

**2-(1*H*-Tetrazol-5-yl)benzonitrile**

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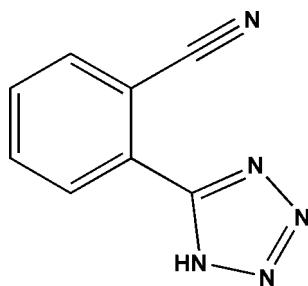
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(C-C) = 0.003$  Å;  $R$  factor = 0.055;  $wR$  factor = 0.138; data-to-parameter ratio = 12.7.

The title compound,  $C_8H_5N_5$ , was synthesized from phthalonitrile. The benzonitrile and tetrazole rings are inclined at a dihedral angle of  $37.14(11)^\circ$ . In the crystal structure, intermolecular N—H···N hydrogen bonds link the tetrazole rings of adjacent molecules, forming chains along the  $a$  axis.

**Related literature**

For background to the chemistry of tetrazoles, see: Bekhit *et al.* (2004); Aykut İkizler & Sancak (1992, 1995, 1998); Rajasekaran & Thampi (2004); Satyanarayana *et al.* (2006); Schmidt & Schieffer (2003); Upadhyayaya *et al.* (2004); Wexler *et al.* (1996).

**Experimental***Crystal data*

$C_8H_5N_5$   
 $M_r = 171.17$   
Monoclinic,  $P2_1/n$   
 $a = 4.9281(10)$  Å

$b = 6.5420(13)$  Å  
 $c = 24.867(5)$  Å  
 $\beta = 95.27(3)^\circ$   
 $V = 798.3(3)$  Å<sup>3</sup>

$Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>

$T = 293(2)$  K  
 $0.25 \times 0.07 \times 0.07$  mm

*Data collection*

Rigaku SCXmini diffractometer  
Absorption correction: multi-scan  
(*CrystalClear*; Rigaku, 2005)  
 $T_{\min} = 0.993$ ,  $T_{\max} = 0.996$

6844 measured reflections  
1553 independent reflections  
1167 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.061$

*Refinement*

$R[F^2 > 2\sigma(F^2)] = 0.055$   
 $wR(F^2) = 0.138$   
 $S = 1.10$   
1553 reflections  
122 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\max} = 0.21$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.24$  e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N2—H2···N5 <sup>i</sup>	0.92 (3)	1.91 (3)	2.820 (2)	172 (2)
N2—H2···N4 <sup>i</sup>	0.92 (3)	2.60 (3)	3.374 (2)	142 (2)

Symmetry code: (i)  $x - 1, y, z$ .

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SJ2471).

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# supporting information

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## **2-(1*H*-Tetrazol-5-yl)benzonitrile**

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### **S1. Comment**

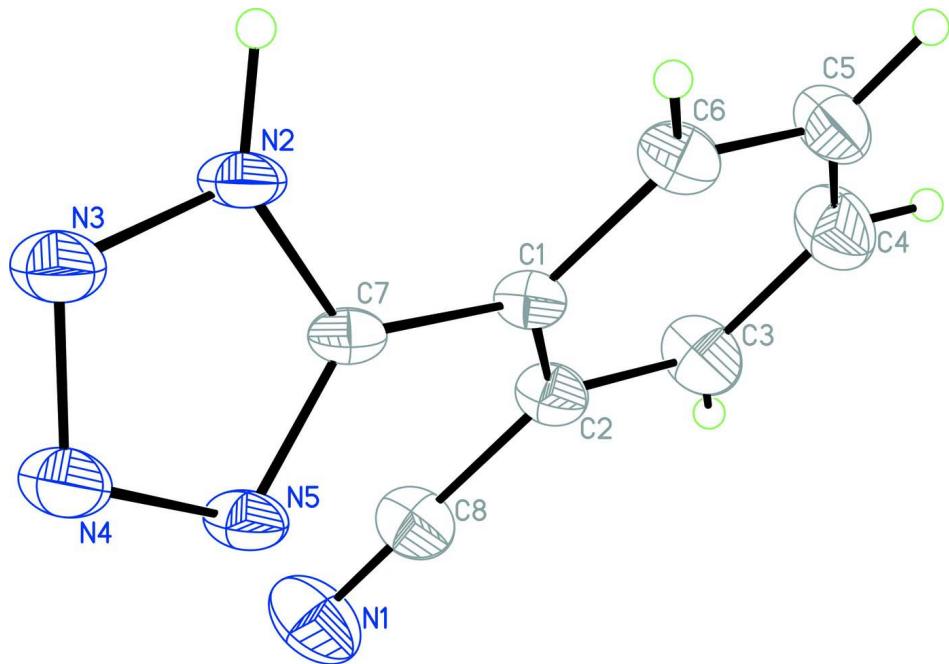
Nitriles are close relatives of azoles and hydrazones and are parent compounds for the preparation of various functional organic materials having triazole, imidazole or tetrazole rings (Aykut İkizler & Sancak, 1992, 1995, 1998). Tetrazoles find wide application in the synthesis of medicinal products such as antihypertensive agents (Wexler *et al.*, 1996; Schmidt & Schieffer, 2003; Satyanarayana *et al.*, 2006), resolvents (Bekhit *et al.*, 2004), anaesthetics (Rajasekaran & Thampi, 2004) and antifungal agents (Upadhayaya *et al.*, 2004). We report herein the crystal structure of the title compound, 2-(1*H*-tetrazol-5-yl)benzonitrile, Fig 1 with its crystal packing shown in Figure 2, Table 1.

### **S2. Experimental**

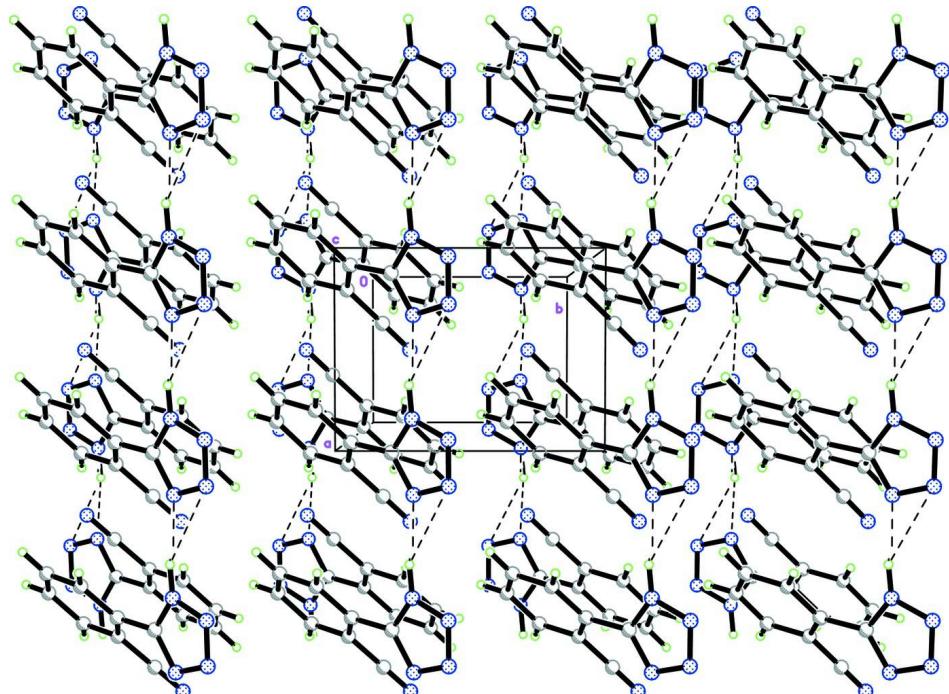
Phthalonitrile (1.28 g, 0.01 mol), sodium azide (0.975 g, 0.015 mol), ammonium chloride (0.605 g, 0.011 mol) and DMF (15 ml) were added in a flask and reacted at 120 °C with stirring for 24 h. A mass of white solid was collected after solvents removed. The crude product was recrystallized by slowly evaporating a mixed solution of ethanol and water (2:1) to yield colorless prism-like crystals, suitable for X-ray analysis.

### **S3. Refinement**

The H2 atom bound to N2 was located in a difference map and was refined freely. Other H atoms were placed in calculated positions, with C—H = 0.93 Å and included in the final cycles of refinement using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}$

**Figure 1**

A view of the title compound with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

**Figure 2**

The crystal packing of the title compound viewed along the *c* axis. Hydrogen bonds are drawn as dashed lines.

(I)

*Crystal data*

$C_8H_5N_5$   
 $M_r = 171.17$   
Monoclinic,  $P2_1/n$   
Hall symbol: -P 2yn  
 $a = 4.9281 (10) \text{ \AA}$   
 $b = 6.5420 (13) \text{ \AA}$   
 $c = 24.867 (5) \text{ \AA}$   
 $\beta = 95.27 (3)^\circ$   
 $V = 798.3 (3) \text{ \AA}^3$   
 $Z = 4$

$F(000) = 352$   
 $D_x = 1.424 \text{ Mg m}^{-3}$   
Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
Cell parameters from 6584 reflections  
 $\theta = 3.1\text{--}28.8^\circ$   
 $\mu = 0.10 \text{ mm}^{-1}$   
 $T = 293 \text{ K}$   
Prism, colorless  
 $0.25 \times 0.07 \times 0.07 \text{ mm}$

*Data collection*

Rigaku SCXmini  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
Detector resolution: 13.6612 pixels  $\text{mm}^{-1}$   
 $\omega$  scans  
Absorption correction: multi-scan  
(*CrystalClear*; Rigaku, 2005)  
 $T_{\min} = 0.993$ ,  $T_{\max} = 0.996$

6844 measured reflections  
1553 independent reflections  
1167 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.061$   
 $\theta_{\max} = 26.0^\circ$ ,  $\theta_{\min} = 3.2^\circ$   
 $h = -6 \rightarrow 6$   
 $k = -8 \rightarrow 8$   
 $l = -30 \rightarrow 30$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.055$   
 $wR(F^2) = 0.138$   
 $S = 1.10$   
1553 reflections  
122 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods  
Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites  
H atoms treated by a mixture of independent  
and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0619P)^2 + 0.1285P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.21 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.24 \text{ e \AA}^{-3}$   
Extinction correction: *SHELXL*  
Extinction coefficient: 0.044 (14)

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.5249 (4)	0.5054 (3)	0.13602 (8)	0.0345 (5)
C2	0.6569 (4)	0.5277 (3)	0.08892 (8)	0.0403 (5)

C3	0.6021 (5)	0.6960 (4)	0.05536 (10)	0.0548 (7)
H3	0.6925	0.7116	0.0244	0.066*
C4	0.4141 (5)	0.8390 (4)	0.06821 (11)	0.0608 (7)
H4	0.3762	0.9509	0.0457	0.073*
C5	0.2821 (5)	0.8174 (4)	0.11410 (11)	0.0539 (6)
H5	0.1537	0.9140	0.1223	0.065*
C6	0.3384 (4)	0.6532 (3)	0.14819 (9)	0.0441 (6)
H6	0.2504	0.6416	0.1796	0.053*
C7	0.5736 (4)	0.3299 (3)	0.17202 (8)	0.0323 (5)
C8	0.8482 (5)	0.3773 (4)	0.07280 (9)	0.0476 (6)
N1	0.9990 (5)	0.2612 (4)	0.05847 (9)	0.0701 (7)
N2	0.3814 (3)	0.2405 (3)	0.19806 (7)	0.0399 (5)
N3	0.4920 (3)	0.0893 (3)	0.22916 (8)	0.0480 (5)
N4	0.7498 (3)	0.0881 (3)	0.22217 (7)	0.0458 (5)
N5	0.8068 (3)	0.2363 (3)	0.18650 (7)	0.0380 (5)
H2	0.195 (6)	0.253 (4)	0.1939 (10)	0.072 (8)*

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0263 (10)	0.0405 (11)	0.0363 (11)	0.0019 (9)	0.0016 (8)	0.0010 (9)
C2	0.0345 (12)	0.0463 (12)	0.0407 (12)	0.0040 (10)	0.0062 (9)	0.0018 (10)
C3	0.0598 (16)	0.0595 (15)	0.0473 (14)	0.0110 (12)	0.0171 (12)	0.0137 (12)
C4	0.0658 (17)	0.0558 (15)	0.0611 (17)	0.0139 (13)	0.0072 (14)	0.0187 (13)
C5	0.0491 (14)	0.0480 (14)	0.0642 (16)	0.0144 (11)	0.0041 (12)	-0.0003 (12)
C6	0.0397 (12)	0.0507 (13)	0.0425 (13)	0.0070 (10)	0.0069 (10)	-0.0014 (10)
C7	0.0221 (9)	0.0434 (11)	0.0319 (11)	0.0004 (8)	0.0055 (8)	-0.0021 (9)
C8	0.0476 (13)	0.0573 (15)	0.0395 (13)	0.0069 (12)	0.0127 (10)	0.0069 (11)
N1	0.0705 (15)	0.0809 (15)	0.0623 (15)	0.0288 (13)	0.0242 (12)	0.0062 (12)
N2	0.0203 (9)	0.0531 (11)	0.0468 (11)	0.0024 (8)	0.0058 (8)	0.0107 (8)
N3	0.0306 (9)	0.0600 (12)	0.0540 (12)	0.0029 (9)	0.0074 (8)	0.0166 (10)
N4	0.0296 (9)	0.0583 (12)	0.0494 (11)	0.0047 (8)	0.0042 (8)	0.0139 (9)
N5	0.0238 (8)	0.0510 (10)	0.0400 (10)	0.0035 (8)	0.0065 (7)	0.0070 (8)

*Geometric parameters ( $\text{\AA}$ ,  $\text{^\circ}$ )*

C1—C6	1.386 (3)	C5—H5	0.9300
C1—C2	1.399 (3)	C6—H6	0.9300
C1—C7	1.462 (3)	C7—N5	1.323 (2)
C2—C3	1.393 (3)	C7—N2	1.331 (2)
C2—C8	1.445 (3)	C8—N1	1.142 (3)
C3—C4	1.375 (3)	N2—N3	1.340 (2)
C3—H3	0.9300	N2—H2	0.92 (3)
C4—C5	1.372 (3)	N3—N4	1.298 (2)
C4—H4	0.9300	N4—N5	1.361 (2)
C5—C6	1.381 (3)		
C6—C1—C2	118.61 (19)	C6—C5—H5	119.8

C6—C1—C7	119.24 (18)	C5—C6—C1	120.5 (2)
C2—C1—C7	122.13 (17)	C5—C6—H6	119.7
C3—C2—C1	120.31 (19)	C1—C6—H6	119.7
C3—C2—C8	117.86 (19)	N5—C7—N2	107.65 (17)
C1—C2—C8	121.81 (19)	N5—C7—C1	128.31 (17)
C4—C3—C2	119.7 (2)	N2—C7—C1	123.99 (17)
C4—C3—H3	120.1	N1—C8—C2	177.8 (2)
C2—C3—H3	120.1	C7—N2—N3	109.61 (16)
C5—C4—C3	120.3 (2)	C7—N2—H2	130.9 (16)
C5—C4—H4	119.9	N3—N2—H2	118.7 (16)
C3—C4—H4	119.9	N4—N3—N2	106.16 (16)
C4—C5—C6	120.5 (2)	N3—N4—N5	110.25 (16)
C4—C5—H5	119.8	C7—N5—N4	106.33 (15)
C6—C1—C2—C3	0.4 (3)	C6—C1—C7—N5	-141.7 (2)
C7—C1—C2—C3	179.22 (19)	C2—C1—C7—N5	39.5 (3)
C6—C1—C2—C8	-178.3 (2)	C6—C1—C7—N2	35.6 (3)
C7—C1—C2—C8	0.5 (3)	C2—C1—C7—N2	-143.2 (2)
C1—C2—C3—C4	-1.1 (4)	N5—C7—N2—N3	-0.1 (2)
C8—C2—C3—C4	177.7 (2)	C1—C7—N2—N3	-177.85 (18)
C2—C3—C4—C5	0.5 (4)	C7—N2—N3—N4	0.3 (2)
C3—C4—C5—C6	0.7 (4)	N2—N3—N4—N5	-0.4 (2)
C4—C5—C6—C1	-1.3 (4)	N2—C7—N5—N4	-0.2 (2)
C2—C1—C6—C5	0.8 (3)	C1—C7—N5—N4	177.48 (19)
C7—C1—C6—C5	-178.05 (19)	N3—N4—N5—C7	0.4 (2)

*Hydrogen-bond geometry (Å, °)*

D—H···A	D—H	H···A	D···A	D—H···A
N2—H2···N5 <sup>i</sup>	0.92 (3)	1.91 (3)	2.820 (2)	172 (2)
N2—H2···N4 <sup>i</sup>	0.92 (3)	2.60 (3)	3.374 (2)	142 (2)

Symmetry code: (i)  $x-1, y, z$ .