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The structure of the complex between the arsenite oxidase from *Pseudorhizobium banfieldiae* sp. strain NT-26 and its native electron acceptor cytochrome c_{552}

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The arsenite oxidase (AioAB) from Pseudorhizobium banfieldiae sp. strain NT-26 catalyzes the oxidation of arsenite to arsenate and transfers electrons to its cognate electron acceptor cytochrome c_{552} (cyt c_{552}). This activity underpins the ability of this organism to respire using arsenite present in contaminated environments. The crystal structure of the AioAB/cytc552 electron transfer complex reveals two $A_2B_2/(cytc_{552})_2$ assemblies per asymmetric unit. Three of the four cytc₅₅₂ molecules in the asymmetric unit dock to AioAB in a cleft at the interface between the AioA and AioB subunits, with an edge-to-edge distance of 7.5 A between the heme of cyt_{252} and the [2Fe–2S] Rieske cluster in the AioB subunit. The interface between the AioAB and cytc552 proteins features electrostatic and nonpolar interactions and is stabilized by two salt bridges. A modest number of hydrogen bonds, salt bridges and relatively small, buried surface areas between protein partners are typical features of transient electron transfer complexes. Interestingly, the fourth $cytc_{552}$ molecule is positioned differently between two AioAB heterodimers, with distances between its heme and the AioAB redox active cofactors that are outside the acceptable range for fast electron transfer. This unique $cytc_{552}$ molecule appears to be positioned to facilitate crystal packing rather than reflecting a functional complex.

1. Introduction

Interprotein electron transfer processes are key biochemical events which play critical roles in fundamental biological processes such as photosynthesis, respiration and nitrogen fixation (Berg et al., 2007). The structural characterization of protein-protein complexes that participate in electron transfer is challenging due to their weak and transient nature (Supplementary Table S1; Antonyuk et al., 2013). The formation of electron transfer complexes requires efficient and finely tuned docking and dynamics at the protein-protein interface (Bendall, 2020; Moser et al., 1992), with complexes being guided together by complementary electrostatic steering and the relative positions of the proteins being tuned through hydrophobic/van der Waals interactions (Leys & Scrutton, 2004). The distance between the redox centers within these complexes has been shown to influence the rate of electron transfer (Davidson, 2000; Marcus & Sutin, 1985; Moser et al., 1992).

Interprotein electron transfer processes are crucial for the survival of arsenic respiring organisms cultured from arsenic contaminated environments (Santini *et al.*, 2007). Arsenic is toxic to most organisms in its inorganic forms arsenite (AsO_4^{3-}) and arsenate (AsO_4^{3-}) (Bissen & Frimmel, 2003;

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Domingo, 1995); however, prokaryotes such as *Pseudorhizo-bium banfieldiae* sp. strain NT-26 can catalyze the aerobic oxidation of arsenite (to arsenate) through the action of the enzyme arsenite oxidase (AioAB). The physiological electron acceptor for the AioAB enzyme has been shown to be a soluble *c*-type cytochrome (cyt c_{552} ; Santini *et al.*, 2007; Santini & vanden Hoven, 2004).

The AioAB complex consists of two subunits: a large subunit, AioA (93 kDa), which contains a molybdenum cofactor (Moco) at the active site and a [3Fe-4S] cluster, and a small subunit, AioB (14 kDa), which contains a Rieske [2Fe-2S] cluster. AioAB is a member of the dimethyl sulfoxide (DMSO) reductase family of molybdoenzymes. The crystal structure of AioAB from P. banfieldiae sp. strain NT-26 has been determined and refined to 2.7 Å resolution (PDB 4aay; Warelow et al., 2013). The proposed reaction mechanism of AioAB comprises oxidation of arsenite at the molybdenum site [reducing molybdenum(VI) to molybdenum(IV)], which releases two electrons that transfer one at a time to the [3Fe-4S] cluster of the AioA subunit and then to the [2Fe-2S] Rieske cluster of the AioB subunit. The electrons are then received by the electron acceptor cytc552 (Bernhardt & Santini, 2006; Santini & Ward, 2018; Supplementary Fig. S1).

To investigate the structural basis of the electron transfer process that underpins respiration using arsenite, here we report the crystal structure of the AioAB/cytc₅₅₂ complex from *P. banfieldiae* sp. strain NT-26. We show that $cytc_{552}$ sits within a cleft at the interface between the AioA and AioB subunits, with a relatively short distance between redox-active cofactors. The crystals show an interesting arrangement, with three of the four $cytc_{552}$ molecules located in a 'functional' location. The positioning of the fourth $cytc_{552}$ seems to be 'nonfunctional' and presumably facilitates crystal packing.

2. Materials and methods

The pPROEX-HTb-AioBA and pET-22b(+)-cyt c_{552} plasmids were prepared as described previously (Santini *et al.*, 2007; Warelow *et al.*, 2013). In these constructs the AioA subunit is composed of residues 2–845, the AioB subunit is composed of residues 41–175 and cyt c_{552} is composed of residues 21–127, in addition to residues derived from affinity tags (Supplementary Table S4). Residue numbering corresponds to the respective UniProt entries (Q6VAL8, Q6VAL9 and Q2TV05).

2.1. Protein overexpression and purification

The pPROEX-HTb-AioBA plasmid was transformed into *Escherichia coli* strain DH5 α (New England Biolabs). Cultures were grown aerobically at 21°C in Luria broth (LB) with 1 mM sodium molybdate and supplemented with ampicillin (100 µg ml⁻¹) with slow shaking (~50 rev min⁻¹). The cultures were induced with 40 µM isopropyl β -D-thiogalacto-pyranoside (IPTG) and were harvested after 24 h.

His-AioBA was purified by immobilized metal-affinity chromatography (IMAC) followed by size-exclusion chromatography (SEC). Frozen cell pellets were thawed and resuspended in binding buffer (20 mM potassium phosphate, 500 mM NaCl, 20 mM imidazole pH 7.3). The cells were lysed using a TS series bench-top cell disruptor (Constant Systems) at 241 MPa and insoluble debris was removed by centrifugation (Beckman JLA-25.50; 30 000g, 1 h, 4°C). The soluble fraction was incubated with 5 ml Ni-Sepharose 6 Fast Flow resin (Cytiva; 4°C; 1 h stirring) that had been pre-equilibrated with binding buffer (20 mM potassium phosphate, 500 mM NaCl, 20 mM imidazole pH 7.3). The resin was washed with ten column volumes (CV) of binding buffer, followed by elution of bound protein with elution buffer (20 mM potassium phosphate, 500 mM NaCl, 500 mM imidazole pH 7.3; 5 CV). The eluent was dialyzed (3000 molecular-weight cutoff SnakeSkin Dialysis Tubing, ThermoScientific) against 2.01 dialysis buffer (50 mM MES pH 5.5) overnight. This resulted in the precipitation of contaminating proteins, which were removed by centrifugation (30 000g, 30 min). The supernatant was concentrated by centrifugal ultrafiltration (10 000 molecular weight cutoff, Millipore Amicon Ultra) and further purified by SEC (HiLoad 16/600 Superdex 200 pg, Cytiva; 4°C) in 50 mM MES, 100 mM NaCl pH 5.5. The purest fractions as determined by SDS-PAGE were pooled and concentrated to approximately 10 mg ml^{-1} by centrifugal ultrafiltration. Aliquots of the purified protein (which will be referred to as AioAB in the following) were snap-frozen and stored at -80° C until further use. The concentration of the AioAB enzyme was measured spectrophotometrically at 682 nm with $\varepsilon_{682} = 5.6 \text{ m}M^{-1} \text{ cm}^{-1}$ (Watson *et al.*, 2017).

The pET-22b(+)-cytc₅₅₂ and pEC86 plasmids were cotransformed into *E. coli* strain BL21(DE3) (New England Biolabs). Cultures were grown at 30°C in LB supplemented with ampicillin (100 µg ml⁻¹), chloramphenicol (60 µg ml⁻¹) and a 1:100 dilution of a trace-metal solution (Ihssen & Egli, 2004; Santini *et al.*, 2007). The cells were induced with 20 µM IPTG at an OD₆₀₀ value of between 1.2 and 1.5 and were harvested after 16 h of shaking at 30°C.

The His-cyt c_{552} protein was purified by cation-exchange chromatography, IMAC and SEC. Frozen cell pellets were thawed at room temperature and resuspended in cell-lysis buffer (20 mM MES pH 5.5). The cells were disrupted by passage through a TS series bench-top cell disruptor (Constant Systems) at 241 MPa. Cell debris was removed by centrifugation (Beckman JLA-25.50; 30 000g, 1 h, 4°C) and the soluble fraction was loaded onto a 5 ml HiTrap SP Sepharose Fast Flow column (Cytiva; 4°C), washed with buffer consisting of 70 mM NaCl, 20 mM MES pH 5.5 and eluted with a linear NaCl gradient (0.07-0.45 M in 50 mM MES pH 5.5). The eluent was dialyzed (3000 molecular weight cutoff SnakeSkin Dialysis Tubing, ThermoScientific) against 2.01 dialysis buffer (50 mM potassium phosphate, 0.5 M NaCl pH 7.4) overnight. The protein was loaded onto a 5 ml HisTrap column (Cytiva) and eluted using an imidazole gradient (0-0.5 M in 50 mM potassium phosphate, 0.5 M NaCl pH 7.4) followed by SEC (HiLoad 16/600 Superdex 75 pg, Cytiva; 4°C; 20 mM Tris, 150 mM NaCl pH 7.8). The purified protein (which will be referred to as $cytc_{552}$) was concentrated to 10 mg ml⁻¹ and stored at -80° C until further use. The concentration of

Table 1

Ci ystanization.	
Method	Hanging-drop vapor diffusion
Plate type	24-well VDX plates
Temperature (K)	293
Protein concentration (mg ml $^{-1}$)	5
Buffer composition of protein solution	50 m <i>M</i> Tris, 100 m <i>M</i> NaCl pH 7.8
Composition of reservoir solution	0.2 <i>M</i> sodium chloride, 0.1 <i>M</i> HEPES pH 7.3, 18%(<i>w</i> / <i>v</i>) PEG 3350
Volume and ratio of drop	2 µl, 1:1
Volume of reservoir (µl)	500

oxidized cyt c_{552} was determined spectrophotometrically at 550 nm using $\varepsilon_{550} = 8.7 \text{ m}M^{-1} \text{ cm}^{-1}$ (Santini *et al.*, 2007).

2.2. Enzyme kinetics

AioAB activity assays were carried out as described previously (Watson et al., 2017). The reduced-oxidized extinction coefficient for cytochrome c_{552} at 550 nm is $23 \text{ m}M^{-1} \text{ cm}^{-1}$ (Santini *et al.*, 2007) and that at 416 nm is $59 \text{ m}M^{-1} \text{ cm}^{-1}$ (Santini *et al.*, 2007). Purified AioAB enzyme (2 nM) was incubated with fully oxidized cytochrome c_{552} (20 µM) in 50 mM Tris-HCl pH 8.0 (Watson et al., 2017) with increasing concentrations of arsenite (0-1 mM). The reaction was followed at 550 nm. The steady-state kinetics with cyt_{552} as the substrate were determined using an excess of arsenite (2.5 mM) pre-incubated with purified AioAB enzyme (2 nM), followed by the addition of various concentrations of cyt_{552} $(0-10 \ \mu M)$. In this case, the reaction was followed at 416 nm (Watson et al., 2017). Kinetic experiments were performed using triplicate measurements and data fitting was carried out using the Michaelis-Menten function with GraphPad Prism version 7.0 for Mac OS X (GraphPad Software, La Jolla, California, USA).

2.3. Protein crystallization and data collection

Purified AioAB (in 50 mM Tris, 100 mM NaCl pH 7.8) and cytc₅₅₂ (in 50 mM Tris, 100 mM NaCl pH 7.8) were mixed and incubated on ice at a molar ratio of 1:1.5 AioAB:cytc₅₅₂ (total protein concentration 5 mg ml⁻¹) before crystallization via sitting-drop vapor diffusion in 96-well plates (Molecular Dimensions). The stoichiometry of the mixture (AioAB: $cytc_{552}$) was based on our previous experience in crystallizing the SorT/SorU complex from Sinorhizobium meliloti (McGrath et al., 2015). Initial crystallization trials for the AioAB/cytc₅₅₂ complex were conducted using the Index HT (Hampton Research) and ProPlex (Molecular Dimensions) screens. Drops consisting of equal volumes (0.2 µl) of reservoir solution and protein solution were dispensed by a Crystal Gryphon liquid-handling system (Art Robbins Instruments) and were equilibrated against a 50 µl reservoir of screen solution at 20°C. Multiple plate-like crystals of AioAB/cytc₅₅₂ were observed within one week in conditions A4 (0.1 M bis-Tris pH 6.5, 2.0 M ammonium sulfate), E7 [0.05 M magnesium chloride hexahydrate, 0.1 M HEPES pH 7.5, 30%(v/v) PEG 3350], F12 [0.2 M sodium chloride, 0.1 M HEPES pH 7.5, 25%(w/v) PEG 3350], G1 [0.2 M sodium chloride, 0.1 M TrisTable 2Data collection and processing.

Values in parentheses are for the highest resolution shell.

Diffraction source	MX2, Australian Synchrotron
Wavelength (Å)	0.953
Temperature (K)	100
Detector	EIGER 16M
Crystal-to-detector distance (mm)	249
Total rotation range (°)	360
Space group	P2 ₁
a, b, c (Å)	129.4, 126.6, 148.0
α, β, γ (°)	90.0, 107.8, 90.0
Mosaicity (°)	0.09
Resolution range (Å)	49.19-2.25 (2.29-2.25)
Total No. of reflections	1510469 (77078)
No. of unique reflections	214836 (10599)
Completeness (%)	99.9 (99.9)
Multiplicity	7.0 (7.3)
$\langle I/\sigma(I)\rangle$	11.4 (2.5)
CC _{1/2}	0.998 (0.915)
R _{merge}	0.112 (0.806)
R _{nim}	0.045 (0.320)
Overall <i>B</i> factor from Wilson plot (Å ²)	29.5

HCl pH 8.5, 25%(w/v) PEG 3350], H1 [0.2 *M* magnesium chloride hexahydrate, 0.1 *M* Tris–HCl pH 8.5, 25%(w/v) PEG 3350] and H6 [0.2 *M* sodium formate, 20%(w/v) PEG 3350] of the Index HT screen. Crystals were also observed within one week in conditions C3 [0.2 *M* ammonium acetate, 0.1 *M* sodium citrate, 20%(w/v) PEG 4000], C8 [0.2 *M* sodium chloride, 0.1 *M* Tris–HCl pH 8.0, 20%(w/v) PEG 4000], C12 [0.2 *M* potassium iodide, 0.1 *M* MES pH 6.5, 25%(w/v) PEG 4000] and D1 [0.2 *M* sodium chloride, 0.1 *M* sodium HEPES pH 7.5, 25%(w/v) PEG 4000] of the ProPlex screen.

Optimization of these conditions was carried out by hanging-drop vapor diffusion in 24-well VDX plates (Hampton Research), varying the concentrations of sodium chloride (0.1–0.2 *M*), HEPES (0.05–0.1 *M*) and PEG 3350 (15–25%) and the pH (6.5–7.5). Diffraction-quality crystals of AioAB/cytc₅₅₂ grew after two weeks in drops consisting of equal volumes (2 μ l; 1:1) of the AioAB/cytc₅₅₂ preparation and reservoir solution [0.2 *M* sodium chloride, 0.1 *M* HEPES pH 7.3, 18%(*w*/*v*) PEG 3350] equilibrated against 500 μ l reservoir solution at 20°C. Crystals were cryoprotected in reservoir solution containing 25%(*w*/*v*) glycerol before flash-cooling in liquid nitrogen. Crystallization conditions are given in Table 1.

2.4. Data collection, structure solution and refinement

Diffraction data were collected from the AioAB/cyt c_{552} crystals using an EIGER 16M detector on beamline MX2 at 13 000 eV at the Australian Synchrotron. All data were collected at 100 K, processed with *XDS* (Kabsch, 2010) and merged and scaled with *AIMLESS* (Evans & Murshudov, 2013). Unit-cell parameters and data-collection statistics are presented in Table 2.

The crystal structure of the AioAB/cyt c_{552} complex was solved by molecular replacement with *MOLREP* (Vagin & Teplyakov, 2010) from the *CCP*4 suite (Winn *et al.*, 2011), using a search model composed of the coordinates of the AioAB structure (PDB entry 4aay; Warelow *et al.*, 2013) with

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the water molecules removed. Initial rounds of refinement of a model with four AioAB complexes per asymmetric unit yielded a difference Fourier electron-density map which showed positive difference density for the location of four molecules of $cytc_{552}$ per asymmetric unit. These were placed by phased molecular replacement with *MOLREP* using a search model generated from the structure of ferrocytochrome c_2 (PDB entry 1co6; Badilla *et al.*, 2018) modified by *CHAINSAW* (Stein, 2008). Manual model building and the addition of water molecules were carried out in *Coot* (Emsley *et al.*, 2010) with iterative cycles of refinement using *REFMAC5* (Murshudov *et al.*, 2011). The geometry of the final model was determined with *MolProbity* (Chen *et al.*, 2010). Refinement statistics are summarized in Table 3.

3. Results and discussion

The structure of the AioAB/cytc₅₅₂ complex was solved and refined to 2.25 Å resolution (Tables 2 and 3). The structure includes four copies of the AioAB assembly per asymmetric unit, arranged as two A₂B₂ heterotetramers. The AioA subunits include residues 2–844 and are composed of four domains (domains I, II, III and IV). Domain I is composed of three antiparallel β -sheets, domain II and domain III have similar $\alpha\beta\alpha$ -sandwich topologies and domain IV predominantly consists of six antiparallel β -sheets flanked by five small α -helices. The AioB subunits include residues 44–175 and have a fold consisting of a six-stranded antiparallel β -barrel and a four-stranded antiparallel β -sheet.

The AioA subunit houses the Moco site, which is a common feature of the DMSO reductase family of molybdenumcontaining enzymes, and the [3Fe–4S] cluster. The Mo atom is coordinated by one oxo ligand and the thiol groups of the two pterin cofactors in an approximate square-pyramidal geometry, with an average Mo=O distance across all four copies per asymmetric unit of 1.8 ± 0.1 Å. The [3Fe–4S] cluster is coordinated by a conserved cysteine-rich motif (Cys24-X₂-Cys27-X₃-Cys31-X₇₀-Ser102) and the AioB subunit houses the [2Fe–2S] Rieske cluster, which is coordinated by two cysteine residues and two histidine residues (Cys103-X-His105-X₁₅-Cys121-X₂-His124).

In addition to the two $AioA_2B_2$ complexes, there are four molecules of cyt_{552} per asymmetric unit. The cyt_{552} protomers are composed of four α -helices arranged to form a bundle that frames a heme-binding site. His38 and Met103 are axial ligands of the central Fe atom and the porphyrin ring is covalently attached to Cys34 and Cys37 (Fig. 2c and Supplementary Fig. S6). This is the first reported crystal structure of cytc₅₅₂ from P. banfieldiae sp. strain NT-26. A search of the coordinates of cyt_{552} against the Protein Data Bank (PDB) using PDBeFold (Krissinel & Henrick, 2004) reveals similarity to the structures of cytc552 from Paracoccus denitrificans (PDB entry 1ql4; Harrenga et al., 2000) and cytc₂ from Rhodopseudomonas viridis (PDB entry 1co6; Sogabe & Miki, 1995), with root-mean-square deviation (r.m.s.d.) values of 0.6-0.7 Å (over 92 and 79 common C^{α} positions), indicating similar structures.

Table 3

Structure refinement.

Values in parentheses are for the highest resolution shell.

Resolution range (Å)	49.23-2.25 (2.31-2.25)
Completeness (%)	92.2
No. of reflections, working set	188208
No. of reflections, test set	10062
Final R _{crvst}	0.183 (0.208)
Final R _{free}	0.230 (0.293)
No. of non-H atoms	· · · ·
Protein	34093
Waters	1751
Total	35834
R.m.s.d.s	
Bond lengths (Å)	0.007
Angles (°)	1.606
Average <i>B</i> factors [†] ($Å^2$)	
Protein	22.9
Waters	9.2
Other	9.9
Ramachandran plot‡	
Most favored (%)	94.2
Allowed (%)	99.8
Outliers§ (%)	0.2
PDB code	8ed4

[†] Calculated by *BAVERAGE* from the *CCP*4 suite (Winn *et al.*, 2011). [‡] Calculated using *MolProbity* (Chen *et al.*, 2010). [§] The outliers were AspA613, IleA811, AspC668, IleC811, AlaG304, IleG811, PheI105, PheJ105 and PheK105. The $2F_o - F_c$ electron-density maps were observed clearly for these residues at 1.1σ and most are consistent between copies in the asymmetric unit and (for AioAB) with the previously published structure.

3.1. Two different $AioA_2B_2/(cytc_{552})_2$ complexes are present in the crystal

In the asymmetric unit, two molecules of $cytc_{552}$ are associated with each of the two AioA₂B₂ assemblies, so there are two AioA₂B₂/($cytc_{552}$)₂ complexes per asymmetric unit (Fig. 1). In one complex (chains *ABI* and *CDJ*) the two $cytc_{552}$ molecules (chains *I* and *J*) are located at similar relative positions in a cleft near the AioA/AioB interface (Fig. 1*a*). In the other AioA₂B₂/($cytc_{552}$)₂ unit (chains *EFK* and *GHL*) the relative positions of the two $cytc_{552}$ molecules are different. One (chain *K*) is consistent with that described above, sitting between the AioA and AioB subunits, while the other (chain *L*) associates with AioA (chain *G*) from one AioAB heterodimer and AioB (chain *F*) from the neighboring heterodimer (Fig. 1*b*).

The three $cytc_{552}$ molecules that lie at the AioA/AioB interface (chains I, J and K) are located such that the edge-toedge distance between the [2Fe-2S] Rieske cluster in AioB and the heme in cyt_{552} is 7.5 Å, which is consistent with fast electron transfer (discussed further below; Page et al., 1999). The unique cyt_{552} (chain L) that associates between heterodimers shows edge-to-edge distances between the cytc552 heme, the Moco (Mo atom) and the [3Fe-4S] cluster (residue Cys24) of AioA of 25 and 29 Å, respectively. The distance between the cytc₅₅₂ heme and the [2Fe-2S] cluster (residue Cys103) in AioB of the neighboring heterodimer is 38 Å (Supplementary Fig. S2). These distances are outside the accepted range for fast electron transfer. The positioning of the unique cyt_{552} in the complex therefore does not represent an electron transfer complex. This positioning of cytc₅₅₂ presumably facilitates crystallization (Supplementary Fig. S3) but does not represent the complexes present in solution and/ or a functional assembly. This is reminiscent of a previously reported structure of chicken liver sulfite oxidase (*Gallus gallus*; PDB entry 1sox; Kisker *et al.*, 1997). This enzyme contains three domains: an N-terminal cytochrome domain, a Moco domain and a C-terminal domain. Interestingly, in this structure the cytochrome domain is positioned so that the edge-to-edge distance is 32 Å between the Mo atom and the heme cofactor, which is also outside the range for fast electron transfer. In solution, a flexible linker between the Moco and cytochrome domains allows the cofactors to approach at proximity. The following discussion will therefore describe the AioAB/cytc₅₅₂ complex with cytc₅₅₂ positioned at the AioA/ AioB interface.

3.2. The AioAB/cytc₅₅₂ interface

As described above, $cytc_{552}$ sits within a cleft near the AioA/ AioB interface and interacts with both subunits of the AioAB heterodimer. Surface areas of 680 and 660 Å² are buried on complex formation for the AioAB (350 Å² for AioA and



Figure 1

Two different AioA₂B₂/(cytc₅₅₂)₂ complexes are present in the asymmetric unit of the AioAB/cytc₅₅₂ crystal structure. In both of the structures the AioA subunits are shown as gray surfaces, the cytc₅₅₂ molecules are in salmon and the AioB subunits are in cyan. (*a*) The AioA₂B₂ complex represented by chains *ABI* and *CDJ*. The two cytc₅₅₂ molecules (chains *I* and *J*) are located at similar positions in a cleft near the AioA/AioB interface. (*b*) The AioA₂B₂ complex represented by chains *EFK* and *GHL*. One molecule of cytc₅₅₂ (chain *K*) is located at the AioA/AioB interface. The second molecule of cytc₅₅₂ (chain *L*) is unique in that it sits between AioAB heterodimers.

330 Å² for AioB) and cytc₅₅₂ proteins, respectively. The contacts between AioAB and cytc₅₅₂ are mediated by three regions of the cytc₅₅₂ structure (residues 32–37, 45–48 and 102–106; Fig. 2*d*). Two salt bridges between Asp67 and Glu73 from the AioB subunit, and Lys95 and Lys110 from cytc₅₅₂ complete the interface (Figs. 2*a* and 2*b*). Notably, Asp67 and Glu73 from the AioB subunit are not conserved in the sequences of comparable Rieske proteins (Supplementary Fig. S5*a*), whereas Lys95 of cytc₅₅₂ is conserved in the sequences of cytc₅₅₂ from *P. denitrificans* (*Pd*cytc₅₅₂; PDB entry 1ql4; Harrenga *et al.*, 2000) and cytc₂ from *R. viridis* (*Rv*cytc₂; PDB entry 1co6; Sogabe & Miki, 1995) and Lys110 is conserved in the sequence of *Pd*cytc₅₅₂ (PDB entry 1ql4; Harrenga *et al.*, 2000) (Supplementary Fig. S5*b*).

The AioAB/cytc₅₅₂ interaction shows significant charge complementarity, with negative charge on the AioAB complex correlating with a concentration of positive charge on the surface of cytc₅₅₂ (Fig. 2*e*). These charged areas encircle neutral surfaces that correlate with the 'footprints' of each electron transfer partner on the other. There are between four and 11 water molecules (over the three AioAB/cytc₅₅₂ complexes per asymmetric unit) that sit between AioAB and cytc₅₅₂ and which interact with polar and charged surface residues. Superposition of the coordinates of the AioA and AioB subunits from this work with those of AioAB alone (PDB entry 4aay; Warelow *et al.*, 2013), yields r.m.s.d. values of 0.20 and 0.28 Å, respectively (over 832 and 132 common C^{α} positions), indicating minimal changes on association with cytc₅₅₂ (Supplementary Fig. S4*b*).

Within the AioAB/cyt c_{552} complex, the cyt c_{552} protein shows an average *B* factor of 44.0 Å², which is significantly higher than that of AioAB (23.6 Å). In addition, the relative temperature factor per residue for cyt c_{552} increases with increasing distance from the AioAB/cyt c_{552} interface (Supplementary Fig. S4*a*), indicating that the cyt c_{552} molecule is dynamic relative to AioAB within the crystalline lattice. This has been observed previously for related electron transfer complexes, such as the SorT/SorU complex from *S. meliloti* (McGrath *et al.*, 2015). Flexibility in the interactions between protein partners has been proposed to be necessary to achieve optimal orientations for efficient electron transfer (Leys & Scrutton, 2004; van Amsterdam *et al.*, 2002).

Taken together, these observations are consistent with the structures of other comparable electron transfer complexes, which typically feature a modest number of hydrogen bonds and salt bridges at the protein–protein interface and relatively small, buried surface areas between protein partners (Supplementary Table S1). The AioAB/cytc₅₅₂ structure is therefore typical of transient complex formation for fast electron transfer (Miyashita *et al.*, 2003).

3.3. The electron transfer pathway between AioAB and cytc₅₅₂

The vinyl groups on the porphyrin ring of $cytc_{552}$ contribute to the AioAB/cyt c_{552} interface. AioB residues Phe108, Pro109 and Pro122 lie closest to the $cytc_{552}$ heme (Fig. 3*a*). As

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mentioned previously, the closest edge-to-edge distance between Cys103, which coordinates the Rieske cluster in AioB, and the vinyl group of heme in cyt_{552} is 7.5 Å, which is well within the distance for fast electron transfer through the protein medium (Page et al., 1999). HARLEM analysis of the coordinates of the complex (Kurnikov, 2000) predicts that the dominant electron-tunneling pathway from AioB to cytc₅₅₂ proceeds from the [2Fe-2S] Rieske cluster in the AioB subunit to Pro122 and across the protein-protein interface to the porphyrin ring and onto the heme iron, with Pro122 at a distance of 4.1 Å from the closest heme vinyl group (Fig. 3a, Supplementary Table S2). This suggests a role for Pro122 in the electron transfer process, which is consistent with its conservation in the sequences of Rieske cluster-containing proteins and subunits, including arsenite oxidase from Alcaligenes faecalis (PDB entry 1g8k; Ellis et al., 2001), the Rieske protein II SoxF from Sulfolobus acidocaldarius (SfSoxF; PDB entry 1jm1; Bönisch et al., 2002), the Rieske protein from Thermus thermophilus (TtRp; PDB entry 1nyk; Hunsicker-Wang et al., 2003) and the Rieske protein involved in photosynthetic and respiratory electron transport in Synechocystis PCC 6803 (SyPetC3; PDB entry 5cxm; Veit et al., 2016; Fig. 3b). The involvement of proline residues in interprotein electron transfer has previously been proposed for the complex between the nitrite reductase from Achromobacter xylos*oxidans* GIFU 1051 and its electron acceptor cytochrome *c* (the AxgNIR/cytc₅₅₁ complex; PDB entry 2zon; Nojiri *et al.*, 2009).

Previous studies examining the kinetics of arsenite oxidation catalyzed by AioAB with horse heart cytochrome c(hhcytc) as the electron acceptor revealed that mutation of Phe108 in AioB to Ala led to a 30-fold decrease in the rate of the reaction relative to the native enzyme (Warelow, 2015). These data are consistent with the observation from the present structure that Phe108 lies at the interface between the AioB and $cytc_{552}$ proteins (the Phe108-heme distance is 5.2 Å). Aromatic residues such as phenylalanine have been shown to be involved in electron transfer in other complexes (Hirasawa et al., 1998; Liang et al., 1987), including in the structure of caa3-type cytochrome oxidase from T. thermophilus (Lyons et al., 2012). Whether this residue is part of the electron transfer pathway to $cytc_{552}$ or facilitates and/or stabilizes complex formation requires further investigation. Notably, this residue is not conserved in the sequences of comparable Rieske proteins (Fig. 3b).

3.4. AioAB is catalytically efficient in the presence of cytc_{552} in solution

Previous studies have reported the kinetics of arsenite oxidation catalyzed by AioAB using the artificial electron



Figure 2

The structure of the AioAB/cytc₅₅₂ electron transfer complex. (*a*) The AioAB/cytc₅₅₂ complex as observed for chains *ABI*, *CDJ* and *EFK* (AioA in blue, AioB in cyan and cytc₅₅₂ in salmon). (*b*) The interface between AioAB and cytc₅₅₂. Residues that participate in the two salt bridges are shown. (*c*) The structure of cytochrome cytc₅₅₂. The four helices are labeled and the heme cofactor is shown in salmon. The heme Fe atom is coordinated by His38 and Met103. The protoporphyrin ring is covalently attached to Cys34 and Cys37. (*d*) 'Open-book' representation of the AioAB/cytc₅₅₂ complex (AioAB is in red and cytc₅₅₂ is in blue) indicating the 'footprint' of interacting residues for each protein. (*e*) The same view as in (*d*) colored according to the electrostatic surfaces (positive charge in blue, negative charge in red and neutral in white).



Figure 3

The electron transfer pathway between AioAB and cytc₅₅₂. (a) Pathway for electron transfer (Kurnikov, 2000). (b) Secondary-structure-based sequence alignment of Rieske proteins. Conserved residues near the Rieske cluster are colored blue and Phe108, which is unique to AioB, is in red. Residue numbers in purple correspond to the AioB sequence. The alignment was generated using *Clustal Omega* (Sievers *et al.*, 2011). Abbreviations used are as follows: AioB, *P. banfieldiae* sp. strain NT-26 (this work); *Af*AioB, AioB subunit of arsenite oxidase from *A. faecalis*; *Sf*SoxF, Rieske protein II from *S. acidocaldarius*; *Tt*Rp, Rieske protein from *T. thermophilus*; *Sy*PetC3, Rieske protein from *Synechocystis* PCC 6803.

acceptors DCPIP and hhcytc. To complement our structural analyses, we determined the activity of the AioAB enzyme with its native electron acceptor cytc₅₅₂. In the presence of an excess of cytc₅₅₂, Michaelis–Menten analysis of arsenite oxidation by AioAB monitored spectrophotometrically yielded a $K_{m(arsenite)}$ of 9.06 ± 1.3 µM and a turnover number of 205 ± 19 s⁻¹. These values are similar to those reported with hhcytc as the electron acceptor (13.0 ± 0.15 µM and 211.2 ± 0.15 s⁻¹, respectively). Analyses in the presence of an excess concentration of arsenite yielded $K_{m(cytc552)} = 2.9 \pm 0.2 \mu M$ and $k_{cat} = 390 \pm 25 \text{ s}^{-1}$ (Supplementary Table S3).

The similarity of the turnover numbers for arsenite oxidation with $cytc_{552}$ and hhcytc as electron acceptors is interesting given that one protein is a native partner and the other is not. Both proteins show positively charged surfaces and basic pI values (8.7 and 10.0, respectively) and have similar redox potentials [cytc₅₅₂, 275 mV (Kalimuthu et al., 2014); hhcytc, 256-266 mV (Weber et al., 1987)]. Presumably, these features allow hhcytc to substitute for $cytc_{552}$ in the *in vitro* assay. The electrostatic surfaces of other *c*-type cytochrome electron acceptors which serve as electron transfer partners to members of the molybdenum protein family also have an overall positive charge. The value of $K_{m(cvtc552)}$ determined here is similar to those reported for these systems, including chicken liver sulfite oxidase (CSO; G. gallus; Kisker et al., 1997) and the sulfite dehydrogenase SorAB complex from Starkeya novella, with $K_{m(cvtc)}$ values between 2 and 4 μM (Kappler et al., 2006; Kappler & Enemark, 2015). A notable exception is the SorT/SorU complex from S. meliloti (McGrath *et al.*, 2015). The electron acceptor SorU has an overall negative charge (pI of ~4) and a higher $K_{m(SorU)}$ ($32 \pm 5 \mu M$).

The k_{cat} values for these systems vary (Supplementary Table S3, with the AioAB/cyt c_{552} system apparently being particularly efficient (Brody & Hille, 1999; Kappler *et al.*, 2006). The docking and dissociation of AioAB/cyt c_{552} before and after electron transfer, respectively, presumably play a significant role in the rate of turnover and can be influenced by the electrostatic complementarity and the number of interactions at the protein–protein interface (Leys & Scrutton, 2004).

4. Conclusion

The structure of the AioAB/cyt c_{552} complex reported here shows an interesting combination of 'functional' and 'nonfunctional' assemblies within the crystals. The positioning of the unique cyt c_{552} molecule between AioAB heterodimers presumably facilitates crystallization but does not represent a fast electron transfer complex. The remaining three AioAB/ cyt c_{552} modules per asymmetric unit show the cyt c_{552} molecules positioned in a cleft between the AioA and AioB subunits, with close association between the redox-active cofactors for fast electron transfer.

5. Related literature

The following references are cited in the supporting information for this article: Axelrod *et al.* (2002), Kappler & Bailey (2005), Krissinel & Henrick (2005) and Kurisu *et al.* (2001).

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References

- Amsterdam, I. M. C. van, Ubbink, M., Einsle, O., Messerschmidt, A., Merli, A., Cavazzini, D., Rossi, G. L. & Canters, G. W. (2002). *Nat. Struct. Biol.* 9, 48–52.
- Antonyuk, S. V., Han, C., Eady, R. R. & Hasnain, S. S. (2013). *Nature*, **496**, 123–126.
- Axelrod, H. L., Abresch, E. C., Okamura, M. Y., Yeh, A. P., Rees, D. C. & Feher, G. (2002). J. Mol. Biol. 319, 501–515.
- Badilla, C., Osborne, T. H., Cole, A., Watson, C., Djordjevic, S. & Santini, J. M. (2018). *Sci. Rep.* **8**, 6282.
- Bendall, D. S. (2020). *Protein Electron Transfer*, edited by D. S. Bendall, pp. 43–68. New York: Garland Science.
- Berg, J. M., Tymoczko, J. L. & Stryer, L. (2007). *Biochemistry*, 6th ed. New York: W. H. Freeman & Co.
- Bernhardt, P. V. & Santini, J. M. (2006). Biochemistry, 45, 2804-2809.
- Bissen, M. & Frimmel, F. H. (2003). Acta Hydrochim. Hydrobiol. 31, 9–18.
- Bönisch, H., Schmidt, C. L., Schäfer, G. & Ladenstein, R. (2002). J. Mol. Biol. 319, 791–805.
- Brody, M. S. & Hille, R. (1999). Biochemistry, 38, 6668-6677.
- Chen, V. B., Arendall, W. B., Headd, J. J., Keedy, D. A., Immormino, R. M., Kapral, G. J., Murray, L. W., Richardson, J. S. & Richardson, D. C. (2010). Acta Cryst. D66, 12–21.
- Davidson, V. L. (2000). Acc. Chem. Res. 33, 87-93.
- Domingo, J. L. (1995). Reprod. Toxicol. 9, 105-113.
- Ellis, P. J., Conrads, T., Hille, R. & Kuhn, P. (2001). Structure, 9, 125–132.
- Emsley, P., Lohkamp, B., Scott, W. G. & Cowtan, K. (2010). Acta Cryst. D66, 486–501.
- Evans, P. R. & Murshudov, G. N. (2013). Acta Cryst. D69, 1204-1214.
- Harrenga, A., Reincke, B., Rüterjans, H., Ludwig, B. & Michel, H. (2000). J. Mol. Biol. 295, 667–678.
- Hirasawa, M., Hurley, J. K., Salamon, Z., Tollin, G., Markley, J. L., Cheng, H., Xia, B. & Knaff, D. B. (1998). *Biochim. Biophys. Acta*, 1363, 134–146.
- Hunsicker-Wang, L. M., Heine, A., Chen, Y., Luna, E. P., Todaro, T., Zhang, Y. M., Williams, P. A., McRee, D. E., Hirst, J., Stout, C. D. & Fee, J. A. (2003). *Biochemistry*, **42**, 7303–7317.
- Ihssen, J. & Egli, T. (2004). Microbiology, 150, 1637-1648.
- Kabsch, W. (2010). Acta Cryst. D66, 133-144.
- Kalimuthu, P., Heath, M. D., Santini, J. M., Kappler, U. & Bernhardt, P. V. (2014). *Biochim. Biophys. Acta*, **1837**, 112–120.
- Kappler, U. & Bailey, S. (2005). J. Biol. Chem. 280, 24999-25007.
- Kappler, U., Bailey, S., Feng, C., Honeychurch, M. J., Hanson, G. R., Bernhardt, P. V., Tollin, G. & Enemark, J. H. (2006). *Biochemistry*, 45, 9696–9705.

- Kappler, U. & Enemark, J. H. (2015). J. Biol. Inorg. Chem. 20, 253–264.
- Kisker, C., Schindelin, H., Pacheco, A., Wehbi, W. A., Garrett, R. M., Rajagopalan, K., Enemark, J. H. & Rees, D. C. (1997). *Cell*, **91**, 973– 983.
- Krissinel, E. & Henrick, K. (2004). Acta Cryst. D60, 2256-2268.
- Krissinel, E. & Henrick, K. (2005). CompLife 2005: Computational Life Sciences, edited by M. R. Berthold, R. C. Glen, K. Diederichs, O. Kohlbacher & I. Fischer, pp. 163–174 Berlin, Heidelberg: Springer.
- Kurisu, G., Kusunoki, M., Katoh, E., Yamazaki, T., Teshima, K., Onda, Y., Kimata-Ariga, Y. & Hase, T. (2001). *Nat. Struct. Biol.* 8, 117–121.
- Kurnikov, I. (2000). HARLEM. https://harlem.chem.cmu.edu/index.php/ software/2-uncategorised/18-harlem.
- Leys, D. & Scrutton, N. S. (2004). Curr. Opin. Struct. Biol. 14, 642-647.
- Liang, N., Pielak, G. J., Mauk, A. G., Smith, M. & Hoffman, B. M. (1987). Proc. Natl Acad. Sci. USA, 84, 1249–1252.
- Lyons, J. A., Aragão, D., Slattery, O., Pisliakov, A. V., Soulimane, T. & Caffrey, M. (2012). *Nature*, **487**, 514–518.
- Marcus, R. & Sutin, N. (1985). Biochim. Biophys. Acta, 811, 265-322.
- McGrath, A. P., Laming, E. L., Casas Garcia, G. P., Kvansakul, M., Guss, J. M., Trewhella, J., Calmes, B., Bernhardt, P. V., Hanson, G. R., Kappler, U. & Maher, M. J. (2015). *eLife*, 4, e09066.
- Miyashita, O., Onuchic, J. N. & Okamura, M. Y. (2003). *Biochemistry*, **42**, 11651–11660.
- Moser, C. C., Keske, J. M., Warncke, K., Farid, R. S. & Dutton, P. L. (1992). *Nature*, **355**, 796–802.
- Murshudov, G. N., Skubák, P., Lebedev, A. A., Pannu, N. S., Steiner, R. A., Nicholls, R. A., Winn, M. D., Long, F. & Vagin, A. A. (2011). *Acta Cryst.* D67, 355–367.
- Nojiri, M., Koteishi, H., Nakagami, T., Kobayashi, K., Inoue, T., Yamaguchi, K. & Suzuki, S. (2009). *Nature*, **462**, 117–120.
- Page, C. C., Moser, C. C., Chen, X. & Dutton, P. L. (1999). *Nature*, **402**, 47–52.
- Santini, J. M., Kappler, U., Ward, S. A., Honeychurch, M. J., vanden Hoven, R. N. & Bernhardt, P. V. (2007). *Biochim. Biophys. Acta*, 1767, 189–196.
- Santini, J. M. & vanden Hoven, R. N. (2004). J. Bacteriol. 186, 1614– 1619.
- Santini, J. M. & Ward, S. A. (2018). *The Metabolism of Arsenite*. Boca Raton: CRC Press.
- Sievers, F., Wilm, A., Dineen, D., Gibson, T. J., Karplus, K., Li, W., Lopez, R., McWilliam, H., Remmert, M., Söding, J., Thompson, J. D. & Higgins, D. G. (2011). *Mol. Syst. Biol.* 7, 539.
- Sogabe, S. & Miki, K. (1995). J. Mol. Biol. 252, 235-247.
- Stein, N. (2008). J. Appl. Cryst. 41, 641-643.
- Vagin, A. & Teplyakov, A. (2010). Acta Cryst. D66, 22-25.
- Veit, S., Takeda, K., Tsunoyama, Y., Baymann, F., Nevo, R., Reich, Z., Rögner, M., Miki, K. & Rexroth, S. (2016). *Biochim. Biophys. Acta*, 1857, 1879–1891.
- Warelow, T. P. (2015). Arsenite Oxidase as a Novel Biosensor for Arsenite. PhD thesis. University College London, United Kingdom.
- Warelow, T. P., Oke, M., Schoepp-Cothenet, B., Dahl, J. U., Bruselat, N., Sivalingam, G. N., Leimkühler, S., Thalassinos, K., Kappler, U., Naismith, J. H. & Santini, J. M. (2013). *PLoS One*, 8, e72535.
- Watson, C., Niks, D., Hille, R., Vieira, M., Schoepp-Cothenet, B., Marques, A. T., Romão, M. J., Santos-Silva, T. & Santini, J. M. (2017). Biochim. Biophys. Acta, 1858, 865–872.
- Weber, C., Michel, B. & Bosshard, H. R. (1987). Proc. Natl Acad. Sci. USA, 84, 6687–6691.
- Winn, M. D., Ballard, C. C., Cowtan, K. D., Dodson, E. J., Emsley, P., Evans, P. R., Keegan, R. M., Krissinel, E. B., Leslie, A. G. W., McCoy, A., McNicholas, S. J., Murshudov, G. N., Pannu, N. S., Potterton, E. A., Powell, H. R., Read, R. J., Vagin, A. & Wilson, K. S. (2011). Acta Cryst. D67, 235–242.