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## Structure Reports

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# N-{3-[Bis(2-hydroxyethyl)aminomethyl]-5-nitrophenyl}benzamide

Anna Mai, Gul S. Khan, George R. Clark and David Barker\*

Department of Chemistry, The University of Auckland, Private Bag 92019, Auckland, New Zealand

Correspondence e-mail: g.clark@auckland.ac.nz

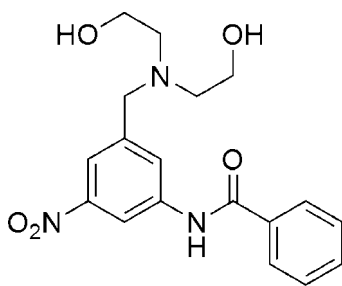
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Key indicators: single-crystal X-ray study;  $T = 90$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.059;  $wR$  factor = 0.134; data-to-parameter ratio = 14.3.

The title compound,  $\text{C}_{18}\text{H}_{21}\text{N}_3\text{O}_5$ , was prepared by the reaction of 3-benzamido-5-nitrobenzyl methanesulfonate with diethanolamine and is an intermediate in the synthesis of DNA minor-groove-binding polybenzamide agents capable of being conjugated to additional biologically active species. The asymmetric unit contains two independent molecules, which differ only in the orientations of the hydroxyethyl groups. In the crystal structure, intermolecular  $\text{N}-\text{H}\cdots\text{O}$  and  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds link molecules into one-dimensional chains.

## Related literature

For related literature on the biological activity of polybenzamide DNA binding agents, see: Storl *et al.* (1993). For related literature on natural and synthetic minor-groove binding agents, including agents containing conjugates, see: Arcamone *et al.* (1964); Atwell *et al.* (1995); Baraldi *et al.* (1999, 2004, 2007); Kumar *et al.* (2004); Sengupta *et al.* (1996); Stafford *et al.* (2007); Turner *et al.* (1999); Wemmer (2000); Yan *et al.* (1997). For related literature, see: Barker *et al.* (2008).



## Experimental

### Crystal data

$\text{C}_{18}\text{H}_{21}\text{N}_3\text{O}_5$   
 $M_r = 359.38$   
 Monoclinic,  $P2_1/c$   
 $a = 22.7867$  (3) Å  
 $b = 11.0879$  (1) Å  
 $c = 13.5106$  (1) Å  
 $\beta = 90.114$  (1)°  
 $V = 3413.54$  (6) Å<sup>3</sup>  
 $Z = 8$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 90$  (2) K  
 $0.34 \times 0.22 \times 0.20$  mm

### Data collection

Bruker SMART CCD diffractometer  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1997)  
 $T_{\min} = 0.858$ ,  $T_{\max} = 0.978$   
 20279 measured reflections  
 6944 independent reflections  
 5142 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.039$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.058$   
 $wR(F^2) = 0.134$   
 $S = 1.04$   
 6944 reflections  
 485 parameters  
 H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.25$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.29$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O4A}-\text{HO4A}\cdots\text{O1A}^i$	0.93 (4)	1.82 (4)	2.736 (2)	168 (3)
$\text{O5A}-\text{HO5A}\cdots\text{O4B}$	0.89 (3)	1.85 (3)	2.738 (2)	178 (3)
$\text{N1A}-\text{H1A}\cdots\text{O5B}$	0.86	2.27	3.089 (2)	159
$\text{O5B}-\text{HO5B}\cdots\text{O4B}$	0.90 (4)	2.30 (4)	3.130 (3)	153 (3)
$\text{O4B}-\text{HO4B}\cdots\text{O4A}$	1.03 (4)	1.75 (4)	2.762 (3)	169 (3)
$\text{N1B}-\text{H1B}\cdots\text{O5A}$	0.86	2.49	3.332 (2)	167

Symmetry code: (i)  $x, y - 1, z$ .

Data collection: SMART (Bruker, 1995); cell refinement: SAINT (Bruker, 1995); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP III (Burnett & Johnson, 1996); software used to prepare material for publication: SHELXTL (Sheldrick, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2602).

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**supplementary materials**

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## *N*-{3-[Bis(2-hydroxyethyl)aminomethyl]-5-nitrophenyl}benzamide

A. Mai, G. S. Khan, G. R. Clark and D. Barker

### Comment

The naturally occurring antibiotic oligopeptides distamycin A, isolated from *Streptomyces Distallicus*, and netropsin, from *Streptomyces netropsis*, are powerful DNA minor groove-binding agents but their cytotoxicity precludes their use as medicines (Arcamone *et al.*, 1964, Baraldi *et al.*, 2004, Wemmer *et al.*, 2000, Storl *et al.*, 1993). In order to increase the DNA binding affinity and sequence specificity along with minimizing the unwanted physiological activities associated with these natural DNA binders, many synthetic oligopeptides have been prepared (Baraldi *et al.*, 2007). In addition, numerous biologically active species have been conjugated to natural and synthetic DNA binding oligopeptides with the purpose of increasing the concentration of these species near DNA (Kumar *et al.*, 2004, Stafford *et al.*, 2007). The title compound is a key intermediate required in the synthesis of a novel polybenzamide DNA minor groove-binding agent.

### Experimental

*N,N*-Bis(2-hydroxyethyl)-3-benzamido-5-nitrobenzylamine was prepared using the method of Barker *et al.* (2008). To a solution of 3-benzamido-5-nitrobenzyl methanesulfonate (0.129 g, 0.368 mmol) in dry THF (1 ml) was added dropwise to a stirred suspension of diethanolamine (0.387 g, 3.68 mmol) in dry THF (2 ml) at 273 K. The mixture was then stirred under an atmosphere of nitrogen overnight before being concentrated *in vacuo* to give a crude residue. This residue was diluted with ethyl acetate (10 ml) and extracted with 2M HCl (2 x 10 ml). The combined acidic extracts were neutralized with 4M NaOH and then extracted with ethyl acetate (2 x 15 ml). The combined organic extracts were dried (MgSO<sub>4</sub>), filtered and the solvent removed *in vacuo*, to afford the title compound (0.128 g, 97%), as a yellow solid, which was recrystallized from ethyl acetate to give yellow crystals (m.p. 385–387 K) suitable for X-ray crystallography. Spectroscopic analysis: IR ( $\nu_{\max}$ , thin film, cm<sup>-1</sup>) 2906, 1680, 1527, 1377. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>,  $\delta$ , p.p.m.) 2.60 (4H, m, N(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>), 3.61 (2H, s, Ar—CH<sub>2</sub>N), 3.68 (4H, m, N(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>), 7.32 (3H, m, Ar—H), 7.46 (1H, m, Ar—H), 7.59 (1H, s, Ar—H), 7.67 (1H, m, Ar—H), 8.10 (1H, br s, Ar—H), 8.35 (1H, m, Ar—H) and 9.03 (NH). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>,  $\delta$ , p.p.m.) 55.9 (CH<sub>2</sub>, N(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>), 58.4 (CH<sub>2</sub>, Ar—CH<sub>2</sub>N), 59.2 (CH<sub>2</sub>, N(CH<sub>2</sub>CH<sub>2</sub>OH)<sub>2</sub>), 114.5 (CH, Ar—C), 118.6 (CH, Ar—C), 125.9 (CH, Ar—C), 127.2 (CH, Ar—C), 128.4 (CH, Ar—C), 132.3 (CH, Ar—C), 133.2 (quat. Ar—C), 139.2 (quat. Ar—C), 142.3 (quat. Ar—C), 148.1 (quat. Ar—C) and 166.4 (C=O) MS *m/z* (FAB) 360 (*M*<sup>+</sup>, 9%), 219 (4), 154 (100), and 120 (NHCOC<sub>6</sub>H<sub>5</sub>, 8). HRMS (FAB), found: *MH*<sup>+</sup> 360.15572. C<sub>18</sub>H<sub>22</sub>N<sub>3</sub>O<sub>5</sub> requires: 360.15595.

### Refinement

Most hydrogen atoms were placed in calculated positions and refined using the riding model with C—H 0.93–0.97 Å and N—H = 0.86 Å, with  $U_{\text{iso}}(\text{H}) = 1.2$  or  $1.5$  times  $U_{\text{eq}}(\text{C})$ . H atoms bonded to O atoms were located in a difference map and refined independently with isotropic displacement parameters.

## Figures

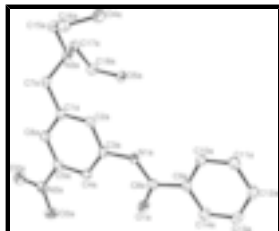


Fig. 1. Molecular structure of one of the independent molecules showing 50% probability displacement ellipsoids for non-hydrogen atoms and hydrogen atoms as arbitrary spheres (Burnett & Johnson, 1996).

## N-{3-[Bis(2-hydroxyethyl)aminomethyl]-5-nitrophenyl}benzamide

### Crystal data

$C_{18}H_{21}N_3O_5$

$M_r = 359.38$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 22.7867$  (3) Å

$b = 11.0879$  (1) Å

$c = 13.5106$  (1) Å

$\beta = 90.114$  (1)°

$V = 3413.54$  (6) Å<sup>3</sup>

$Z = 8$

$F_{000} = 1520$

$D_x = 1.399$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 8192 reflections

$\theta = 0.9$ – $26.4$ °

$\mu = 0.10$  mm<sup>-1</sup>

$T = 90$  (2) K

Needle, yellow

$0.34 \times 0.22 \times 0.20$  mm

### Data collection

Bruker SMART CCD diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 90$ (2) K

area-detector  $\omega$  scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1997)

$T_{\min} = 0.858$ ,  $T_{\max} = 0.978$

20279 measured reflections

6944 independent reflections

5142 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.039$

$\theta_{\max} = 26.4$ °

$\theta_{\min} = 0.9$ °

$h = -28 \rightarrow 25$

$k = -12 \rightarrow 13$

$l = -16 \rightarrow 16$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.058$

$wR(F^2) = 0.134$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0471P)^2 + 3.1255P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$S = 1.04$   $(\Delta/\sigma)_{\max} < 0.001$   
 6944 reflections  $\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$   
 485 parameters  $\Delta\rho_{\min} = -0.29 \text{ e } \text{\AA}^{-3}$   
 Primary atom site location: structure-invariant direct methods Extinction correction: none

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O1A	0.75115 (7)	0.40282 (14)	0.38171 (13)	0.0273 (4)
O2A	0.49791 (7)	0.27802 (15)	0.31865 (13)	0.0281 (4)
O3A	0.56436 (7)	0.41672 (15)	0.30441 (13)	0.0313 (4)
O4A	0.74524 (7)	-0.35100 (16)	0.37188 (13)	0.0298 (4)
HO4A	0.7493 (15)	-0.433 (3)	0.384 (2)	0.066 (11)*
O5A	0.67682 (7)	-0.03575 (15)	0.58394 (12)	0.0220 (4)
HO5A	0.7008 (14)	-0.098 (3)	0.574 (2)	0.051 (9)*
N1A	0.75628 (7)	0.19893 (16)	0.37317 (13)	0.0169 (4)
H1A	0.7800	0.1387	0.3734	0.020*
N2A	0.54971 (8)	0.31179 (18)	0.31848 (14)	0.0218 (4)
N3A	0.63691 (8)	-0.19644 (16)	0.41918 (13)	0.0186 (4)
C1A	0.62210 (9)	0.0133 (2)	0.35151 (15)	0.0185 (5)
C2A	0.68029 (9)	0.0499 (2)	0.36241 (15)	0.0176 (5)
H2A	0.7094	-0.0081	0.3708	0.021*
C3A	0.69585 (9)	0.1717 (2)	0.36096 (15)	0.0167 (5)
C4A	0.65270 (9)	0.2599 (2)	0.34618 (15)	0.0179 (5)
H4A	0.6621	0.3415	0.3440	0.021*
C5A	0.59536 (9)	0.2201 (2)	0.33495 (15)	0.0186 (5)
C6A	0.57860 (9)	0.1005 (2)	0.33811 (15)	0.0179 (5)
H6A	0.5394	0.0786	0.3315	0.022*
C7A	0.60623 (9)	-0.1185 (2)	0.34772 (16)	0.0199 (5)
H7A1	0.5643	-0.1261	0.3587	0.024*
H7A2	0.6143	-0.1482	0.2816	0.024*
C8A	0.78069 (9)	0.3098 (2)	0.38447 (15)	0.0178 (5)
C9A	0.84602 (9)	0.3155 (2)	0.39933 (14)	0.0170 (5)
C10A	0.88212 (9)	0.2143 (2)	0.40848 (15)	0.0188 (5)
H10A	0.8659	0.1374	0.4061	0.023*

## supplementary materials

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C11A	0.94201 (10)	0.2282 (2)	0.42108 (16)	0.0223 (5)
H11A	0.9658	0.1603	0.4269	0.027*
C12A	0.96711 (10)	0.3422 (2)	0.42520 (16)	0.0234 (5)
H12A	1.0074	0.3508	0.4332	0.028*
C13A	0.93159 (10)	0.4429 (2)	0.41722 (17)	0.0244 (5)
H13A	0.9481	0.5195	0.4202	0.029*
C14A	0.87147 (10)	0.4300 (2)	0.40483 (16)	0.0222 (5)
H14A	0.8479	0.4982	0.4001	0.027*
C15A	0.63902 (10)	-0.3212 (2)	0.38016 (16)	0.0217 (5)
H15A	0.6036	-0.3371	0.3424	0.026*
H15B	0.6402	-0.3775	0.4351	0.026*
C16A	0.69204 (10)	-0.3417 (2)	0.31475 (17)	0.0250 (5)
H16A	0.6865	-0.4152	0.2769	0.030*
H16B	0.6956	-0.2753	0.2684	0.030*
C17A	0.61045 (10)	-0.1931 (2)	0.51788 (16)	0.0222 (5)
H17A	0.6280	-0.2564	0.5578	0.027*
H17B	0.5689	-0.2110	0.5116	0.027*
C18A	0.61714 (10)	-0.0742 (2)	0.57241 (17)	0.0225 (5)
H18A	0.5957	-0.0124	0.5367	0.027*
H18B	0.5995	-0.0819	0.6374	0.027*
O1B	0.75199 (7)	0.42678 (15)	0.63678 (14)	0.0322 (4)
O2B	1.00570 (7)	0.29674 (15)	0.68363 (12)	0.0273 (4)
O3B	0.93973 (7)	0.43732 (15)	0.69510 (13)	0.0297 (4)
O4B	0.75112 (7)	-0.22469 (16)	0.54787 (15)	0.0344 (4)
HO4B	0.7437 (17)	-0.269 (4)	0.482 (3)	0.097 (14)*
O5B	0.81384 (7)	-0.05141 (15)	0.39894 (13)	0.0241 (4)
HO5B	0.7984 (15)	-0.080 (3)	0.455 (3)	0.066 (11)*
N1B	0.74752 (7)	0.22275 (16)	0.62409 (13)	0.0172 (4)
H1B	0.7240	0.1631	0.6150	0.021*
N2B	0.95413 (8)	0.33162 (17)	0.68277 (13)	0.0207 (4)
N3B	0.87113 (8)	-0.15414 (16)	0.55811 (13)	0.0189 (4)
C1B	0.88035 (9)	0.0342 (2)	0.65113 (15)	0.0171 (5)
C2B	0.82249 (9)	0.0722 (2)	0.63819 (15)	0.0175 (4)
H2B	0.7933	0.0147	0.6286	0.021*
C3B	0.80713 (9)	0.1941 (2)	0.63923 (14)	0.0164 (4)
C4B	0.85090 (9)	0.2812 (2)	0.65427 (15)	0.0182 (5)
H4B	0.8421	0.3631	0.6555	0.022*
C5B	0.90790 (9)	0.2400 (2)	0.66727 (15)	0.0181 (5)
C6B	0.92402 (9)	0.1200 (2)	0.66600 (15)	0.0182 (5)
H6B	0.9630	0.0972	0.6748	0.022*
C7B	0.89400 (10)	-0.0993 (2)	0.64869 (16)	0.0201 (5)
H7B1	0.9361	-0.1111	0.6520	0.024*
H7B2	0.8765	-0.1383	0.7058	0.024*
C8B	0.72308 (9)	0.3357 (2)	0.62235 (16)	0.0183 (5)
C9B	0.65804 (9)	0.3421 (2)	0.60460 (15)	0.0174 (5)
C10B	0.62171 (9)	0.2426 (2)	0.59274 (16)	0.0211 (5)
H10B	0.6376	0.1653	0.5935	0.025*
C11B	0.56166 (10)	0.2582 (2)	0.57970 (17)	0.0230 (5)
H11B	0.5375	0.1912	0.5718	0.028*

C12B	0.53755 (10)	0.3727 (2)	0.57847 (16)	0.0230 (5)
H12B	0.4973	0.3826	0.5699	0.028*
C13B	0.57351 (10)	0.4725 (2)	0.59003 (17)	0.0244 (5)
H13B	0.5574	0.5496	0.5894	0.029*
C14B	0.63358 (10)	0.4578 (2)	0.60249 (16)	0.0217 (5)
H14B	0.6576	0.5251	0.6095	0.026*
C15B	0.85390 (10)	-0.2800 (2)	0.56934 (18)	0.0241 (5)
H15C	0.8810	-0.3205	0.6138	0.029*
H15D	0.8556	-0.3202	0.5056	0.029*
C16B	0.79190 (10)	-0.2881 (2)	0.6105 (2)	0.0295 (6)
H16C	0.7803	-0.3721	0.6152	0.035*
H16D	0.7910	-0.2537	0.6765	0.035*
C17B	0.90493 (9)	-0.1255 (2)	0.46962 (16)	0.0219 (5)
H17C	0.9354	-0.1856	0.4603	0.026*
H17D	0.9237	-0.0476	0.4777	0.026*
C18B	0.86478 (10)	-0.1229 (2)	0.37859 (17)	0.0256 (5)
H18C	0.8858	-0.0891	0.3227	0.031*
H18D	0.8530	-0.2044	0.3616	0.031*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1A	0.0197 (8)	0.0177 (9)	0.0446 (10)	0.0000 (7)	-0.0021 (7)	-0.0027 (7)
O2A	0.0161 (8)	0.0301 (10)	0.0381 (10)	0.0037 (7)	-0.0001 (7)	0.0003 (8)
O3A	0.0264 (9)	0.0175 (9)	0.0500 (11)	0.0047 (7)	-0.0008 (8)	0.0038 (8)
O4A	0.0260 (9)	0.0202 (10)	0.0433 (10)	0.0004 (7)	-0.0023 (8)	-0.0025 (8)
O5A	0.0189 (8)	0.0200 (9)	0.0269 (9)	-0.0015 (7)	-0.0029 (6)	-0.0022 (7)
N1A	0.0124 (9)	0.0179 (10)	0.0203 (9)	0.0004 (7)	0.0003 (7)	0.0007 (7)
N2A	0.0177 (10)	0.0245 (11)	0.0231 (10)	0.0042 (8)	0.0009 (7)	-0.0007 (8)
N3A	0.0184 (9)	0.0159 (10)	0.0216 (9)	-0.0025 (7)	-0.0008 (7)	0.0015 (7)
C1A	0.0160 (11)	0.0223 (12)	0.0172 (11)	-0.0010 (9)	-0.0012 (8)	0.0007 (9)
C2A	0.0165 (11)	0.0197 (12)	0.0166 (10)	0.0022 (9)	-0.0004 (8)	0.0013 (8)
C3A	0.0140 (10)	0.0212 (12)	0.0148 (10)	-0.0024 (9)	0.0008 (8)	0.0000 (8)
C4A	0.0172 (11)	0.0184 (12)	0.0180 (11)	-0.0006 (9)	0.0019 (8)	0.0009 (8)
C5A	0.0162 (11)	0.0225 (12)	0.0171 (10)	0.0037 (9)	0.0016 (8)	0.0010 (9)
C6A	0.0134 (10)	0.0231 (12)	0.0174 (10)	-0.0027 (9)	0.0019 (8)	0.0011 (9)
C7A	0.0151 (10)	0.0193 (12)	0.0252 (12)	-0.0027 (9)	-0.0041 (9)	0.0002 (9)
C8A	0.0195 (11)	0.0184 (12)	0.0156 (10)	0.0003 (9)	0.0014 (8)	0.0006 (8)
C9A	0.0145 (10)	0.0248 (12)	0.0116 (10)	-0.0020 (9)	0.0011 (8)	-0.0003 (8)
C10A	0.0176 (11)	0.0198 (12)	0.0188 (11)	-0.0031 (9)	0.0007 (8)	0.0002 (9)
C11A	0.0191 (11)	0.0266 (13)	0.0213 (11)	0.0031 (10)	0.0001 (9)	0.0023 (9)
C12A	0.0152 (11)	0.0326 (14)	0.0223 (12)	-0.0056 (10)	-0.0007 (9)	0.0000 (10)
C13A	0.0216 (12)	0.0227 (13)	0.0290 (12)	-0.0079 (10)	0.0007 (10)	-0.0021 (10)
C14A	0.0205 (12)	0.0219 (12)	0.0242 (12)	-0.0008 (9)	-0.0004 (9)	-0.0022 (9)
C15A	0.0234 (12)	0.0193 (12)	0.0226 (11)	-0.0049 (9)	-0.0001 (9)	-0.0001 (9)
C16A	0.0262 (12)	0.0216 (13)	0.0271 (12)	0.0016 (10)	0.0019 (10)	-0.0004 (9)
C17A	0.0201 (11)	0.0217 (12)	0.0250 (12)	-0.0027 (9)	0.0020 (9)	-0.0009 (9)
C18A	0.0195 (11)	0.0245 (13)	0.0233 (12)	-0.0001 (10)	0.0031 (9)	-0.0017 (9)

## supplementary materials

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O1B	0.0197 (9)	0.0183 (9)	0.0585 (12)	-0.0010 (7)	-0.0027 (8)	-0.0001 (8)
O2B	0.0166 (8)	0.0315 (10)	0.0340 (9)	-0.0032 (7)	0.0004 (7)	-0.0044 (8)
O3B	0.0248 (9)	0.0210 (10)	0.0432 (10)	-0.0052 (7)	0.0037 (8)	-0.0076 (7)
O4B	0.0218 (9)	0.0294 (10)	0.0519 (12)	0.0049 (8)	-0.0131 (8)	-0.0103 (9)
O5B	0.0194 (8)	0.0239 (9)	0.0289 (9)	0.0039 (7)	-0.0015 (7)	0.0039 (7)
N1B	0.0120 (9)	0.0178 (10)	0.0217 (9)	-0.0008 (7)	-0.0002 (7)	-0.0017 (7)
N2B	0.0178 (10)	0.0248 (11)	0.0195 (9)	-0.0052 (8)	0.0029 (7)	-0.0023 (8)
N3B	0.0186 (9)	0.0159 (10)	0.0223 (10)	0.0001 (7)	0.0000 (7)	-0.0012 (7)
C1B	0.0164 (11)	0.0211 (12)	0.0136 (10)	0.0001 (9)	-0.0007 (8)	0.0007 (8)
C2B	0.0171 (11)	0.0183 (11)	0.0170 (10)	-0.0029 (9)	-0.0002 (8)	-0.0019 (8)
C3B	0.0154 (10)	0.0214 (12)	0.0124 (10)	-0.0005 (9)	0.0017 (8)	-0.0020 (8)
C4B	0.0188 (11)	0.0191 (12)	0.0168 (10)	0.0013 (9)	0.0031 (8)	-0.0018 (8)
C5B	0.0174 (11)	0.0234 (12)	0.0134 (10)	-0.0053 (9)	0.0032 (8)	-0.0020 (8)
C6B	0.0142 (10)	0.0251 (12)	0.0154 (10)	0.0009 (9)	0.0005 (8)	-0.0023 (9)
C7B	0.0170 (11)	0.0194 (12)	0.0238 (11)	0.0026 (9)	-0.0029 (9)	0.0016 (9)
C8B	0.0190 (11)	0.0170 (12)	0.0189 (11)	0.0001 (9)	0.0012 (8)	0.0012 (8)
C9B	0.0168 (11)	0.0209 (12)	0.0145 (10)	0.0019 (9)	0.0018 (8)	0.0011 (8)
C10B	0.0187 (11)	0.0185 (12)	0.0260 (12)	0.0020 (9)	0.0003 (9)	-0.0020 (9)
C11B	0.0184 (11)	0.0246 (13)	0.0262 (12)	-0.0016 (10)	0.0007 (9)	-0.0012 (9)
C12B	0.0169 (11)	0.0306 (14)	0.0216 (11)	0.0041 (10)	-0.0006 (9)	0.0006 (10)
C13B	0.0255 (12)	0.0209 (13)	0.0267 (12)	0.0086 (10)	-0.0011 (10)	0.0016 (9)
C14B	0.0228 (12)	0.0191 (12)	0.0232 (11)	-0.0002 (9)	0.0003 (9)	0.0022 (9)
C15B	0.0229 (12)	0.0172 (12)	0.0322 (13)	0.0012 (10)	-0.0023 (10)	-0.0003 (9)
C16B	0.0217 (12)	0.0218 (13)	0.0449 (15)	0.0001 (10)	-0.0013 (11)	0.0010 (11)
C17B	0.0169 (11)	0.0228 (12)	0.0259 (12)	0.0029 (9)	0.0018 (9)	-0.0008 (9)
C18B	0.0261 (12)	0.0266 (13)	0.0243 (12)	0.0027 (10)	-0.0003 (10)	-0.0021 (10)

### *Geometric parameters (Å, °)*

O1A—C8A	1.232 (3)	O1B—C8B	1.221 (3)
O2A—N2A	1.238 (2)	O2B—N2B	1.237 (2)
O3A—N2A	1.225 (3)	O3B—N2B	1.229 (2)
O4A—C16A	1.440 (3)	O4B—C16B	1.439 (3)
O4A—HO4A	0.93 (4)	O4B—HO4B	1.03 (4)
O5A—C18A	1.433 (3)	O5B—C18B	1.433 (3)
O5A—HO5A	0.89 (3)	O5B—HO5B	0.90 (4)
N1A—C8A	1.357 (3)	N1B—C8B	1.371 (3)
N1A—C3A	1.419 (3)	N1B—C3B	1.410 (3)
N1A—H1A	0.8600	N1B—H1B	0.8600
N2A—C5A	1.471 (3)	N2B—C5B	1.478 (3)
N3A—C17A	1.465 (3)	N3B—C15B	1.458 (3)
N3A—C7A	1.472 (3)	N3B—C17B	1.458 (3)
N3A—C15A	1.481 (3)	N3B—C7B	1.462 (3)
C1A—C2A	1.394 (3)	C1B—C6B	1.391 (3)
C1A—C6A	1.396 (3)	C1B—C2B	1.394 (3)
C1A—C7A	1.506 (3)	C1B—C7B	1.514 (3)
C2A—C3A	1.396 (3)	C2B—C3B	1.396 (3)
C2A—H2A	0.9300	C2B—H2B	0.9300
C3A—C4A	1.401 (3)	C3B—C4B	1.403 (3)

C4A—C5A	1.387 (3)	C4B—C5B	1.388 (3)
C4A—H4A	0.9300	C4B—H4B	0.9300
C5A—C6A	1.381 (3)	C5B—C6B	1.380 (3)
C6A—H6A	0.9300	C6B—H6B	0.9300
C7A—H7A1	0.9700	C7B—H7B1	0.9700
C7A—H7A2	0.9700	C7B—H7B2	0.9700
C8A—C9A	1.503 (3)	C8B—C9B	1.503 (3)
C9A—C10A	1.396 (3)	C9B—C10B	1.389 (3)
C9A—C14A	1.398 (3)	C9B—C14B	1.399 (3)
C10A—C11A	1.383 (3)	C10B—C11B	1.390 (3)
C10A—H10A	0.9300	C10B—H10B	0.9300
C11A—C12A	1.389 (3)	C11B—C12B	1.384 (3)
C11A—H11A	0.9300	C11B—H11B	0.9300
C12A—C13A	1.383 (3)	C12B—C13B	1.386 (3)
C12A—H12A	0.9300	C12B—H12B	0.9300
C13A—C14A	1.387 (3)	C13B—C14B	1.388 (3)
C13A—H13A	0.9300	C13B—H13B	0.9300
C14A—H14A	0.9300	C14B—H14B	0.9300
C15A—C16A	1.515 (3)	C15B—C16B	1.522 (3)
C15A—H15A	0.9700	C15B—H15C	0.9700
C15A—H15B	0.9700	C15B—H15D	0.9700
C16A—H16A	0.9700	C16B—H16C	0.9700
C16A—H16B	0.9700	C16B—H16D	0.9700
C17A—C18A	1.517 (3)	C17B—C18B	1.532 (3)
C17A—H17A	0.9700	C17B—H17C	0.9700
C17A—H17B	0.9700	C17B—H17D	0.9700
C18A—H18A	0.9700	C18B—H18C	0.9700
C18A—H18B	0.9700	C18B—H18D	0.9700
C16A—O4A—HO4A	104 (2)	C16B—O4B—HO4B	112 (2)
C18A—O5A—HO5A	110 (2)	C18B—O5B—HO5B	107 (2)
C8A—N1A—C3A	127.10 (19)	C8B—N1B—C3B	126.87 (19)
C8A—N1A—H1A	116.5	C8B—N1B—H1B	116.6
C3A—N1A—H1A	116.5	C3B—N1B—H1B	116.6
O3A—N2A—O2A	123.19 (19)	O3B—N2B—O2B	123.43 (19)
O3A—N2A—C5A	119.14 (18)	O3B—N2B—C5B	118.97 (18)
O2A—N2A—C5A	117.67 (19)	O2B—N2B—C5B	117.59 (19)
C17A—N3A—C7A	112.73 (17)	C15B—N3B—C17B	115.88 (18)
C17A—N3A—C15A	111.18 (17)	C15B—N3B—C7B	113.99 (18)
C7A—N3A—C15A	109.31 (17)	C17B—N3B—C7B	114.08 (17)
C2A—C1A—C6A	119.1 (2)	C6B—C1B—C2B	119.2 (2)
C2A—C1A—C7A	120.9 (2)	C6B—C1B—C7B	121.68 (19)
C6A—C1A—C7A	119.81 (19)	C2B—C1B—C7B	119.12 (19)
C1A—C2A—C3A	121.4 (2)	C1B—C2B—C3B	121.9 (2)
C1A—C2A—H2A	119.3	C1B—C2B—H2B	119.1
C3A—C2A—H2A	119.3	C3B—C2B—H2B	119.1
C2A—C3A—C4A	119.95 (19)	C2B—C3B—C4B	119.32 (19)
C2A—C3A—N1A	116.78 (19)	C2B—C3B—N1B	117.30 (19)
C4A—C3A—N1A	123.26 (19)	C4B—C3B—N1B	123.4 (2)
C5A—C4A—C3A	117.0 (2)	C5B—C4B—C3B	117.2 (2)

## supplementary materials

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C5A—C4A—H4A	121.5	C5B—C4B—H4B	121.4
C3A—C4A—H4A	121.5	C3B—C4B—H4B	121.4
C6A—C5A—C4A	124.3 (2)	C6B—C5B—C4B	124.4 (2)
C6A—C5A—N2A	118.23 (19)	C6B—C5B—N2B	118.33 (19)
C4A—C5A—N2A	117.5 (2)	C4B—C5B—N2B	117.3 (2)
C5A—C6A—C1A	118.21 (19)	C5B—C6B—C1B	118.1 (2)
C5A—C6A—H6A	120.9	C5B—C6B—H6B	121.0
C1A—C6A—H6A	120.9	C1B—C6B—H6B	121.0
N3A—C7A—C1A	115.71 (17)	N3B—C7B—C1B	110.62 (17)
N3A—C7A—H7A1	108.4	N3B—C7B—H7B1	109.5
C1A—C7A—H7A1	108.4	C1B—C7B—H7B1	109.5
N3A—C7A—H7A2	108.4	N3B—C7B—H7B2	109.5
C1A—C7A—H7A2	108.4	C1B—C7B—H7B2	109.5
H7A1—C7A—H7A2	107.4	H7B1—C7B—H7B2	108.1
O1A—C8A—N1A	122.1 (2)	O1B—C8B—N1B	122.3 (2)
O1A—C8A—C9A	120.6 (2)	O1B—C8B—C9B	121.2 (2)
N1A—C8A—C9A	117.29 (19)	N1B—C8B—C9B	116.54 (19)
C10A—C9A—C14A	118.73 (19)	C10B—C9B—C14B	119.3 (2)
C10A—C9A—C8A	124.1 (2)	C10B—C9B—C8B	124.6 (2)
C14A—C9A—C8A	117.1 (2)	C14B—C9B—C8B	116.1 (2)
C11A—C10A—C9A	120.1 (2)	C11B—C10B—C9B	120.1 (2)
C11A—C10A—H10A	119.9	C11B—C10B—H10B	119.9
C9A—C10A—H10A	119.9	C9B—C10B—H10B	119.9
C10A—C11A—C12A	120.8 (2)	C12B—C11B—C10B	120.4 (2)
C10A—C11A—H11A	119.6	C12B—C11B—H11B	119.8
C12A—C11A—H11A	119.6	C10B—C11B—H11B	119.8
C13A—C12A—C11A	119.4 (2)	C11B—C12B—C13B	119.8 (2)
C13A—C12A—H12A	120.3	C11B—C12B—H12B	120.1
C11A—C12A—H12A	120.3	C13B—C12B—H12B	120.1
C12A—C13A—C14A	120.3 (2)	C12B—C13B—C14B	120.1 (2)
C12A—C13A—H13A	119.9	C12B—C13B—H13B	119.9
C14A—C13A—H13A	119.9	C14B—C13B—H13B	119.9
C13A—C14A—C9A	120.6 (2)	C13B—C14B—C9B	120.2 (2)
C13A—C14A—H14A	119.7	C13B—C14B—H14B	119.9
C9A—C14A—H14A	119.7	C9B—C14B—H14B	119.9
N3A—C15A—C16A	111.96 (18)	N3B—C15B—C16B	110.18 (19)
N3A—C15A—H15A	109.2	N3B—C15B—H15C	109.6
C16A—C15A—H15A	109.2	C16B—C15B—H15C	109.6
N3A—C15A—H15B	109.2	N3B—C15B—H15D	109.6
C16A—C15A—H15B	109.2	C16B—C15B—H15D	109.6
H15A—C15A—H15B	107.9	H15C—C15B—H15D	108.1
O4A—C16A—C15A	111.69 (19)	O4B—C16B—C15B	110.8 (2)
O4A—C16A—H16A	109.3	O4B—C16B—H16C	109.5
C15A—C16A—H16A	109.3	C15B—C16B—H16C	109.5
O4A—C16A—H16B	109.3	O4B—C16B—H16D	109.5
C15A—C16A—H16B	109.3	C15B—C16B—H16D	109.5
H16A—C16A—H16B	107.9	H16C—C16B—H16D	108.1
N3A—C17A—C18A	115.01 (19)	N3B—C17B—C18B	110.30 (18)
N3A—C17A—H17A	108.5	N3B—C17B—H17C	109.6

C18A—C17A—H17A	108.5	C18B—C17B—H17C	109.6
N3A—C17A—H17B	108.5	N3B—C17B—H17D	109.6
C18A—C17A—H17B	108.5	C18B—C17B—H17D	109.6
H17A—C17A—H17B	107.5	H17C—C17B—H17D	108.1
O5A—C18A—C17A	113.93 (18)	O5B—C18B—C17B	109.84 (18)
O5A—C18A—H18A	108.8	O5B—C18B—H18C	109.7
C17A—C18A—H18A	108.8	C17B—C18B—H18C	109.7
O5A—C18A—H18B	108.8	O5B—C18B—H18D	109.7
C17A—C18A—H18B	108.8	C17B—C18B—H18D	109.7
H18A—C18A—H18B	107.7	H18C—C18B—H18D	108.2

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O4A—HO4A $\cdots$ O1A <sup>i</sup>	0.93 (4)	1.82 (4)	2.736 (2)	168 (3)
O5A—HO5A $\cdots$ O4B	0.89 (3)	1.85 (3)	2.738 (2)	178 (3)
N1A—H1A $\cdots$ O5B	0.86	2.27	3.089 (2)	159
O5B—HO5B $\cdots$ O4B	0.90 (4)	2.30 (4)	3.130 (3)	153 (3)
O4B—HO4B $\cdots$ O4A	1.03 (4)	1.75 (4)	2.762 (3)	169 (3)
N1B—H1B $\cdots$ O5A	0.86	2.49	3.332 (2)	167

Symmetry codes: (i)  $x, y-1, z$ .

