for biological compounds, such as sugars and polysaccharides. The molecule of EG is a triple rotor about the O-C, C-C and C-O bonds. Following the common assumption that each of the rotors can assume 3 positions, the molecule can exist as $3^3=27$ conformers of which 10 are unique [1]. We have looked at the crystal structures of EG frozen at varied thermodynamical conditions. This study also pertains to a basic problem of the relation between the conformational preferences of isolated molecules and their conformation adopted in the crystalline state [2]. Till now one EG structure crystallized by temperature freezing was determined at 130 K: crystal was orthorhombic, in $P2_12_12_1$ (phase I) with $Z^2=1$ [3]. Two new crystal phases of EG have been obtained by *in-situ* pressure-freezing in a Merrill-Bassett DAC [4]. These new crystal structures were solved straightforwardly by direct methods in space group $P2_1$ at 1.2 and 2.0 GPa (phase II), and in space group *Pbcn* at 2.2 GPa (phase III).

[1] Howard, D. L., Jorgensten, P., Kjaergaard, H. G. J. Am Chem. Soc. **2005**, 127, 17096-17103

[2] Bürgi, H. B., Dunitz, J. D. Acta Cryst. 1988, B44, 445-448.

[3] Thalladi, V. R.; Boese, R.; Weiss, H. C.: Angew. Chem. Int. Ed. **2000**, 39, 918 – 922.

[4] Budzianowski, A., Katrusiak A. *In High-Pressure Crystallography*; A. Katrusiak, P. F. McMillan Eds.; Dordrecht: Kluwer Academic Publisher, **2004**, 157-168.

Keywords: high-pressure phases, hydrogen bonds in organic crystals, molecular interactions

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The structure and nature of ice VII to 20 GPa

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Ice VII has a simple average structure in which the oxygens form a body-centred cubic (bcc) arrangement and the protons are disordered over two off-centre sites in H-bonds that connect the oxygens into two interpenetrating tetrahedral networks. However, there is a complication in that the oxygens are disordered over a number of sites that are displaced ~ 0.1 A from the mean bcc positions [1]. The protons must also be disordered over several sites around their mean position. The direction of these displacements affects both the number of sites and hence the number of structural configurations - and the geometry of the H-bonds. Some possible displacement directions would require there to be a range of quite different bondlengths, which could have implications for understanding the bond centring to ice X at much higher pressures. Because of the small magnitude of the displacements, it is difficult to obtain a decisive determination of their direction. But overcoming this problem is crucial to a full understanding of the ice VII phase. There are also other questions about this phase, including the nature of its short-range structure, the possibility of incommensuration [2] and some form of discontinuity around 14 GPa (~18 GPa in the deuterated form) [3]. We will report a range of studies, using both neutron and x-ray diffraction techniques, that provide new information on the structural disorder and also on the evidence for or against any change in structure in the range up to 20 GPa.

- [1] R.J.Nelmes et al, Phys. Rev. Lett. 81 (1998) 2719-2722 .
- [2] P.Loubeyre et al, Nature 397 (1999) 503-506.
- [3] P.Pruzan et al, J. Raman Spec., 34 (2003) 591-610.

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High pressure X-ray cell for soft matter

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There is currently high demand for development of techniques that will allow probing of the time-dependent structural behaviour of outof-equilibrium soft condensed matter systems. An ideal technique for monitoring and triggering the structural evolution of such systems within the millisecond regime is time resolved X-ray diffraction, employing pressure as the triggering variable. Using pressure jumps to trigger structural change has several major advantages over other methods, e.g. temperature jumps: The pressure step propagates at the speed of sound and thermodynamic equilibrium is achieved far more rapidly than the rate of acquisition, bi-directional pressure jumps propagate at the same speed, if structural changes are fully reversible, pressure jumps can be repeated with identical experimental parameters to allow for data averaging, pressure does not normally disrupt intramolecular bonds below 2GPa (20 kbar) While changing pressure is an ideal technique for studying phase changes, it is technically challenging which has limited its use as a common laboratory technique. We have recently used synchrotron light and the pressure jump technique to elucidate and investigate a novel liquid crystalline mesostructure in a hydrated lipid system and are now developing a high performance, robust and user friendly facility for investigation of soft condensed matter systems at high pressure. This facility will be freely available to users at Diamond Light Source and it is hoped that this will allow this highly advantageous technique to become accessible to a much wider section of the community than is currently possible.

Keywords: soft condensed matter, high pressure X-ray diffraction, lipid mesophases

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High-pressure structural studies of pharmaceutical materials

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The phenomenon of polymorphism is of crucial importance to many industries including the pharmaceutical, pigments and energetics industries. Polymorphism is an extremely interesting research topic for the structural scientist, however, the appearance of an unidentified polymorph during the production of an active pharmaceutical ingredient (API) can cause serious problems for both the production and processing of the pharmaceutical. The physical properties of an API e.g. solubility and hence bioavailability can vary dramatically depending on the polymorph that is isolated and so the emergence of a new polymorph requires a cessation of production together with a full investigation of the polymorphism exhibited by the API. For this reason the pharmaceutical industry places great