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Key indicators

Single-crystal X-ray study
 $T = 120$ K
 Mean $\sigma(C-C) = 0.003$ Å
 R factor = 0.043
 wR factor = 0.125
 Data-to-parameter ratio = 18.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

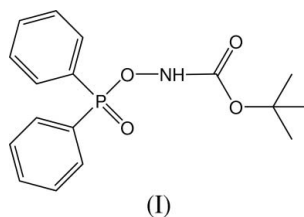
tert-Butyl *N*-(phosphinoyloxy)carbamate

The title compound, $C_{17}H_{20}NO_4P$, contains pyramidal N atoms and adopts similar conformations in its three independent molecules (*A*, *B* and *C*). Molecules *A* and *B* form a dimer in the crystal structure by way of a pair of $N-H \cdots O$ hydrogen bonds, as does *C* with its inversion-generated partner.

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Comment

N-Alkyl-*O*-(diphenylphosphoryl)hydroxylamines have been reported to be potentially useful electrophilic aminating agents, for example in Schmidt reactions (Harger, 1981; Boche & Schrott, 1982*a,b*; Boche *et al.*, 1988). It was not unreasonable to infer that such systems would act as nitrogen peracid equivalents and could be widely applicable reagents in organic synthesis (Masse & Sturtz, 1988*a,b*). In the course of exploring this possibility, we prepared the title compound, (I).



The asymmetric unit of (I) comprises three independent molecules (Fig. 1, Table 1). Molecules *A* and *B* are related by an approximate local inversion centre and are linked into a dimer by a pair of $N-H \cdots O$ hydrogen bonds (Table 2). The third independent molecule (*C*) forms a similar hydrogen-bonded dimer with its symmetry-equivalent partner, generated by a crystallographic inversion centre. The geometries and conformations of all three molecules are essentially the same. The P atoms have tetrahedral geometry. The P–O single bond in (I) is substantially longer [mean 1.616 (2) Å] than in diphenylphosphinic acid $Ph_2P(=O)OH$ [1.550 (1) Å; Lyssenko *et al.*, 2002] or in $Ph_2P(=O)OBu-t$ [1.569 (3) Å; Grice *et al.*, 2004], but comparable to the values in $Ph_2P(=O)ONEt_2$ [1.599 (1) Å; Spek & Veldman, 1999] or $Ph_2P(=O)ON=C(Cl)Pr-i$ [1.613 (3) Å; Martynov *et al.*, 1988]. It is noteworthy that the difference between the $P=O$ and the P–OR bond lengths is increased from 0.048 (1) Å in $Ph_2P(=O)OH$ and 0.093 (4) Å in $Ph_2P(=O)OBu-t$ to 0.136 (2) Å in (I), 0.128 (1) Å in $Ph_2P(=O)ONEt_2$ and 0.151 (4) Å in $Ph_2P(=O)ON=C(Cl)Pr-i$. Thus, we may conclude that bonding to an N atom reduces the additional π -component of the (formally) single P–O bond. The P–C(Ph) distances in (I) [mean 1.786 (2) Å] are as expected.

The N atoms are substantially pyramidal (sp^3) in all three independent molecules in (I): atoms H1N, H2N and H3N

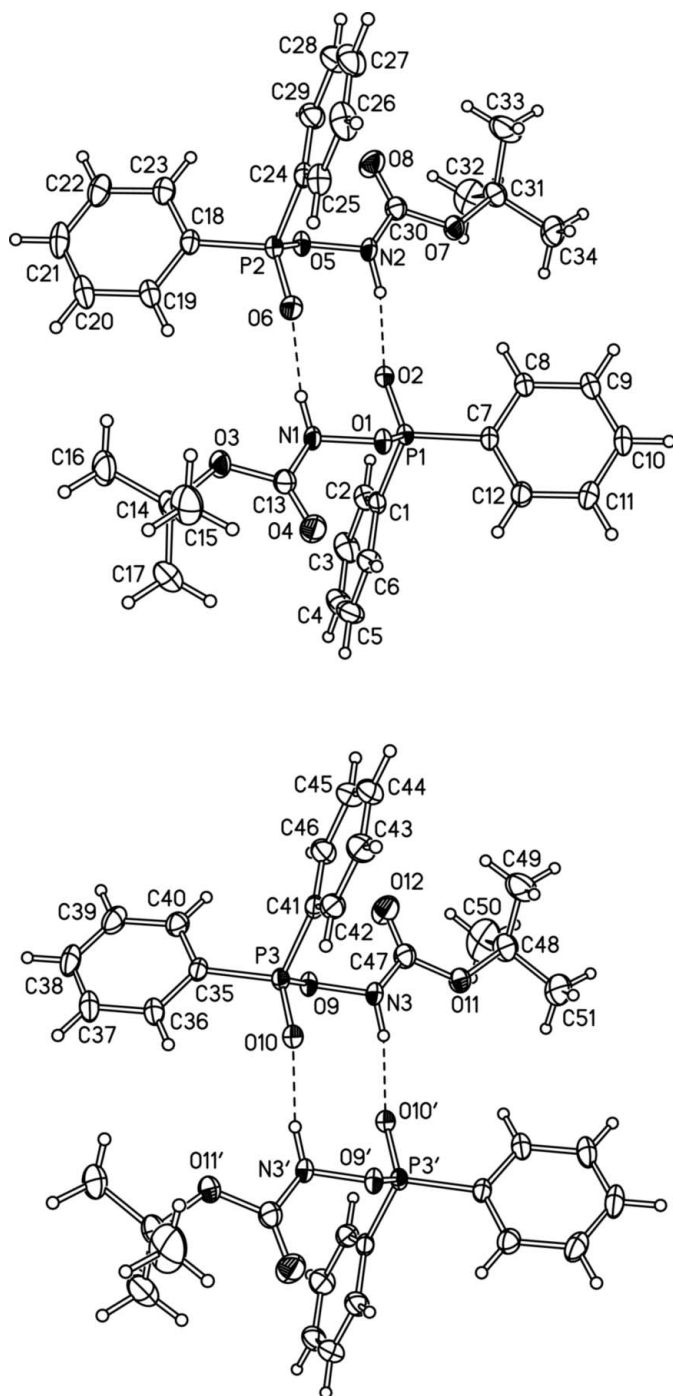


Figure 1

The structures of the constituent molecules of (I) showing 50% displacement ellipsoids (arbitrary spheres for the H atoms) and hydrogen bonding (dashed lines) interactions. Top: molecules A and B. Bottom: molecule C and its symmetry equivalent by inversion (symmetry code for primed atom labels: $1 - x, 2 - y, 1 - z$).

deviate from the corresponding O—N—C planes by 0.68 (2), 0.70 (2) and 0.63 (2) Å, respectively (see also Table 1). Similar geometry has been observed in TosOONHC(=O)OBU-*t* (Tos = *p*-toluenesulfonyl; Boche *et al.*, 1994; Blatch *et al.*, 2006) and in (O₂N)Cl₂C₆H₂—ONHC(=O)OBU-*t* (Boyles *et al.*, 2002), whereas in Me₃ONHC(=O)OBU-*t* the N atom has a planar configuration (Mortl *et al.*, 1995).

Experimental

A solution of *t*-BuOC(O)NHOH (6.39 g, 0.048 mol) in an anhydrous mixture of CH₂Cl₂ (160 ml) and toluene (320 ml) was cooled to 273 K under argon, treated with 1.2 equivalents of triethylamine (6.021 g, 0.0595 mol) and stirred for 30 min. Diphenylphosphinyl chloride (9.16 ml, 0.048 mol) was added; the solution was warmed to room temperature, stirred for another 72 h and evaporated. The solid residue was recrystallized from CH₂Cl₂ to give colourless crystals of (I) (12.14 g, 76%). Mass-spectrum, *m/z* ES (+): 233.3 (Ph₂PO₂NH₂, 82%) and 356.33 (45%). The melting point was not measured because of the thermal instability and explosive nature of (I).

Crystal data

C ₁₇ H ₂₀ NO ₄ P	$V = 2687.3 (3) \text{ \AA}^3$
$M_r = 333.31$	$Z = 6$
Triclinic, $P\bar{1}$	$D_x = 1.236 \text{ Mg m}^{-3}$
$a = 12.181 (1) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 13.734 (1) \text{ \AA}$	$\mu = 0.17 \text{ mm}^{-1}$
$c = 16.532 (1) \text{ \AA}$	$T = 120 (2) \text{ K}$
$\alpha = 101.10 (1)^\circ$	Block, colourless
$\beta = 96.77 (1)^\circ$	$0.23 \times 0.21 \times 0.19 \text{ mm}$
$\gamma = 92.93 (1)^\circ$	

Data collection

Bruker SMART 6K	11796 independent reflections
CCD diffractometer	9155 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.040$
Absorption correction: none	$\theta_{\text{max}} = 30.0^\circ$
11843 measured reflections	

Refinement

Refinement on F^2	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.043$	$w = 1/[\sigma^2(F_o^2) + (0.0824P)^2]$
$wR(F^2) = 0.125$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.00$	$(\Delta/\sigma)_{\text{max}} = 0.001$
11796 reflections	$\Delta\rho_{\text{max}} = 0.55 \text{ e \AA}^{-3}$
652 parameters	$\Delta\rho_{\text{min}} = -0.53 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

P1—O2	1.4808 (9)	P2—O5	1.6162 (11)
P1—O1	1.6144 (11)	P3—O10	1.4813 (10)
P2—O6	1.4783 (10)	P3—O9	1.6187 (11)
C13—N1—O1	111.09 (10)	O5—N2—H2N	109.6 (11)
C13—N1—H1N	114.4 (11)	C47—N3—O9	111.97 (11)
O1—N1—H1N	107.8 (11)	C47—N3—H3N	117.6 (11)
C30—N2—O5	111.24 (10)	O9—N3—H3N	110.6 (12)
C30—N2—H2N	115.3 (12)		

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1N \cdots O6	0.886 (17)	1.836 (17)	2.7138 (16)	170.0 (15)
N2—H2N \cdots O2	0.95 (2)	1.83 (2)	2.7688 (16)	168.4 (17)
N3—H3N \cdots O10 ⁱ	0.923 (18)	1.855 (19)	2.7666 (16)	169.0 (16)

Symmetry code: (i) $-x + 1, -y + 2, -z + 1$.

The amino H atoms were located in difference maps and their positions and U_{iso} values were freely refined. The phenyl H atoms were treated as riding on their C atoms with C–H = 0.95 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The methyl groups were refined as rigid bodies (C–H = 0.98 Å) allowed to rotate around their linking C–C bonds, with a common refined U_{iso} for the three H atoms.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Bruker, 2001); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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tert*-Butyl *N*-(phosphinoyloxy)carbamateCrystal data*

$C_{17}H_{20}NO_4P$

$M_r = 333.31$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 12.181\ (1)\ \text{\AA}$

$b = 13.734\ (1)\ \text{\AA}$

$c = 16.532\ (1)\ \text{\AA}$

$\alpha = 101.10\ (1)^\circ$

$\beta = 96.77\ (1)^\circ$

$\gamma = 92.93\ (1)^\circ$

$V = 2687.3\ (3)\ \text{\AA}^3$

$Z = 6$

$F(000) = 1056$

$D_x = 1.236\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 5811 reflections

$\theta = 2.2\text{--}34.4^\circ$

$\mu = 0.17\ \text{mm}^{-1}$

$T = 120\ \text{K}$

Block, colourless

$0.23 \times 0.21 \times 0.19\ \text{mm}$

Data collection

Bruker SMART CCD 6K

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 8 pixels mm^{-1}

ω scans

11843 measured reflections

11796 independent reflections

9155 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.040$

$\theta_{\text{max}} = 30.0^\circ$, $\theta_{\text{min}} = 1.5^\circ$

$h = -16 \rightarrow 13$

$k = -16 \rightarrow 19$

$l = -23 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.125$

$S = 1.00$

11796 reflections

652 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0824P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.55\ \text{e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.53\ \text{e \AA}^{-3}$

Special details

Experimental. The data collection nominally covered a hemisphere of reciprocal Space, by 1 run of ω scans each set at different φ and/or 2θ angles and each scan (10 s exposure) covering 0.3° in ω . Crystal to detector distance 4.85 cm. ^1H NMR (200 MHz, CDCl_3): δ 1.39 (s, 9 H, CMe_3), 7.44–7.48 (m, 4 H, ArH) 7.53–7.57 (m, 2 H, ArH), 7.94–7.99 (m, 4 H, ArH) and 8.63 (broad s, 1 H, NH). ^{13}C NMR (100 MHz, CDCl_3): δ 27.9 (CH_3), 83.1 (CMe_3), 128.4 (ArH), 128.5 (ArH), 129.5 [$\text{ARP}(\text{O})$], 132.4 (ArH), 132.8 (ArH) and 156.2 (ArH). ^{31}P NMR (81 MHz, CDCl_3): δ 40.96. Mass-spectrum, m/z ES (+): 233.3 ($\text{Ph}_2\text{PO}_2\text{NH}_2$, 82%) and 356.33 (45%).

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
P1	0.10055 (3)	0.26988 (3)	0.27389 (2)	0.01775 (10)
O1	0.18559 (8)	0.36743 (7)	0.30630 (6)	0.0210 (2)
O2	0.08405 (8)	0.23290 (7)	0.18273 (6)	0.0207 (2)
O4	0.18353 (11)	0.55106 (8)	0.39437 (6)	0.0342 (3)
O3	0.14394 (10)	0.60494 (7)	0.27240 (6)	0.0274 (3)
N1	0.14437 (11)	0.44524 (9)	0.26621 (7)	0.0209 (3)
H1N	0.1747 (14)	0.4421 (12)	0.2197 (10)	0.022 (4)*
C1	−0.02620 (12)	0.30120 (11)	0.31342 (8)	0.0210 (3)
C2	−0.12504 (14)	0.25301 (12)	0.26970 (10)	0.0290 (4)
H2	−0.1243	0.2066	0.2191	0.035*
C3	−0.22496 (15)	0.27315 (14)	0.30047 (12)	0.0372 (4)
H3	−0.2926	0.2408	0.2706	0.045*
C4	−0.22558 (16)	0.33991 (14)	0.37421 (13)	0.0411 (5)
H4	−0.2938	0.3523	0.3954	0.049*
C5	−0.12866 (16)	0.38862 (13)	0.41741 (11)	0.0376 (4)
H5	−0.1302	0.4349	0.4679	0.045*
C6	−0.02786 (14)	0.37012 (12)	0.38722 (9)	0.0289 (4)
H6	0.0392	0.4043	0.4167	0.035*
C7	0.16770 (12)	0.18519 (11)	0.32898 (8)	0.0196 (3)
C8	0.20979 (12)	0.10244 (11)	0.28261 (9)	0.0211 (3)
H8	0.2023	0.0926	0.2237	0.025*
C9	0.26281 (12)	0.03471 (11)	0.32361 (9)	0.0239 (3)
H9	0.2920	−0.0213	0.2925	0.029*
C10	0.27309 (13)	0.04875 (11)	0.40974 (9)	0.0261 (4)
H10	0.3091	0.0022	0.4373	0.031*
C11	0.23058 (13)	0.13071 (12)	0.45561 (9)	0.0270 (4)
H11	0.2374	0.1400	0.5145	0.032*
C12	0.17830 (13)	0.19906 (11)	0.41571 (9)	0.0248 (3)
H12	0.1497	0.2552	0.4472	0.030*
C13	0.16195 (13)	0.53720 (11)	0.32002 (9)	0.0230 (3)
C14	0.14467 (16)	0.71156 (11)	0.31001 (10)	0.0310 (4)

C15	0.25948 (18)	0.74960 (13)	0.35381 (12)	0.0452 (5)
H151	0.3140	0.7340	0.3150	0.055 (4)*
H152	0.2769	0.7176	0.4014	0.055 (4)*
H153	0.2619	0.8218	0.3734	0.055 (4)*
C16	0.11550 (18)	0.75684 (13)	0.23410 (11)	0.0417 (5)
H161	0.1721	0.7435	0.1967	0.052 (4)*
H162	0.1124	0.8289	0.2516	0.052 (4)*
H163	0.0432	0.7274	0.2050	0.052 (4)*
C17	0.05589 (19)	0.72724 (14)	0.36741 (12)	0.0479 (5)
H171	-0.0145	0.6928	0.3385	0.060 (4)*
H172	0.0467	0.7986	0.3836	0.060 (4)*
H173	0.0779	0.7004	0.4172	0.060 (4)*
P2	0.24725 (3)	0.38635 (3)	0.04178 (2)	0.01939 (10)
O5	0.16717 (9)	0.28576 (7)	0.00742 (6)	0.0223 (2)
O6	0.25685 (9)	0.42489 (8)	0.13240 (6)	0.0246 (2)
O7	0.21538 (10)	0.05117 (7)	0.04486 (6)	0.0263 (3)
O8	0.19732 (11)	0.10585 (9)	-0.07767 (7)	0.0375 (3)
N2	0.20689 (11)	0.21047 (9)	0.05057 (7)	0.0210 (3)
H2N	0.1694 (16)	0.2106 (14)	0.0977 (12)	0.039 (5)*
C18	0.17804 (12)	0.46636 (11)	-0.01717 (9)	0.0211 (3)
C19	0.14058 (13)	0.55285 (11)	0.02770 (10)	0.0250 (3)
H19	0.1523	0.5666	0.0868	0.030*
C20	0.08625 (14)	0.61845 (12)	-0.01490 (11)	0.0310 (4)
H20	0.0606	0.6771	0.0153	0.037*
C21	0.06930 (14)	0.59849 (13)	-0.10092 (11)	0.0346 (4)
H21	0.0326	0.6438	-0.1295	0.042*
C22	0.10585 (15)	0.51247 (14)	-0.14566 (10)	0.0344 (4)
H22	0.0937	0.4990	-0.2047	0.041*
C23	0.16013 (13)	0.44597 (12)	-0.10403 (9)	0.0281 (4)
H23	0.1849	0.3871	-0.1345	0.034*
C24	0.38035 (12)	0.36019 (11)	0.01099 (9)	0.0214 (3)
C25	0.47290 (14)	0.40633 (12)	0.06499 (10)	0.0288 (4)
H25	0.4639	0.4494	0.1157	0.035*
C26	0.57862 (15)	0.38912 (13)	0.04411 (12)	0.0378 (4)
H26	0.6421	0.4199	0.0809	0.045*
C27	0.59084 (15)	0.32707 (14)	-0.03009 (13)	0.0405 (5)
H27	0.6631	0.3162	-0.0444	0.049*
C28	0.49962 (15)	0.28076 (14)	-0.08368 (11)	0.0377 (4)
H28	0.5092	0.2380	-0.1344	0.045*
C29	0.39359 (14)	0.29679 (12)	-0.06343 (9)	0.0295 (4)
H29	0.3305	0.2647	-0.1000	0.035*
C30	0.20366 (13)	0.11913 (11)	-0.00321 (9)	0.0234 (3)
C31	0.21882 (16)	-0.05516 (12)	0.00805 (10)	0.0333 (4)
C32	0.1093 (2)	-0.09396 (15)	-0.04397 (13)	0.0540 (6)
H321	0.0489	-0.0800	-0.0101	0.079 (5)*
H322	0.1093	-0.1659	-0.0638	0.079 (5)*
H323	0.0986	-0.0611	-0.0917	0.079 (5)*
C33	0.3164 (2)	-0.07041 (16)	-0.04069 (15)	0.0645 (7)

H331	0.3839	-0.0374	-0.0060	0.064 (4)*
H332	0.3033	-0.0418	-0.0906	0.064 (4)*
H333	0.3253	-0.1418	-0.0570	0.064 (4)*
C34	0.2348 (2)	-0.10051 (13)	0.08505 (12)	0.0458 (5)
H341	0.1730	-0.0859	0.1173	0.060 (4)*
H342	0.3046	-0.0723	0.1192	0.060 (4)*
H343	0.2371	-0.1728	0.0684	0.060 (4)*
P3	0.58759 (3)	1.05685 (3)	0.38744 (2)	0.01867 (10)
O9	0.51367 (8)	0.95294 (7)	0.35097 (6)	0.0216 (2)
O10	0.58442 (8)	1.09867 (7)	0.47662 (6)	0.0214 (2)
O11	0.56502 (10)	0.72252 (8)	0.39681 (7)	0.0302 (3)
O12	0.57720 (12)	0.77787 (9)	0.27653 (7)	0.0420 (3)
N3	0.55055 (11)	0.88095 (9)	0.39871 (7)	0.0215 (3)
H3N	0.5092 (15)	0.8805 (14)	0.4420 (11)	0.034 (5)*
C35	0.52414 (12)	1.13203 (11)	0.32102 (9)	0.0206 (3)
C36	0.48845 (13)	1.22263 (11)	0.36013 (9)	0.0252 (3)
H36	0.4984	1.2414	0.4191	0.030*
C37	0.43820 (14)	1.28491 (12)	0.31187 (11)	0.0307 (4)
H37	0.4139	1.3464	0.3380	0.037*
C38	0.42358 (14)	1.25756 (14)	0.22619 (11)	0.0344 (4)
H38	0.3891	1.3003	0.1937	0.041*
C39	0.45897 (14)	1.16801 (14)	0.18714 (10)	0.0340 (4)
H39	0.4490	1.1499	0.1282	0.041*
C40	0.50918 (13)	1.10465 (12)	0.23440 (9)	0.0272 (4)
H40	0.5330	1.0432	0.2078	0.033*
C41	0.72615 (12)	1.03390 (11)	0.36752 (8)	0.0209 (3)
C42	0.81257 (13)	1.08015 (12)	0.42822 (9)	0.0260 (3)
H42	0.7964	1.1220	0.4775	0.031*
C43	0.92164 (13)	1.06508 (13)	0.41655 (10)	0.0322 (4)
H43	0.9802	1.0958	0.4582	0.039*
C44	0.94527 (14)	1.00512 (13)	0.34412 (10)	0.0328 (4)
H44	1.0202	0.9948	0.3364	0.039*
C45	0.86058 (14)	0.96007 (13)	0.28310 (10)	0.0328 (4)
H45	0.8774	0.9198	0.2332	0.039*
C46	0.75090 (13)	0.97382 (12)	0.29476 (9)	0.0282 (4)
H46	0.6926	0.9423	0.2532	0.034*
C47	0.56398 (14)	0.79048 (11)	0.34854 (9)	0.0265 (4)
C48	0.59500 (19)	0.62056 (12)	0.36590 (12)	0.0426 (5)
C49	0.7133 (2)	0.62491 (16)	0.34395 (15)	0.0647 (7)
H491	0.7159	0.6575	0.2963	0.088 (5)*
H492	0.7366	0.5572	0.3297	0.088 (5)*
H493	0.7633	0.6627	0.3917	0.088 (5)*
C50	0.5102 (3)	0.56827 (15)	0.29476 (15)	0.0728 (8)
H501	0.4358	0.5733	0.3115	0.079 (5)*
H502	0.5249	0.4980	0.2799	0.079 (5)*
H503	0.5148	0.5996	0.2467	0.079 (5)*
C51	0.5900 (2)	0.57414 (14)	0.44159 (14)	0.0577 (7)
H511	0.5159	0.5790	0.4590	0.068 (4)*

H512	0.6453	0.6095	0.4870	0.068 (4)*
H513	0.6056	0.5040	0.4279	0.068 (4)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
P1	0.0196 (2)	0.01637 (19)	0.01840 (16)	0.00518 (17)	0.00190 (13)	0.00567 (13)
O1	0.0230 (6)	0.0164 (5)	0.0252 (5)	0.0036 (5)	0.0011 (4)	0.0088 (4)
O2	0.0238 (6)	0.0196 (5)	0.0199 (4)	0.0056 (5)	0.0029 (4)	0.0059 (4)
O4	0.0536 (8)	0.0245 (6)	0.0235 (5)	0.0055 (6)	0.0024 (5)	0.0031 (4)
O3	0.0420 (7)	0.0143 (5)	0.0257 (5)	0.0028 (5)	0.0027 (5)	0.0045 (4)
N1	0.0293 (7)	0.0143 (6)	0.0206 (5)	0.0050 (6)	0.0024 (5)	0.0068 (4)
C1	0.0230 (8)	0.0187 (7)	0.0246 (6)	0.0066 (7)	0.0055 (5)	0.0093 (5)
C2	0.0270 (9)	0.0265 (8)	0.0344 (8)	0.0044 (8)	0.0038 (6)	0.0080 (6)
C3	0.0231 (9)	0.0338 (10)	0.0584 (11)	0.0032 (9)	0.0102 (8)	0.0151 (8)
C4	0.0325 (10)	0.0335 (10)	0.0659 (12)	0.0109 (9)	0.0270 (9)	0.0169 (9)
C5	0.0482 (12)	0.0278 (9)	0.0425 (9)	0.0108 (9)	0.0259 (8)	0.0073 (7)
C6	0.0338 (9)	0.0258 (8)	0.0286 (7)	0.0063 (8)	0.0086 (6)	0.0051 (6)
C7	0.0187 (7)	0.0186 (7)	0.0231 (6)	0.0032 (7)	0.0009 (5)	0.0085 (5)
C8	0.0201 (8)	0.0187 (7)	0.0254 (6)	0.0030 (7)	0.0021 (5)	0.0071 (5)
C9	0.0189 (8)	0.0192 (7)	0.0348 (8)	0.0037 (7)	0.0023 (6)	0.0088 (6)
C10	0.0201 (8)	0.0244 (8)	0.0364 (8)	0.0030 (7)	-0.0021 (6)	0.0157 (6)
C11	0.0264 (8)	0.0299 (8)	0.0265 (7)	0.0032 (7)	-0.0021 (6)	0.0130 (6)
C12	0.0256 (8)	0.0250 (8)	0.0244 (7)	0.0048 (7)	0.0002 (6)	0.0078 (6)
C13	0.0263 (8)	0.0182 (7)	0.0254 (7)	0.0014 (7)	0.0051 (6)	0.0054 (5)
C14	0.0455 (11)	0.0134 (7)	0.0332 (8)	0.0051 (8)	0.0038 (7)	0.0031 (6)
C15	0.0547 (13)	0.0258 (9)	0.0499 (11)	-0.0040 (10)	-0.0042 (9)	0.0032 (8)
C16	0.0629 (14)	0.0209 (8)	0.0424 (10)	0.0056 (9)	0.0029 (9)	0.0106 (7)
C17	0.0632 (14)	0.0338 (10)	0.0518 (11)	0.0211 (10)	0.0206 (10)	0.0083 (8)
P2	0.0207 (2)	0.0181 (2)	0.02074 (16)	0.00477 (18)	0.00170 (13)	0.00693 (13)
O5	0.0228 (6)	0.0195 (5)	0.0265 (5)	0.0044 (5)	0.0007 (4)	0.0101 (4)
O6	0.0302 (6)	0.0220 (5)	0.0228 (5)	0.0041 (5)	0.0044 (4)	0.0060 (4)
O7	0.0382 (7)	0.0150 (5)	0.0262 (5)	0.0041 (5)	0.0048 (4)	0.0045 (4)
O8	0.0616 (9)	0.0279 (6)	0.0230 (5)	0.0044 (6)	0.0055 (5)	0.0044 (4)
N2	0.0278 (7)	0.0155 (6)	0.0220 (5)	0.0047 (6)	0.0034 (5)	0.0088 (4)
C18	0.0173 (7)	0.0204 (7)	0.0276 (7)	0.0009 (7)	0.0017 (5)	0.0108 (6)
C19	0.0198 (8)	0.0199 (8)	0.0367 (8)	0.0026 (7)	0.0025 (6)	0.0098 (6)
C20	0.0224 (9)	0.0208 (8)	0.0515 (10)	0.0034 (8)	-0.0006 (7)	0.0139 (7)
C21	0.0245 (9)	0.0302 (9)	0.0526 (10)	0.0003 (8)	-0.0067 (7)	0.0245 (8)
C22	0.0311 (9)	0.0400 (10)	0.0340 (8)	0.0011 (9)	-0.0067 (7)	0.0188 (7)
C23	0.0270 (9)	0.0288 (8)	0.0297 (7)	0.0039 (8)	-0.0007 (6)	0.0108 (6)
C24	0.0219 (8)	0.0192 (7)	0.0252 (6)	0.0034 (7)	0.0016 (5)	0.0102 (5)
C25	0.0265 (9)	0.0251 (8)	0.0347 (8)	0.0009 (8)	-0.0022 (6)	0.0100 (6)
C26	0.0214 (9)	0.0312 (10)	0.0607 (12)	-0.0020 (8)	-0.0033 (8)	0.0152 (8)
C27	0.0235 (9)	0.0393 (11)	0.0653 (12)	0.0088 (9)	0.0141 (8)	0.0204 (9)
C28	0.0336 (10)	0.0390 (10)	0.0439 (10)	0.0122 (9)	0.0139 (8)	0.0090 (8)
C29	0.0275 (9)	0.0328 (9)	0.0283 (7)	0.0083 (8)	0.0028 (6)	0.0051 (6)
C30	0.0245 (8)	0.0201 (7)	0.0257 (7)	0.0007 (7)	0.0031 (6)	0.0052 (5)

C31	0.0506 (11)	0.0162 (7)	0.0346 (8)	0.0081 (8)	0.0127 (7)	0.0034 (6)
C32	0.0709 (16)	0.0287 (10)	0.0549 (12)	-0.0107 (11)	-0.0033 (11)	0.0008 (9)
C33	0.0868 (18)	0.0432 (12)	0.0805 (16)	0.0340 (13)	0.0522 (14)	0.0213 (11)
C34	0.0753 (15)	0.0216 (9)	0.0448 (10)	0.0126 (10)	0.0128 (10)	0.0122 (7)
P3	0.0186 (2)	0.01860 (19)	0.02017 (16)	0.00437 (17)	0.00114 (13)	0.00722 (13)
O9	0.0215 (6)	0.0196 (5)	0.0252 (5)	0.0028 (5)	-0.0006 (4)	0.0098 (4)
O10	0.0218 (5)	0.0215 (5)	0.0219 (5)	0.0041 (5)	0.0022 (4)	0.0067 (4)
O11	0.0423 (7)	0.0173 (5)	0.0343 (6)	0.0073 (5)	0.0120 (5)	0.0078 (4)
O12	0.0706 (10)	0.0278 (6)	0.0289 (6)	0.0067 (7)	0.0131 (6)	0.0042 (5)
N3	0.0261 (7)	0.0170 (6)	0.0239 (6)	0.0046 (6)	0.0038 (5)	0.0090 (5)
C35	0.0161 (7)	0.0222 (7)	0.0259 (7)	0.0008 (7)	0.0012 (5)	0.0121 (5)
C36	0.0219 (8)	0.0223 (8)	0.0327 (7)	0.0031 (7)	-0.0002 (6)	0.0106 (6)
C37	0.0233 (9)	0.0239 (8)	0.0481 (10)	0.0021 (8)	-0.0002 (7)	0.0180 (7)
C38	0.0241 (9)	0.0386 (10)	0.0465 (10)	0.0015 (8)	-0.0030 (7)	0.0284 (8)
C39	0.0282 (9)	0.0477 (11)	0.0302 (8)	0.0017 (9)	-0.0015 (6)	0.0213 (7)
C40	0.0239 (8)	0.0325 (9)	0.0267 (7)	0.0032 (8)	0.0016 (6)	0.0105 (6)
C41	0.0205 (8)	0.0208 (7)	0.0236 (6)	0.0038 (7)	0.0024 (5)	0.0100 (5)
C42	0.0239 (8)	0.0286 (8)	0.0262 (7)	0.0022 (7)	0.0030 (6)	0.0069 (6)
C43	0.0205 (8)	0.0391 (10)	0.0365 (8)	0.0010 (8)	0.0006 (6)	0.0088 (7)
C44	0.0233 (9)	0.0362 (10)	0.0432 (9)	0.0085 (8)	0.0104 (7)	0.0135 (7)
C45	0.0321 (9)	0.0342 (9)	0.0348 (8)	0.0105 (8)	0.0127 (7)	0.0063 (7)
C46	0.0262 (8)	0.0305 (9)	0.0277 (7)	0.0065 (8)	0.0027 (6)	0.0046 (6)
C47	0.0307 (9)	0.0201 (8)	0.0291 (7)	0.0021 (7)	0.0043 (6)	0.0055 (6)
C48	0.0671 (14)	0.0170 (8)	0.0494 (10)	0.0134 (9)	0.0228 (9)	0.0092 (7)
C49	0.0791 (17)	0.0465 (13)	0.0849 (17)	0.0339 (13)	0.0472 (14)	0.0244 (12)
C50	0.124 (3)	0.0217 (10)	0.0651 (15)	-0.0023 (14)	0.0042 (15)	-0.0025 (10)
C51	0.0944 (19)	0.0274 (10)	0.0640 (13)	0.0241 (12)	0.0337 (13)	0.0215 (9)

Geometric parameters (Å, °)

P1—O2	1.4808 (9)	C24—C29	1.395 (2)
P1—O1	1.6144 (11)	C25—C26	1.392 (2)
P1—C7	1.7835 (13)	C25—H25	0.9500
P1—C1	1.7871 (15)	C26—C27	1.381 (3)
O1—N1	1.4430 (14)	C26—H26	0.9500
O4—C13	1.2020 (17)	C27—C28	1.379 (3)
O3—C13	1.3400 (17)	C27—H27	0.9501
O3—C14	1.4757 (17)	C28—C29	1.390 (2)
N1—C13	1.3883 (18)	C28—H28	0.9501
N1—H1N	0.886 (17)	C29—H29	0.9500
C1—C2	1.392 (2)	C31—C32	1.511 (3)
C1—C6	1.396 (2)	C31—C33	1.514 (3)
C2—C3	1.394 (2)	C31—C34	1.518 (2)
C2—H2	0.9500	C32—H321	0.9801
C3—C4	1.378 (3)	C32—H322	0.9800
C3—H3	0.9500	C32—H323	0.9799
C4—C5	1.374 (3)	C33—H331	0.9803
C4—H4	0.9500	C33—H332	0.9802

C5—C6	1.396 (2)	C33—H333	0.9804
C5—H5	0.9501	C34—H341	0.9802
C6—H6	0.9501	C34—H342	0.9801
C7—C12	1.3992 (18)	C34—H343	0.9801
C7—C8	1.3995 (18)	P3—O10	1.4813 (10)
C8—C9	1.3939 (18)	P3—O9	1.6187 (11)
C8—H8	0.9502	P3—C35	1.7853 (13)
C9—C10	1.389 (2)	P3—C41	1.7895 (15)
C9—H9	0.9500	O9—N3	1.4367 (14)
C10—C11	1.391 (2)	O11—C47	1.3393 (18)
C10—H10	0.9500	O11—C48	1.4752 (18)
C11—C12	1.3864 (18)	O12—C47	1.2007 (18)
C11—H11	0.9500	N3—C47	1.3837 (18)
C12—H12	0.9501	N3—H3N	0.923 (18)
C14—C15	1.515 (3)	C35—C40	1.3961 (19)
C14—C17	1.518 (3)	C35—C36	1.4026 (19)
C14—C16	1.518 (2)	C36—C37	1.3942 (19)
C15—H151	0.9800	C36—H36	0.9500
C15—H152	0.9802	C37—C38	1.381 (2)
C15—H153	0.9802	C37—H37	0.9500
C16—H161	0.9802	C38—C39	1.389 (2)
C16—H162	0.9801	C38—H38	0.9500
C16—H163	0.9800	C39—C40	1.395 (2)
C17—H171	0.9805	C39—H39	0.9499
C17—H172	0.9807	C40—H40	0.9502
C17—H173	0.9804	C41—C46	1.3954 (19)
P2—O6	1.4783 (10)	C41—C42	1.399 (2)
P2—O5	1.6162 (11)	C42—C43	1.385 (2)
P2—C18	1.7836 (14)	C42—H42	0.9500
P2—C24	1.7890 (15)	C43—C44	1.386 (2)
O5—N2	1.4374 (14)	C43—H43	0.9500
O7—C30	1.3400 (17)	C44—C45	1.383 (2)
O7—C31	1.4748 (18)	C44—H44	0.9501
O8—C30	1.2018 (17)	C45—C46	1.389 (2)
N2—C30	1.3867 (18)	C45—H45	0.9501
N2—H2N	0.95 (2)	C46—H46	0.9499
C18—C23	1.3976 (19)	C48—C50	1.509 (3)
C18—C19	1.4019 (19)	C48—C51	1.516 (3)
C19—C20	1.3932 (19)	C48—C49	1.528 (3)
C19—H19	0.9501	C49—H491	0.9801
C20—C21	1.384 (2)	C49—H492	0.9800
C20—H20	0.9500	C49—H493	0.9801
C21—C22	1.391 (2)	C50—H501	0.9803
C21—H21	0.9500	C50—H502	0.9802
C22—C23	1.393 (2)	C50—H503	0.9801
C22—H22	0.9500	C51—H511	0.9804
C23—H23	0.9500	C51—H512	0.9803
C24—C25	1.393 (2)	C51—H513	0.9803

O2—P1—O1	115.19 (6)	C25—C26—H26	120.1
O2—P1—C7	112.68 (6)	C28—C27—C26	120.93 (16)
O1—P1—C7	99.91 (6)	C28—C27—H27	119.5
O2—P1—C1	112.18 (6)	C26—C27—H27	119.5
O1—P1—C1	105.79 (6)	C27—C28—C29	119.93 (16)
C7—P1—C1	110.27 (7)	C27—C28—H28	120.1
N1—O1—P1	107.68 (8)	C29—C28—H28	120.0
C13—O3—C14	120.86 (11)	C28—C29—C24	119.56 (16)
C13—N1—O1	111.09 (10)	C28—C29—H29	120.2
C13—N1—H1N	114.4 (11)	C24—C29—H29	120.2
O1—N1—H1N	107.8 (11)	O8—C30—O7	127.79 (14)
C2—C1—C6	119.87 (14)	O8—C30—N2	125.97 (14)
C2—C1—P1	118.31 (11)	O7—C30—N2	106.14 (11)
C6—C1—P1	121.80 (12)	O7—C31—C32	109.70 (14)
C1—C2—C3	119.72 (15)	O7—C31—C33	110.27 (16)
C1—C2—H2	120.1	C32—C31—C33	112.80 (18)
C3—C2—H2	120.2	O7—C31—C34	101.63 (12)
C4—C3—C2	120.01 (17)	C32—C31—C34	110.71 (18)
C4—C3—H3	120.1	C33—C31—C34	111.16 (16)
C2—C3—H3	119.9	C31—C32—H321	109.5
C5—C4—C3	120.74 (16)	C31—C32—H322	109.4
C5—C4—H4	119.7	H321—C32—H322	109.5
C3—C4—H4	119.6	C31—C32—H323	109.5
C4—C5—C6	120.10 (16)	H321—C32—H323	109.5
C4—C5—H5	120.0	H322—C32—H323	109.5
C6—C5—H5	119.9	C31—C33—H331	109.4
C1—C6—C5	119.55 (17)	C31—C33—H332	109.5
C1—C6—H6	120.2	H331—C33—H332	109.5
C5—C6—H6	120.3	C31—C33—H333	109.6
C12—C7—C8	120.03 (12)	H331—C33—H333	109.4
C12—C7—P1	122.08 (10)	H332—C33—H333	109.5
C8—C7—P1	117.89 (10)	C31—C34—H341	109.5
C9—C8—C7	119.44 (13)	C31—C34—H342	109.5
C9—C8—H8	120.3	H341—C34—H342	109.5
C7—C8—H8	120.2	C31—C34—H343	109.4
C10—C9—C8	120.33 (13)	H341—C34—H343	109.5
C10—C9—H9	119.9	H342—C34—H343	109.5
C8—C9—H9	119.8	O10—P3—O9	115.30 (6)
C9—C10—C11	120.09 (13)	O10—P3—C35	112.54 (6)
C9—C10—H10	119.9	O9—P3—C35	99.52 (6)
C11—C10—H10	120.0	O10—P3—C41	111.02 (6)
C12—C11—C10	120.22 (14)	O9—P3—C41	106.13 (6)
C12—C11—H11	119.9	C35—P3—C41	111.72 (7)
C10—C11—H11	119.9	N3—O9—P3	108.44 (8)
C11—C12—C7	119.88 (13)	C47—O11—C48	120.81 (12)
C11—C12—H12	120.1	C47—N3—O9	111.97 (11)
C7—C12—H12	120.0	C47—N3—H3N	117.6 (12)

O4—C13—O3	128.10 (14)	O9—N3—H3N	110.6 (11)
O4—C13—N1	126.01 (14)	C40—C35—C36	120.06 (13)
O3—C13—N1	105.80 (11)	C40—C35—P3	123.37 (11)
O3—C14—C15	109.68 (13)	C36—C35—P3	116.57 (11)
O3—C14—C17	109.88 (15)	C37—C36—C35	119.50 (14)
C15—C14—C17	112.75 (16)	C37—C36—H36	120.2
O3—C14—C16	101.85 (12)	C35—C36—H36	120.3
C15—C14—C16	111.39 (16)	C38—C37—C36	120.25 (15)
C17—C14—C16	110.74 (15)	C38—C37—H37	119.9
C14—C15—H151	109.4	C36—C37—H37	119.9
C14—C15—H152	109.5	C37—C38—C39	120.48 (14)
H151—C15—H152	109.5	C37—C38—H38	119.8
C14—C15—H153	109.5	C39—C38—H38	119.8
H151—C15—H153	109.5	C38—C39—C40	120.07 (15)
H152—C15—H153	109.5	C38—C39—H39	120.0
C14—C16—H161	109.5	C40—C39—H39	120.0
C14—C16—H162	109.5	C39—C40—C35	119.63 (15)
H161—C16—H162	109.5	C39—C40—H40	120.2
C14—C16—H163	109.5	C35—C40—H40	120.2
H161—C16—H163	109.5	C46—C41—C42	119.44 (14)
H162—C16—H163	109.5	C46—C41—P3	123.19 (11)
C14—C17—H171	109.5	C42—C41—P3	117.37 (11)
C14—C17—H172	109.6	C43—C42—C41	120.11 (14)
H171—C17—H172	109.4	C43—C42—H42	120.0
C14—C17—H173	109.5	C41—C42—H42	119.9
H171—C17—H173	109.4	C42—C43—C44	119.96 (15)
H172—C17—H173	109.4	C42—C43—H43	120.0
O6—P2—O5	115.40 (6)	C44—C43—H43	120.0
O6—P2—C18	112.66 (6)	C45—C44—C43	120.48 (15)
O5—P2—C18	99.07 (6)	C45—C44—H44	119.8
O6—P2—C24	110.56 (6)	C43—C44—H44	119.8
O5—P2—C24	106.56 (6)	C44—C45—C46	119.92 (15)
C18—P2—C24	112.02 (7)	C44—C45—H45	120.1
N2—O5—P2	108.19 (8)	C46—C45—H45	120.0
C30—O7—C31	121.02 (11)	C45—C46—C41	120.08 (15)
C30—N2—O5	111.24 (10)	C45—C46—H46	120.0
C30—N2—H2N	115.3 (12)	C41—C46—H46	119.9
O5—N2—H2N	109.6 (11)	O12—C47—O11	127.51 (14)
C23—C18—C19	120.06 (13)	O12—C47—N3	125.58 (15)
C23—C18—P2	123.04 (11)	O11—C47—N3	106.82 (12)
C19—C18—P2	116.90 (11)	O11—C48—C50	110.23 (16)
C20—C19—C18	119.55 (14)	O11—C48—C51	101.75 (14)
C20—C19—H19	120.3	C50—C48—C51	110.8 (2)
C18—C19—H19	120.2	O11—C48—C49	109.18 (16)
C21—C20—C19	120.30 (15)	C50—C48—C49	113.8 (2)
C21—C20—H20	119.9	C51—C48—C49	110.39 (18)
C19—C20—H20	119.8	C48—C49—H491	109.5
C20—C21—C22	120.31 (14)	C48—C49—H492	109.4

C20—C21—H21	119.8	H491—C49—H492	109.5
C22—C21—H21	119.9	C48—C49—H493	109.5
C21—C22—C23	120.13 (15)	H491—C49—H493	109.5
C21—C22—H22	119.9	H492—C49—H493	109.5
C23—C22—H22	119.9	C48—C50—H501	109.4
C22—C23—C18	119.64 (15)	C48—C50—H502	109.6
C22—C23—H23	120.2	H501—C50—H502	109.5
C18—C23—H23	120.2	C48—C50—H503	109.4
C25—C24—C29	120.18 (14)	H501—C50—H503	109.5
C25—C24—P2	117.08 (11)	H502—C50—H503	109.5
C29—C24—P2	122.74 (12)	C48—C51—H511	109.6
C26—C25—C24	119.62 (16)	C48—C51—H512	109.4
C26—C25—H25	120.2	H511—C51—H512	109.4
C24—C25—H25	120.2	C48—C51—H513	109.5
C27—C26—C25	119.77 (16)	H511—C51—H513	109.4
C27—C26—H26	120.2	H512—C51—H513	109.5
O2—P1—O1—N1	58.18 (9)	C18—P2—C24—C25	106.65 (13)
C7—P1—O1—N1	179.16 (8)	O6—P2—C24—C29	160.14 (12)
C1—P1—O1—N1	-66.33 (9)	O5—P2—C24—C29	34.01 (14)
P1—O1—N1—C13	142.58 (11)	C18—P2—C24—C29	-73.30 (14)
O2—P1—C1—C2	26.47 (14)	C31—O7—C30—O8	-1.8 (3)
O1—P1—C1—C2	152.85 (12)	C31—O7—C30—N2	-178.31 (14)
C7—P1—C1—C2	-100.00 (13)	O5—N2—C30—O8	18.3 (2)
O2—P1—C1—C6	-155.13 (12)	O5—N2—C30—O7	-165.15 (12)
O1—P1—C1—C6	-28.76 (14)	C30—O7—C31—C32	-63.8 (2)
C7—P1—C1—C6	78.40 (14)	C30—O7—C31—C33	61.0 (2)
O2—P1—C7—C12	-171.40 (12)	C30—O7—C31—C34	179.00 (15)
O1—P1—C7—C12	65.83 (14)	O10—P3—O9—N3	-57.91 (9)
C1—P1—C7—C12	-45.21 (15)	C35—P3—O9—N3	-178.52 (8)
O2—P1—C7—C8	8.16 (15)	C41—P3—O9—N3	65.44 (9)
O1—P1—C7—C8	-114.61 (12)	P3—O9—N3—C47	-133.94 (12)
C1—P1—C7—C8	134.35 (12)	O10—P3—C35—C40	-174.95 (12)
C14—O3—C13—O4	-2.6 (3)	O9—P3—C35—C40	-52.35 (14)
C14—O3—C13—N1	174.07 (14)	C41—P3—C35—C40	59.37 (15)
O1—N1—C13—O4	-16.6 (2)	O10—P3—C35—C36	4.97 (15)
O1—N1—C13—O3	166.69 (11)	O9—P3—C35—C36	127.57 (12)
C13—O3—C14—C15	64.6 (2)	C41—P3—C35—C36	-120.72 (13)
C13—O3—C14—C17	-59.9 (2)	O10—P3—C41—C46	164.23 (12)
C13—O3—C14—C16	-177.33 (15)	O9—P3—C41—C46	38.23 (14)
O6—P2—O5—N2	-57.27 (9)	C35—P3—C41—C46	-69.26 (15)
C18—P2—O5—N2	-177.79 (8)	O10—P3—C41—C42	-16.56 (14)
C24—P2—O5—N2	65.88 (9)	O9—P3—C41—C42	-142.56 (12)
P2—O5—N2—C30	-138.18 (11)	C35—P3—C41—C42	109.96 (13)
O6—P2—C18—C23	179.38 (13)	C48—O11—C47—O12	6.2 (3)
O5—P2—C18—C23	-58.11 (14)	C48—O11—C47—N3	-170.61 (15)
C24—P2—C18—C23	53.97 (16)	O9—N3—C47—O12	21.4 (2)
O6—P2—C18—C19	-0.87 (15)	O9—N3—C47—O11	-161.75 (12)

O5—P2—C18—C19	121.64 (13)	C47—O11—C48—C50	-64.9 (2)
C24—P2—C18—C19	-126.28 (13)	C47—O11—C48—C51	177.42 (17)
O6—P2—C24—C25	-19.90 (14)	C47—O11—C48—C49	60.7 (2)
O5—P2—C24—C25	-146.03 (12)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1N \cdots O6	0.886 (17)	1.836 (17)	2.7138 (16)	170.0 (15)
N2—H2N \cdots O2	0.95 (2)	1.83 (2)	2.7688 (16)	168.4 (17)
N3—H3N \cdots O10 ⁱ	0.923 (18)	1.855 (19)	2.7666 (16)	169.0 (16)

Symmetry code: (i) $-x+1, -y+2, -z+1$.