08-Inorganic and Mineralogical Crystallography

ATOM	POSITION	X	Y	Z	N	$\mathbf{B}[A^2]$
\mathbf{La}	6a	0	0	0.25	1.0	0.25(2)
Al	6b	0				0.43(5)
0	18e	0.474(1)	0	0.25	1.0	0.57(2)

Table 1 Structural parameters of LaAlO₃.

PS-08.04.19 A SINGLE-CRYSTAL X-RAY-DIFFRACTION STUDY OF THE ORTHORHOMBIC Al₃Mn PHASE. By N.C.Shi[†], X.Z.Li[‡], Z.S.Ma[†] and K.H.Kuo[‡], [†] X-Ray Laboratory, China University of Geosciences, Beijing 100083, China. [‡] Beijing Laboratory of Electron Microscopy, Chinese Academy of Sciences, P.O.Box 2724, Beijing, China.

Al₃Mn (with a stoichiometrical composition Al₂₉Mn₁₀), M_r=1331.8, orthorhombic, Pn2₁a, a=14.837(4), b=12.457(2), c=12.505(2)Å, V=2311.4(8)Å³, Z=4, D_x=3.56, Mo K α =0.71069Å, μ =5.62mm⁻¹, F(000)=2355.02, T=room temperature, R=0.068 for 1550 reflections. About 2/3 Mn atoms have icosahedral coordination. In a repeat unit b, there are 4 interlocked icosahedra (two Mn and two Al) consisting of 1 pentagonal prism and 4 antiprisms. The icosahedral subunits form a network on (010) constituting a layer structure, consisting of an almost "flat" layer F sandwiched between two puckered P and p layers in the sequence of PFpP'F'p' ..., where P'F'p' is rotated from PFp by 180° around the layer normal. A similar layer structure has also been found in the π-AlMnM (M=Ni,Cu,Zn) phases. Both these two crystalline structures can be obtained from that of the Al-Mn decagonal quasicrystal with a periodicity of 12.4Å along the tenfold axis by substituting a rational ratio of two consecutive Fibonacci numbers F_{n+1}/F_n for the irrational $\tau = (1+\sqrt{5})/2$ in two quasiperiodic directions on the plane normal to the tenfold axis.

In recent years, a number of intriguing physical phenomena related to the charge density wave (CDW) have been observed in some low dimensional materials. Because most of these materials were found to be of the early transition metal chalcogenides, we have studied the ternary system R-Nb-Se (R=In,Sb).

InNb₃(Se₂)₆ was synthesized by the reaction of Nb₂Se₉ and In with approximately stoichiometric proportions at 773-873K.

The preparation of SbNb₃(Se₂)₆ succeeded by an analogous procedure with the stoichiometric elements as the starting reactant; at 973k.

The structures are determined by single crystal X-ray diffraction method. The results showed that $InNb_3(Se_2)_6$ (a=9.450(1).

- c=19.068(9)A. P4/mnc.z=4) and SbNb₃(Se₂)₆ (a=9.466(2), c= 19.075(3)A, P4/mnc,z=4) are very similar in structural parameters. Both of the structures are composed of the one dimensional infinite chains $[Nb_3(Se_2)_b]_{\infty}$ along the c axis with the third component R (R=In,Sb) intercalated between them. In one
- [Nb3(Se2)6]∞ chain, the distances between the niobium atoms are arranged in a periodic sequence of …Nb-3.037-Nb-3.248-Nb -3.248A-Nb… for InNb3(Se2)6. According to the Nb Nb distances

the building block of a chain. In comparison with the Nb-Nb bond lengths in (MX4)nY(Gressier, Meerschaut, Guemas,Rouxel & Monceau, J. Solid State Chem., 1984, 51, 141) and a-Nbla (Dahl & Wampler, Acta Cryst., 1962, 15, 903), the Nb-Nb distances in the above sequence should also involve chemical bonding. Therefore the Nb3(Se2)6 units in one chain may be viewed as combined together by the bridging $(Se_2)^{2-}$ ligands as well as the Nb-Nb bonding. Considering the long distance (6.69A) between the center lines of different chains, the metal-metal interaction is very anisotropic and typical of the one dimensional character. The bond lengths of Nb-Se (2.558(1)~2.721 (2)A) and Se-Se (2.325(2)~2.356(2)A) are all in agreement with those in other known niobium selenides such as NbSeg (Hodeau, et al., J. Phys. C., 11, 4117). The In atoms located in the one dimensional channels between different chains form the tetragonal pyramid coordination geometry with four equidistant Se atoms from four different chains. The height of the tetragonal pyramid, which shows the position of the intercalated atom in the channel, for InSe4 is 0.364A, and which reveals the main difference between the structures of InNb3(Se2)6 and SbNb₃(Se₂)₆ (0.589A for SbSe₄). The large distance of In…Se (3.233(1),3.263(3)A for Sb...Se) suggests that the In(Sb) atoms are inserted as in In~3Mo15Se19 (Gruttner & Yvon, Acta Cryst., 1979, B35, 285). Unusual high values of $\mathbb{U}_{33} \nearrow \mathbb{U}_{11}$ (2.42 and 4.51 for In and Sb, respectively) are indicative of the large thermal vibrations of the intercalated atoms along the c Electrical resistivity measurements indicate that $InNb_3(Se_2)_6$: $\rho_n=19.2(3)$, $\rho_b=19.4(4)$, $\rho_c=0.74(1)\Omega cm$ (298K); SbNb₃(Se₂)₆ : ρ_{a} =26.0(5), ρ_{b} =25.5(5), ρ_{c} =0.11(2) Ω cm (298K) and the dash rises of $\rho_{\,\rm c}$ at ~ 165 and 64K for $\rm InNb_3(Se_2)_6;\ \sim 185$, 70 and 30K for SbNb₃(Se₂)₆ in the range of 289K-298K.

in the sequence, the Nb₃(Se₂)₆ unit can be formally viewed as

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PS-08.04.21 A PATTERSON SYNTHESIS WITH PROFILES OF DIFFRACTION INTENSITIES. By Y.Ono* and M.Tokonami, Mineralogical Institute,

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It has been shown previously (Kaplow, R., Rowe, T.A. & Averbach, B.L.(1968). Phys. Rev. 168, 1068-1079) that the 'Coupling coefficient' of atomic interaction to be defined in study of a radial distribution function of hexagonal selenium by the powder method. As for single crystals, with a conventional structure analysis by integrated diffraction intensities, we can

only obtain average atomic interactions, such as thermal atomic vibration, of all unit cells in the crystal. In this work, to obtain the coefficients also in similar

In this work, to obtain the coefficients also in single crystals, a 'direct' Patterson synthesis which is unusually calculated with a data set of profiled intensities collected by step-scan is investigated. The sample was a monoclinic \triangleleft -selenium crystal and the

In the sample was a monochrine of setting direction direction of $29 < 45^{\circ}$ with wave length of 0.7Å. The scan width and step width of the peak profile data were $+0.5^{\circ}$ and 0.01° in ω and total data of hkl used were 778.

In the Patterson map with profiled intensity data, the periodicity of unit cells have disappeared:Remarking the origin peak, usually 'equivalent' peaks' height and widths become the lower and the broader as far from the origin. This means that the stronger interactions of

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