relationship for compounds containing only these elements. For $C_a H_b$ compounds this expression estimates 75.0% of entries with $E \le 5\%$, the highest value for all families. However, deviations are marked for aromatic compounds in which there are important $\pi - \pi$ interactions between the delocalized π -systems (Hunter & Sanders, 1990). On the other hand, for compounds containing N and/or O atoms, the number of entries with $E \le 5\%$ decreases, possibly due to the existence of C-H-O and N—H…O contacts in the crystal structures (Desiraju, 1991; Taylor & Kennard, 1984), which produces a more compact crystal packing and a decrease in the average volume per atom. This effect also occurs in compounds containing F atoms in which there are contacts of the types C-H...F and N-H...F (Taylor & Kennard, 1982; Emerson, Román, Luque, Guitérrez-Zorrilla & Martínez-Ripoll, 1991). In this latter case the average volume per atom decreases and the more accurate formula is $F_{1/2}$ in which the average volume (12 Å³) is smaller than that in $F_{1/3}$ (14 Å³).

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