MS106.P15

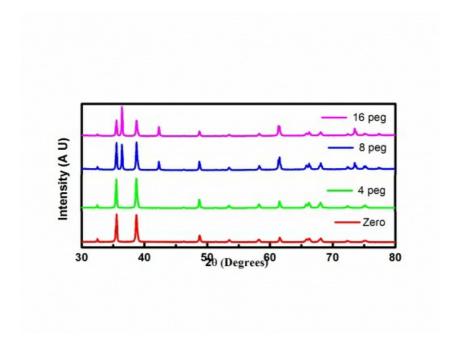
Phase Transformation of Cu₂O from CuO in Presence of PEG

<u>Sangishetti Uday Kumar</u>¹, Ajay Kumar Vaddepally¹, Pavani Katikaneani²

¹Metallurgical And Materials Engineering , Rajiv Gandhi University Of Knowledge T , Nirmal, India, ²Chemistry , Rajiv Gandhi University of Knowledge Technologies-Basar , NIrmal , India
E-mail: uday.mme@gmail.com

This work proposes a simple method for the copper oxide nanostructures based on simple heating method under ambient conditions. Polyethylene glycol (PEG) is employed as a structure directing agent in driving the morphology and phase transformation. Typically, Copper oxide nanoparicles of size below 100 nm were synthesized at temperature around 600°C. The morphology and mechanism were studied by scanning electron microscopy and energy dispersive X-ray spectroscopy. Interestingly this work demonstrates the structural phase transformation of tenorite (CuO) to cuprite (Cu2O) upon addition of different amount of PEG (say 4 g, 8 g, 12 g) and then heat treating at 600°C.

- [1] Nadia Nasihat Sheno, Ali Morsali, Sang Woo Joo, (2014). Materials Letters, 117, 31-33.
- [2] Sawsan Dagher, Yousef Haik, Ahmad I. Ayesh, Nacir Tit, (2014). Journal of Luminescence, 151, 149-154.
- [3] A. El-Trass, H. ElShamy, I. El-Mehasseb, M. El-Kemary, (2012). Applied Surface Science, 258, 2997-3001.



Keywords: nanoparticles; tenorite; cuprite; phase transformation; structure directing agent.