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Keywords: powder diffractometry; thermal motion in crystal; TLS matrices

FA5-MS01-P26

Substitution Effects in Some Bismuth Based Rare Earth Manganites. <u>Kiril Krezhov</u>^a, Stefan Kovachev^a, Daniela Kovacheva^b, Erzhebet Svab^c, Francoise Bourée^d, Gilles André^d. ^aInstitute for Nuclear Research and Nuclear Energy. ^bInstitute of General and Inorganic Chemistry, Sofia, Bulgaria. ^cResearch Institute for Solid State Physics and Optics, Budapest, Hungary. ^dLaboratoire Leon Brillouin, CEA/Saclay, Gif-sur-Yvette, France. E-mail: <u>krezhov@inrne.bas.bg</u>

In consistence with other authors we established that the characteristic signs attributed to the charge ordering (CO) and orbital ordering (OO) in the parent manganites $Bi_{0.5}Ca_{0.5}MnO_3$ and $Bi_{0.5}Sr_{0.5}MnO_3$ appear at $T_{CO} \approx 320$ K and $T_{\rm co} \approx 525$ K, respectively [1]. We investigated the magnetic ordering and CO/OO phenomena with respect to temperature and chemical doping in $Bi_{0.25}R_{0.25}AA_{0.5}MnO_3$ (R = La, Nd, Ho, Er, Tm) (A-site doping [1-3]) and $Bi_{0.5}AA_{0.5}Fe_xMn_{1.x}O_3$ (B-site doping [4-6]), where AA stands for Ca or Sr. For both sets of Bi-based manganites prepared as polycrystalline powders by controlled solid-state reaction we correlated the structural findings from high-resolution X-ray and neutron powder diffraction with results for transport and magnetic properties. At 295 K Bi_{0.25}R_{0.25}Ca_{0.5}MnO₃ are in paramagnetic phase of small monoclinic distortion allowing for structural description in orthorhombic Pnma symmetry. A significant anisotropic distortion of the MnO₆ octahedra occurs in the characteristic manner associated with the particular orbital order of manganese ions at half doping level. The findings corroborate the conclusion that the effective Bi3+ ionic radius in Bi-Ca manganites is close in value to that of Nd³⁺ indicating that the lone pair character of 6s² Bi³⁺ orbitals is rather constrained than dominant. Except for

 $Bi_{0.25}La_{0.25}Ca_{0.5}MnO_3$ which undergoes a ferromagnetic transition at $T_c = 281$ K followed by a transition to an antiferromagnetic state at lower temperatures, the rest compounds become antiferromagnetic. At 10 K the ground magnetic state of most Ca-Bi compounds could be acceptably described by the insulating antiferromagnetic CE ordering. The RT cell parameters of Bi_{0.5}Ca_{0.5}Fe_xMn_{1.5} $_{\rm x}O_3$ (0 $\leq x \leq 0.6$) change monotonously with increasing x. The compounds are antiferromagnetic with T_N which decreases with increasing Fe-substitution. For $x \ge 0.1$ the antiferromagnetism is accompanied by a weak ferromagnetism. The CO-OO order in the neat form is suppressed at very low Fe-substitution ($x \le 0.05$). Room temperature diffraction data of $Bi_{0.5}Sr_{0.5}Fe_xMn_{1,x}O_3$ ($0 \le x \le x$ 0.3 and cubic for x > 0.4. Unit cell volume increases with x. The charge/orbital order was observed for the compound without Fe. The magnetic state of the compounds with different Fe-content changes from paramagnetic to weak antiferromagnetic below T_N, in the range 116 K - 155 K. All Ca-Fe and Sr-Fe compounds are semiconductors between 100 K and 600 K and there was no magnetoresistivity effect in fields up to 7 kOe.

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Keywords: charge order; magnetic structure; manganites

FA5-MS01-P27

Molecular and Crystalline Structures of Three (s)-4-alkoxycarbonyl-2-azetidinones Containing Long Alkyl Side Chains From Synchrotron X-ray Powder Diffraction Data. Luis E. Seijas^a, Asiloé J. Moraª, Gerzon E. Delgadoª, Francisco Lopéz-Carrasquero^b, Michela Brunelli^c, Andrew N. Fitch^d. ^aLaboratorio de Cristalografía, Departamento de Ouímica, Facultad de Ciencias, Universidad de Los Andes, Mérida, 5101, Venezuela. ^bLaboratorio de Polímeros, Departamento de Química, Facultad de Ciencias, Universidad de Los Andes, Mérida, 5101, Venezuela. °ILL Institut Laue-Langevin, BP 156, 38042 Grenoble cedex 9, France. dEuropean Synchrotron Radiation Facility, BP 220, 38043 Grenoble cedex, France. E-mail: asiloe@ula.ve

The (S)-4-alkoxycarbonyl-2-azetidinones are optically active β -lactam derivatives of aspartic acid, which are used as precursors of carbapenems-type antibiotics and poly- β -aspartates. The crystal structures of three (S)-4-alkoxycarbonyl-2-azetidinone with alkyl chains with

^{25&}lt;sup>th</sup> European Crystallographic Meeting, ECM 25, İstanbul, 2009 *Acta Cryst.* (2009). A**65**, s 327