

Supplementary Material for

[Ni(MeCN)(H₂O)₂(NO₃)₂] (15-crown-5) MeCN: Detailed Study of a Four-Phase Sequence

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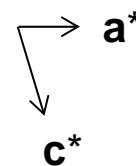
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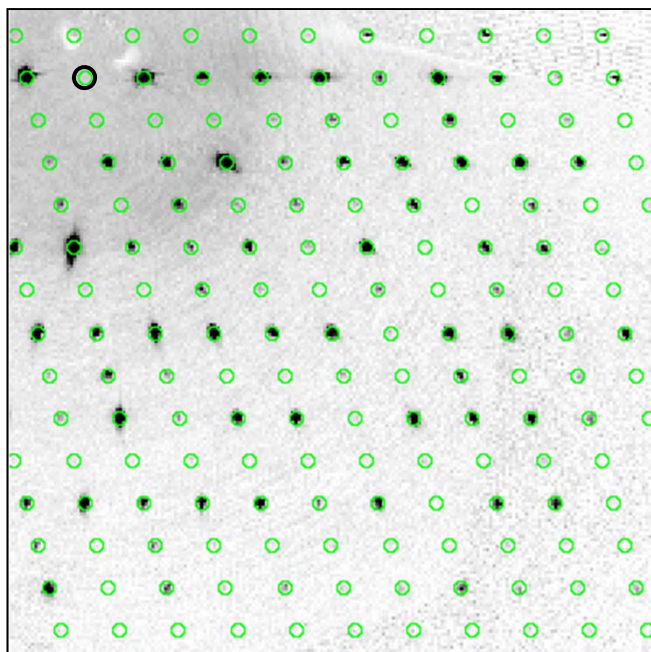
Synopsis: The structure of single-crystal acetonitrilediaqua- $\kappa^1 O$ -nitrate- $\kappa^2 O$ -nitratonickel(II) 15-crown-5 acetonitrile solvate was studied at intervals of *ca.* 10 K in the range 90 – 273 K as the crystal passed, without obvious damage, through three phase transitions. The phase sequence, which includes an intermediate, modulated phase, is very similar to that found for [Ni(H₂O)₆](NO₃)₂·(15-crown-5)·2H₂O.

1. Parts of *h1l* reciprocal-lattice slices at 230, 233, 236, and 239 K showing the diffraction pattern during the transition into and out of phase II (1 pg.)
2. Parts of *h0l* reciprocal-lattice slices in the range 90 – 140 K (10 K intervals) showing the decrease in intensity of the *h0l*, *l* odd reflections with the approach of the transition from phase III to phase IV (1 pg.)
3. Reciprocal-lattice slices for the *B2₁* phase at 90 K and phase II at 233 K that show the two structures are probably the same (8 pp.)
4. Reciprocal-lattice slices *nkl*, *n* = 0 – 3 for the *P2₁* cell at 90 K that show that there is no **b*** component to the streaks (4 pp.)
4. Ellipsoid plot for the *B2₁* structure (assumed to be phase II) at 90 K (1 pg.)
5. Detailed description of the disorder in the *B2₁* structure (assumed to be phase II) at 90 K (3 pp.)
6. Cell dimensions determined for the basic cell (*P2₁/m*, *Z'* = ½) of phase II within its range of stability (1 pg.)

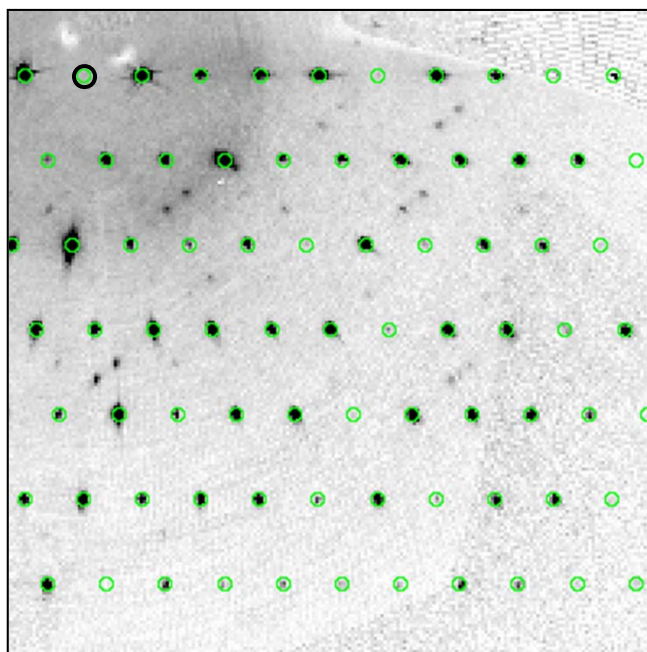
$h1l$ reciprocal-lattice slices at four temperatures within and near the stability range of phase II (indexing in the $P2_1/m$ phase I cell = basic cell)



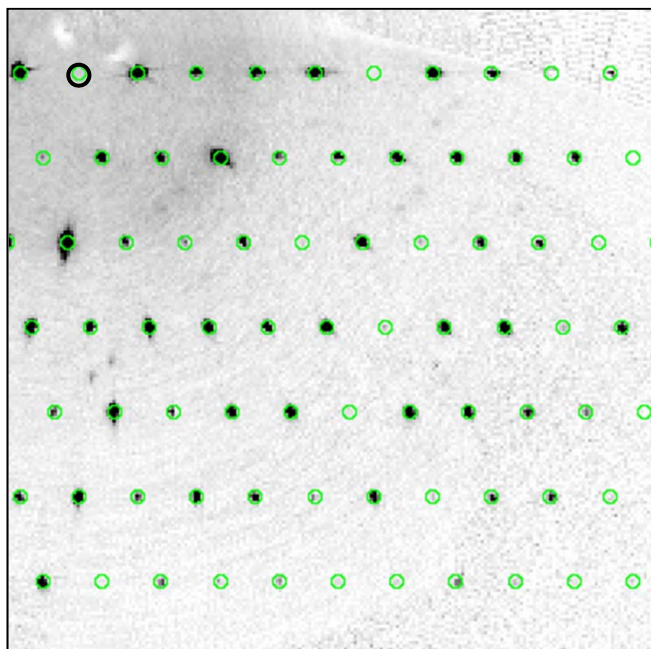
230 K (phase III)



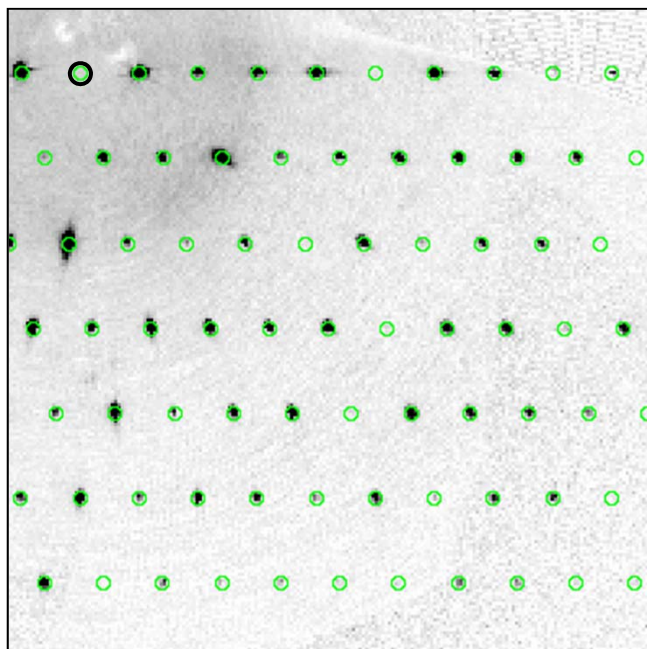
233 K (phase II)



236 K (phase II)

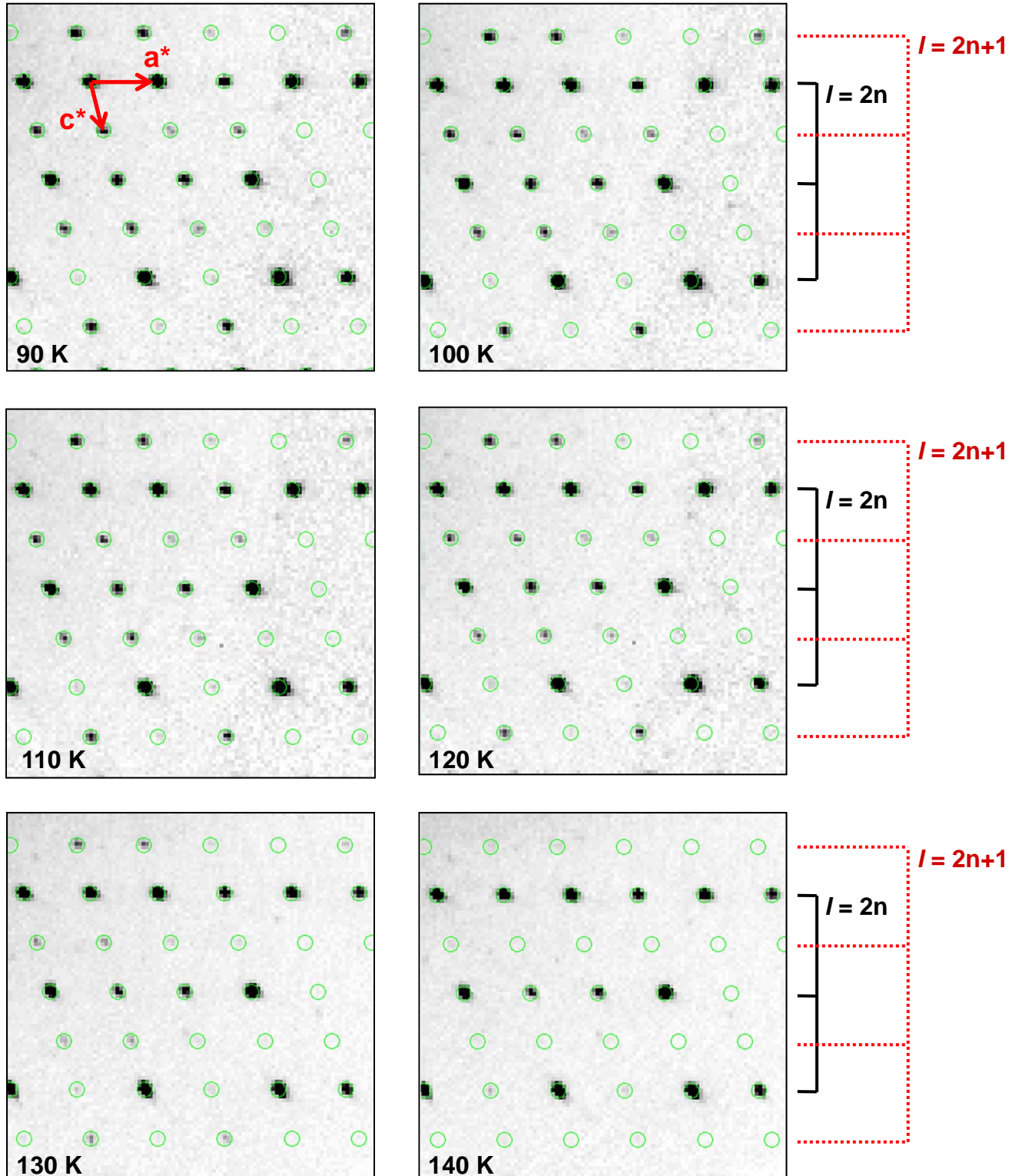


239 K (phase I?)



(The green circles mark calculated positions of reflections)

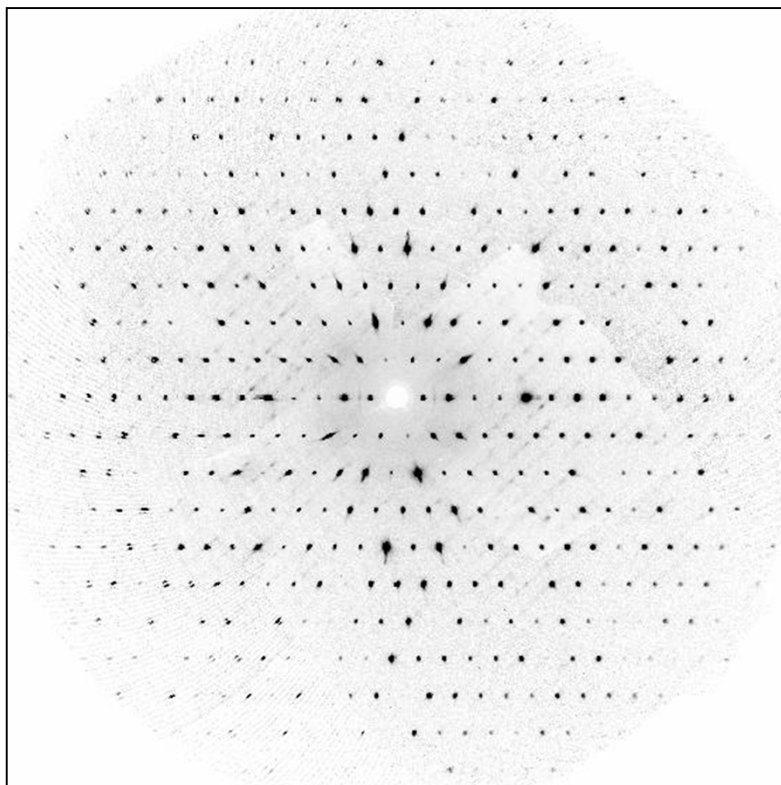
Parts of $h0l$ reciprocal-lattice slices as the transition IV \rightarrow III is approached from below
(axial system originates at reflection 4 0 2)



(The green circles mark calculated positions of reflections)

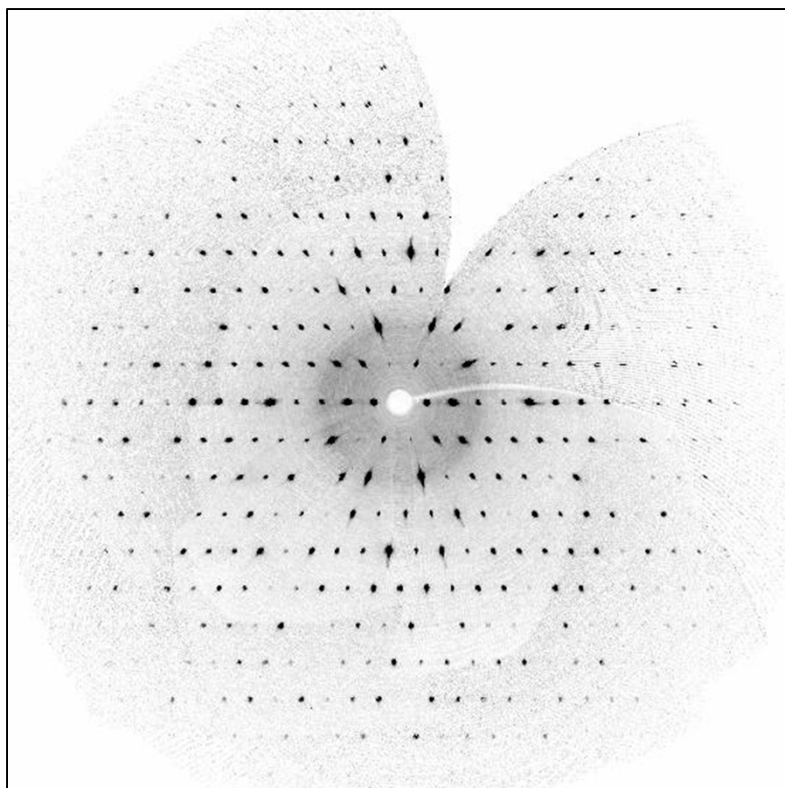
Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

$h0l$



$B2_1$ at 90 K

→ a^*

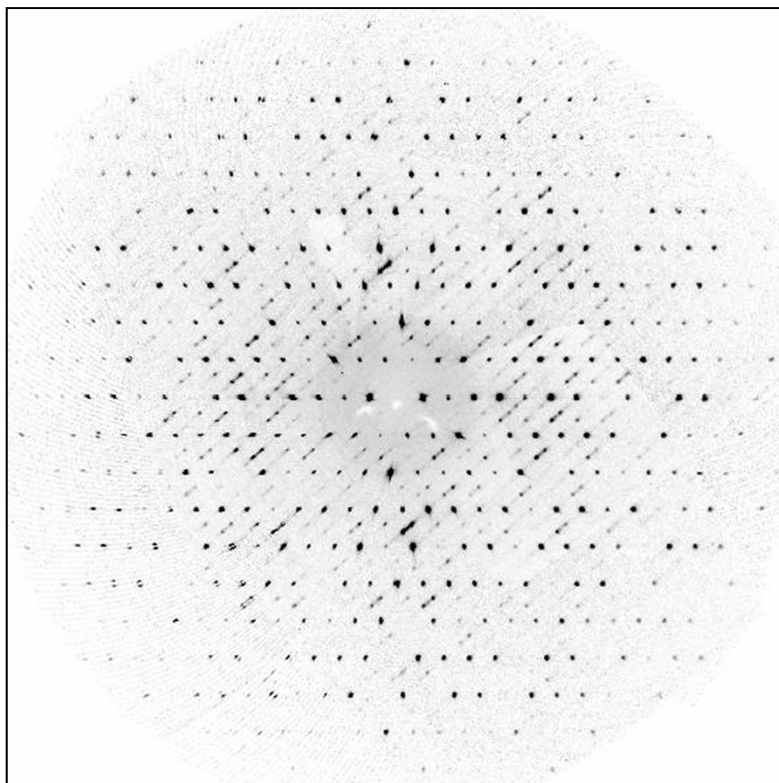


phase II at 233 K

→ a^*

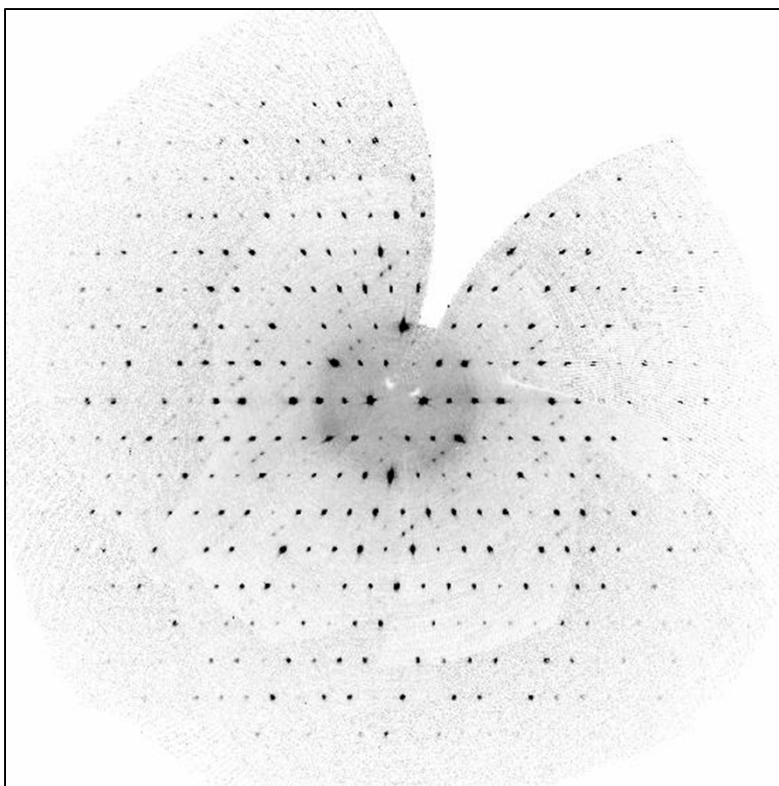
Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

$h1l$



$B2_1$ at 90 K

→ a^*

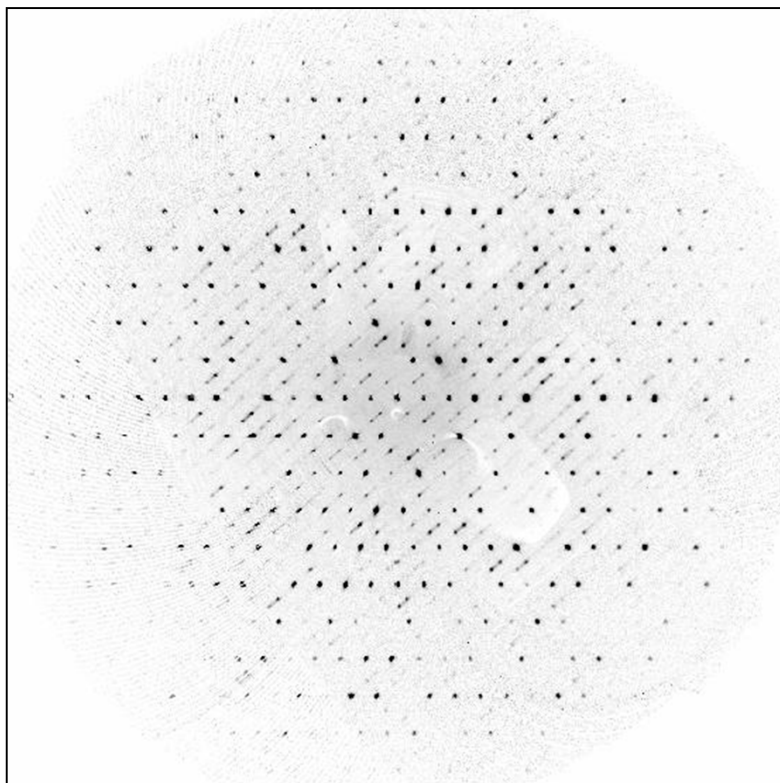


phase II at 233 K

→ a^*

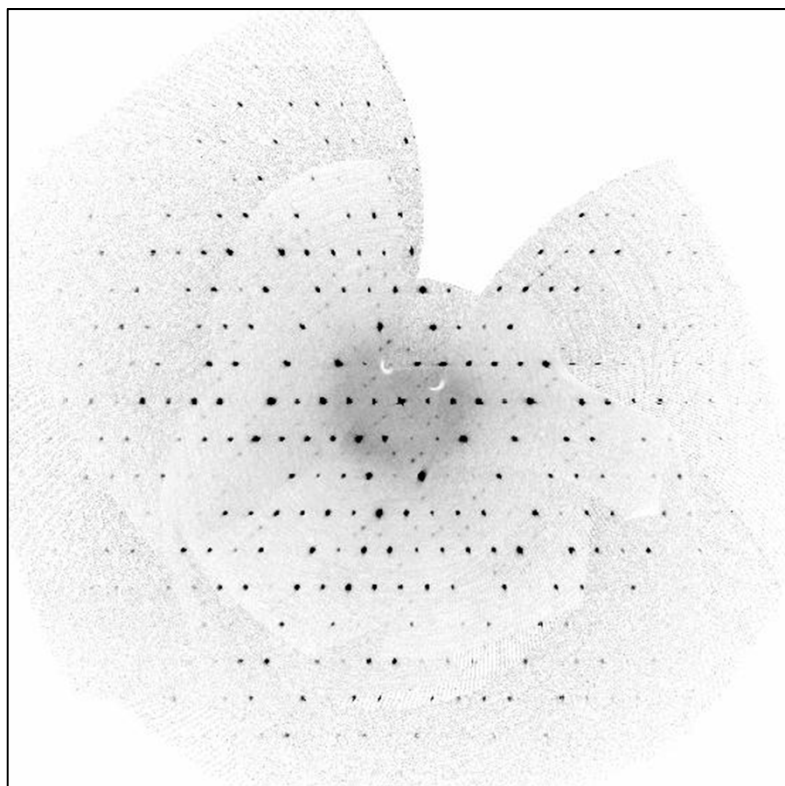
Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

$h2l$



$B2_1$ at 90 K

→ a^*

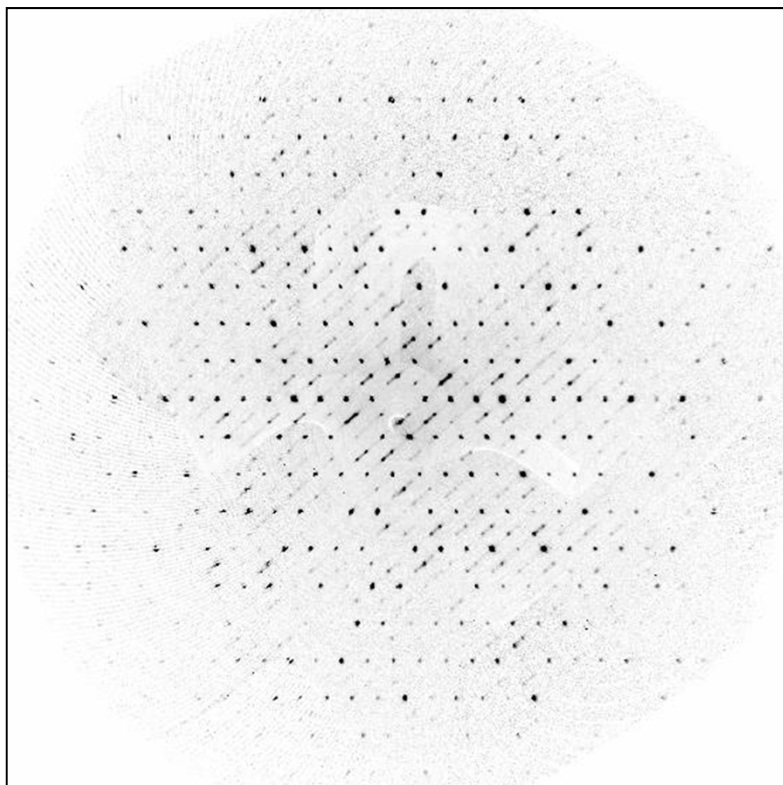


phase II at 233 K

→ a^*

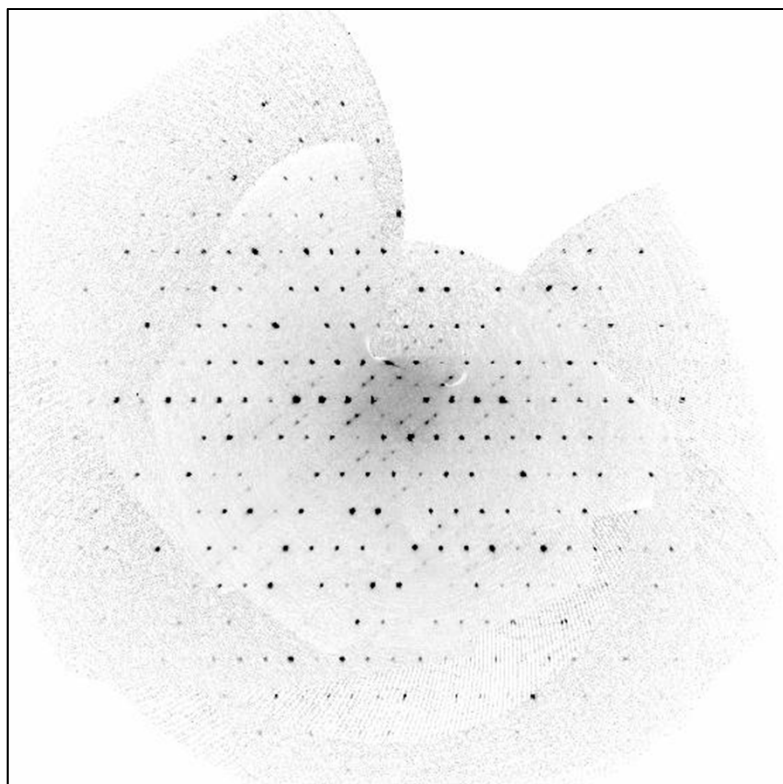
Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

$h3l$



$B2_1$ at 90 K

→ a^*

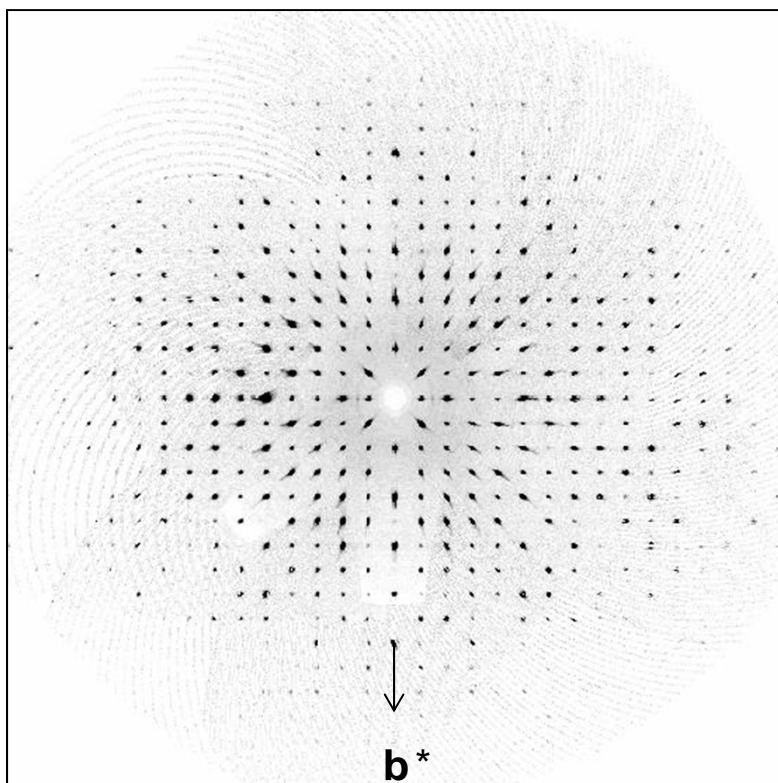


phase II at 233 K

→ a^*

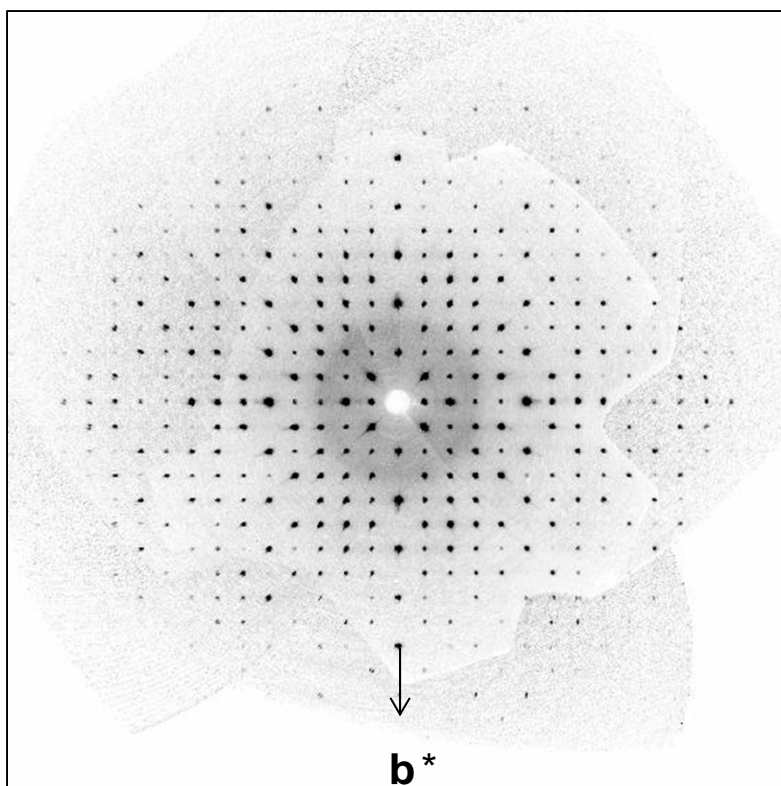
Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

$hk0$



$B2_1$ at 90 K

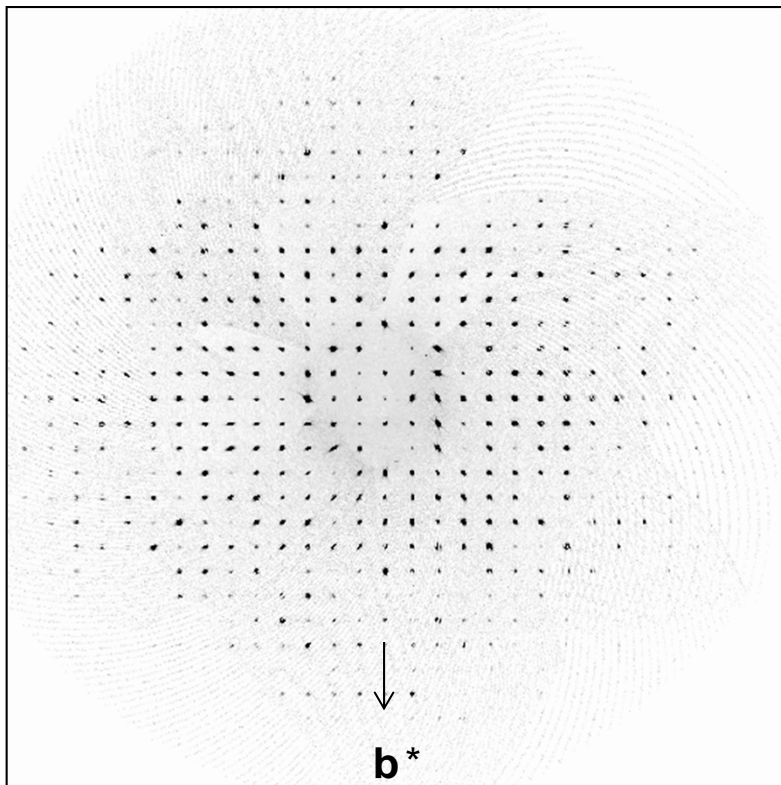
→ a^*



phase II at 233 K

→ a^*

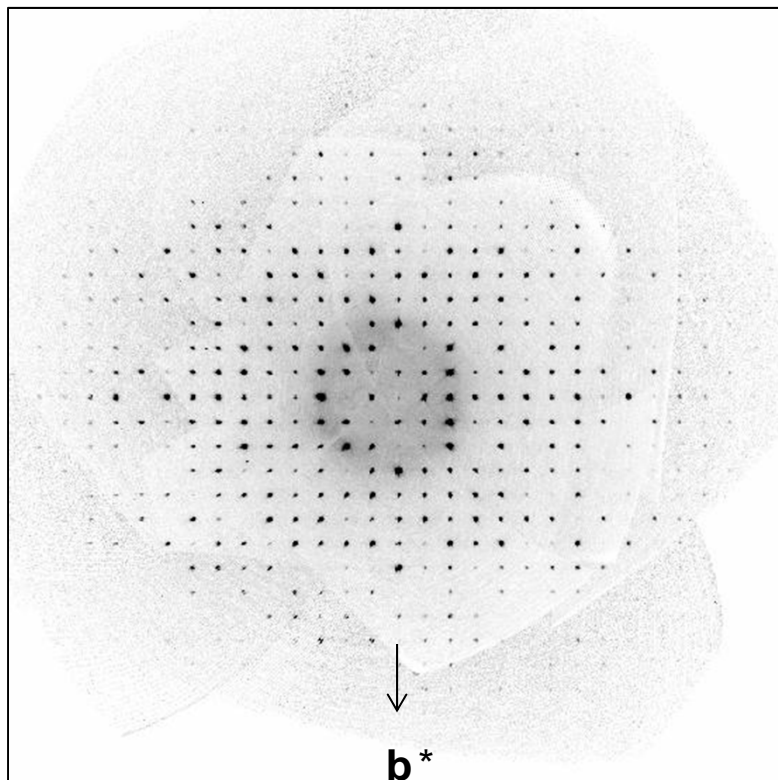
Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K
 $hk5$ ($B2_1$) and $hk1$ ($P2_1/m$)



$B2_1$ at 90 K

→ a^*

↓ b^*



phase II at 233 K

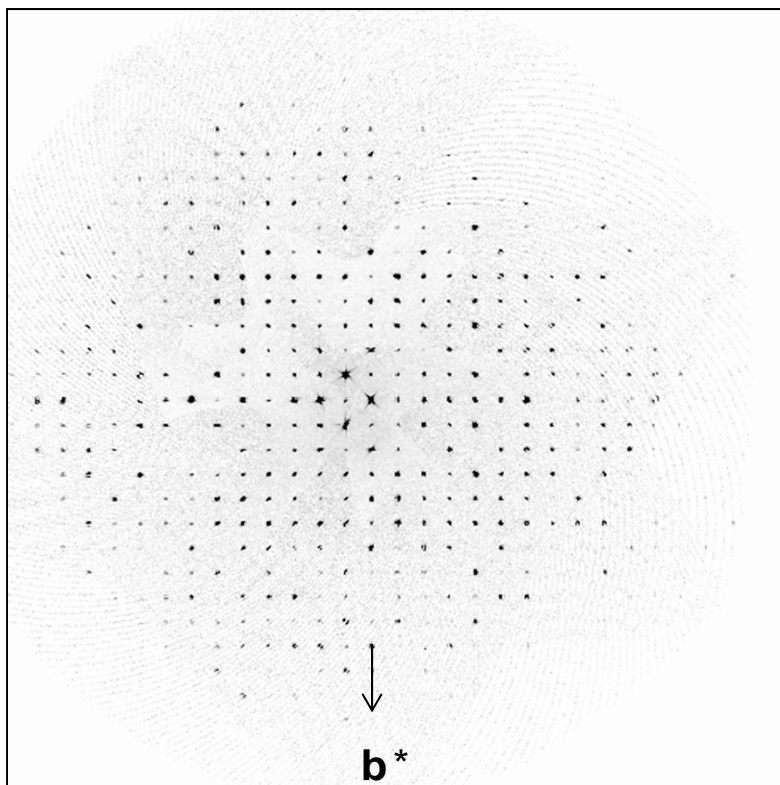
→ a^*

↓ b^*

(There is an origin shift
along a^* between the
two cells because the
two c^* axes are not
parallel)

Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

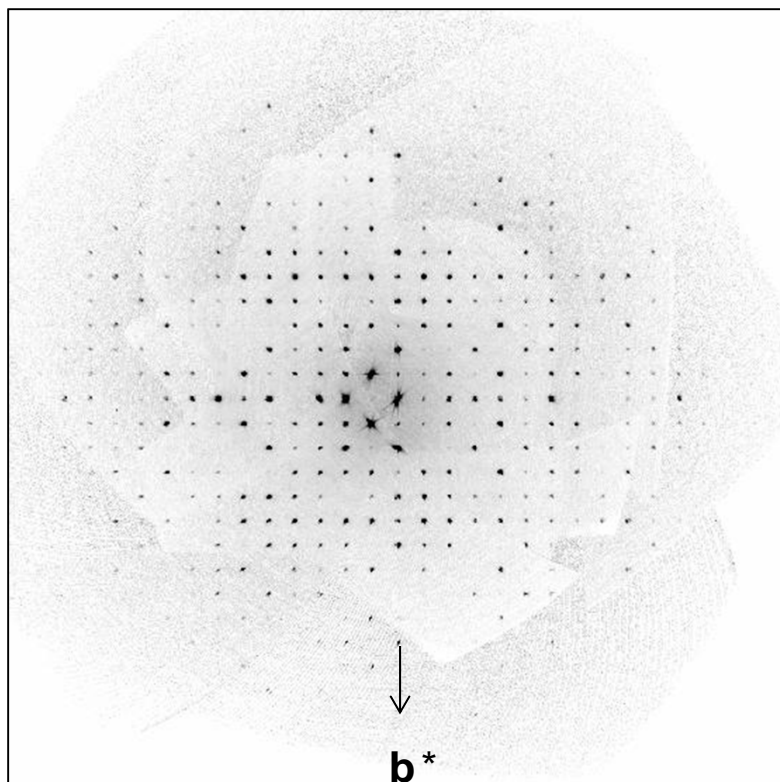
$hk 10$ ($B2_1$) and $hk2$ ($P2_1/m$)



$B2_1$ at 90 K

→ a^*

b^*



phase II at 233 K

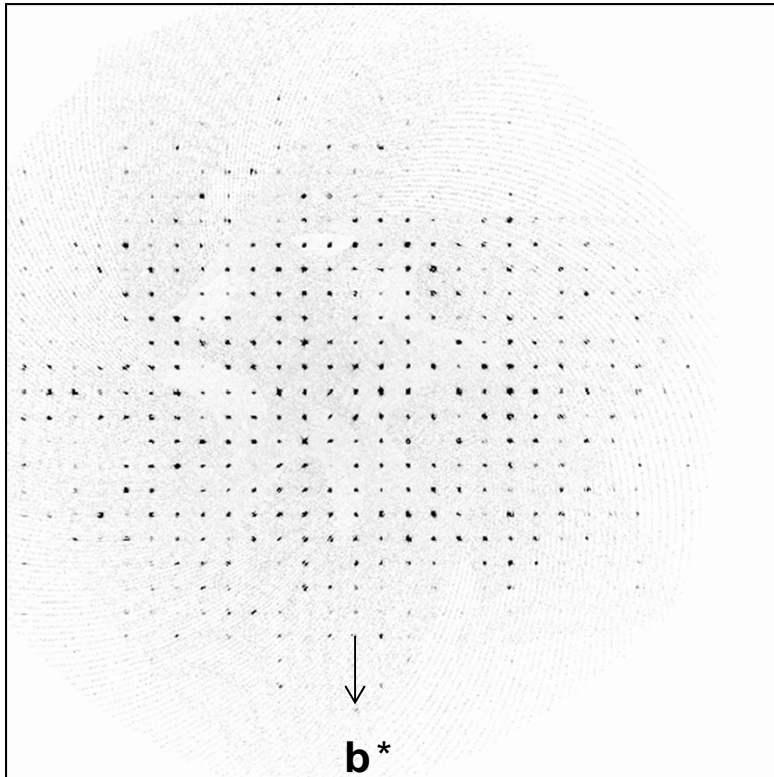
→ a^*

b^*

(There is an origin shift
along a^* between the
two cells because the
two c^* axes are not
parallel)

Reciprocal lattice slices for $B2_1$ at 90 K and phase II
(indexed in $P2_1/m$) at 233 K

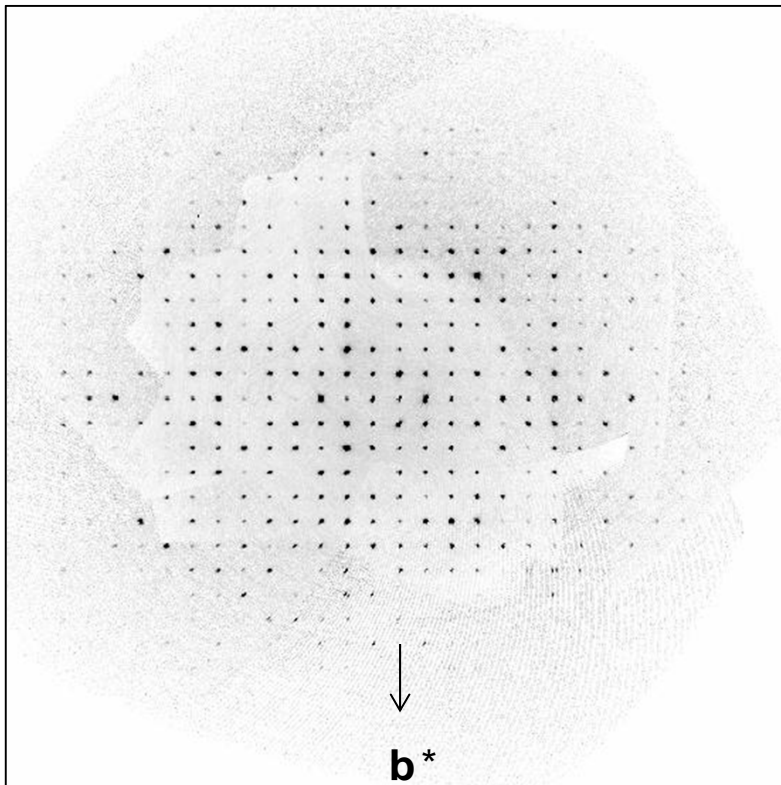
$hk 15$ ($B2_1$) and $hk3$ ($P2_1/m$)



$B2_1$ at 90 K

→ a^*

b^*



phase II at 233 K

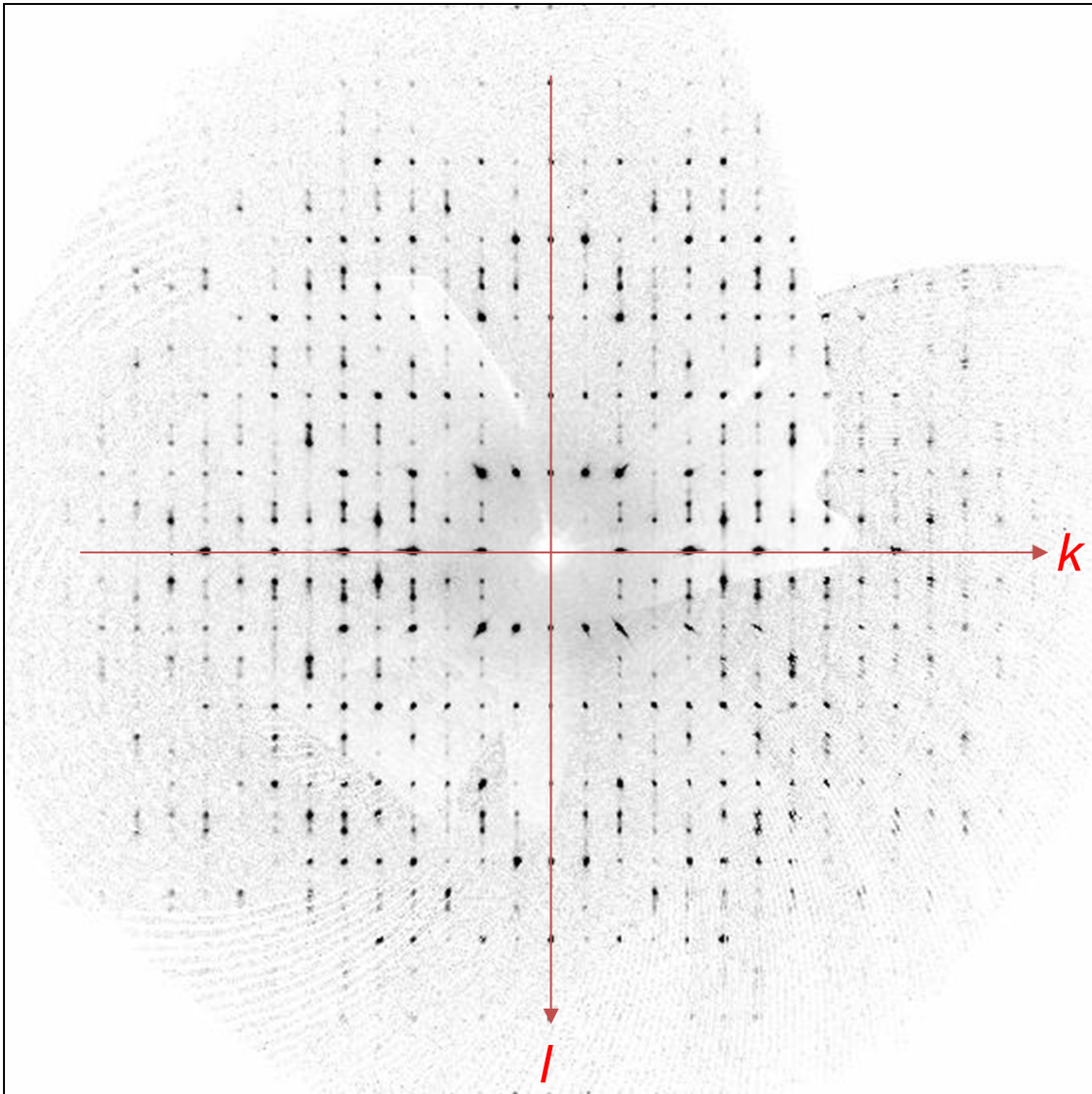
→ a^*

b^*

(There is an origin shift
along a^* between the
two cells because the
two c^* axes are not
parallel)

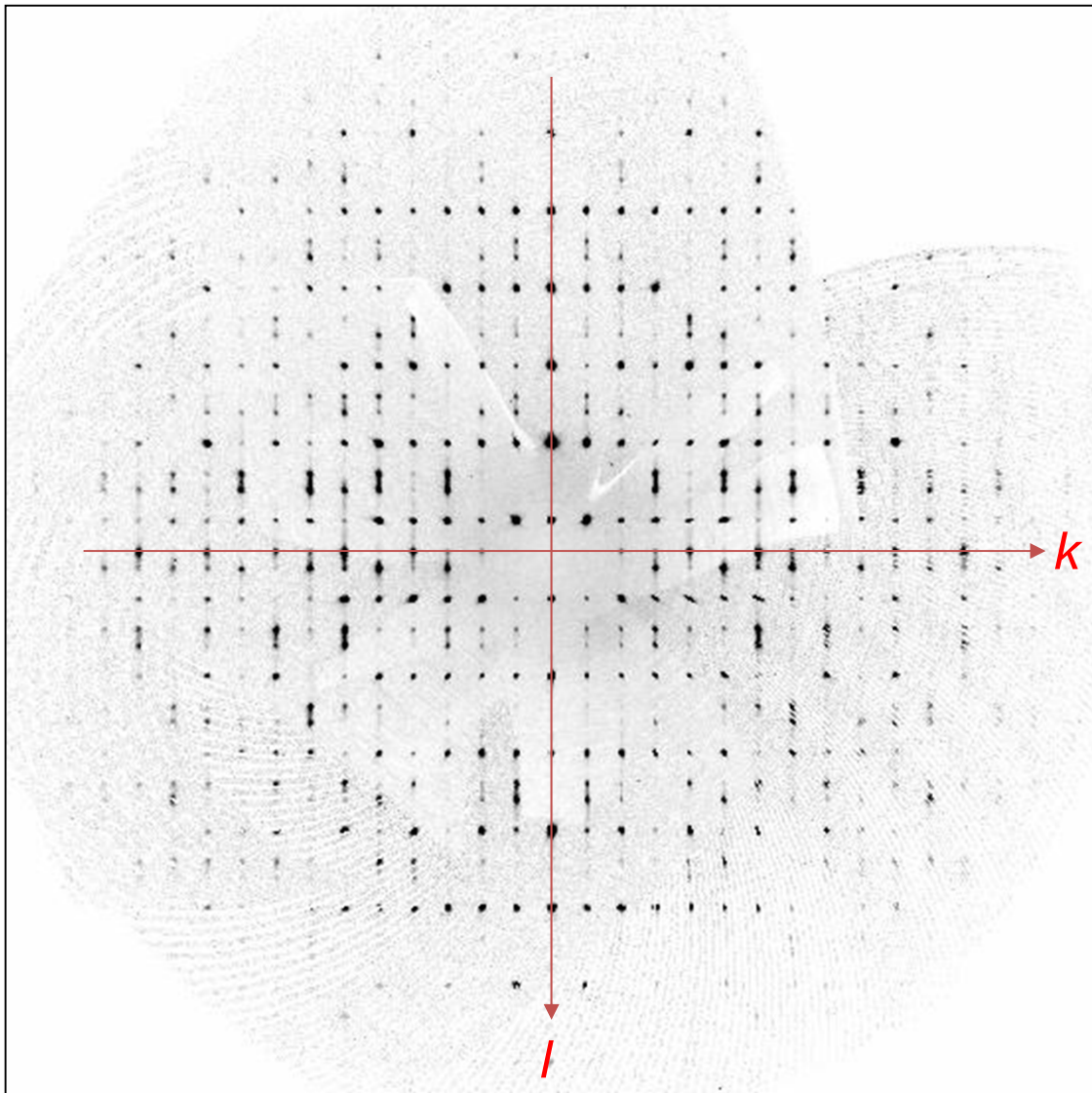
Reciprocal Lattice Slice for $P2_1$ (standard Cell) at 90 K

$0kl$ ($P2_1$ at 90 K)



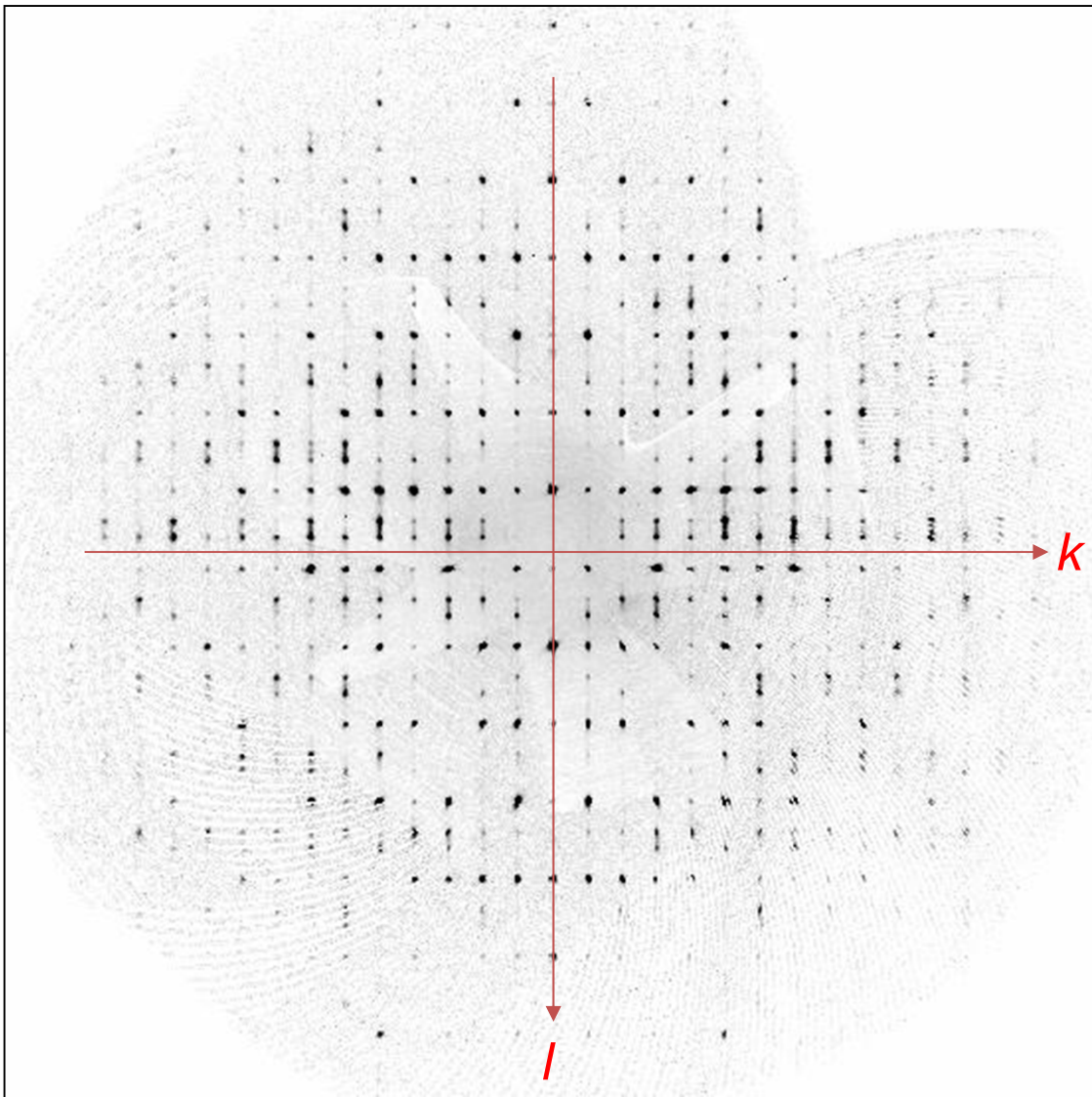
Reciprocal Lattice Slice for $P2_1$ (standard Cell) at 90 K

$1kl$ ($P2_1$ at 90 K)



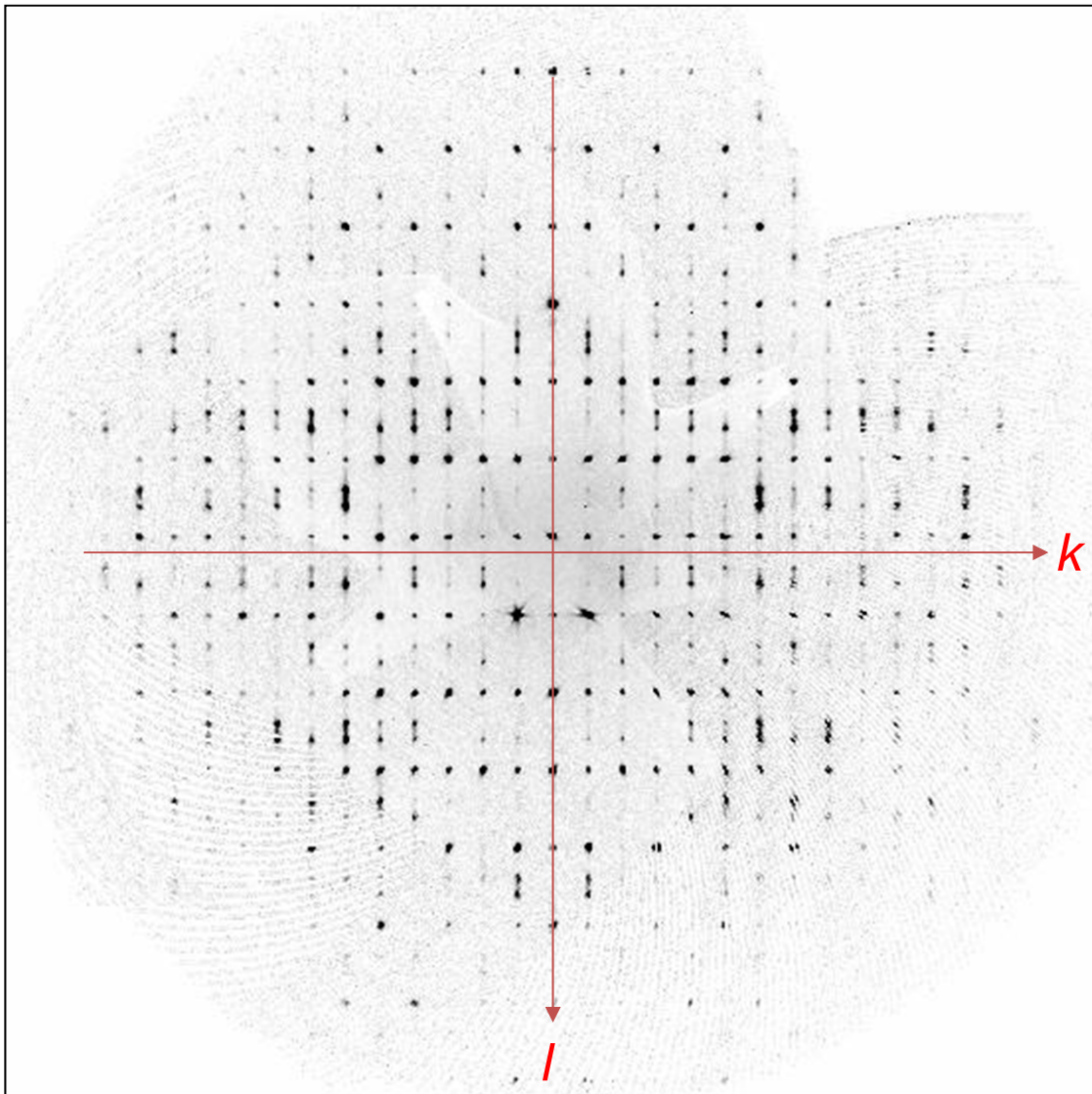
Reciprocal Lattice Slice for $P2_1$ (standard Cell) at 90 K

$2kl$ ($P2_1$ at 90 K)

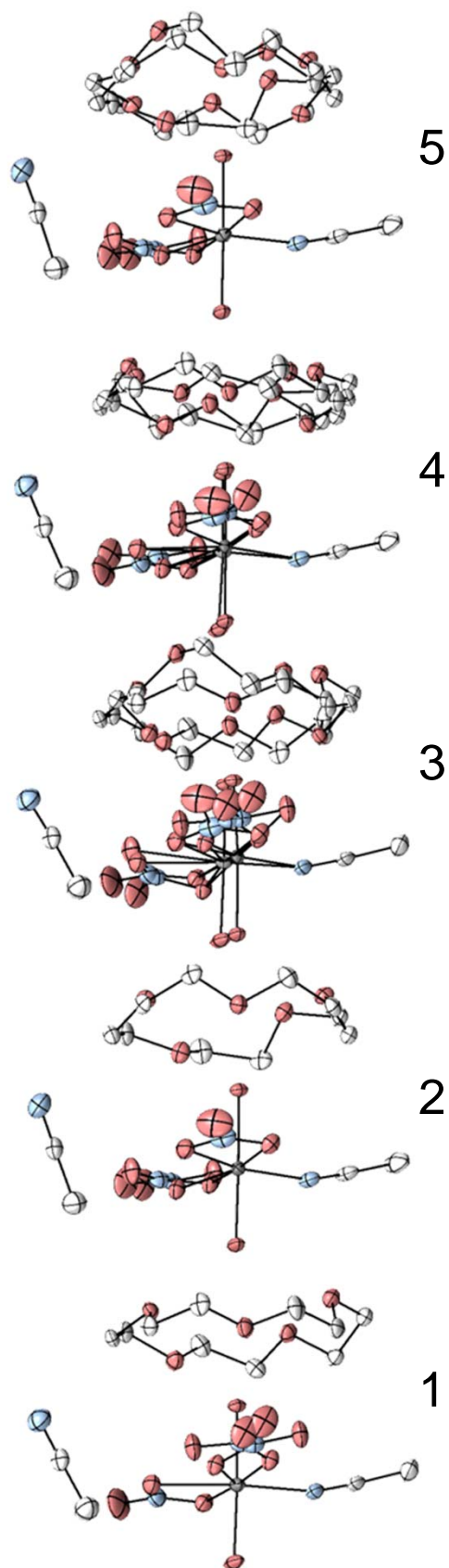


Reciprocal Lattice Slice for $P2_1$ (standard Cell) at 90 K

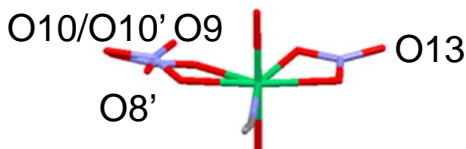
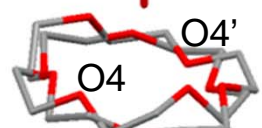
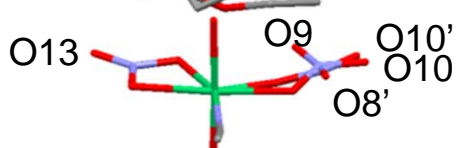
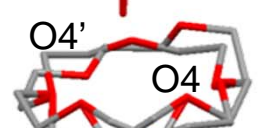
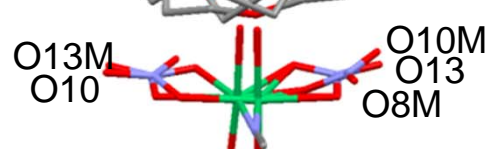
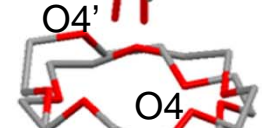
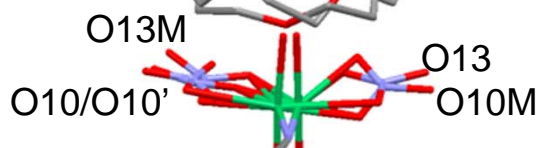
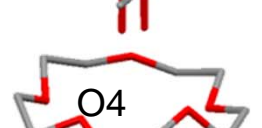

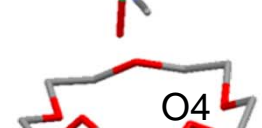
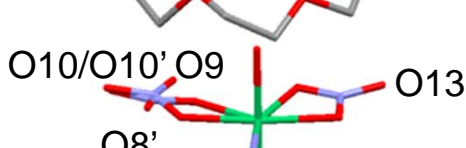
$3kl$ ($P2_1$ at 90 K)



Ellipsoid Plot for
 $B2_1$ Structure at 90 K



$B2_1$ Structure at 90 K

		Occupancy Factors for the major components
1		
		O4 0.859(3)
5		O9 0.785(5)
		O4 0.812(3)
4		O13 0.573(3) O9 (same)
		O4 0.853(2)
3		O13 0.852(3) O9 0.312(3) O8' 0.569(4) O9M 0.129(4)
		
2		O9 0.459(5)
		
1		O9 0.707(5)

(All atoms in a $C_{10}H_{20}O_5$ molecule or a NO_3^- ion have the same occupancy factor)

Description of the disorder in the $B2_1$ structure (assumed to be phase II) at 90 K, where the phase is metastable:

Ni Complex #1 (contains Ni_1)

Ordered except for the presence of two orientations of the κ^1O nitrate ligand. In phase III these two orientations are nearly equally probable and in phase I the probabilities are similar (0.56 vs. 0.44), but in phase II at 90 K the antiperiplanar orientation (*ap*; N3-Ni-O8-N1 and N3-Ni...N1-O9 torsion angles near 180°) has an occupancy factor of 0.707(5). The occupancy factor for the synperiplanar orientation (*sp*; N3-Ni-O9'-N1' and N3-Ni...N1'-O8' torsion angles < 30°) is 0.293(5).

15-Crown-5 #1 (i.e., 15C5_1)

Ordered.

Ni Complex #2 (contains Ni_2)

Ordered except for the presence of two orientations of the κ^1O nitrate ligand. The *ap* orientation has an occupancy factor of 0.459(5).

15-Crown-5 #2 (i.e., 15C5_2)

Ordered

Ni Complex #3 (contains Ni_3)

Disorder of the κ^1O and κ^2O nitrate ligands; two positions of the H₂O-Ni-OH₂ unit; occupancy factor for the κ^2O nitrate related to the corresponding ligand in Ni complex #2 by the pseudoglide is 0.852(3). Occupancy factors for the two H₂O-Ni-OH₂ units were tied to those of κ^1O/κ^2O nitrate disorder so as to preserve the Ni-O distances found in ordered complexes. Two orientations of the κ^1O nitrate ligand were found on the side of greater overall occupancy [0.312 and 0.560(3) for *ap* and *sp*] but only one orientation (*ap*) could be identified on the other. The occupancy factors for the three κ^1O nitrate ligands sum to 1.001, but they do not quite match the occupancy factors for the two positions of the κ^2O ligand [0.312+0.560 = 0.872(4) vs. 0.852(3)]. This complicated model was, however, the best that could be found.

15-Crown-5 #3 (i.e., 15C5_3)

Two orientations of the ring as in phase I. The occupancy factor for the major component is 0.853(2).

Ni Complex #4 (contains Ni_4)

Disorder of the κ^1O and κ^2O nitrate ligands; two positions of the H₂O-Ni-OH₂ unit; occupancy factors are close to ½ [i.e., 0.573/0.427(3)]. The occupancy factors of the *ap* and *sp* orientations

of the κ^1O nitrate ligand (one found on one side of the pseudomirror and one on the other) were set equal to the occupancy factor for the corresponding κ^2O nitrate ligand.

15-Crown-5 #4 (i.e., 15C5_4)

Two orientations of the ring as in phase I. The occupancy factor for the major component is 0.812(3).

Ni Complex #5 (contains Ni_5)

Ordered except for the presence of two orientations of the κ^1O nitrate ligand. The *ap* orientation has an occupancy factor of 0.785(4).

15-Crown-5 #5 (i.e., 15C5_5)

Two orientations of the ring as in phase I. The occupancy factor for the major component is 0.859(3).

The alternation of the κ^1O/κ^2O nitrate ligands found in phase III is found in Ni complexes #5, #1, and #2. It is less perfect in #3 [occupancy factors 0.852/0.148(3) rather than 1/0], and is essentially absent in #4 [occupancy factors 0.573/0.427(3)]. An odd number of independent complexes in the chain (here, 5) guarantees that the alternation must fail at some point (here, at Ni complex #4).

In phases I and III the occupancy factors for the two orientations of the κ^1O nitrate ligand are roughly equal. In phase II the occupancies are quite near $\frac{1}{2}$ for complexes #2, #3, and #4, but the *ap* orientation is clearly more favorable for complexes #5 and #1 [occupancy factors 0.785(4) and 0.707(5)].

Two of the 15-crown-5 molecules (#1 and #2) are completely ordered, as they are in phase III. The other three are more ordered than in phase I [occupancy factors for major component 0.852(2), 0.812(3), and 0.859(3) in phase II vs. 0.667(3) in phase I].

Cell constants determined for the basic cell ($P2_1/m$, $Z' = 1/2$) of phase II within its range of stability:

T (K)	a (Å)	b (Å)	c (Å)	β (°)	V (Å ³)
233 (1 st T sequence)	12.133	12.321	8.133	105.51	1171.4
239 (2 nd T sequence)	12.131	12.334	8.140	105.51	1173.5
236 (2 nd T sequence)	12.132	12.327	8.137	105.51	1172.6
233 (2 nd T sequence)	12.136	12.316	8.131	105.50	1171.0