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New $M_{1/3}Re_{2/3}O_2$ (M = Fe, Co, Ni) oxides with rutile-like structures

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For the system $M_{1/3}Re_{2/3}O_2$ (M = Fe, Co, Ni) different rutile or rutile-derivated structural modifications were found depending on the temperature during high-pressure high-temperature syntheses. The crystal structures have been determined by single crystal X-ray diffraction. For synthesis temperature higher than 1300 $^{\circ}$ C, phases with tetragonal rutile aristotype structure (S. G. P4₂/mnm) and a statistical distribution of M- and Re-atoms on the metal site in the structure were obtained. At lower synthesis temperature, rutile-like structures with orthorhombic or monoclinic symmetry, partial order on M- and Re-sites and metallic Re-Re bonds were observed for Co1/3Re2/3O2 and Ni1/3Re2/3O2. A monoclinic CrWO4type structure (S. G. C2/m) with only Re-atoms on the 4g site and statistically distributed Ni- and Re-atoms on the 4i site was adopted by Ni_{1/3}Re_{2/3}O₂. Single crystals Co_{1/3}Re_{2/3}O₂ with orthorhombic CrWO₄-type structure (S. G. F222) and also partially mixed Co,Repositions were found together with a monoclinic modification, which is isotypic to monoclinic Ni_{1/3}Re_{2/3}O₂. Note, that another orhorhombic modification (S. G. Cmmm) was reported for the composition Co_{1/2}Re_{1/2}O₂ with distinct positions of Co and Re and without any metallic Re-Re-bond [1]. Unfortunately, the synthesis conditions were not described in sufficient detail to allow a comparison with our preparation method. Ni_{1/3}Re_{2/3}O₂ with the rutile-structure orders ferroor ferrimagnetically at 23.5 K, whereas tetragonal Co_{1/3}Re_{2/3}O₂ shows two magnetic tranitions: below 17.5 K the ferromagnetic component dominates in contrast to an antiferromagnetic order between 27.0 and 17.5 K.

[1] W. H. Baur, W. Joswig, G. Pieper, D. Kassner, J. Solid State Chem. 99 (1992) 207-211.

Keywords: high-pressure high-temperature synthesis, rutilelike structure, metallic Re-Re bond

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Compression and thermal behavior of $GeO_2\ glass$

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One fascinating high-pressure behavior of tetrahedral glasses and melts is the local coordination change with increasing pressure, which provides a structural basis for understanding numerous anomalies in their high-pressure properties. Because the coordination change is often not retained upon decompression, studies must be conducted in situ. Previous in situ studies have revealed that the short-range order of tetrahedrally structured glasses and melts changes above a threshold pressure and gradually transforms to an octahedral form with further pressure increase. We report compression behavior of GeO₂ glass up to 35 GPa, and a thermal effect associated with the coordination change at given pressures. The results show a clear anomaly in compression curve associated with the coordination change, as well as distinct thermal behaviors of GeO₂ glass in tetrahedral, octahedral, and their intermediate forms. An unusual thermally induced densification, as large as 16%, was observed on a GeO₂ glass at a pressure of 5.5 GPa. The large thermal densification at high pressure was found to be associated with the 4-to 6-fold coordination increase. Experiments at other pressures show that the tetrahedral GeO₂ glass displayed small thermal densification at 3.3 GPa arising from the relaxation of intermediate range structure, whereas the octahedral glass at 12.3 GPa did not display any detectable thermal effects.

Keywords: high pressure, coordination change, compression

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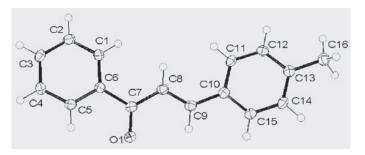
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Disappearing and reappearing polymorphism in *p*-methylchalcone

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The polymorphism of *p*-methylchalcone (p-MC) has been the subject of interest for about 80 years. One stable form (Form I) and two metastable forms (Form II and Form IV) of *p*-methylchalcone have been obtained, with Form II having previously been reported to be a disappearing polymorph. To prevent the conversion of Form II, which has a melting point of 89-91 °C, into the stable form it was produced in conditions free from seeds of the stable form. Form IV was discovered serendipitously as a single crystal that appeared in the preparation of a powder of Form II. Forms I and II have also been characterized by optical microscopy, differential scanning calorimetry (DSC), variable temperature powder X-ray diffraction and spectroscopic analysis. The crystal structures of these three forms have been solved, Forms I and IV by single-crystal methods and Form II from synchrotron powder diffraction.



Keywords: p-Methylchalcone, polymorphism, X-ray structure determination

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Switchable polymorph-dependent luminescence of terpyridine

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