## Oral Contributions

## [MS13-05] Dithiazyl Radicals -Structures and Charge Densities of their Crystals and Cocrystals

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Purely organic systems can exhibit conductivity, superconductivity or magnetically ordered phases - properties usually thought as restricted only for crystals containing metallic centres (Cu, Mn, etc.). One of the most intriguing groups of such systems is a family of thiazyl radicals.[1] These radicals are chemically stable, so they can be arranged in closely packed structures. Relatively high electrostatic polarization allows for inter- and intramolecular S...N interactions. In the crystal phase, the thiazyl group often coexists with a phenyl molecular fragment, which further stabilizes the crystal lattice by introducing an intermolecular  $\pi$ ... $\pi$  aromatic interactions (e.g. phenyl...perfluorophenyl stack interaction). The spin structure of these compounds is strongly coupled to the crystal structure. One can then try to adjust the magnetic properties (e.g. FM-AFM ordering) of such systems by small changes of structural parameters (e.g. distances between molecules in stack). This and other intriguing properties suggest that the dithiazyl radicals are promising candidates for the construction of molecular devices. No experimental electron density distribution of such system in the crystal phase has been accomplished so far. The scope of the work was to determine the quantitative electron density distribution and its parameters ( $\rho$  and  $\Delta \rho$  in critical points, integrated charges, etc.) for the three very interesting model crystals of thiazyl radicals belonging to dithiadiazolyl Hansen-Coppens multipole family.[2] The expansion of electron density model was refined

against the high resolution (sin $\theta / \lambda > 0.7$ Å-1) X-ray diffraction data to obtain the best models of the electron density distribution in given crystals. These models were then used to calculate quantitative electron density properties using the Bader's Quantum Theory of Atoms in Molecules (QTAIM) such as critical points parameters  $\rho$ CP,  $\Delta \rho CP$ , bond paths), atomic basins or integrated parameters electron densitv (integrated charges, atomic multipoles and volumes, etc). The obtained results and detailed analysis of dithiadiazolyl radicals should hopefully help in a better understanding of the magnetic phenomena in organic systems.

[1] Rawson, J. M., Alberola, A., Whalley, A., J. Mater. Chem. 16, (2006), 2560-2575.

[2] C. Allen, D.A. Haynes, C.M. Pask, J.M. Rawson, CrystEngComm 11, (2009), 2048-2050.

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