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New electronic perovskite oxides from beyond high pressure synthesis

J. Attfield¹

¹*University of Edinburgh, Centre for Science at Extreme Conditions and School of Chemistry, Edinburgh, UK*

Materials synthesised from high pressure (HP) conditions offer a variety of possibilities for the discovery of new electronic and magnetic phenomena. Recovery of a solid to ambient pressure (AP) from its HP thermodynamic stability field effectively introduces a large negative pressure that can drive the material into an unusual electronic ground state; as illustrated by HP structural and property studies of PbRuO₃ and BiNiO₃. PbRuO₃ is synthesised at 10 GPa and shows an orbitally ordered phase at AP and low temperatures that is suppressed at an apparent quantum critical point near 6 GPa [1]. A new structural phase emerges at much higher pressures ~30 GPa. BiNiO₃ synthesised at 6 GPa shows an unusual charge order that is suppressed at 3.4 GPa; the transition is associated with a volume collapse leading to colossal negative thermal expansion [2]. Materials recovered from HP are also precursors for 'hard-soft' chemistry, where the instability of a dense precursor from 'hard' HP conditions is relieved through 'soft' post-synthesis modification. We recently demonstrated this concept for the HP phase SrCrO₃ which on reduction gives two new phases SrCrO_{2.80} and SrCrO_{2.75} with long period oxygen vacancy, charge and spin ordered superstructures [3]. These studies have been carried out in collaboration with co-authors of the papers below.

[1] S. A. J. Kimber, J. A. Rodgers, W. H. Hua, C. A. Murray, D. N. Argyriou, A. N. Fitch, D. I. Khomskii, J. P. Attfield. *Phys. Rev. Lett.* 102, 046409 (2009); A.F. Kusmartseva, A. Sinclair, J.A. Rodgers, S.A.J. Kimber and J.P. Attfield. *Phys. Rev. B* 87, 1651, [2] M. Azuma, S. Carlsson, J. Rodgers, M.G. Tucker, M. Tsujimoto, S. Ishiwata, S. Isoda, Y. Shimakawa, M. Takano and J. P. Attfield. *J. Am. Chem. Soc.*, 129, 14433 (2007); M. Azuma, W.T. Chen, H. Seki, M. Czapski, O. Smirnova, K. Oka, M. Mizumaki, T. Watanuki,, [3] A.M. Arevalo-Lopez, J.A. Rodgers, M.S. Senn, F. Sher, J. Farnham, W. Gibbs and J.P. Attfield. *Angew. Chem.* 51, 10791, (2012); A.M. Arevalo-Lopez, F. Sher, J. Farnham, A.J. Watson and J.P. Attfield. *Chem. Mat.* 25, 2346, (2013).

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