Microsymposium

Crystal Structure and Dielectric Property of Endohedral Lithium Fullerene

<u>S. Aoyagi</u>¹, K. Sugimoto², H. Okada³, N. Hoshino⁴, T. Akutagawa⁴

¹Nagoya City University, Department of Information and Biological Sciences, Nagoya, Japan, ²SPring-8/JASRI, Kouto, Sayo, Hyogo, Japan, ³The University of Tokyo, Department of Chemistry, Tokyo, Japan, ⁴Tohoku University, IMRAM and Graduate School of Engineering, Sendai, Japan

Endohedral lithium fullerene Li+@C60 can have a dielectric polarization by the off-centered location of the Li+ cation inside the C60 cage. The x-ray structure analysis of the PF6- salt [Li@C60](PF6) revealed that the Li+ cation occupies two off-centered equivalent positions at 20 K and hence the crystal is non-polar [1]. The disordered structure at low temperature is explained by a static orientation disorder of polar Li+@C60 cations and/or a dynamic tunneling of the Li+ cation inside the C60 cage. The Li+ tunneling would be suppressed by an intermolecular interaction at lower temperature and a dielectric phase transition might be induced. We reveal the dielectric property and crystal structure of [Li@C60](PF6) below 20 K in this study. The temperature dependence of the dielectric permittivity was measured for the single crystal down to 9 K. The dielectric permittivity increases with decreasing temperature according to the Curie-Weiss law. Such a behavior was also observed in H2O@C60 crystal but not in empty C60 crystal [2]. No dielectric phase transition was observed in H2O@C60](PF6) around 18 K. The single-crystal x-ray diffraction experiment below 20 K was also performed at SPring-8 BL02B1. The crystal has a cubic structure at 20 K [1]. The temperature dependence of the cubic lattice constant shows no anomaly around 18 K. However, diffraction peaks that are forbidden for the given structure appear below 18 K. Thus the crystal symmetry is lowered by the dielectric phase transition. We present the result of the crystal structure analysis of the newly discovered low-temperature phase.

[1] S. Aoyagi, Y. Sado, et al., Angew. Chem. Int. Ed., 2012, 51, 3377-3381, [2] S. Aoyagi, N. Hoshino, et al., Chem. Commun., 2014, 50, 524-526

Keywords: Fullerene, Endohedral Metallofullerene, Crystal Structure Analysis