MS30-P31 Coordination polymers of the types $[MX_2(4-\text{cypy})_x]_n$ and $[MX_2\text{py}_x]_n$: syntheses, polymorphism and structure relations

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Syntheses and structural characterizations of coordination polymers have constituted a rapidly expanding field of research in the last decades, due to their interesting physical properties and potential applications. [1]

We report on the polymorphism of $[CoBr_2(4-cypy)_2]_n$ and $[NiCl_2[4-cypy)_2]_n$ (4-cypy = 4-cyanopyridine). Both phases show octahedral coordination of the M^{II} ions, which is built up by two N-donor ligands and completed by bridging halide ions in the equatorial plane. This motif typically leads to chains in the $[MX_2(4-cypy)_2]_n$ and the $[MX_2py_2]_n$ phases with $M^{II} = Mn$, Co, Ni, Cu, Pd, X = Cl, Br.

For the phases of the types $[MX_2(4\text{-cypy})_x]_n$ and $[MX_2\text{py}_x]_n$, (x < 2), the degree of polymerization increases with decreasing x which results in the constitution of double chains, bands or planes, depending on the structural requirements of the bridging halide ions and the potential bidentate co-ligand 4-cyanopyridine.

The structure relations of the above mentioned coordination polymers with different values of x are discussed. [2-5]

References:

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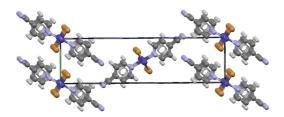


Figure 1. Crystal structure of $[CoBr_2(4-cypy)_2]_n$ projected along a. Monoclinic space group P 1 2 /n 1. Pseudo orthorhombic, minimal non-isomorphic supergroup P 2/n 2 /n 2 /n 2.

Keywords: Coordination polymers, structure relations, crystal engineering, polymorphism, condensed networks, substituted pyridine ligands

MS30-P32 A row of (XeF5+)M2+(SbF6)3 (M=Mg, Ni, Mn, Co, Cu, Zn) compounds: peculiarities of crystal structure, limit of isomorphic substitution, merohedral twinning

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Mixed-cation compounds of general (XeF5+)M2+(SbF6)3 (M=Mg, Ni, Mn, Co, Cu1, Zn) have been synthesized and structurally investigated using single crystal diffraction technique. A wide row (Mg, Mn, Co, Cu, Zn) derivatives are isotypic and crystallize in a monoclinic P21/n space group with a β angle varying from 90.064(3) (Zn) to 90.432(2) (Co). Apex-shared MF6 and SbF6 octahedra are interconnected into infinite tridimensional framework with cavities, occupied by XeF5+ cations (Fig. 1). All investigated (XeF5+)M2+(SbF6)3 salts demonstrate more or less pronounced merohedral twinning with the same twinning law. Increasing of ionic radii (Hg) has led to a formation of completely different (XeF5)3[Hg(HF)]2(SbF6)7 compound.

Keywords: Mixed-cation compounds