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## Structure Reports

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## 4-Bromo-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-5-(4-methoxybenzylideneamino)-1*H*-pyrazole-3-carbonitrile

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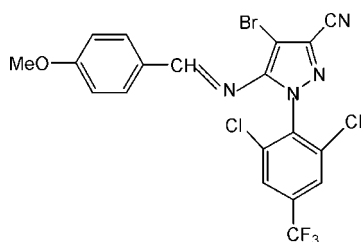
Received 12 February 2008; accepted 27 February 2008

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.050;  $wR$  factor = 0.161; data-to-parameter ratio = 17.3.

The title compound,  $\text{C}_{19}\text{H}_{10}\text{BrCl}_2\text{F}_3\text{N}_4\text{O}$ , is an imine with an overall Y shape. The dihedral angles between the pyrazole ring and the methoxy- and trifluoromethyl-substituted benzene ring planes are  $88.4$  (2) and  $65.8$  (2)°, respectively.

### Related literature

For the insecticidal properties of similar compounds, see: Philippe (1997, 2000). For a related structure, see: Zhong *et al.* (2005).



### Experimental

#### Crystal data

 $\text{C}_{19}\text{H}_{10}\text{BrCl}_2\text{F}_3\text{N}_4\text{O}$  $M_r = 518.12$ Monoclinic,  $P2_1/c$  $a = 10.215$  (7) Å $b = 13.407$  (9) Å $c = 15.108$  (11) Å $\beta = 94.634$  (13)° $V = 2062$  (2) Å<sup>3</sup> $Z = 4$ Mo  $K\alpha$  radiation $\mu = 2.30$  mm<sup>-1</sup> $T = 293$  (2) K $0.28 \times 0.25 \times 0.23$  mm

#### Data collection

Bruker P4/SMART CCD

diffractometer

Absorption correction: multi-scan

(SADABS; Bruker, 2002)

 $T_{\min} = 0.568$ ,  $T_{\max} = 0.618$ 

15375 measured reflections

4681 independent reflections

4061 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.028$ 

#### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.049$  $wR(F^2) = 0.160$  $S = 1.02$ 

4681 reflections

271 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.87$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -1.22$  e Å<sup>-3</sup>

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SJ2465).

### References

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**supplementary materials**

*Acta Cryst.* (2008). E64, o646 [ doi:10.1107/S1600536808005539 ]

## 4-Bromo-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-5-(4-methoxybenzylideneamino)-1H-pyrazole-3-carbonitrile

G. Zhu, S. Li and Z. Yang

### Comment

The title compound, (I), (Fig. 1) is similar to the very effective insecticides used to treat animals such as cows and sheep (Philippe, 1997, 2000) and its structure is reported here, Fig 1. The molecule contains three essentially planar rings and the dihedral angles between the pyrazole ring (C1—C3, N2, N3) and the ring planes of the (C5—C7) and (C13—C17) benzene rings are 88.4 (2)° and 65.8 (2)°, respectively.

### Experimental

The title compound was prepared by literature methods (Zhong *et al.*, 2005, Philippe, 2000). Colourless single crystals suitable for X-ray analysis were obtained by slow evaporation of an anhydrous ethanol-acetone (2:1) solution of (I) (m.p. 435–437 K).

### Refinement

All H atoms were initially located in a difference Fourier map but were eventually placed in their geometrically idealized positions and constrained to ride on their parent atoms, with C—H = 0.93 Å, and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  and 0.96 Å,  $U_{\text{iso}} = 1.5U_{\text{eq}}(\text{C})$  for CH<sub>3</sub> atoms. The low  $U_{\text{eq}}$  of C11 as compared to its neighbours may be attributed to the high displacement parameters for atoms F1, F2 and F3, indicating either large thermal motion or rotational disorder of the trifluoromethyl group. However, attempts to represent the CF<sub>3</sub> group using a disorder model were unsuccessful.

### Figures

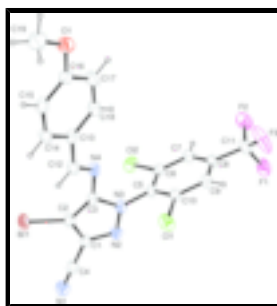


Fig. 1. The molecular structure of (I) showing the atom numbering scheme and displacement ellipsoids at the 30% probability level.

## 4-Bromo-1-[2,6-dichloro-4-(trifluoromethyl)phenyl]-5-(4-methoxybenzylideneamino)-1H-pyrazole-3-carbonitrile

### Crystal data

C<sub>19</sub>H<sub>10</sub>BrCl<sub>2</sub>F<sub>3</sub>N<sub>4</sub>O

$M_r = 518.12$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 10.215 (7) \text{ \AA}$

$b = 13.407 (9) \text{ \AA}$

$c = 15.108 (11) \text{ \AA}$

$\beta = 94.634 (13)^\circ$

$V = 2062 (2) \text{ \AA}^3$

$Z = 4$

$F_{000} = 1024$

$D_x = 1.669 \text{ Mg m}^{-3}$

Melting point: 435-437 K

Mo  $K\alpha$  radiation

$\lambda = 0.71073 \text{ \AA}$

Cell parameters from 3780 reflections

$\theta = 2.6\text{--}24.2^\circ$

$\mu = 2.30 \text{ mm}^{-1}$

$T = 293 (2) \text{ K}$

Block, colorless

$0.28 \times 0.25 \times 0.23 \text{ mm}$

### Data collection

Bruker P4 CCD  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 293(2) \text{ K}$

$\omega$  scans

Absorption correction: multi-scan  
(SADABS; Bruker, 2002)

$T_{\min} = 0.568$ ,  $T_{\max} = 0.618$

15375 measured reflections

4681 independent reflections

4061 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$

$\theta_{\max} = 27.6^\circ$

$\theta_{\min} = 2.8^\circ$

$h = -13 \rightarrow 10$

$k = -17 \rightarrow 17$

$l = -19 \rightarrow 19$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.049$

$wR(F^2) = 0.160$

$S = 1.02$

4681 reflections

271 parameters

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.1058P)^2 + 1.2139P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.87 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -1.22 \text{ e \AA}^{-3}$

Extinction correction: none

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on All data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Br1	0.47949 (4)	0.47599 (2)	0.18806 (3)	0.06204 (17)
Cl1	0.47167 (9)	0.15617 (7)	-0.07867 (6)	0.0635 (3)
Cl2	0.21780 (9)	0.09133 (6)	0.21206 (5)	0.0598 (2)
O1	-0.1317 (3)	0.5592 (2)	-0.2291 (2)	0.0705 (7)
N1	0.7437 (3)	0.2820 (3)	0.3107 (3)	0.0722 (9)
N2	0.5089 (2)	0.17560 (18)	0.16494 (16)	0.0440 (5)
N3	0.4063 (2)	0.19538 (16)	0.10558 (15)	0.0401 (5)
N4	0.2718 (2)	0.32184 (17)	0.03993 (16)	0.0440 (5)
C1	0.5457 (3)	0.2656 (2)	0.19544 (18)	0.0426 (6)
C2	0.4669 (3)	0.34164 (19)	0.1561 (2)	0.0431 (6)
C3	0.3773 (3)	0.29447 (18)	0.09722 (18)	0.0393 (5)
C4	0.6555 (3)	0.2745 (2)	0.2597 (2)	0.0510 (7)
C5	0.3380 (3)	0.11449 (18)	0.06234 (17)	0.0372 (5)
C6	0.2496 (3)	0.0587 (2)	0.10637 (18)	0.0404 (5)
C7	0.1868 (3)	-0.02304 (19)	0.0660 (2)	0.0455 (6)
H7	0.1275	-0.0603	0.0959	0.055*
C8	0.2134 (3)	-0.0482 (2)	-0.0186 (2)	0.0444 (6)
C9	0.2998 (3)	0.0065 (2)	-0.0649 (2)	0.0467 (6)
H9	0.3161	-0.0110	-0.1225	0.056*
C10	0.3616 (3)	0.0880 (2)	-0.02374 (18)	0.0412 (5)
C11	0.1513 (4)	-0.1404 (2)	-0.0601 (3)	0.0606 (8)
C12	0.2714 (3)	0.4090 (2)	0.0053 (2)	0.0468 (6)
H12	0.3433	0.4503	0.0194	0.056*
C13	0.1648 (3)	0.4469 (2)	-0.05455 (19)	0.0423 (6)
C14	0.1740 (3)	0.5440 (2)	-0.0862 (2)	0.0499 (7)
H14	0.2473	0.5823	-0.0682	0.060*
C15	0.0759 (3)	0.5846 (2)	-0.1438 (2)	0.0485 (6)
H15	0.0826	0.6499	-0.1637	0.058*
C16	-0.0316 (3)	0.5275 (2)	-0.1712 (2)	0.0474 (6)
C17	-0.0425 (3)	0.4296 (3)	-0.1403 (2)	0.0556 (7)
H17	-0.1157	0.3913	-0.1586	0.067*
C18	0.0549 (3)	0.3903 (2)	-0.0829 (2)	0.0499 (7)
H18	0.0476	0.3252	-0.0627	0.060*

## supplementary materials

C19	-0.1250 (5)	0.6581 (3)	-0.2612 (4)	0.0878 (15)
H19C	-0.2005	0.6714	-0.3015	0.132*
H19B	-0.0467	0.6660	-0.2916	0.132*
H19A	-0.1233	0.7039	-0.2123	0.132*
F1	0.2269 (3)	-0.21908 (16)	-0.0449 (2)	0.0972 (10)
F2	0.0391 (3)	-0.1633 (2)	-0.0287 (3)	0.1064 (11)
F3	0.1310 (4)	-0.1354 (3)	-0.1460 (2)	0.1269 (15)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Br1	0.0722 (3)	0.0340 (2)	0.0776 (3)	-0.00455 (12)	-0.00746 (19)	-0.01343 (13)
Cl1	0.0740 (5)	0.0690 (5)	0.0497 (4)	-0.0276 (4)	0.0184 (4)	-0.0037 (3)
Cl2	0.0782 (5)	0.0583 (5)	0.0453 (4)	0.0063 (4)	0.0197 (4)	0.0037 (3)
O1	0.0646 (15)	0.0602 (15)	0.0822 (18)	0.0061 (12)	-0.0225 (13)	0.0004 (13)
N1	0.0669 (19)	0.0670 (19)	0.079 (2)	0.0077 (15)	-0.0188 (17)	-0.0201 (16)
N2	0.0514 (13)	0.0380 (12)	0.0416 (11)	0.0037 (10)	-0.0028 (10)	-0.0017 (9)
N3	0.0520 (12)	0.0308 (10)	0.0366 (10)	-0.0008 (9)	-0.0025 (9)	-0.0019 (8)
N4	0.0547 (13)	0.0327 (11)	0.0434 (12)	0.0007 (9)	-0.0037 (10)	-0.0006 (9)
C1	0.0483 (14)	0.0377 (13)	0.0412 (13)	0.0013 (10)	0.0008 (11)	-0.0060 (10)
C2	0.0517 (15)	0.0309 (12)	0.0460 (14)	-0.0012 (10)	-0.0002 (11)	-0.0049 (10)
C3	0.0478 (13)	0.0294 (11)	0.0407 (13)	-0.0014 (10)	0.0045 (10)	-0.0017 (9)
C4	0.0555 (17)	0.0464 (15)	0.0501 (16)	0.0048 (12)	-0.0012 (13)	-0.0126 (12)
C5	0.0458 (13)	0.0271 (11)	0.0381 (12)	0.0005 (9)	0.0002 (10)	-0.0006 (9)
C6	0.0481 (14)	0.0325 (12)	0.0410 (13)	0.0050 (10)	0.0052 (11)	0.0046 (10)
C7	0.0440 (14)	0.0331 (13)	0.0596 (17)	-0.0016 (10)	0.0057 (12)	0.0084 (11)
C8	0.0461 (14)	0.0305 (12)	0.0551 (16)	-0.0028 (10)	-0.0045 (12)	-0.0017 (11)
C9	0.0571 (16)	0.0409 (13)	0.0419 (14)	-0.0053 (12)	0.0019 (12)	-0.0070 (11)
C10	0.0477 (14)	0.0372 (13)	0.0392 (13)	-0.0059 (10)	0.0053 (10)	-0.0023 (10)
C11	0.0627 (19)	0.0413 (16)	0.075 (2)	-0.0116 (14)	-0.0131 (16)	-0.0049 (15)
C12	0.0539 (16)	0.0358 (13)	0.0498 (15)	-0.0028 (11)	-0.0012 (12)	0.0000 (11)
C13	0.0527 (15)	0.0331 (12)	0.0407 (13)	-0.0004 (11)	0.0027 (11)	-0.0002 (10)
C14	0.0570 (17)	0.0344 (13)	0.0567 (17)	-0.0059 (12)	-0.0056 (13)	0.0023 (12)
C15	0.0604 (17)	0.0320 (12)	0.0522 (16)	0.0032 (11)	-0.0007 (13)	0.0019 (11)
C16	0.0513 (16)	0.0440 (16)	0.0461 (15)	0.0069 (11)	-0.0005 (12)	-0.0038 (11)
C17	0.0546 (17)	0.0481 (17)	0.0628 (19)	-0.0122 (13)	-0.0031 (14)	-0.0011 (14)
C18	0.0614 (17)	0.0363 (13)	0.0522 (16)	-0.0082 (12)	0.0061 (13)	0.0024 (11)
C19	0.091 (3)	0.059 (2)	0.106 (4)	0.020 (2)	-0.035 (3)	0.006 (2)
F1	0.0977 (18)	0.0389 (11)	0.150 (3)	0.0004 (11)	-0.0223 (18)	-0.0259 (14)
F2	0.0762 (16)	0.0812 (18)	0.164 (3)	-0.0395 (14)	0.0210 (18)	-0.0324 (19)
F3	0.198 (4)	0.095 (2)	0.0784 (18)	-0.074 (2)	-0.044 (2)	-0.0037 (16)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

Br1—C2	1.867 (3)	C8—C11	1.503 (4)
Cl1—C10	1.714 (3)	C9—C10	1.384 (4)
Cl2—C6	1.712 (3)	C9—H9	0.9300
O1—C16	1.359 (4)	C11—F3	1.299 (5)
O1—C19	1.416 (5)	C11—F2	1.311 (5)

N1—C4	1.142 (5)	C11—F1	1.317 (4)
N2—C1	1.334 (4)	C12—C13	1.450 (4)
N2—N3	1.349 (3)	C12—H12	0.9300
N3—C3	1.365 (3)	C13—C18	1.393 (4)
N3—C5	1.420 (3)	C13—C14	1.393 (4)
N4—C12	1.280 (4)	C14—C15	1.385 (4)
N4—C3	1.376 (4)	C14—H14	0.9300
C1—C2	1.402 (4)	C15—C16	1.374 (5)
C1—C4	1.428 (4)	C15—H15	0.9300
C2—C3	1.377 (4)	C16—C17	1.401 (5)
C5—C6	1.383 (4)	C17—C18	1.370 (5)
C5—C10	1.388 (4)	C17—H17	0.9300
C6—C7	1.386 (4)	C18—H18	0.9300
C7—C8	1.371 (5)	C19—H19C	0.9600
C7—H7	0.9300	C19—H19B	0.9600
C8—C9	1.380 (4)	C19—H19A	0.9600
C16—O1—C19	117.3 (3)	F3—C11—F2	107.5 (3)
C1—N2—N3	103.5 (2)	F3—C11—F1	105.0 (4)
N2—N3—C3	113.8 (2)	F2—C11—F1	105.5 (3)
N2—N3—C5	118.8 (2)	F3—C11—C8	113.7 (3)
C3—N3—C5	127.3 (2)	F2—C11—C8	113.2 (3)
C12—N4—C3	118.5 (3)	F1—C11—C8	111.4 (3)
N2—C1—C2	112.0 (2)	N4—C12—C13	123.5 (3)
N2—C1—C4	119.7 (3)	N4—C12—H12	118.3
C2—C1—C4	128.3 (3)	C13—C12—H12	118.3
C3—C2—C1	105.6 (2)	C18—C13—C14	118.6 (3)
C3—C2—Br1	129.7 (2)	C18—C13—C12	123.2 (3)
C1—C2—Br1	124.6 (2)	C14—C13—C12	118.2 (3)
N3—C3—N4	118.1 (2)	C15—C14—C13	121.1 (3)
N3—C3—C2	105.0 (2)	C15—C14—H14	119.4
N4—C3—C2	136.8 (2)	C13—C14—H14	119.4
N1—C4—C1	179.6 (5)	C16—C15—C14	119.4 (3)
C6—C5—C10	118.8 (2)	C16—C15—H15	120.3
C6—C5—N3	120.5 (2)	C14—C15—H15	120.3
C10—C5—N3	120.7 (2)	O1—C16—C15	124.1 (3)
C5—C6—C7	120.8 (3)	O1—C16—C17	115.5 (3)
C5—C6—C12	119.4 (2)	C15—C16—C17	120.3 (3)
C7—C6—C12	119.8 (2)	C18—C17—C16	119.8 (3)
C8—C7—C6	119.1 (3)	C18—C17—H17	120.1
C8—C7—H7	120.5	C16—C17—H17	120.1
C6—C7—H7	120.5	C17—C18—C13	120.8 (3)
C7—C8—C9	121.6 (3)	C17—C18—H18	119.6
C7—C8—C11	118.9 (3)	C13—C18—H18	119.6
C9—C8—C11	119.5 (3)	O1—C19—H19C	109.5
C8—C9—C10	118.7 (3)	O1—C19—H19B	109.5
C8—C9—H9	120.7	H19C—C19—H19B	109.5
C10—C9—H9	120.7	O1—C19—H19A	109.5
C9—C10—C5	121.0 (2)	H19C—C19—H19A	109.5
C9—C10—C11	119.8 (2)	H19B—C19—H19A	109.5

Fig. 1

