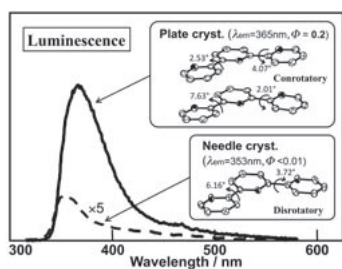


Nagasaki, 2-1303-8, Ikeda, Nagasaki, 856-0026, Japan, E-mail : mutai@iis.u-tokyo.ac.jp

2,2':6',2''-Terpyridine (tpy), which is practically non-fluorescent ($\Phi < 0.01$) in organic solutions, in an amorphous solid and a needle crystal ($P2_12_12_1$, $Z=4$), was found to show an efficient solid-state luminescence (365 nm, $\Phi = 0.2$) upon formation of a plate crystal ($P2_1/c$, $Z=8$). Furthermore, effective on-off switching of this solid-state luminescence was realized by heat-mode interconversion between the plate and needle crystals. It was shown that the rate constant of the non-radiative relaxation of the needle crystal ($k=1.2 \times 10^9 \text{ s}^{-1}$) was an order of magnitude larger than the plate crystal, while that of the emissive process were comparable ($k \sim 10^7 \text{ s}^{-1}$). In contrast to a frozen solution, both polymorphs did not exhibit phosphorescence at 77 K, suggesting that the triplet state might not be involved in the non-radiative process. Comparing both polymorphs, tpy molecules were in different conformation, conrotated and disrotated, which might be due to a different molecular packing. Relation between the difference in the crystal structure and the solid-state luminescence property was further studied.



Keywords: luminescence, polymorphism, heterocyclic compounds

P08.11.114

Acta Cryst. (2008). A64, C453

Polymorphism of chlorpropamide: Structure and transitions

Tatiana N. Drebuschak^{1,2}, Nikita V. Chukanov^{2,3}, Valeri A. Drebuschak^{2,4}, Yurii A. Chesalov^{2,5}, Elena V. Boldyreva^{1,2}
¹Institute of Solid State Chemistry and Mechanochemistry SB RAS, Kutateladze 18, Novosibirsk, NSO, 630128, Russia, ²Novosibirsk State University, Ul. Pirogova 2, Novosibirsk, 630090, Russia, ³Novosibirsk Institute of Organic Chemistry SB RAS, Pr. Ak. Lavrentieva, 9, Novosibirsk, 630090, Russia, ⁴Institute of Geology and Mineralogy SB RAS, Pr. Ak. Koptuyuga, 3, Novosibirsk, 630090, Russia, ⁵Boreskov Institute of Catalysis SB RAS, Pr. Ak. Lavrentieva, 5, Novosibirsk, 630090, Russia, E-mail: tanya@xray.nsu.ru

The importance of the polymorphism of drugs can hardly be overestimated. Detailed structural studies not only allow to identify a selected form reliably, but also provide a valuable insight into the structure-properties relations, especially, when combined with vibrational spectroscopy, thermal analysis and calorimetry. Chlorpropamide an antidiabetic drug, has been known to be highly polymorphic since a very long time, but the crystal structures of all the claimed forms but the one produced commercially remained unsolved. Structure of only one polymorph was known when we started our investigation. We solved three new ones. The molecule is very flexible and different parts of the molecule can freely rotate and bend. Having crystal structure data, we defined quantitatively the values of torsion angles and variations in hydrogen bonds among them. Motif of hydrogen bonds is really the same, and the polymorphism results mainly from the difference in intramolecular (conformational polymorphism). Packing of neighbouring molecules in the structure is similar for all polymorphs but one (beta). Hydrogen bonds in that polymorph differ from those in the rest ones. We investigated polymorph transitions near the melting point by using DSC, IR

spectroscopy, and X-ray powder diffraction. All the polymorphs are proved to transform into the fifth polymorph, epsilon, which melts then. Kinetics of polymorph transition is inhibited for beta polymorph. This work was supported by the grants from BRHE (RUX0-008-NO-06/BP2M08, Y3-C-08-01), Innovation Project "Education" from the Russian Ministry of Education and Science #456, and Integration Projects #49 and #110 of SB RAS.

Keywords: polymorphism, drug structures, hydrogen bonds

P08.14.115

Acta Cryst. (2008). A64, C453

Structure formation effects on electro-optics of montmorillonite clay-5CB liquid crystal composites

Tamara Bezrodna¹, Irina Chashechnikova¹, Galyna Puchkovska¹, Yevgeniy Shaydyuk¹, Anatoliy Tolochko¹, Jan Baran², Marek Drozd²
¹Institute of Physics NAS Ukraine, 46 Nauki prosp., Kyiv, Kyiv, 03022, Ukraine, ²Institute of Low Temperature and Structure Research, PAS, 2 Okolna st., Wroclaw 50-950, Poland, E-mail: tomaalone@yahoo.com

Nanomaterials based on liquid crystals (LCs) and montmorillonite (MMT) clay are perspective due to their applications in the information record and storage devices, etc. The nematic LC, pentylcyanobiphenyl (5CB) was used in our composites. The MMT was modified with octadecylbenzyltrimethylammonium chloride (OBDM) and dioctadecyltrimethylammonium chloride (DODM), which leads to the better chemical affinity of the clay to LCs and increase of MMT basal spacings for the penetration of 5CB dimers to the interlayer space. The nature of the surfactant is responsible for chemical affinity between the clay and LC, affecting the structure formation in the composites. These properties explain their different electro-optical (EO) characteristics. In the case of OBDM-MMT, well separated small clay particles are uniformly distributed in the 5CB medium. Under applied electric field, this mobile system is easily oriented along the field direction resulting in significant increase in optical transparency, when the applied voltage increases. Due to the fairly strong molecular interactions between the composite components, the field-induced alignment of 5CB domains remains the same even when the electric field is removed (considerable EO memory). Contrary, due to the smaller affinity of DODM to 5CB molecules, DODM-MMT particles are only partially covered by LC layer, which allows them to aggregate. These aggregates form a stable colloidal network with 5CB dimers. In EO measurements, when an electric field is applied to the composite, its transparency increases, but less, than for the former one. So the EO memory of the composite with DODM-MMT is low, but the contrast is high. These investigations will assist in the development of the composites with desired optical characteristics.

Keywords: LC-clay composites, X-ray diffraction, electro-optical properties

P08.14.116

Acta Cryst. (2008). A64, C453-454

Crystal structure of metastable cubic B₁-phase of the La₂Mo₂O₉ single crystal

Olga A. Alekseeva¹, Natalia I. Sorokina¹, Igor A. Verin¹, Elena P. Kharitonova², Valentina I. Voronkova²

¹Shubnikov Institute of Crystallography RAS, Leninskii pr., 59, Moscow, Moscow, 119333, Russia, ²Moscow State University, Leninskie Gory, 1, Moscow, 119991, Russia, E-mail: olalex@ns.crys.ras.ru