Keywords: X-ray; benzimidazole; 2-benzoylbenzoate copper complex

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Synthesis, Spectrothermal and Structural Characterization of Saccharinatobis(trishydroxymethylaminomethane)zinc(II) saccharinate Complex, [Zn(sac)(tham),](sac). Humeyra Pasaoglu<sup>a</sup>, Zerrin Heren<sup>b</sup>, Gokhan Kastas<sup>a</sup>, Kamber Akdag<sup>a</sup>. \*aDepartment of Physics, Faculty of Arts and Sciences, Ondokuz Mayis University, Samsun, Turkey. \*bDepartment of Chemistry, Faculty of Arts and Sciences, Ondokuz Mayis University, Samsun, Turkey.

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A novel ZnII complex of the saccharinate ligand (sac) with tris-hydroxymethylaminomethane (tham) was synthesized and characterized by elemental analysis, FT-IR spectroscopy, simultaneous TG and DTA techniques, and X-ray diffraction. The complex, [Zn(sac)(tham),] (sac), crystallizes in monoclinic system with space group P21/c [a = 7.55954(3) Å, b = 13.0532(6) Å, c = 27.7777(10)Å,  $\beta = 100.539(3)^{\circ}$ , Z = 4]. The Zn<sup>II</sup> ion has a distorted octahedral coordination. The tham ligand has chemically different functions in the structure, acting as both bidentate and tridentate ligands. There are also sac moieties to act as N-bonded ligand and as a counter anion. The molecular packing of the complex is provided by moderate hydrogen bonds as well as  $\pi \cdots \pi$  interactions between the sac moieties. The IR spectra and the thermal decomposition of the complex are also discussed.

Keywords: zinc; saccharinato complex; trishydroxymethylaminomethane; crystal structure; thermal decomposition