

(Anthracen-9-ylmethyl)dimethylamine at 120 K

R. Alan Howie^{a*} and
Solange M. S. V. Wardell^b^aDepartment of Chemistry, University of Aberdeen, Meston Walk, Aberdeen AB24 3UE, Scotland, and ^bFundação Oswaldo Cruz, Farmanguinhos, Rua Sizenando Nabuco 100, Manguinhos, 21041250 Rio de Janeiro, RJ, Brazil

Correspondence e-mail: r.a.howie@abdn.ac.uk

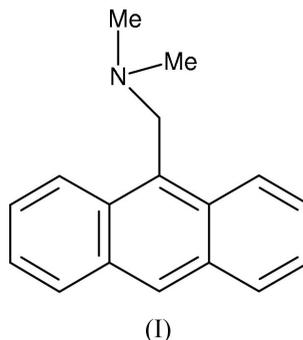
Key indicators

Single-crystal X-ray study
 $T = 120\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.055
 wR factor = 0.140
Data-to-parameter ratio = 18.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the structure of the title compound, $\text{C}_{17}\text{H}_{17}\text{N}$, the two molecules in the asymmetric unit are confined to distinct layers, one for each type of molecule. The layers differ in the orientation, relative to the edges of the unit cell, of the molecules within them.

Comment

The determination of the structure of the title compound, (I), reported here, follows on from the recent report of the structure of (anthracen-9-ylmethyl)diethylamine, (II) (Howie *et al.*, 2005). Compound (I) was unexpectedly isolated from a reaction mixture of 9-(chloromethyl)anthracene and 1,4,8,11-tetraazacyclotetradecane (cyclam) in *N,N*-dimethylformamide (DMF). Clearly, DMF had acted as a dimethylaminating reagent in the preparation of (I). There are scattered reports in the literature of DMF acting as a dimethylaminating agent in reactions with organic halides, activated for nucleophilic attack. Some examples include reactions with haloheteroarenes, such as chloropyridazines (Lee, Yoon & Kim, 2000) and bromopyridines (Watanabe *et al.*, 1980), acyl chlorides (Lee, Park & Yoon, 2000; Knunyants *et al.*, 1966), and (chloromethyl)arenes (Min'kov & Kravtsov, 1976). Subsequently, (I) was synthesized successfully by the reaction of 9-(chloromethyl)anthracene with excess Me_2NH in CH_2Cl_2 with a procedure similar to that used for (II) (Howie *et al.*, 2005), except that triethylamine was not added to the reaction mixture. The title compound has also been reported as the product of the Leuckart reaction between 9-anthracene-carboxaldehyde and DMF in 90% formic acid (Marcus & Fitzpatrick, 1959).



The asymmetric unit of (I) contains two molecules, which have been labelled in an identical manner (Fig. 1) and are distinguished by suffixes *A* and *B*. Leaving aside the difference in methyl and ethyl *N*-substituents, the molecular geometries of the molecules *A* and *B* of (I) and the molecule of (II) are, as would be expected, virtually identical. For the molecules of

Received 26 April 2005

Accepted 3 May 2005

Online 14 May 2005

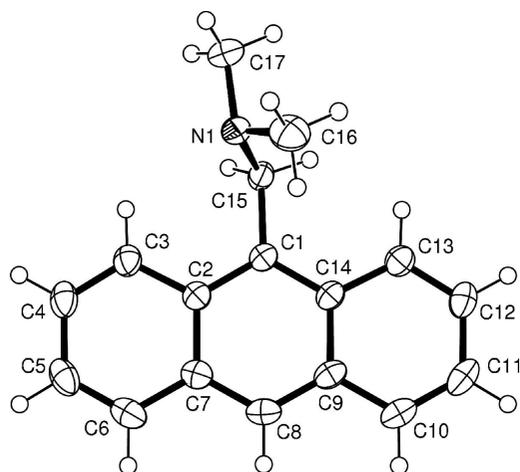


Figure 1
Molecule *A* of (I), showing the labelling scheme used for both molecules. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small circles of arbitrary radii.

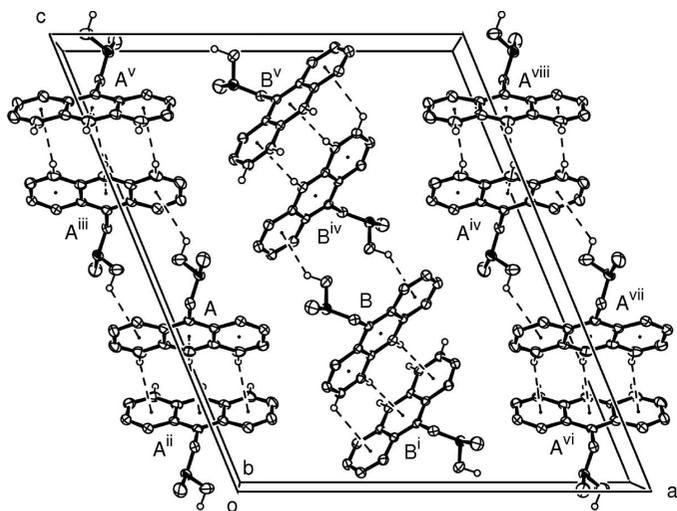


Figure 2
A view of the unit cell contents of (I). Displacement ellipsoids are drawn at the 50% probability level and H atoms involved in C—H... π interactions (dashed lines) are shown as small circles of arbitrary radii. Labels indicate molecule type. [Symmetry codes: (i) $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$; (ii) $-x, \frac{1}{2} + y, \frac{1}{2} - z$; (iii) $-x, 1 - y, 1 - z$; (iv) $1 - x, 1 - y, 1 - z$; (v) $x, \frac{1}{2} - y, \frac{1}{2} + z$; (vi) $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$; (vii) $1 + x, y, z$; (viii) $1 + x, \frac{1}{2} - y, \frac{1}{2} + z$.]

(I), the C—N distances and the C—N—C angles lie in the ranges 1.454 (3)–1.470 (2) Å and 109.43 (14)–111.58 (14)°, respectively; the C—C bond lengths and internal angles of the essentially planar anthracene ring systems (r.m.s. displacements for the atoms C1–C14 defining them of 0.0263 and 0.0376 Å) are in the ranges 1.352 (3)–1.446 (2) Å and 116.91 (15)–123.56 (16)°, respectively, and, finally, the C1—C15 bond length is 1.512 (2) Å in both molecules. It is noticeable that the pairs of values, one from each of the molecules in the bimolecular asymmetric unit from which the limiting values in the ranges given above are selected, always have the same designations, e.g. C5—C6 is the shortest bond in the anthracene ring system for both molecules. This fact provides a crude indication of the close similarity of the molecular geometries, as well as confirming the conformity of

the labelling scheme as applied to the two molecules. As shown by the torsion angles given in Table 1, the representative molecules *A* and *B* of the asymmetric unit of (I) are enantiomers. This arises purely from the choice of molecules because the centrosymmetric space group requires that the structure be completely racemic. In (I), the displacements of the atoms of the methylamino substituent from the least-squares plane defined by C1–C14, with the values for molecule *B* in square brackets, are 0.033 (2) [0.024 (2)], 2.387 (2) [2.383 (2)], 1.201 (4) [1.067 (3)] and 1.322 (2) Å [1.257 (2) Å], respectively for the atoms in the order C15, C16, C17 and N1, and are very similar to the displacements of the corresponding atoms in the molecule of (II). The anthracene moieties of molecules *A* and *B* of (I), as is the case for the molecule of (II), are in fact very slightly U shaped, as shown by the dihedral angles between the outer and inner rings, which are in the range 1.38 (10)–2.30 (10)°. Molecules *A* and *B* of (I) are found in separate layers parallel to (100), which differ (Fig. 2) in the orientation of the molecules within the unit cell. As a consequence, the type *A* and type *B* molecules differ slightly in the C—H... π intermolecular interactions (see later) in which they participate. For the choice of origin used in the refinement of the structure, the layers of type *A* molecules are centred on $x = 0$ and 1 and alternate with layers of type *B* molecules at $x = \frac{1}{2}$. Contacts between the molecules take the form of the C—H... π interactions given in Table 2 and occur entirely within the layers, as shown for type *A* molecules in Fig. 3. The connectivity within the layers in (I) is identical in form to that observed in (II). However, in comparing (I) and (II), the cell edges *b* and *c* are interchanged in length, as is the orientation of the molecules and therefore of the intermolecular connectivity within the layers relative to the symmetry elements of the space group $P2_1/c$, which is common to both structures. Moreover, in (II), neighbouring layers are related by cell translation in the direction of *a*, whereas in (I) they are not, and the cell edge *a* is therefore doubled in (I) compared with (II). Overall, the structures of (I) and (II) are closely related but the compounds are not isostructural.

Experimental

A solution of 9-chloromethylanthracene and cyclam (each 2 mmol) in dry DMF (20 ml) was refluxed for 6 h. Much of the solvent was then removed under high vacuum and the residue was chromatographed on a silica column, using as eluant hexane/ethyl acetate (ethyl acetate increasing from 5 to 100%). The pure title compound was obtained from intermediate fractions and was recrystallized from EtOH (m. p. 348–350 K).

Crystal data

$C_{17}H_{17}N$
 $M_r = 235.32$
 Monoclinic, $P2_1/c$
 $a = 19.6924$ (4) Å
 $b = 6.2383$ (1) Å
 $c = 23.4415$ (7) Å
 $\beta = 112.4743$ (10)°
 $V = 2661.01$ (11) Å³
 $Z = 8$

$D_x = 1.175$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 28 370 reflections
 $\theta = 2.9$ – 27.5 °
 $\mu = 0.07$ mm⁻¹
 $T = 120$ (2) K
 Plate, pale yellow
 $0.36 \times 0.16 \times 0.08$ mm

Data collection

Enraf–Nonius KappaCCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SORTAV; Blessing, 1995, 1997)
 $T_{\min} = 0.936$, $T_{\max} = 0.995$
 28 370 measured reflections

6034 independent reflections
 3938 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.085$
 $\theta_{\text{max}} = 27.5^\circ$
 $h = -23 \rightarrow 25$
 $k = -7 \rightarrow 8$
 $l = -30 \rightarrow 30$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.140$
 $S = 1.03$
 6034 reflections
 329 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.062P)^2 + 0.4189P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.18 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.22 \text{ e } \text{\AA}^{-3}$

Table 1

Selected torsion angles ($^\circ$).

C2A–C1A–C15A–N1A	71.54 (19)
C2B–C1B–C15B–N1B	–68.23 (19)
C14A–C1A–C15A–N1A	–109.24 (17)
C14B–C1B–C15B–N1B	114.10 (16)
C1A–C15A–N1A–C16A	65.22 (19)
C1B–C15B–N1B–C16B	–66.82 (18)
C1A–C15A–N1A–C17A	–172.65 (15)
C1B–C15B–N1B–C17B	170.50 (14)

Table 2

Geometry (\AA , $^\circ$) of the C–H... π contacts in (I).

C–H...Cg ^a	H...Cg	H _{perp} ^b	γ^c	C–H...Cg	C...Cg
C5B–H5B...Cg6 ⁱ	3.114	3.087	8	125	3.745
C6A–H6A...Cg5 ⁱⁱ	2.726	2.645	14	139	3.501
C6B–H6B...Cg2 ⁱ	2.537	2.480	12	138	3.306
C8A–H8A...Cg1 ⁱⁱ	2.879	2.819	12	144	3.688
C8B–H8B...Cg4 ⁱ	2.858	2.858	1	148	3.701
C10A–H10A...Cg3 ⁱⁱⁱ	2.933	2.884	10	147	3.763
C16A–H16A...Cg5 ⁱⁱⁱ	3.176	2.850	26	128	3.861
C16B–H16B...Cg6 ^{iv}	2.734	2.687	11	137	3.516

Notes: (a) Cg_n, n = 1–6, are the centroids of the rings C1A/C2A/C7A–C9A/C14A, C1B/C2B/C7B–C9B/C14B, C2A–C7A, C2B–C7B, C9A–C14A and C9B–C14B, respectively; (b) H_{perp} is the perpendicular distance of the H atom from the mean plane of the ring; (c) γ is the angle at the H atom between H...Cg and H_{perp}. Symmetry codes: (i) $1 - x, 1/2 + y, 1/2 - z$; (ii) $-x, 1/2 + y, 1/2 - z$; (iii) $-x, 1 - y, 1 - z$; (iv) $1 - x, 1 - y, 1 - z$.

In the final stages of refinement, H atoms were placed in calculated positions, with C–H = 0.95, 0.98 and 0.99 \AA for aryl, methyl and methylene H atoms, respectively, and refined with a riding model with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H atoms and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ otherwise. The rotational orientation of the methyl groups was also refined.

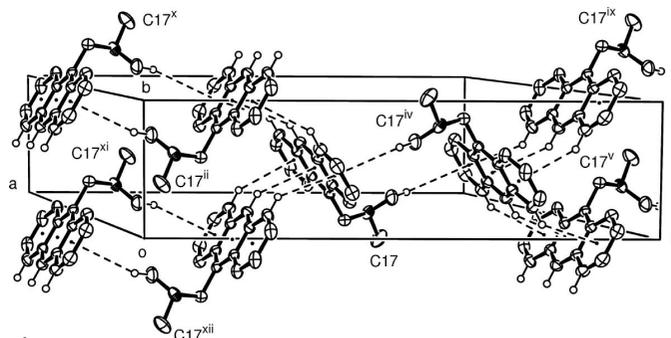


Figure 3

A layer of type A molecules of (I). Displacement ellipsoids are drawn at the 50% probability level and H atoms involved in C–H... π interactions (dashed lines) are shown as small circles of arbitrary radii. Selected atoms are labelled. [Symmetry codes: (ii) $-x, \frac{1}{2} + y, \frac{1}{2} - z$; (iv) $1 - x, 1 - y, 1 - z$; (v) $x, \frac{1}{2} - y, \frac{1}{2} + z$; (ix) $x, \frac{3}{2} - y, \frac{1}{2} + z$; (x) $x, \frac{3}{2} - y, z - \frac{1}{2}$; (xi) $x, \frac{1}{2} - y, z - \frac{1}{2}$; (xii) $-x, y - \frac{1}{2}, \frac{1}{2} - z$.]

Data collection: COLLECT (Hooft, 1998); cell refinement: DENZO (Otwinowski & Minor, 1997) and COLLECT; data reduction: DENZO and COLLECT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2003).

The use of the EPSRC X-ray crystallographic service at Southampton and the valuable assistance of the staff there is gratefully acknowledged.

References

Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–37.
 Blessing, R. H. (1997). *J. Appl. Cryst.* **30**, 421–426.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Hooft, R. W. W. (1998). COLLECT. Nonius BV, Delft, The Netherlands.
 Howie, R. A., Kindness, A., McKay, M. G. & Maguire, G. E. M. (2005). *Acta Cryst.* **E61**, o52–o54.
 Knunyants, Yu. A., Cheburkov, Yu. A. & Aronov, Yu. E. (1966). *Izv. Akad. Nauk SSSR Ser. Khim.* pp. 1038–1047.
 Lee, W. S., Park, K. H. & Yoon, Y. J. (2000). *Synth. Commun.* **30**, 4241–4245.
 Lee, W. S., Yoon, Y. J. & Kim, S. K. (2000). *J. Heterocycl. Chem.* **37**, 1591–1595.
 Marcus, E. & Fitzpatrick, J. T. (1959). *J. Org. Chem.* pp. 1031–1032.
 Min'kov, V. A. & Kravtsov, V. S. (1976). *Vopr. Khim. Khim. Tekhnol.* **43**, 1213–126.
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
 Watanabe, T., Tanaka, Y., Sekiya, K., Akita, Y. & Ohta, A. (1980). *Synthesis*, pp. 39–43.

supporting information

Acta Cryst. (2005). E61, o1686–o1688 [https://doi.org/10.1107/S1600536805014200]

(Anthracen-9-ylmethyl)dimethylamine at 120 K

R. Alan Howie and Solange M. S. V. Wardell

(Anthracen-9-ylmethyl)dimethylamine

Crystal data

$C_{17}H_{17}N$

$M_r = 235.32$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2ybc$

$a = 19.6924$ (4) Å

$b = 6.2383$ (1) Å

$c = 23.4415$ (7) Å

$\beta = 112.4743$ (10)°

$V = 2661.01$ (11) Å³

$Z = 8$

$F(000) = 1008$

$D_x = 1.175$ Mg m⁻³

Melting point = 348–350 K

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 28370 reflections

$\theta = 2.9$ – 27.5 °

$\mu = 0.07$ mm⁻¹

$T = 120$ K

Plate, pale yellow

$0.36 \times 0.16 \times 0.08$ mm

Data collection

Enraf–Nonius KappaCCD area-detector diffractometer

Radiation source: Enraf–Nonius FR591 rotating anode

10 cm confocal mirrors monochromator

Detector resolution: 9.091 pixels mm⁻¹

φ and ω scans

Absorption correction: multi-scan (SORTAV; Blessing, 1995, 1997)

$T_{\min} = 0.936$, $T_{\max} = 0.995$

28370 measured reflections

6034 independent reflections

3938 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.085$

$\theta_{\max} = 27.5$ °, $\theta_{\min} = 3.2$ °

$h = -23 \rightarrow 25$

$k = -7 \rightarrow 8$

$l = -30 \rightarrow 30$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.055$

$wR(F^2) = 0.140$

$S = 1.03$

6034 reflections

329 parameters

0 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.062P)^2 + 0.4189P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.18$ e Å⁻³

$\Delta\rho_{\min} = -0.22$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Least-squares planes (x, y, z in crystal coordinates) and deviations from them (* indicates atom used to define plane)

- 0.6341 (40) x + 3.3132 (23) y + 18.6289 (52) z = 7.9481 (8)

* 0.0299 (0.0014) C1A * 0.0255 (0.0015) C2A * 0.0098 (0.0015) C3A * -0.0317 (0.0016) C4A * -0.0446 (0.0016) C5A * -0.0060 (0.0015) C6A * 0.0213 (0.0015) C7A * 0.0251 (0.0015) C8A * 0.0283 (0.0016) C9A * 0.0121 (0.0015) C10A * -0.0325 (0.0016) C11A * -0.0386 (0.0016) C12A * -0.0134 (0.0015) C13A * 0.0147 (0.0015) C14A 0.0330 (0.0022) C15A 2.3866 (0.0024) C16A 1.2006 (0.0035) C17A 1.3221 (0.0022) N1A

Rms deviation of fitted atoms = 0.0263

- 11.7825 (45) x + 3.1263 (22) y + 18.9037 (43) z = 2.1947 (26)

Angle to previous plane (with approximate e.s.d.) = 35.44 (0.03)

* 0.0433 (0.0014) C1B * 0.0372 (0.0014) C2B * 0.0182 (0.0014) C3B * -0.0396 (0.0015) C4B * -0.0617 (0.0015) C5B * -0.0248 (0.0014) C6B * 0.0301 (0.0014) C7B * 0.0484 (0.0014) C8B * 0.0379 (0.0015) C9B * 0.0106 (0.0015) C10B * -0.0318 (0.0015) C11B * -0.0564 (0.0015) C12B * -0.0300 (0.0014) C13B * 0.0186 (0.0014) C14B 0.0236 (0.0021) C15B 2.3831 (0.0023) C16B 1.0672 (0.0032) C17B 1.2574 (0.0021) N1B

Rms deviation of fitted atoms = 0.0376

- 0.1452 (139) x + 3.3520 (37) y + 18.3338 (102) z = 7.9399 (11)

Angle to previous plane (with approximate e.s.d.) = 36.78 (0.05)

* -0.0055 (0.0011) C2A * 0.0068 (0.0012) C3A * -0.0012 (0.0013) C4A * -0.0060 (0.0013) C5A * 0.0070 (0.0012) C6A * -0.0011 (0.0012) C7A

Rms deviation of fitted atoms = 0.0052

- 0.6494 (132) x + 3.3130 (35) y + 18.6357 (94) z = 7.9740 (18)

Angle to previous plane (with approximate e.s.d.) = 1.52 (0.10)

* 0.0057 (0.0011) C1A * 0.0004 (0.0011) C2A * -0.0038 (0.0012) C7A * 0.0010 (0.0012) C8A * 0.0051 (0.0012) C9A * -0.0085 (0.0011) C14A

Rms deviation of fitted atoms = 0.0049

- 1.0916 (142) x + 3.2627 (38) y + 18.9201 (99) z = 8.0600 (33)

Angle to previous plane (with approximate e.s.d.) = 1.38 (0.10)

* -0.0015 (0.0012) C9A * 0.0059 (0.0013) C10A * -0.0054 (0.0013) C11A * 0.0003 (0.0013) C12A * 0.0040 (0.0012) C13A * -0.0033 (0.0012) C14A

Rms deviation of fitted atoms = 0.0040

- 12.0887 (105) x + 3.2295 (36) y + 18.4099 (97) z = 2.0021 (58)

Angle to previous plane (with approximate e.s.d.) = 35.74 (0.07)

* -0.0085 (0.0011) C2B * 0.0070 (0.0012) C3B * -0.0009 (0.0012) C4B * -0.0037 (0.0012) C5B * 0.0018 (0.0012) C6B * 0.0043 (0.0011) C7B

Rms deviation of fitted atoms = 0.0051

- 11.7773 (98) x + 3.1106 (33) y + 18.9504 (85) z = 2.2407 (66)

Angle to previous plane (with approximate e.s.d.) = 2.30 (0.10)

* 0.0111 (0.0011) C1B * 0.0004 (0.0011) C2B * -0.0104 (0.0011) C7B * 0.0089 (0.0011) C8B * 0.0028 (0.0011) C9B * -0.0127 (0.0011) C14B

Rms deviation of fitted atoms = 0.0090

- 11.5170 (109) x + 3.0215 (38) y + 19.3350 (92) z = 2.4863 (87)

Angle to previous plane (with approximate e.s.d.) = 1.74 (0.10)

* -0.0007 (0.0011) C9B * -0.0008 (0.0012) C10B * 0.0021 (0.0013) C11B * -0.0018 (0.0013) C12B * 0.0003 (0.0012) C13B * 0.0010 (0.0011) C14B

Rms deviation of fitted atoms = 0.0013

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1A	0.12180 (8)	0.1488 (2)	0.47531 (7)	0.0333 (4)
C1A	0.04902 (9)	0.3139 (3)	0.37409 (7)	0.0247 (4)
C2A	0.10193 (9)	0.4194 (3)	0.35690 (7)	0.0257 (4)
C3A	0.17505 (9)	0.3397 (3)	0.37272 (8)	0.0317 (4)
H3A	0.1896	0.2129	0.3968	0.038*
C4A	0.22424 (10)	0.4418 (3)	0.35401 (8)	0.0373 (5)
H4A	0.2722	0.3845	0.3647	0.045*
C5A	0.20426 (11)	0.6328 (3)	0.31867 (8)	0.0385 (5)
H5A	0.2388	0.7022	0.3056	0.046*
C6A	0.13624 (11)	0.7163 (3)	0.30357 (8)	0.0357 (5)
H6A	0.1239	0.8459	0.2806	0.043*
C7A	0.08265 (9)	0.6138 (3)	0.32144 (7)	0.0287 (4)
C8A	0.01177 (10)	0.6966 (3)	0.30452 (8)	0.0312 (4)
H8A	-0.0007	0.8254	0.2812	0.037*
C9A	-0.04113 (9)	0.5958 (3)	0.32082 (8)	0.0290 (4)
C10A	-0.11383 (10)	0.6799 (3)	0.30250 (8)	0.0353 (4)
H10A	-0.1258	0.8105	0.2800	0.042*
C11A	-0.16610 (10)	0.5773 (3)	0.31658 (9)	0.0396 (5)
H11A	-0.2144	0.6347	0.3033	0.048*
C12A	-0.14879 (10)	0.3849 (3)	0.35107 (9)	0.0365 (4)
H12A	-0.1856	0.3147	0.3611	0.044*
C13A	-0.08020 (9)	0.2988 (3)	0.37006 (8)	0.0318 (4)
H13A	-0.0700	0.1697	0.3933	0.038*
C14A	-0.02282 (9)	0.3985 (3)	0.35580 (8)	0.0264 (4)
C15A	0.07100 (9)	0.1096 (3)	0.41135 (8)	0.0277 (4)
H15A	0.0263	0.0377	0.4117	0.033*
H15B	0.0947	0.0118	0.3912	0.033*
C16A	0.08701 (12)	0.2721 (3)	0.50934 (9)	0.0450 (5)
H16A	0.1228	0.3004	0.5511	0.067*
H16B	0.0456	0.1906	0.5117	0.067*
H16C	0.0691	0.4083	0.4881	0.067*
C17A	0.14812 (12)	-0.0559 (4)	0.50609 (11)	0.0544 (6)
H17A	0.1718	-0.1371	0.4830	0.082*
H17B	0.1065	-0.1382	0.5078	0.082*
H17C	0.1837	-0.0298	0.5481	0.082*
N1B	0.39723 (8)	0.1841 (2)	0.39976 (7)	0.0277 (3)
C1B	0.48528 (9)	0.3584 (3)	0.36158 (7)	0.0218 (4)
C2B	0.43857 (9)	0.4833 (2)	0.31150 (7)	0.0215 (3)
C3B	0.36472 (9)	0.4218 (3)	0.27462 (8)	0.0262 (4)
H3B	0.3450	0.2951	0.2848	0.031*
C4B	0.32186 (9)	0.5414 (3)	0.22507 (8)	0.0294 (4)
H4B	0.2733	0.4956	0.2010	0.035*
C5B	0.34912 (9)	0.7332 (3)	0.20918 (8)	0.0290 (4)
H5B	0.3189	0.8152	0.1746	0.035*
C6B	0.41853 (9)	0.7994 (3)	0.24344 (8)	0.0251 (4)

H6B	0.4363	0.9284	0.2326	0.030*
C7B	0.46527 (8)	0.6791 (2)	0.29538 (7)	0.0207 (3)
C8B	0.53689 (9)	0.7461 (3)	0.32991 (7)	0.0244 (4)
H8B	0.5539	0.8777	0.3198	0.029*
C9B	0.58410 (9)	0.6246 (3)	0.37888 (7)	0.0238 (4)
C10B	0.65828 (9)	0.6913 (3)	0.41263 (8)	0.0304 (4)
H10B	0.6751	0.8224	0.4020	0.036*
C11B	0.70529 (10)	0.5711 (3)	0.45957 (9)	0.0362 (5)
H11B	0.7543	0.6182	0.4816	0.043*
C12B	0.68060 (10)	0.3759 (3)	0.47515 (8)	0.0350 (4)
H12B	0.7137	0.2919	0.5078	0.042*
C13B	0.61039 (9)	0.3045 (3)	0.44460 (8)	0.0291 (4)
H13B	0.5956	0.1725	0.4565	0.035*
C14B	0.55856 (9)	0.4252 (3)	0.39490 (7)	0.0231 (4)
C15B	0.45519 (9)	0.1515 (3)	0.37601 (8)	0.0261 (4)
H15C	0.4354	0.0637	0.3380	0.031*
H15D	0.4959	0.0702	0.4069	0.031*
C16B	0.42588 (10)	0.2858 (3)	0.46034 (8)	0.0346 (4)
H16D	0.3858	0.3074	0.4748	0.052*
H16E	0.4476	0.4247	0.4574	0.052*
H16F	0.4635	0.1938	0.4896	0.052*
C17B	0.36320 (12)	-0.0213 (3)	0.40245 (10)	0.0460 (5)
H17D	0.3232	0.0011	0.4170	0.069*
H17E	0.4000	-0.1172	0.4310	0.069*
H17F	0.3435	-0.0857	0.3612	0.069*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1A	0.0254 (8)	0.0407 (9)	0.0295 (8)	0.0005 (7)	0.0056 (6)	0.0096 (7)
C1A	0.0245 (9)	0.0252 (9)	0.0212 (8)	-0.0005 (7)	0.0051 (7)	-0.0035 (7)
C2A	0.0250 (9)	0.0290 (9)	0.0194 (8)	-0.0037 (7)	0.0045 (7)	-0.0035 (7)
C3A	0.0256 (10)	0.0422 (11)	0.0251 (9)	-0.0016 (8)	0.0072 (7)	0.0000 (8)
C4A	0.0244 (10)	0.0582 (13)	0.0265 (9)	-0.0066 (9)	0.0065 (8)	-0.0025 (8)
C5A	0.0342 (11)	0.0516 (12)	0.0288 (10)	-0.0175 (9)	0.0112 (8)	-0.0046 (9)
C6A	0.0441 (12)	0.0344 (10)	0.0248 (9)	-0.0131 (8)	0.0089 (9)	-0.0004 (8)
C7A	0.0320 (10)	0.0279 (9)	0.0221 (8)	-0.0051 (7)	0.0059 (7)	-0.0030 (7)
C8A	0.0384 (11)	0.0249 (9)	0.0249 (9)	-0.0002 (8)	0.0061 (8)	0.0018 (7)
C9A	0.0302 (10)	0.0283 (9)	0.0223 (8)	0.0042 (7)	0.0031 (7)	-0.0030 (7)
C10A	0.0366 (11)	0.0317 (10)	0.0300 (10)	0.0098 (8)	0.0045 (8)	0.0003 (8)
C11A	0.0285 (11)	0.0475 (12)	0.0359 (11)	0.0134 (9)	0.0045 (9)	-0.0040 (9)
C12A	0.0242 (10)	0.0459 (11)	0.0372 (10)	0.0011 (8)	0.0092 (8)	-0.0035 (9)
C13A	0.0277 (10)	0.0344 (10)	0.0297 (9)	-0.0001 (8)	0.0070 (8)	-0.0009 (8)
C14A	0.0243 (9)	0.0279 (9)	0.0232 (8)	0.0008 (7)	0.0048 (7)	-0.0035 (7)
C15A	0.0223 (9)	0.0269 (9)	0.0326 (9)	0.0010 (7)	0.0090 (7)	0.0007 (7)
C16A	0.0511 (13)	0.0529 (13)	0.0279 (10)	-0.0017 (10)	0.0118 (9)	-0.0005 (9)
C17A	0.0436 (13)	0.0628 (15)	0.0560 (14)	0.0198 (11)	0.0182 (11)	0.0327 (12)
N1B	0.0261 (8)	0.0282 (8)	0.0290 (8)	-0.0047 (6)	0.0105 (6)	0.0030 (6)

C1B	0.0242 (9)	0.0223 (8)	0.0220 (8)	0.0030 (6)	0.0122 (7)	-0.0016 (6)
C2B	0.0231 (9)	0.0231 (8)	0.0212 (8)	0.0009 (6)	0.0118 (7)	-0.0022 (6)
C3B	0.0257 (9)	0.0263 (9)	0.0269 (9)	-0.0044 (7)	0.0106 (7)	-0.0008 (7)
C4B	0.0221 (9)	0.0362 (10)	0.0283 (9)	-0.0028 (7)	0.0077 (7)	0.0021 (7)
C5B	0.0267 (10)	0.0358 (10)	0.0251 (9)	0.0046 (7)	0.0104 (8)	0.0079 (7)
C6B	0.0260 (9)	0.0251 (9)	0.0281 (9)	0.0004 (7)	0.0144 (7)	0.0036 (7)
C7B	0.0219 (8)	0.0211 (8)	0.0225 (8)	0.0016 (6)	0.0124 (7)	-0.0006 (6)
C8B	0.0274 (9)	0.0230 (8)	0.0274 (9)	-0.0009 (7)	0.0157 (7)	-0.0019 (7)
C9B	0.0224 (9)	0.0285 (9)	0.0235 (8)	-0.0007 (7)	0.0121 (7)	-0.0038 (7)
C10B	0.0262 (9)	0.0390 (10)	0.0275 (9)	-0.0039 (8)	0.0121 (8)	-0.0047 (8)
C11B	0.0222 (10)	0.0548 (12)	0.0304 (10)	-0.0018 (8)	0.0089 (8)	-0.0058 (9)
C12B	0.0260 (10)	0.0490 (12)	0.0281 (9)	0.0100 (8)	0.0081 (8)	0.0051 (8)
C13B	0.0278 (10)	0.0335 (10)	0.0279 (9)	0.0065 (7)	0.0127 (8)	0.0026 (7)
C14B	0.0226 (9)	0.0281 (9)	0.0211 (8)	0.0036 (7)	0.0110 (7)	-0.0026 (6)
C15B	0.0309 (10)	0.0222 (9)	0.0252 (8)	0.0016 (7)	0.0105 (7)	0.0005 (7)
C16B	0.0361 (11)	0.0409 (11)	0.0321 (10)	-0.0004 (8)	0.0190 (8)	-0.0021 (8)
C17B	0.0471 (13)	0.0416 (12)	0.0471 (12)	-0.0183 (10)	0.0157 (10)	0.0051 (10)

Geometric parameters (Å, °)

N1A—C16A	1.454 (3)	N1B—C16B	1.458 (2)
N1A—C17A	1.461 (2)	N1B—C17B	1.458 (2)
N1A—C15A	1.470 (2)	N1B—C15B	1.463 (2)
C1A—C14A	1.414 (2)	C1B—C14B	1.416 (2)
C1A—C2A	1.415 (2)	C1B—C2B	1.416 (2)
C1A—C15A	1.512 (2)	C1B—C15B	1.512 (2)
C2A—C3A	1.432 (2)	C2B—C3B	1.431 (2)
C2A—C7A	1.437 (2)	C2B—C7B	1.436 (2)
C3A—C4A	1.364 (3)	C3B—C4B	1.367 (2)
C3A—H3A	0.9500	C3B—H3B	0.9500
C4A—C5A	1.418 (3)	C4B—C5B	1.418 (2)
C4A—H4A	0.9500	C4B—H4B	0.9500
C5A—C6A	1.352 (3)	C5B—C6B	1.358 (2)
C5A—H5A	0.9500	C5B—H5B	0.9500
C6A—C7A	1.427 (2)	C6B—C7B	1.427 (2)
C6A—H6A	0.9500	C6B—H6B	0.9500
C7A—C8A	1.396 (2)	C7B—C8B	1.395 (2)
C8A—C9A	1.390 (3)	C8B—C9B	1.393 (2)
C8A—H8A	0.9500	C8B—H8B	0.9500
C9A—C10A	1.428 (2)	C9B—C10B	1.432 (2)
C9A—C14A	1.446 (2)	C9B—C14B	1.444 (2)
C10A—C11A	1.356 (3)	C10B—C11B	1.361 (3)
C10A—H10A	0.9500	C10B—H10B	0.9500
C11A—C12A	1.414 (3)	C11B—C12B	1.410 (3)
C11A—H11A	0.9500	C11B—H11B	0.9500
C12A—C13A	1.361 (2)	C12B—C13B	1.366 (2)
C12A—H12A	0.9500	C12B—H12B	0.9500
C13A—C14A	1.436 (2)	C13B—C14B	1.434 (2)

C13A—H13A	0.9500	C13B—H13B	0.9500
C15A—H15A	0.9900	C15B—H15C	0.9900
C15A—H15B	0.9900	C15B—H15D	0.9900
C16A—H16A	0.9800	C16B—H16D	0.9800
C16A—H16B	0.9800	C16B—H16E	0.9800
C16A—H16C	0.9800	C16B—H16F	0.9800
C17A—H17A	0.9800	C17B—H17D	0.9800
C17A—H17B	0.9800	C17B—H17E	0.9800
C17A—H17C	0.9800	C17B—H17F	0.9800
C16A—N1A—C17A	110.12 (16)	C16B—N1B—C17B	110.70 (15)
C16A—N1A—C15A	111.58 (14)	C16B—N1B—C15B	111.32 (13)
C17A—N1A—C15A	109.45 (16)	C17B—N1B—C15B	109.43 (14)
C14A—C1A—C2A	120.04 (15)	C14B—C1B—C2B	119.84 (14)
C14A—C1A—C15A	121.46 (15)	C14B—C1B—C15B	122.14 (14)
C2A—C1A—C15A	118.50 (15)	C2B—C1B—C15B	117.98 (14)
C1A—C2A—C3A	122.84 (16)	C1B—C2B—C3B	122.79 (15)
C1A—C2A—C7A	119.79 (15)	C1B—C2B—C7B	119.91 (14)
C3A—C2A—C7A	117.37 (16)	C3B—C2B—C7B	117.29 (14)
C4A—C3A—C2A	121.49 (18)	C4B—C3B—C2B	121.46 (16)
C4A—C3A—H3A	119.3	C4B—C3B—H3B	119.3
C2A—C3A—H3A	119.3	C2B—C3B—H3B	119.3
C3A—C4A—C5A	120.49 (18)	C3B—C4B—C5B	120.73 (16)
C3A—C4A—H4A	119.8	C3B—C4B—H4B	119.6
C5A—C4A—H4A	119.8	C5B—C4B—H4B	119.6
C6A—C5A—C4A	120.16 (18)	C6B—C5B—C4B	119.85 (15)
C6A—C5A—H5A	119.9	C6B—C5B—H5B	120.1
C4A—C5A—H5A	119.9	C4B—C5B—H5B	120.1
C5A—C6A—C7A	121.36 (18)	C5B—C6B—C7B	121.30 (15)
C5A—C6A—H6A	119.3	C5B—C6B—H6B	119.4
C7A—C6A—H6A	119.3	C7B—C6B—H6B	119.4
C8A—C7A—C6A	121.49 (17)	C8B—C7B—C6B	121.10 (14)
C8A—C7A—C2A	119.38 (16)	C8B—C7B—C2B	119.52 (14)
C6A—C7A—C2A	119.12 (16)	C6B—C7B—C2B	119.36 (14)
C9A—C8A—C7A	121.85 (16)	C9B—C8B—C7B	121.55 (15)
C9A—C8A—H8A	119.1	C9B—C8B—H8B	119.2
C7A—C8A—H8A	119.1	C7B—C8B—H8B	119.2
C8A—C9A—C10A	121.44 (17)	C8B—C9B—C10B	121.05 (15)
C8A—C9A—C14A	119.38 (16)	C8B—C9B—C14B	119.59 (15)
C10A—C9A—C14A	119.16 (17)	C10B—C9B—C14B	119.34 (15)
C11A—C10A—C9A	121.36 (17)	C11B—C10B—C9B	121.44 (17)
C11A—C10A—H10A	119.3	C11B—C10B—H10B	119.3
C9A—C10A—H10A	119.3	C9B—C10B—H10B	119.3
C10A—C11A—C12A	120.03 (17)	C10B—C11B—C12B	119.35 (17)
C10A—C11A—H11A	120.0	C10B—C11B—H11B	120.3
C12A—C11A—H11A	120.0	C12B—C11B—H11B	120.3
C13A—C12A—C11A	120.95 (19)	C13B—C12B—C11B	121.73 (17)
C13A—C12A—H12A	119.5	C13B—C12B—H12B	119.1

C11A—C12A—H12A	119.5	C11B—C12B—H12B	119.1
C12A—C13A—C14A	121.59 (18)	C12B—C13B—C14B	121.21 (17)
C12A—C13A—H13A	119.2	C12B—C13B—H13B	119.4
C14A—C13A—H13A	119.2	C14B—C13B—H13B	119.4
C1A—C14A—C13A	123.56 (16)	C1B—C14B—C13B	123.54 (15)
C1A—C14A—C9A	119.53 (16)	C1B—C14B—C9B	119.53 (14)
C13A—C14A—C9A	116.91 (15)	C13B—C14B—C9B	116.92 (15)
N1A—C15A—C1A	112.44 (14)	N1B—C15B—C1B	113.32 (13)
N1A—C15A—H15A	109.1	N1B—C15B—H15C	108.9
C1A—C15A—H15A	109.1	C1B—C15B—H15C	108.9
N1A—C15A—H15B	109.1	N1B—C15B—H15D	108.9
C1A—C15A—H15B	109.1	C1B—C15B—H15D	108.9
H15A—C15A—H15B	107.8	H15C—C15B—H15D	107.7
N1A—C16A—H16A	109.5	N1B—C16B—H16D	109.5
N1A—C16A—H16B	109.5	N1B—C16B—H16E	109.5
H16A—C16A—H16B	109.5	H16D—C16B—H16E	109.5
N1A—C16A—H16C	109.5	N1B—C16B—H16F	109.5
H16A—C16A—H16C	109.5	H16D—C16B—H16F	109.5
H16B—C16A—H16C	109.5	H16E—C16B—H16F	109.5
N1A—C17A—H17A	109.5	N1B—C17B—H17D	109.5
N1A—C17A—H17B	109.5	N1B—C17B—H17E	109.5
H17A—C17A—H17B	109.5	H17D—C17B—H17E	109.5
N1A—C17A—H17C	109.5	N1B—C17B—H17F	109.5
H17A—C17A—H17C	109.5	H17D—C17B—H17F	109.5
H17B—C17A—H17C	109.5	H17E—C17B—H17F	109.5
C2A—C1A—C15A—N1A	71.54 (19)	C1A—C15A—N1A—C16A	65.22 (19)
C2B—C1B—C15B—N1B	-68.23 (19)	C1B—C15B—N1B—C16B	-66.82 (18)
C14A—C1A—C15A—N1A	-109.24 (17)	C1A—C15A—N1A—C17A	-172.65 (15)
C14B—C1B—C15B—N1B	114.10 (16)	C1B—C15B—N1B—C17B	170.50 (14)
