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Key indicators

Powder synchrotron study
 $T = 290$ K
 Mean $\sigma(C-C) = 0.006$ Å
 R factor = 0.025
 wR factor = 0.030
 Data-to-parameter ratio = 75.44

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Ampicillin trihydrate from synchrotron powder diffraction data

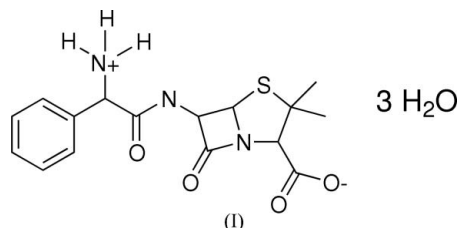
The crystal structure of ampicillin trihydrate {systematic name: 6-[D(-)- α -aminophenylacetamido]penicillanic acid trihydrate}, $C_{16}H_{19}N_3O_4S \cdot 3H_2O$, a broad-spectrum β -lactam antibiotic of the aminopenicillin type, has been determined from synchrotron X-ray powder diffraction data. The three water molecules form an infinite hydrogen-bonded chain through the crystal structure, with hydrogen bonds to the NH_3^+ , COO^- , $C=O$ and NH groups of the ampicillin molecules.

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Comment

The title compound, (I), has been used as a broad-spectrum antibiotic since 1961. The crystal structure was reported in 1968 (James *et al.*, 1968), but no atomic coordinates were given in the paper or deposited. Boles *et al.* (1978) published the crystal structure of a related compound, amoxicillin trihydrate. They apparently had access to the atomic coordinates of the crystal structure of compound (I), because in their paper they show that the two crystal structures are isostructural. However, the atomic coordinates of the title compound have not been published to date. We report the crystal structure here, determined from synchrotron X-ray powder diffraction.



The structural model of compound (I) obtained in the present work (Fig. 1a) is both chemically reasonable and in accord with the figures given by James *et al.* (1968). Selected geometric parameters are given in Table 1. We note, however, that the hydrogen bond $O26''' \cdots O25'''$ in their Fig. 1, which appears to link four water molecules together into a closed tetramer, is spurious, and instead should have formed a chain (Fig. 1b). Both the pattern of hydrogen bonding, and the positions of the H atoms of the water molecules in the structure, are chemically sensible and compare well with those from the crystal structure of the isostructural amoxicillin trihydrate (Boles *et al.*, 1978). Details of the $O-H \cdots O$ and $N-H \cdots O$ hydrogen bonds are given in Table 2 and Fig. 1.

Experimental

The sample of compound (I) was a gift from Setauket Pharmacy, Setauket, New York, USA, in the form of a gelatin capsule of the

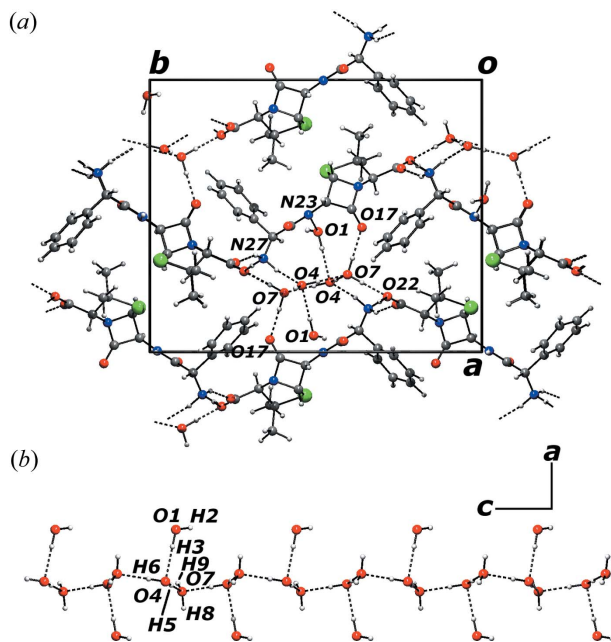


Figure 1
 (a) A view, along the *c* axis, of the crystal structure of compound (I), showing the O—H···O and N—H···O hydrogen bonds as dashed lines (see Table 2 for details). (b) A view along the *b* axis of the hydrogen-bonded (dashed lines) water network in the crystal structure of compound (I). [A screw axis is present at $(\frac{1}{2}, 0, z)$].

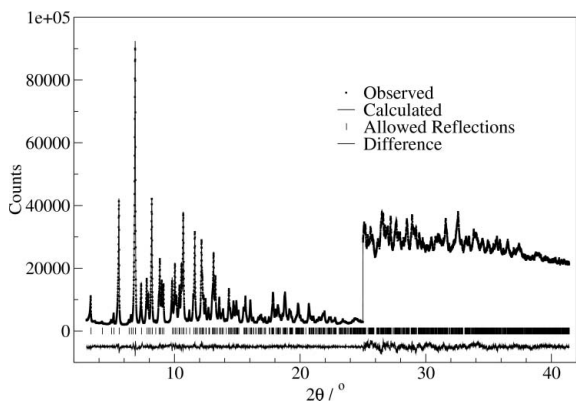


Figure 2
 Observed, calculated and difference X-ray powder diffraction profiles for compound (I). The region 25–42° in 2θ has been magnified 10 times.

compound. Some of the contents were loaded into a thin-walled glass capillary of 1.5 mm nominal diameter. Any excipients that might have been present were not crystalline. A diffraction pattern was collected at the X3B1 beamline of the National Synchrotron Light Source, Brookhaven National Laboratory. The wavelength of 0.7003 (1) Å was selected by a double Si(111) monochromator and the diffracted beam analyzed by a Ge(111) crystal before the detector. The beam on the sample had dimensions 2 mm × 8 mm. Data were collected from $2\theta = 3\text{--}41.6^\circ$ in steps of 0.005° , with counting time increasing quadratically from 1–8 s per point. The incident beam was monitored by an ion chamber, which was used to normalize the data for decay and fluctuations of the intensity.

Crystal data

$C_{16}H_{19}N_3O_4S \cdot 3H_2O$
 $M_r = 403.06$
 Orthorhombic, $P2_12_12_1$
 $a = 15.52275$ (16) Å
 $b = 18.9256$ (3) Å
 $c = 6.67375$ (8) Å
 $V = 1960.60$ (3) Å³
 $Z = 4$

$D_x = 1.367$ Mg m⁻³
 Synchrotron radiation
 $\lambda = 0.7003$ Å
 $T = 290$ K
 Specimen shape: cylinder
 0.7 × 20 mm
 Particle morphology: powder, white

Data collection

Huber 424 Q -2Q diffractometer on X3B1 beamline
 Specimen mounting: Lindemann glass capillary
 Specimen mounted in transmission mode

Scan method: step
 $T = 290$ K
 $2\theta_{min} = 3, 2\theta_{max} = 41.6^\circ$
 Increment in $2\theta = 0.005^\circ$

Refinement

$R_p = 0.025$
 $R_{wp} = 0.030$
 $R_{exp} = 0.013$
 $S = 2.37$
 $2\theta_{min} = 3, 2\theta_{max} = 41.6^\circ$
 Increment in $2\theta = 0.005^\circ$
 Wavelength of incident radiation: 0.7003 Å
 Excluded region(s): none
 Profile function: CW profile function number 3 with 19 terms.
 Pseudo-Voigt profile coefficients as parameterized in Thompson *et*

al. (1987). Asymmetry correction of Finger *et al.* (1994). Peak tails are ignored where the intensity is below 0.0010 times the peak
 10788 reflections
 143 parameters
 H-atom parameters constrained
 Weighting scheme based on measured s.u. values
 $(\Delta/\sigma)_{max} = 0.03$
 Preferred orientation correction: none

Table 1
 Selected geometric parameters (Å, °).

S10—C11	1.859 (4)	N13—C12	1.473 (5)
S10—C14	1.797 (3)	N13—C14	1.506 (5)
O17—C16	1.201 (5)	N13—C16	1.384 (5)
O21—C20	1.222 (6)	N23—C15	1.433 (5)
O22—C20	1.273 (6)	N23—C24	1.348 (7)
O25—C24	1.213 (5)	N27—C26	1.475 (6)
C11—S10—C14	90.10 (18)	N23—C15—C16	115.5 (3)
C12—N13—C14	117.0 (3)	N23—C15—C14	116.5 (3)
C12—N13—C16	128.1 (3)	O17—C16—C15	135.4 (3)
C14—N13—C16	93.1 (3)	O17—C16—N13	131.0 (4)
C15—N23—C24	123.3 (3)	N13—C16—C15	93.0 (3)
S10—C11—C19	107.5 (3)	O21—C20—C12	118.0 (4)
S10—C11—C12	104.6 (2)	O21—C20—O22	126.3 (4)
S10—C11—C18	110.0 (3)	O22—C20—C12	115.6 (4)
N13—C12—C11	105.6 (3)	O25—C24—N23	124.5 (5)
N13—C12—C20	112.8 (3)	O25—C24—C26	122.1 (5)
N13—C14—C15	87.1 (2)	N23—C24—C26	113.2 (3)
S10—C14—N13	103.6 (2)	N27—C26—C24	110.0 (3)
S10—C14—C15	119.5 (2)	N27—C26—C28	112.9 (3)

Table 2
 Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1—H3···O4	0.98	2.02	2.9968	180
O4—H5···O7 ⁱ	0.98	1.80	2.7850	180
O4—H6···O7	0.98	1.82	2.7967	180
O7—H8···O17	0.98	1.84	2.8225	180
O7—H9···O22 ⁱⁱ	0.98	1.74	2.7166	179
N27—H38···O4 ⁱⁱⁱ	1.001 (15)	1.86	2.8272	161
N27—H39···O21 ^{iv}	1.00 (3)	1.83 (3)	2.742 (6)	151.1 (15)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N27–H40 \cdots O22 ^v	1.00 (3)	1.80 (2)	2.688 (6)	147 (2)
N23–H51 \cdots O1	1.002 (10)	1.97	2.9161	156

Symmetry codes: (i) $-x + \frac{3}{2}, -y + 1, z - \frac{1}{2}$; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1$; (iii) $-x + \frac{3}{2}, -y + 1, z + \frac{1}{2}$; (iv) $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$; (v) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$.

The starting model for Rietveld refinement was obtained by solving the crystal structure from the powder diffraction pattern. This also provided an independent check that the published crystal structure is correct. However, with the crystal structure being known, its determination from the powder pattern is mainly academic. The crystal structure was determined with the program *DASH* (David *et al.*, 2004). For the structure solution, the data were truncated at 22.855° in 2θ , corresponding to a real-space resolution of 1.767 \AA . The background was subtracted with a Bayesian high-pass filter (David & Sivia, 2001). Peak positions for indexing were obtained by fitting with an asymmetry-corrected Voigt function, followed by indexing with the program *DICVOL* (Boultif & Louer, 1991). An orthorhombic and several monoclinic unit cells were obtained. However, all the monoclinic unit cells were pseudo-orthorhombic with nearly the same parameters as the orthorhombic cell, indicating that the orthorhombic unit cell is the correct one. The figures of merit given by *DICVOL* were $M(20) = 62.1$ and $F(20) = 337.1$ (0.0014, 42). The space group reported for the single-crystal structure, $P2_12_12_1$, gave an excellent Pawley fit.

Simulated annealing was used to solve the crystal structure of compound (I) from the powder pattern in direct space. The starting molecular geometry was taken from the anhydrate (Boles & Girven, 1976), entry AMCILL in the Cambridge Structural Database (Allen, 2002). The molecule is a zwitterion, in agreement with the single-crystal study. Because H atoms do not contribute significantly to the powder diffraction pattern, due to their low X-ray scattering power, they were ignored during the structure solution process. Hence, the water molecule can be reduced to an O atom, which reduces its number of degrees of freedom from six to three. The molecule has five flexible torsion angles, which, when combined with the three water molecules, give a total of 20 degrees of freedom. In ten simulated annealing runs, the correct crystal structure was found twice, with a profile $\chi^2 = 81.7$, 11 times the Pawley χ^2 . The next-best crystal structure had a profile $\chi^2 = 240$. The low success rate and high profile χ^2 are caused by the high R factor of 10.6% of the crystal structure of AMCILL from which the starting model was taken; when the structure solution was repeated with a better starting model (obtained from Rietveld refinement against the powder data), the correct

structure was found four times in ten runs, with a profile $\chi^2 = 20$, less than three times the Pawley χ^2 .

The background subtraction, peak fitting, indexing, Pawley refinement and simulated-annealing algorithms used are as implemented in the program *DASH*.

For the Rietveld refinement (Fig. 2), H atoms were included in the initial model in calculated positions. Bond lengths, bond angles and planar groupings were subjected to suitable constraints, including bonds to H atoms. Data were included to 41.42° in 2θ , corresponding to a real-space resolution of 0.99 \AA . The refinement was not particularly sensitive to the position of the water H atoms and these were included in calculated positions, with the water molecules being fixed in position for the final refinement cycles. The refinement proceeded smoothly to reach a minimum characterized by an excellent fit to the diffraction profile ($\chi^2 = 5.637$, $R_p = 0.0296$, $R_{wp} = 0.0296$ and $R_{Bragg} = 0.0295$).

Data collection: local software; cell refinement: *GSAS* (Larson & Von Dreele, 2000); data reduction: local software; program(s) used to solve structure: *DASH* (David *et al.*, 2004); program(s) used to refine structure: *GSAS*; molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
 Boles, M. O. & Girven, R. J. (1976). *Acta Cryst.* **B32**, 2279–2284.
 Boles, M. O., Girven, R. J. & Gane, P. A. C. (1978). *Acta Cryst.* **B34**, 461–466.
 Boultif, A. & Louer, D. (1991). *J. Appl. Cryst.* **24**, 987–993.
 David, W. I. F., Shankland, K., Van de Streek, J., Pidcock, E. & Motherwell, S. (2004). *DASH*. Version 3.0. Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, England.
 David, W. I. F. & Sivia, D. S. (2001). *J. Appl. Cryst.* **34**, 318–324.
 Finger, L. W., Cox, D. E. & Jephcoat, A. P. (1994). *J. Appl. Cryst.* **27**, 892–900.
 James, M. N. G., Hall, D. & Hodgkin, D. C. (1968). *Nature*, **220**, 168–170.
 Larson, A. C. & Von Dreele, R. B. (2000). *GSAS*. Report LAUR 86-748. Los Alamos National Laboratory, New Mexico, USA.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
 Thompson, P., Cox, D. E. & Hastings, J. B. (1987). *J. Appl. Cryst.* **20**, 79–83.

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6-[D(-)- α -Aminophenylacetamido]penicillanic acid trihydrate*Crystal data*

$C_{16}H_{19}N_3O_4S \cdot 3H_2O$

$M_r = 403.06$

Orthorhombic, $P2_12_12_1$

$a = 15.52275$ (16) Å

$b = 18.9256$ (3) Å

$c = 6.67375$ (8) Å

$V = 1960.60$ (3) Å³

$Z = 4$

$D_x = 1.367$ Mg m⁻³

Synchrotron radiation, $\lambda = 0.7003$ Å

$T = 290$ K

Particle morphology: powder

white

cylinder, 0.7×20 mm

Data collection

Diffractometer x3b1

Radiation source: Brookhaven NSLS

Specimen mounting: Lindemann glass capillary

Data collection mode: transmission

Scan method: step

Refinement

Least-squares matrix: full

$R_p = 0.025$

$R_{wp} = 0.030$

$R_{exp} = 0.013$

$R(F^2) = 0.02950$

Excluded region(s): none

Profile function: CW Profile function number 3

with 19 terms Pseudovoigt profile coefficients as parameterized in Thompson et al. (1987).

Asymmetry correction of Finger et al. (1994).

#1(GU) = 0.000 #2(GV) = 0.000 #3(GW) =

0.000 #4(GP) = 0.000 #5(LX) = 2.213 #6(LY) =

33.715 #7(S/L) = 0.0130 #8(H/L) = 0.0130

#9(trns) = 0.00 #10(shift) = 0.0832 #11(stec) =

0.00 #12(pte) = 0.00 #13(sfec) = 0.00 #14(L11)

= 0.000 #15(L22) = 0.000 #16(L33) = 0.000

#17(L12) = 0.000 #18(L13) = 0.000 #19(L23) =

0.000 Peak tails are ignored where the intensity

is below 0.0010 times the peak Aniso.

broadening axis 0.0 0.0 1.0

143 parameters

136 restraints

H-atom parameters constrained

Weighting scheme based on measured s.u.'s

$(\Delta/\sigma)_{max} = 0.03$

Background function: GSAS Background

function number 1 with 9 terms. Shifted

Chebyshev function of 1st kind 1: 2261.50 2:

-239.085 3: 47.0827 4: -155.949 5: 139.395 6:

-237.334 7: 231.040 8: -74.9779 9: 19.4239 10:

-0.402815

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S10	0.33689 (14)	0.46707 (11)	0.2754 (3)	0.0350 (4)*
O17	0.5429 (2)	0.3613 (2)	0.5525 (7)	0.0350 (4)*
O21	0.3263 (3)	0.2430 (2)	0.6113 (6)	0.0350 (4)*
O22	0.2965 (3)	0.2156 (2)	0.2943 (6)	0.0350 (4)*
O25	0.5385 (3)	0.5837 (2)	0.7961 (5)	0.0350 (4)*
N13	0.3923 (2)	0.37247 (18)	0.5272 (5)	0.0350 (4)*
N23	0.4964 (3)	0.5225 (2)	0.5209 (5)	0.0350 (4)*
N27	0.6573 (3)	0.6581 (2)	0.5579 (6)	0.0350 (4)*
C11	0.2977 (2)	0.37521 (15)	0.2374 (5)	0.0350 (4)*
C12	0.3623 (2)	0.32784 (17)	0.3604 (6)	0.0350 (4)*
C14	0.3598 (2)	0.4474 (2)	0.5333 (4)	0.0350 (4)*
C15	0.4496 (2)	0.46826 (18)	0.6238 (5)	0.0350 (4)*
C16	0.4759 (2)	0.39222 (19)	0.5728 (8)	0.0350 (4)*
C18	0.2064 (3)	0.3683 (3)	0.3136 (8)	0.0350 (4)*
C19	0.3001 (3)	0.3604 (3)	0.0177 (6)	0.0350 (4)*
C20	0.3239 (4)	0.2563 (2)	0.4322 (6)	0.0350 (4)*
C24	0.5369 (4)	0.5759 (3)	0.6158 (5)	0.0350 (4)*
C26	0.5768 (2)	0.6298 (2)	0.4722 (5)	0.0350 (4)*
C28	0.5114 (3)	0.6874 (3)	0.4222 (6)	0.0350 (4)*
C29	0.4704 (4)	0.6867 (3)	0.2375 (7)	0.0350 (4)*
C30	0.4035 (3)	0.7333 (3)	0.2002 (7)	0.0350 (4)*
C31	0.3851 (4)	0.7862 (3)	0.3350 (8)	0.0350 (4)*
C32	0.4208 (4)	0.7839 (3)	0.5249 (7)	0.0350 (4)*
C33	0.4853 (4)	0.7355 (3)	0.5675 (6)	0.0350 (4)*
O1	0.55980	0.49350	0.11790	0.0350 (4)*
O4	0.74455	0.45750	0.19890	0.0350 (4)*
O7	0.71633	0.40465	0.58470	0.0350 (4)*
H34	0.4160 (4)	0.3174 (3)	0.2832 (9)	0.0350 (4)*
H35	0.3068 (3)	0.4564 (4)	0.6142 (8)	0.0350 (4)*
H36	0.4413 (5)	0.4787 (4)	0.7695 (7)	0.0350 (4)*
H37	0.5926 (5)	0.6037 (4)	0.3330 (8)	0.0350 (4)*
H38	0.6990 (11)	0.6186 (5)	0.581 (6)	0.0350 (4)*
H39	0.6446 (7)	0.682 (2)	0.688 (3)	0.0350 (4)*
H40	0.6835 (16)	0.6927 (18)	0.463 (3)	0.0350 (4)*
H41	0.4940 (15)	0.6530 (13)	0.1193 (16)	0.0350 (4)*
H42	0.3735 (13)	0.7338 (9)	0.0544 (16)	0.0350 (4)*
H43	0.3981 (15)	0.8194 (12)	0.6399 (17)	0.0350 (4)*
H44	0.5112 (14)	0.7321 (11)	0.7174 (14)	0.0350 (4)*
H45	0.1765 (10)	0.4198 (5)	0.317 (6)	0.0350 (4)*
H46	0.1705 (9)	0.3342 (18)	0.214 (4)	0.0350 (4)*

H47	0.2068 (5)	0.346 (2)	0.463 (3)	0.0350 (4)*
H48	0.3658 (5)	0.364 (2)	-0.0353 (16)	0.0350 (4)*
H49	0.275 (2)	0.3082 (8)	-0.0102 (13)	0.0350 (4)*
H50	0.261 (2)	0.3986 (13)	-0.0597 (12)	0.0350 (4)*
H51	0.503 (3)	0.5188 (16)	0.3719 (11)	0.0350 (4)*
H52	0.3394 (16)	0.8272 (10)	0.297 (2)	0.0350 (4)*
H2	0.55279	0.52556	1.00373	0.0350 (4)*
H3	0.62022	0.48173	0.14439	0.0350 (4)*
H5	0.75832	0.50601	0.15871	0.0350 (4)*
H6	0.73466	0.43898	0.33409	0.0350 (4)*
H8	0.65604	0.38978	0.57372	0.0350 (4)*
H9	0.74556	0.36129	0.62735	0.0350 (4)*

Geometric parameters (Å, °)

S10—C11	1.859 (4)	C14—C15	1.570 (4)
S10—C14	1.797 (3)	C15—C16	1.534 (5)
O17—C16	1.201 (5)	C24—C26	1.531 (6)
O21—C20	1.222 (6)	C26—C28	1.527 (6)
O22—C20	1.273 (6)	C28—C29	1.387 (7)
O25—C24	1.213 (5)	C28—C33	1.390 (7)
O1—H3	0.9800	C29—C30	1.385 (8)
O1—H2 ⁱ	0.9800	C30—C31	1.376 (8)
O4—H5	0.9800	C31—C32	1.384 (7)
O4—H6	0.9800	C32—C33	1.387 (8)
O7—H9	0.9800	C12—H34	1.000 (7)
O7—H8	0.9800	C14—H35	0.999 (6)
N13—C12	1.473 (5)	C15—H36	1.001 (6)
N13—C14	1.506 (5)	C18—H47	1.08 (2)
N13—C16	1.384 (5)	C18—H45	1.080 (12)
N23—C15	1.433 (5)	C18—H46	1.08 (3)
N23—C24	1.348 (7)	C19—H49	1.078 (18)
N27—C26	1.475 (6)	C19—H48	1.082 (10)
N23—H51	1.002 (10)	C19—H50	1.08 (2)
N27—H39	1.00 (3)	C26—H37	1.080 (7)
N27—H38	1.001 (15)	C29—H41	1.079 (19)
N27—H40	1.00 (3)	C30—H42	1.079 (14)
C11—C18	1.511 (6)	C31—H52	1.08 (2)
C11—C12	1.576 (5)	C32—H43	1.079 (19)
C11—C19	1.493 (5)	C33—H44	1.080 (13)
C12—C20	1.555 (5)		
S10...N13	2.602 (4)	H2...H36	2.5000
S10...N23	3.149 (5)	H2...H51 ^x	2.5800
S10...H51	2.83 (4)	H2...H3 ^x	1.6300
S10...H35 ⁱⁱ	2.869 (6)	H2...O25	1.7800
S10...H50 ⁱⁱⁱ	3.16 (2)	H2...C24	2.7700
O1...O4	2.9968	H3...O4	2.0200

O1...O25 ⁱ	2.7633	H3...H51	2.4700
O1...N23	2.9161	H3...H5	2.1900
O4...N27 ^{iv}	2.8272	H3...H6	2.3300
O4...O7 ^{iv}	2.7850	H5...O7 ^{iv}	1.8000
O4...O1	2.9968	H5...H6 ^{iv}	2.4100
O4...O7	2.7967	H5...H9 ^{iv}	2.5200
O7...O17	2.8225	H5...H38 ^{iv}	2.5000
O7...O4	2.7967	H5...H3	2.1900
O7...O4 ^v	2.7850	H5...H8 ^{iv}	2.4500
O7...O22 ^{vi}	2.7166	H6...O7	1.8200
O17...N23	3.142 (5)	H6...H3	2.3300
O17...C31 ^{vii}	3.156 (7)	H6...H9	2.4500
O17...C30 ^{vii}	3.067 (7)	H6...H5 ^v	2.4100
O17...O7	2.8225	H6...H8	2.2200
O17...C32 ^{viii}	3.228 (7)	H6...H38 ^{iv}	2.2600
O21...N13	2.715 (5)	H6...H52 ^{vii}	2.5600
O21...N27 ^{viii}	2.742 (6)	H8...O17	1.8400
O22...C26 ^{vii}	3.109 (5)	H8...H43 ^{viii}	2.4800
O22...O7 ^{ix}	2.7166	H8...C16	2.8000
O22...C19	3.305 (6)	H8...H6	2.2200
O22...C28 ^{vii}	3.356 (6)	H8...H5 ^v	2.4500
O22...C18	3.213 (7)	H9...H6	2.4500
O22...N27 ^{vii}	2.688 (6)	H9...H5 ^v	2.5200
O25...C33	3.356 (7)	H9...O21 ^{vi}	2.8300
O25...N27	2.813 (6)	H9...C20 ^{vi}	2.5700
O25...O1 ^x	2.7633	H9...O22 ^{vi}	1.7400
O1...H51	1.9700	H34...H48	2.430 (18)
O1...H37	2.5800	H34...O17	2.793 (7)
O4...H52 ^{vii}	2.7900	H34...C28 ^{vii}	3.034 (8)
O4...H38 ^{iv}	1.8600	H34...C29 ^{vii}	3.041 (8)
O4...H3	2.0200	H35...C18	3.039 (8)
O7...H6	1.8200	H35...S10 ⁱⁱⁱ	2.869 (6)
O7...H5 ^v	1.8000	H35...H50 ^x	2.537 (16)
O17...H8	1.8400	H36...H2	2.5000
O17...H42 ^{vii}	2.831 (18)	H36...O25	2.501 (9)
O17...H34	2.793 (7)	H37...H51	2.14 (4)
O17...H43 ^{viii}	2.384 (16)	H37...O1	2.5800
O21...H44 ^{viii}	2.78 (2)	H37...H41	2.29 (2)
O21...H39 ^{viii}	1.83 (3)	H37...O22 ^{xi}	2.858 (9)
O21...H9 ^{ix}	2.8300	H38...H52 ^{xiii}	2.54 (3)
O21...H47	2.87 (3)	H38...H5 ^v	2.5000
O22...H37 ^{vii}	2.858 (9)	H38...H6 ^v	2.2600
O22...H40 ^{vii}	1.80 (2)	H38...O4 ^v	1.8600
O22...H9 ^{ix}	1.7400	H39...O25	2.59 (3)
O22...H49	2.704 (13)	H39...C33	2.790 (19)
O25...H36	2.501 (9)	H39...H44	2.29 (3)
O25...H39	2.59 (3)	H39...O21 ^{xii}	1.83 (3)
O25...H41 ^x	2.617 (17)	H39...C20 ^{xii}	2.94 (3)

O25...H44	2.89 (2)	H40...H49 ^{xi}	2.30 (4)
O25...H2	1.7800	H40...O22 ^{xi}	1.80 (2)
N13...S10	2.602 (4)	H40...C20 ^{xi}	2.90 (2)
N13...O21	2.715 (5)	H41...O25 ⁱ	2.617 (17)
N13...N23	3.267 (5)	H41...H37	2.29 (2)
N23...O17	3.142 (5)	H42...C18 ⁱⁱ	2.803 (17)
N23...O1	2.9161	H42...O17 ^{xi}	2.831 (18)
N23...S10	3.149 (5)	H42...H47 ⁱⁱ	2.05 (3)
N23...N13	3.267 (5)	H43...C16 ^{xii}	3.07 (2)
N27...O4 ^v	2.8272	H43...H8 ^{xii}	2.4800
N27...O22 ^{xi}	2.688 (6)	H43...O17 ^{xii}	2.384 (16)
N27...O25	2.813 (6)	H44...C24	3.06 (2)
N27...O21 ^{xii}	2.742 (6)	H44...O25	2.89 (2)
N27...H44	2.87 (2)	H44...N27	2.87 (2)
C18...O22	3.213 (7)	H44...O21 ^{xiii}	2.78 (2)
C19...O22	3.305 (6)	H44...H39	2.29 (3)
C26...O22 ^{xi}	3.109 (5)	H46...C32 ⁱⁱ	2.93 (3)
C28...O22 ^{xi}	3.356 (6)	H46...H49	2.26 (3)
C30...O17 ^{xi}	3.067 (7)	H46...C33 ⁱⁱ	2.92 (2)
C31...O17 ^{xi}	3.156 (7)	H47...C14	3.09 (2)
C32...O17 ^{xii}	3.228 (7)	H47...O21	2.87 (3)
C33...O25	3.356 (7)	H47...H42 ⁱⁱⁱ	2.05 (3)
C14...H47	3.09 (2)	H47...C30 ⁱⁱⁱ	2.77 (2)
C16...H43 ^{viii}	3.07 (2)	H47...C20	2.50 (3)
C16...H8	2.8000	H48...H34	2.430 (18)
C18...H42 ⁱⁱⁱ	2.803 (17)	H49...H40 ^{vii}	2.30 (4)
C18...H35	3.039 (8)	H49...H46	2.26 (3)
C20...H47	2.50 (3)	H49...O22	2.704 (13)
C20...H9 ^{ix}	2.5700	H50...S10 ⁱⁱ	3.16 (3)
C20...H39 ^{viii}	2.94 (3)	H50...H35 ⁱ	2.537 (16)
C20...H40 ^{vii}	2.90 (2)	H51...S10	2.83 (4)
C24...H2	2.7700	H51...O1	1.9700
C24...H44	3.06 (2)	H51...H2 ⁱ	2.5800
C28...H34 ^{xi}	3.034 (8)	H51...H3	2.4700
C29...H34 ^{xi}	3.041 (8)	H51...H37	2.14 (4)
C30...H47 ⁱⁱ	2.77 (2)	H52...O4 ^{xi}	2.7900
C32...H46 ⁱⁱⁱ	2.93 (3)	H52...H6 ^{xi}	2.5600
C33...H39	2.790 (19)	H52...H38 ^{xiv}	2.54 (3)
C33...H46 ⁱⁱⁱ	2.92 (2)		
C11—S10—C14	90.10 (18)	C26—C28—C33	120.6 (4)
H2 ⁱ —O1—H3	113.00	C29—C28—C33	119.5 (5)
H5—O4—H6	128.00	C26—C28—C29	119.5 (4)
H8—O7—H9	103.00	C28—C29—C30	119.8 (5)
C12—N13—C14	117.0 (3)	C29—C30—C31	120.1 (5)
C12—N13—C16	128.1 (3)	C30—C31—C32	119.5 (5)
C14—N13—C16	93.1 (3)	C31—C32—C33	119.9 (5)
C15—N23—C24	123.3 (3)	C28—C33—C32	120.0 (4)

C24—N23—H51	118 (2)	N13—C12—H34	103.8 (4)
C15—N23—H51	119 (2)	C11—C12—H34	112.0 (5)
H38—N27—H40	109 (2)	C20—C12—H34	107.8 (5)
H38—N27—H39	109 (3)	C15—C14—H35	118.7 (4)
C26—N27—H38	109.7 (11)	N13—C14—H35	116.8 (5)
C26—N27—H40	109.7 (14)	S10—C14—H35	108.6 (4)
H39—N27—H40	110 (2)	C16—C15—H36	115.8 (5)
C26—N27—H39	109.5 (9)	C14—C15—H36	108.0 (5)
S10—C11—C19	107.5 (3)	N23—C15—H36	112.9 (5)
C12—C11—C18	111.9 (3)	C11—C18—H45	109.4 (11)
S10—C11—C12	104.6 (2)	C11—C18—H46	109.2 (11)
C12—C11—C19	112.9 (3)	C11—C18—H47	109.8 (6)
C18—C11—C19	109.7 (3)	H46—C18—H47	110 (2)
S10—C11—C18	110.0 (3)	H45—C18—H46	109 (2)
C11—C12—C20	114.3 (3)	H45—C18—H47	110 (3)
N13—C12—C11	105.6 (3)	C11—C19—H48	109.4 (8)
N13—C12—C20	112.8 (3)	C11—C19—H49	109.4 (6)
N13—C14—C15	87.1 (2)	C11—C19—H50	109.3 (9)
S10—C14—N13	103.6 (2)	H48—C19—H49	110 (2)
S10—C14—C15	119.5 (2)	H49—C19—H50	109.2 (19)
N23—C15—C16	115.5 (3)	H48—C19—H50	109 (2)
C14—C15—C16	85.1 (3)	C28—C26—H37	106.8 (5)
N23—C15—C14	116.5 (3)	C24—C26—H37	109.0 (5)
O17—C16—C15	135.4 (3)	N27—C26—H37	107.9 (5)
O17—C16—N13	131.0 (4)	C28—C29—H41	120.0 (12)
N13—C16—C15	93.0 (3)	C30—C29—H41	120.0 (11)
O21—C20—C12	118.0 (4)	C29—C30—H42	119.4 (11)
O21—C20—O22	126.3 (4)	C31—C30—H42	119.6 (11)
O22—C20—C12	115.6 (4)	C32—C31—H52	120.0 (9)
O25—C24—N23	124.5 (5)	C30—C31—H52	120.3 (9)
O25—C24—C26	122.1 (5)	C31—C32—H43	120.1 (11)
N23—C24—C26	113.2 (3)	C33—C32—H43	120.1 (11)
C24—C26—C28	110.1 (3)	C28—C33—H44	119.9 (12)
N27—C26—C24	110.0 (3)	C32—C33—H44	119.9 (12)
N27—C26—C28	112.9 (3)		
C14—S10—C11—C12	-39.8 (2)	C11—C12—C20—O21	-124.3 (5)
C14—S10—C11—C18	80.5 (3)	C11—C12—C20—O22	59.0 (6)
C14—S10—C11—C19	-160.1 (3)	N13—C12—C20—O21	-3.6 (6)
C11—S10—C14—N13	37.1 (2)	S10—C14—C15—C16	-95.1 (3)
C11—S10—C14—C15	131.6 (3)	S10—C14—C15—N23	21.1 (4)
C14—N13—C16—C15	10.2 (3)	N13—C14—C15—C16	9.0 (3)
C16—N13—C14—C15	-10.0 (3)	N13—C14—C15—N23	125.1 (3)
C14—N13—C12—C11	-2.6 (4)	N23—C15—C16—O17	44.6 (8)
C12—N13—C14—C15	-146.1 (3)	C14—C15—C16—O17	161.7 (7)
C14—N13—C12—C20	-128.1 (4)	C14—C15—C16—N13	-9.8 (3)
C16—N13—C12—C20	113.4 (5)	N23—C15—C16—N13	-126.9 (3)
C16—N13—C12—C11	-121.1 (4)	N23—C24—C26—C28	89.1 (5)

C12—N13—C14—S10	-26.5 (3)	O25—C24—C26—N27	37.9 (7)
C16—N13—C14—S10	109.7 (3)	O25—C24—C26—C28	-87.2 (6)
C12—N13—C16—O17	-33.5 (8)	N23—C24—C26—N27	-145.8 (4)
C12—N13—C16—C15	138.6 (4)	N27—C26—C28—C29	132.8 (5)
C14—N13—C16—O17	-161.9 (6)	N27—C26—C28—C33	-54.6 (6)
C15—N23—C24—O25	0.3 (9)	C24—C26—C28—C33	68.7 (6)
C15—N23—C24—C26	-175.9 (4)	C24—C26—C28—C29	-103.8 (5)
C24—N23—C15—C14	136.1 (5)	C26—C28—C33—C32	-174.8 (5)
C24—N23—C15—C16	-126.3 (5)	C26—C28—C29—C30	172.0 (5)
S10—C11—C12—C20	154.2 (3)	C33—C28—C29—C30	-0.6 (8)
S10—C11—C12—N13	29.6 (3)	C29—C28—C33—C32	-2.2 (8)
C19—C11—C12—N13	146.2 (3)	C28—C29—C30—C31	8.5 (8)
C19—C11—C12—C20	-89.2 (4)	C29—C30—C31—C32	-13.5 (8)
C18—C11—C12—C20	35.1 (4)	C30—C31—C32—C33	10.6 (9)
C18—C11—C12—N13	-89.5 (4)	C31—C32—C33—C28	-2.8 (9)
N13—C12—C20—O22	179.7 (4)		

Symmetry codes: (i) $x, y, z-1$; (ii) $-x+1/2, -y+1, z-1/2$; (iii) $-x+1/2, -y+1, z+1/2$; (iv) $-x+3/2, -y+1, z-1/2$; (v) $-x+3/2, -y+1, z+1/2$; (vi) $x+1/2, -y+1/2, -z+1$; (vii) $-x+1, y-1/2, -z+1/2$; (viii) $-x+1, y-1/2, -z+3/2$; (ix) $x-1/2, -y+1/2, -z+1$; (x) $x, y, z+1$; (xi) $-x+1, y+1/2, -z+1/2$; (xii) $-x+1, y+1/2, -z+3/2$; (xiii) $x+1/2, -y+3/2, -z+1$; (xiv) $x-1/2, -y+3/2, -z+1$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H3 \cdots O4	0.98	2.02	2.9968	180
O4—H5 \cdots O7 ^{iv}	0.98	1.80	2.7850	180
O4—H6 \cdots O7	0.98	1.82	2.7967	180
O7—H8 \cdots O17	0.98	1.84	2.8225	180
O7—H9 \cdots O22 ^{vi}	0.98	1.74	2.7166	179
N27—H38 \cdots O4 ^v	1.001 (15)	1.86	2.8272	161
N27—H39 \cdots O21 ^{xii}	1.00 (3)	1.83 (3)	2.742 (6)	151.1 (15)
N27—H40 \cdots O22 ^{xi}	1.00 (3)	1.80 (2)	2.688 (6)	147 (2)
N23—H51 \cdots O1	1.002 (10)	1.97	2.9161	156
C14—H35 \cdots S10 ⁱⁱⁱ	0.999 (6)	2.869 (6)	3.815 (4)	158.3 (6)
C26—H37 \cdots O1	1.080 (7)	2.58	3.5092	143
C32—H43 \cdots O17 ^{xii}	1.079 (19)	2.384 (16)	3.228 (7)	134.0 (16)

Symmetry codes: (iii) $-x+1/2, -y+1, z+1/2$; (iv) $-x+3/2, -y+1, z-1/2$; (v) $-x+3/2, -y+1, z+1/2$; (vi) $x+1/2, -y+1/2, -z+1$; (xi) $-x+1, y+1/2, -z+1/2$; (xii) $-x+1, y+1/2, -z+3/2$.