Aqueous routes used for the synthesis of nanoscale particles of LiFePO_4 may lead, under specific conditions, to a full solid solution behavior upon Li^+ extraction at ambient temperature, hence favoring an easier monitoring of the state of charge/discharge of the electrode [1]. We demonstrated through neutron diffraction that this behavior is strongly connected with the presence of significant amounts of structural defects within the crystallites (Li \Leftrightarrow Fe exchange, vacancies) [2]. We demonstrated as well that it was possible to monitor and adjust the amount of structural defects through careful annealing in air between 140°C and 500°C, up to a composition close to $\text{LiFe}_{0.67}\text{PO}_4$ [3]. An important feature is that these findings suggest unusually high mobility of Fe within the triphylite framework



Figure 1: Proposed model for the superstructure of $\text{LiFe}_{23}\text{PO}_4$ built by analogy with the structure of laihunite $\text{Fe}_{23}\text{SiO}_4$

The resulting powders show complete new electrochemical features (fig. 1) at various redox steps between 3.5 V and 2.8 V vs Li, associated with partial redistribution of Fe and Li within the octahedral sites, followed by *in situ* Mossbauer spectroscopy and X-ray diffraction. New crystal super-structures [4] with long range ordering of Fe, vacancies and/or Li (fig. 2), determined through electron diffraction and high resolution Synchrotron X-Ray and neutron diffraction will be presented.

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A new homologous series of oxygen vacancy ordered strontium manganese perovskite

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Perovskite manganites have been extensively studied during last several decades due to a wide range of functional properties like colossal magnetoresistance, mixed ionic-electronic conductivity, and large magnetocaloric effect. Additionally oxygen deficient manganites are relevant as Solid Oxide Fuel Cell cathode materials.

During a systematic study of structural aspects of oxygen deficient $La_xSr_{1-x}MnO_{3-\delta}$ compounds we determined the structure of a new compound $Sr_7Mn_7O_{19}$ (SrMnO_{2.714}) [1] showing a new oxygen vacancy ordering arrangement. We also found that the previously described vacancy ordered phases $Sr_2Mn_2O_5$ (SrMnO_{2.5}), $Sr_5Mn_5O_{13}$ (SrMnO_{2.6}) and the new phase form a new homes series of compounds with general formula $Sr_{4+n}Mn^{3+}_{4}Mn^{4+}_{n}O_{10-3n}$ that in certain conditions may accept La substitution [2] and may also be extended to well-known compounds in the CaMnO_{3- δ}, LaCuO_{3- δ} and NdCuO_{3- δ} series [3,4,5].

Starting with the n=0 member of the series (Sr₄Mn₄O₁₀), containing pyramidal Jahn-Teller Mn³⁺ cations, a basic building block for the homologous series is built based on four pyramids lying on the 001 plane with the apex of the pyramids pointing alternatively in the in the [100], [010], [0-10] and [-100] directions in a $p_x p_y p_{-x}$ arrangement. This arrangement repeats in the normal direction of the plane replacing x by y and viceversa. The nth member of the series is built by adding n Mn⁴⁺ octahedra to the four pyramids keeping the order of the pyramids and the 2/m symmetry of the original block. The n=1 member Sr₅Mn₅O₁₃ is formed with the building block $p_x p_y Op_{-y} p_{-x}$ while the n=3 member $Sr_7Mn_7O_{19}$ is formed by a combination of $p_xp_yOOOp_yp_{-x}$ and $p_yOp_xOp_{-x}$ $_{x}Op_{y}$ blocks in the x and y direction. The n=0 and 1 members of the series are also found in $LaCuO_{3-\delta}$ and $NdCuO_{3-\delta}$ systems, while only n=0 members are found for SrCuO3-8 and CaMnO3-8 systems, although many stable compounds with stoichiometries compatible with n=2 and 4 have been studied in this latter system.

The lack of observation of n=2 and 4 members of the $Sr_{4+n}Mn^{3+}_{4}Mn^{4+}_{n}O_{10-3n}$ series is supposed to be related to the geometrical incompatibility among the two possible $p_xp_yOOp_{-y}p_x$ and $p_xOp_yp_{-y}Op_{-x}$ building blocks (different from the combination of them in n=3 member) and the reduction of stability of structures where two pyramids are linked by the apical oxygen resulting from these blocks. Additionally, A cation size seems to correlate with the lack of observation of other members of the series in the CaMnO₃₋₆ system.



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Polyene-diphenylaniline D5 dyes and their role in the efficiency of DSSC solar cells

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