to this fractional oriented anions streaked along c have been used. The progress of refinement has been monitored by assessing data of different index condition. Refinement was kept in control using RAELS89 to refine rigid body thermal parameters and impose localised symmetry constraints. The anions appear to have restricted orientation and be clustered around the same z heights as the cations, 1/2c apart in the average structure. Refinement is continuing. The observed incommensurate modulation appears to be the result of the c axial length being longer than required to have anions touching. This allows extra species, presumably solvent, to be fitted into the channels.

We wish to thank Professor W. Robert Scheidt and Dr. Hungsun Song of The University of Notre Dame, IN for providing the data for this problem.

PS-07.05.13 STRUCTURAL CHEMISTRY STUDIES ON RARE EARTH AMINOACID COMPLEXES By Y.H.Lin, Y. Xin, A.Z.Ma, L.M.Li and S.Q.Xi, Changchun Institute of Applied Chemistry, Academia Sinica, Changchun, China

This report gives a brief account of some of structural chemistry studies on rare earth aminoacid (REA). During the past years, our research group has succeeded in synthesizing series of REA. These complexes respectively belong to ([Er(pro),(H,O),Cl,])n (Ma,Li,Lin and Xi, Acta cryst. (1993). C49) Sm(C10,),(Val)n · 4H,0, Sm(C10,),(ASP)n · mH,0, Sm(C10,),(Gly), 4H.O, YbC1,(Ata) . 4H.O, YbC1,(pro) . 6H.O, Er(C10,),(pro) . nH.O, La(ClO<sub>4</sub>),(Ala), 4H<sub>2</sub>O, La(ClO<sub>4</sub>),(Gly), 4H<sub>2</sub>O, Sm(ClO<sub>4</sub>),(Ala), 3H<sub>2</sub>O and Er(C(O,),(His), . 5H,O. The chemical structures of these complexes have been identified by elemental analyses. IR and mass spectra. Among them, the molecular configuration and the bond parameters of eleven typic complexes have been determined by careful X-ray structural analyses. The characteristics of the motecular structure of REA have been summarized. First, the amino groups in REA are protonated and not coordinated. Secondly, rare earth atoms are separated from each other by a distance about 4Å, so that there are no metal-metal bonds between them. Thirdly, the coordination number of rare earth metals are 8-10, their coordinate configuration forming square antiprism, tricapped trigonalprism and bicapped square antiprism respectively. Finally, these structure can be divied into infinite

PS-07.05.14 CRYSTAL STRUCTURES OF (AdeH<sub>2</sub>)SbCl<sub>5</sub>H<sub>2</sub>O AND (AdeH)<sub>2</sub>SbCl<sub>5</sub>H<sub>2</sub>O (Ade = ADENINE). By Tu Lian-Dong\* and Hu Sheng-Zhi, Chemistry Department, Xiamen University, Xiamen, China.

three-dimensional network, one-dimensional chain and

binuctear bridge three classes.