## MS30-P4 Synthesis and crystal structures of new di- and polynuclear silver(I) saccharinate complexes with tertiary monophosphanes

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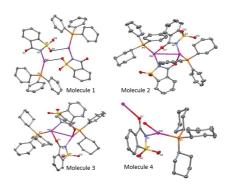
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Over the years there has been a continuous interest in the biological activity of silver(I) metal complexes. Recently, we reported a number of silver(I) sac and found that these complexes have significant antibacterial and anticancer activities [1]. As an extension of our studies, in this work we synthesized and studied the crystal structures of four new silver(I) sac complexes containing the monophosphane ligands, namely  $[Ag(\mu-sac)(PPh_2)]_2$ (1),  $[Ag(\mu-sac)(PPh_2Cy)]_2$  (2),  $[Ag(\mu-sac)(PPhCy_2)]_2$  (3)  $[Ag(\mu-sac)(PCy_3)]_n$ where and **(4)**, PPh,Cy triphenylphosphane, cyclohexyldiphenylphosphane, PPhCy. dicyclohexylphenylphosphane and tricyclohexylphosphane (see Figure.) The silver(Ĭ) ion in complex 1 is coordinated by a sac ligand via the N atom and a PPh, ligand, forming a [Ag(sac)(PPh<sub>2</sub>)] unit and the [Ag(sac)(PPh<sub>3</sub>)] units are further doubly bridged by relatively long Ag-O<sub>sulf</sub> bonds, leading to a dinuclear structure [Ag(sac)(PPh<sub>3</sub>)]<sub>2</sub>. The dimeric units of complex 1 are connected by weak intermolecular interactions such as C-H...O,  $\pi(\text{sac})$ ...  $\pi$  (PPh<sub>2</sub>) and CH...  $\pi$  (sac and PPh<sub>3</sub>). Complexes 2 and 3 show a structural similarity. Each complex also exhibits significant interactions between the silver(I) centers. The Ag-Ag distances are comparable with literature but significantly smaller than that found in other dinuclear sac complexes showing strong argentophilic interactions with Ag - Ag distances around 2.90 Å. Complex 4 is a one-dimensional metallopolymer in which the silver(I) ions are bridged by the sac ligands through the imino N and the carbonyl O atoms, leading to a linear one dimensional polymeric chain running along the b-axis. In all four crystal structures, the di- and polynuclear structures are achieved by the N/O bridging mode of the sac ligand. All phosphane ligands act as a monodentate donor through the P atom. Complexes 2 and 3 exhibit strong argentophilic interactions.

## References:

[1] V.T. Yilmaz, E. Gocmen, C. Icsel, M. Cengiz, S.Y. Susluer, O. Buyukgungor, J. Biol. Inorg. Chem. 19 (2014)



**Figure 1.** View of the molecular structures. Hydrogen atoms were omitted for clarity. The probability level of thermal ellipsoids are 30%.

Keywords: Crystal structure, Saccharinate, Phosphine