conserved water molecule W-1, the order of preference for binding to lactoperoxidase appears to be as Br first, Cl⁻second, SCN⁻ third and I⁻ the last. The positions of these anions in the substrate binding site are further defined in terms of subsites where Br is located in subsite 1, Cl⁻ in subsite 2, SCN⁻ in subsite 3 and I⁻ in subsite 4.

Keywords: antimicrobial, heme, oxidation

MS93.P04

Acta Cryst. (2011) A67, C769

Structure and mechanism of chloromomuconolactone dehalogenase from *Rhodococcus opacus* 1CP

Christian Roth,^a Stefan Kaschabek,^b Janosch A. D. Gröning,^b Michael Schlömann,^b Norbert Sträter^a ^aCenter for Biotechnology and Biomedicine, Institute for Structural Analytics of Biopolymers, Faculty of Chemistry and Mineralogy, University of Leipzig, Deutscher Platz 5, 04103 Leipzig (Germany). ^bEnvironmental Microbiology, TU Bergakademie Freiberg, Leipziger Str. 29, 09599 Freiberg (Germany). E-mail: christian.roth@bbz.uni-leipzig.de

Chloroaromatic compounds are often very persistent environmental pollutants. Nevertheless numerous bacteria are able to metabolise these compounds and use them as the sole energy and carbon source. *Rhodococcus opacus* 1CP is able to degrade 3-chlorocatechol via a unique variant of the modified *ortho*-pathway. This pathway involves chloromuconolactone dehalogenase which dehalogenates the 5-chloromuconolactone to *cis*-dienelactone. The enzyme shows a high similarity to muconolactone isomerases, but is not able to catalyse the isomerisation reaction. In order to characterize the catalytic mechanism of this unusal dehalogenase, we crystallised the enzyme and subjected it to X-ray structural analysis. Datasets of up to 1.65 Å resolution were collected from two different crystal forms using synchrotron radiation. Cocrystallisation with substrate analogs yielded to crystals which are currently analysed for binding, to characterise the mechanism of dehalogenation.

Keywords: biotechnology, biocrystallography, enzyme

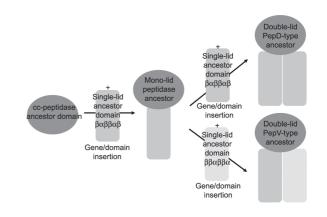
MS93.P05

Acta Cryst. (2011) A67, C769

L-N-carbamoylase structure suggests a striking case of protein evolution

Sergio Martínez-Rodríguez,^a Abel García-Pino,^{c.d} Josefa María Clemente-Jiménez,^a Remy Loris,^{c.d} Felipe Rodríguez-Vico,^a Juan Ma. García-Ruiz,^b Fco. Javier Las Heras-Vázquez,^a Jose Antonio Gavira,^b ^aDpto. Química Física, Bioquímica y Química Inorgánica, Universidad de Almería, Almería, (Spain). ^bLaboratorio de Estudios Cristalográficos. CSIC-Universidad de Granada, Granada, (Spain). ^cDepartment of Molecular and Cellular Interactions, VIB, Brussels, (Belgium). ^dStructural Biology Brussels, Vrije Universiteit Brussels, (Belgium). E-mail: srodrig@ual.es

N-Carbamoyl-L-amino acid amidohydrolases (L-carbamoylases, E.C. 3.5.1.87) are important industrial enzymes used in the kinetic resolution of racemic mixtures of *N*-carbamoyl-amino acids due to their strict enantiospecificity [1]. This work reports the first Lcarbamoylase structure belonging to *Geobacillus stearothermophilus* CECT43 (BsLcar) at 2.7 Å resolution showing the typical lung-shaped scaffold from the peptidase M20/M25/M40 family [2]. This common architecture is formed by two domains which for other homologs are known as catalytic and lid domains, respectively. Structural analysis of BsLcar and several members of the peptidase M20/M25/M40 family reveals several positionally conserved residues in this family, which can be hypothesized to have a critical role in substrate binding for the whole group of enzymes. An unexpected molecule bound in the catalytic cleft confirms large-scale rearrangements toward the lid domain for substrate hydrolysis, and allows us to infer the mechanism governing domain motion of L-carbamoylases. Comparative studies of the lid domain reveal a new "small molecule binding domain" (SMBD) belonging to the ACT-like superfamily. Based on BsLcar structure we propose an evolutionary pathway for all members of the peptidase M20/M25/M40 family through domain fusion and convergent/divergent evolution from common peptidase and SMBD ancestors.



[1] S. Martínez-Rodríguez, A.I. Martínez-Gómez, F. Rodríguez-Vico, J.M. Clemente-Jiménez, F.J. Las Heras-Vázquez, *Applied microbiology and biotechnology* 2010, *85*, 441-458. [2] S. Martínez-Rodríguez, M. Andújar-Sánchez, J.M. Clemente-Jiménez, V. Jara-Pérez, F. Rodríguez-Vico, F.J. Las Heras-Vázquez, *Biochimie* 2006, *88*, 837-847.

Keywords: peptidase, evolution, amidohydrolase

MS93.P06

Acta Cryst. (2011) A67, C769-C770

Crystal structures of cytochrome c peroxidase from *Pseudomonas* stutzeri in active and inactive forms.

<u>A. Mukhopadhyay</u>^a C. Trimoteo,^a C. Bonifacio,^a I. Moura,^a M. J. Romao,^a J. Trincao,^a *aREQUIMTE/CQFB*, Departamento de Química, FCT, Universidade Nova de Lisboa, 2829-516 Caparica, (Portugal).

Bacterial di-heme Cytochrome c Peroxidase (CCP) is essential to maintain H₂O₂ below toxic levels by catalyzing its reduction to H₂O. Bacterial CCP consists of two heme domains, one harboring the electron transfer heme (E heme) and the other the peroxidatic heme (P heme). The crystal structure of the CCP from Pseudomonas stutzeri was obtained in two different redox states. The oxidized form (inactive) crystal structure was refined to 1.6 Å resolution[1]. In this structure the peroxidatic heme is coordinated to six ligands. The reduced form, in the active mixed valence state, was refined to 2.02 Å resolution and has a water molecule bound to the peroxidatic heme. These two structures have significant conformational differences in some regions, in particular around the P heme and the interface between the two domains. Previous studies indicate that CCP from P. stutzeri has a very high affinity for calcium [2]. This property has been addressed from a structural point of view. Structural data will also be obtained for the CCP from the same organism in the calcium free state in the oxidized form. These structures, along with the IN and OUT forms of P. Nautica [3] will help to obtain a better understanding of the electron transfer mechanism within di-heme CCP and also of the role of the calcium in its activation and mechanism.