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N-(2-Methoxyethyl)phthalimide

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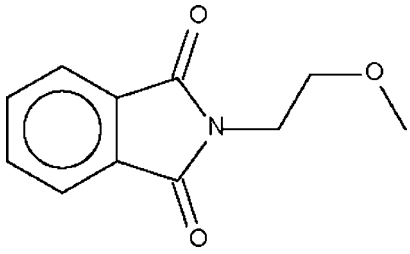
Key indicators: single-crystal X-ray study; T = 100 K; mean σ(C–C) = 0.002 Å; R factor = 0.060; wR factor = 0.206; data-to-parameter ratio = 15.1.

The title molecule, $C_{11}H_{11}NO_3$, lies on a crystallographic mirror plane which bisects the plane of the phthalimide unit and contains the C and O atoms of the 2-methoxyethyl group.

Related literature

For medicinal properties of the title compound, see: Chapman *et al.* (1989); Hall *et al.* (1994). For a kinetic study of the reaction that yields the title compound, see: Khan (1994).



Experimental

Crystal data

$C_{11}H_{11}NO_3$ M_r = 205.21
Orthorhombic, *Pnma*
a = 7.0514 (2) Å
b = 9.3852 (2) Å
c = 14.6024 (4) Å
V = 966.37 (4) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.10 mm^{-1}
T = 100 (2) K
0.30 × 0.20 × 0.10 mm

Data collection

Bruker SMART APEX diffractometer
Absorption correction: none
7349 measured reflections
1164 independent reflections
986 reflections with $I > 2\sigma(I)$
R_{int} = 0.039

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.059
$wR(F^2)$ = 0.206
S = 1.11
1164 reflections
77 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.50 e Å^{-3}
$\Delta\rho_{min}$ = −0.50 e Å^{-3}

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2624).

References

Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Chapman, J. M., Sowell, J. W., Abdalla, G., Hall, I. H. & Wong, O. T. (1989). *J. Pharm. Sci.* **78**, 903–909.
Hall, I. H., Chapman, J. M. & Wong, O. T. (1994). *Anti-Cancer Drugs*, **5**, 75–82.
Khan, M. N. (1994). *Indian J. Chem.* B**33**, 646–650.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Westrip, S. P. (2008). *publCIF*. In preparation.

supporting information

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S1. Comment

The title compound was previously reported in a kinetic study (Khan, 1994). We intend to carry out studies on the medicinal properties of the compound; some such properties have been reported (Chapman *et al.*, 1989; Hall *et al.*, 1994). The molecule of *N*-(2-methoxyethyl)phthalimide lies on a mirror plane that relates one half of the phthalamido portion of the molecule to the other; the 2-methoxyethyl substituent lies on the mirror plane itself (Fig. 1).

S2. Experimental

Phthalic anhydride (2.59 g, 17.5 mmol) and 2-methoxyethylamine (1.50 ml, 17.5 mmol) were dissolved in acetic acid (25 ml). The mixture was heated at 393–413 K for 4 h; the reaction was monitored by TLC. Water was added to precipitate the product, which was collected (80% yield.) Crystals were obtained upon recrystallization from water.

S3. Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 0.99 Å) and were included in the refinement in the riding model approximation, with *U*(H) set to 1.2–1.5 *U*(C).

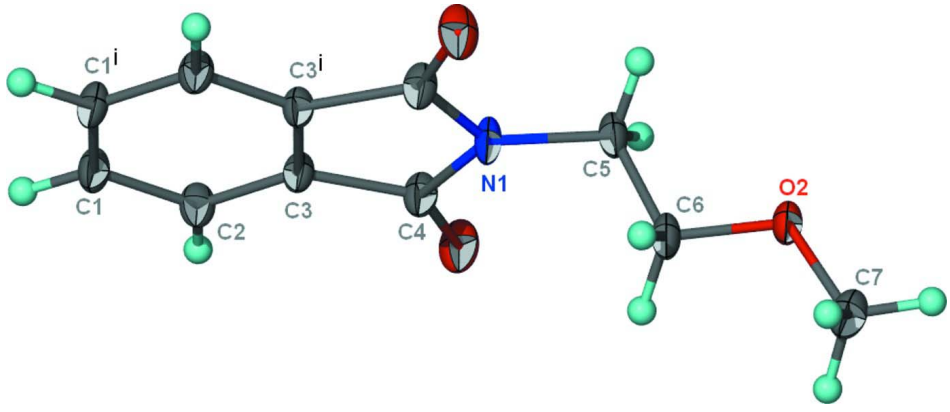

Figure 1

Thermal ellipsoid plot (Barbour, 2001) of $C_{11}H_{11}NO_3$ at the 70% probability level. Hydrogen atoms are drawn as spheres of arbitrary radiius. Symmetry transformation (i): *x*, 1/2 – *y*, *z*.

N-(2-Methoxyethyl)phthalimide

Crystal data

$C_{11}H_{11}NO_3$
M_r = 205.21
Orthorhombic, *Pnma*
Hall symbol: -P 2ac 2n
a = 7.0514 (2) Å
b = 9.3852 (2) Å

c = 14.6024 (4) Å
V = 966.37 (4) Å^3
Z = 4
F(000) = 432
D_x = 1.410 Mg m^{-3}
Mo $K\alpha$ radiation, λ = 0.71073 Å

Cell parameters from 2463 reflections
θ = 2.6–28.3°
μ = 0.10 mm^{-1}
T = 100 K
Prism, colorless
0.30 × 0.20 × 0.10 mm

Data collection

Bruker SMART APEX diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
ω scans
7349 measured reflections
1164 independent reflections

986 reflections with $I > 2\sigma(I)$
R_{int} = 0.039
θ_{max} = 27.5°, θ_{min} = 2.6°
h = −8→9
k = −12→12
l = −12→18

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.059
$wR(F^2)$ = 0.206
S = 1.11
1164 reflections
77 parameters
0 restraints
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.1433P)^2 + 0.309P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max}$ = 0.001
$\Delta\rho_{max}$ = 0.50 e Å^{-3}
$\Delta\rho_{min}$ = −0.50 e Å^{-3}

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	U_{iso}*/U_{eq}	Occ. (<1)
O1	0.2012 (2)	0.00647 (14)	0.60172 (9)	0.0229 (5)	
O2	0.3277 (3)	0.2500	0.33669 (12)	0.0161 (5)	
N1	0.2007 (4)	0.2500	0.57847 (14)	0.0167 (6)	
C1	0.2372 (3)	0.17575 (18)	0.89087 (12)	0.0176 (5)	
H1	0.2434	0.1260	0.9475	0.021*	
C2	0.2281 (3)	0.09866 (19)	0.80904 (12)	0.0168 (5)	
H2	0.2280	−0.0026	0.8088	0.020*	
C3	0.2194 (3)	0.17598 (18)	0.72848 (11)	0.0144 (5)	
C4	0.2072 (3)	0.12658 (18)	0.63197 (13)	0.0171 (5)	
C5	0.1763 (4)	0.2500	0.47961 (16)	0.0176 (6)	
H5A	0.1034	0.3354	0.4610	0.021*	0.50
H5B	0.1034	0.1646	0.4610	0.021*	0.50
C6	0.3665 (4)	0.2500	0.43155 (16)	0.0175 (6)	
H6A	0.4404	0.3357	0.4486	0.021*	0.50
H6B	0.4404	0.1643	0.4486	0.021*	0.50
C7	0.4974 (4)	0.2500	0.28287 (17)	0.0210 (7)	
H7A	0.4642	0.2500	0.2177	0.031*	
H7B	0.5721	0.1647	0.2970	0.031*	0.50
H7C	0.5721	0.3353	0.2970	0.031*	0.50

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0377 (10)	0.0142 (8)	0.0166 (8)	0.0009 (6)	−0.0011 (6)	−0.0037 (5)
O2	0.0210 (11)	0.0178 (9)	0.0093 (9)	0.000	0.0003 (7)	0.000
N1	0.0282 (14)	0.0143 (11)	0.0075 (10)	0.000	0.0000 (8)	0.000
C1	0.0240 (10)	0.0185 (10)	0.0102 (9)	0.0008 (7)	0.0016 (6)	0.0018 (6)
C2	0.0237 (11)	0.0134 (8)	0.0133 (9)	−0.0006 (7)	0.0001 (7)	0.0017 (6)
C3	0.0189 (10)	0.0142 (9)	0.0101 (9)	0.0001 (7)	0.0003 (6)	−0.0012 (6)
C4	0.0258 (11)	0.0133 (9)	0.0121 (9)	0.0011 (7)	0.0006 (7)	0.0004 (6)
C5	0.0220 (14)	0.0216 (12)	0.0091 (12)	0.000	−0.0024 (9)	0.000
C6	0.0239 (15)	0.0197 (11)	0.0088 (12)	0.000	−0.0011 (9)	0.000
C7	0.0284 (17)	0.0196 (12)	0.0149 (12)	0.000	0.0043 (11)	0.000

Geometric parameters (Å, º)

O1—C4	1.211 (2)	C3—C3^{i}	1.389 (3)
O2—C6	1.412 (3)	C3—C4	1.486 (2)
O2—C7	1.432 (3)	C5—C6	1.514 (4)
N1—C4	1.398 (2)	C5—H5A	0.9900
N1—C4^{i}	1.398 (2)	C5—H5B	0.9900
N1—C5	1.454 (3)	C6—H6A	0.9900
C1—C1^{i}	1.394 (3)	C6—H6B	0.9900
C1—C2	1.398 (2)	C7—H7A	0.9800
C1—H1	0.9500	C7—H7B	0.9800
C2—C3	1.384 (2)	C7—H7C	0.9800
C2—H2	0.9500		
C6—O2—C7	112.1 (2)	N1—C5—H5A	109.5
C4—N1—C4^{i}	111.9 (2)	C6—C5—H5A	109.5
C4—N1—C5	123.97 (11)	N1—C5—H5B	109.5
C4^{i}—N1—C5	123.97 (11)	C6—C5—H5B	109.5
C1^{i}—C1—C2	121.16 (10)	H5A—C5—H5B	108.1
C1^{i}—C1—H1	119.4	O2—C6—C5	106.5 (2)
C2—C1—H1	119.4	O2—C6—H6A	110.4
C3—C2—C1	117.21 (17)	C5—C6—H6A	110.4
C3—C2—H2	121.4	O2—C6—H6B	110.4
C1—C2—H2	121.4	C5—C6—H6B	110.4
C2—C3—C3^{i}	121.63 (11)	H6A—C6—H6B	108.6
C2—C3—C4	130.19 (16)	O2—C7—H7A	109.5
C3^{i}—C3—C4	108.18 (9)	O2—C7—H7B	109.5
O1—C4—N1	124.48 (17)	H7A—C7—H7B	109.5
O1—C4—C3	129.64 (16)	O2—C7—H7C	109.5
N1—C4—C3	105.87 (15)	H7A—C7—H7C	109.5
N1—C5—C6	110.8 (2)	H7B—C7—H7C	109.5
C1^{i}—C1—C2—C3	0.1 (2)	C3^{i}—C3—C4—O1	−179.02 (19)
C1—C2—C3—C3^{i}	−0.1 (2)	C2—C3—C4—N1	179.5 (2)

C1—C2—C3—C4	−179.47 (19)	C3^{i}—C3—C4—N1	0.07 (17)
C4^{i}—N1—C4—O1	179.03 (13)	C4—N1—C5—C6	−92.3 (2)
C5—N1—C4—O1	3.2 (4)	C4^{i}—N1—C5—C6	92.3 (2)
C4^{i}—N1—C4—C3	−0.1 (3)	C7—O2—C6—C5	180.0
C5—N1—C4—C3	−176.0 (2)	N1—C5—C6—O2	180.0
C2—C3—C4—O1	0.5 (4)		

Symmetry code: (i) x, $-y+1/2$, z.