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Tetrakis(4-aminopyridine- κ N¹)-dichloridocopper(II) monohydrate

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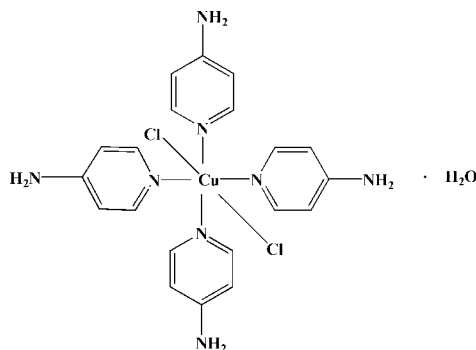
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.042; wR factor = 0.104; data-to-parameter ratio = 41.3.

The asymmetric unit of the title compound, $[\text{CuCl}_2(\text{C}_5\text{H}_6\text{N}_2)_4]\cdot\text{H}_2\text{O}$, contains two crystallographically independent complex molecules and two water molecules. The Cu^{II} ion in each molecule is six-coordinated in an elongated octahedral geometry, with the equatorial plane defined by four pyridine N atoms of four aminopyridine ligands and the axial positions occupied by two Cl atoms. In the crystal structure, molecules are linked into a three-dimensional framework by $\text{C}-\text{H}\cdots\text{Cl}$, $\text{O}-\text{H}\cdots\text{Cl}$, $\text{N}-\text{H}\cdots\text{O}$, $\text{N}-\text{H}\cdots\text{Cl}$ and $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds and $\text{C}/\text{N}-\text{H}\cdots\pi$ interactions involving the pyridine rings.

Related literature

For related literature on 4-aminopyridine, see: Judge & Bever (2006); Schwid *et al.* (1997); Strupp *et al.* (2004). For bond lengths, see: Moncol *et al.* (2004); Zaleski *et al.* (2005); Anderson *et al.* (2005).



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Experimental

Crystal data

$[\text{CuCl}_2(\text{C}_5\text{H}_6\text{N}_2)_4]\cdot\text{H}_2\text{O}$
 $M_r = 528.93$
 Triclinic, $P\bar{1}$
 $a = 9.5430$ (2) Å
 $b = 14.1606$ (2) Å
 $c = 17.4662$ (3) Å
 $\alpha = 88.463$ (1)°
 $\beta = 86.075$ (1)°

$\gamma = 85.781$ (1)°
 $V = 2347.81$ (7) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 1.19$ mm⁻¹
 $T = 100.0$ (1) K
 $0.51 \times 0.40 \times 0.12$ mm

Data collection

Bruker SMART APEXII CCD
 area-detector diffractometer
 Absorption correction: multi-scan
 (SADABS; Bruker, 2005)
 $T_{\text{min}} = 0.583$, $T_{\text{max}} = 0.871$

106511 measured reflections
 24484 independent reflections
 17417 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.047$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.104$
 $S = 1.07$
 24484 reflections
 593 parameters
 6 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.92$ e Å⁻³
 $\Delta\rho_{\text{min}} = -1.11$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$\text{Cg}1$, $\text{Cg}2$ and $\text{Cg}3$ are centroids of the $\text{N}1/\text{C}1-\text{C}5$, $\text{N}9/\text{C}21-\text{C}25$ and $\text{N}11/\text{C}26-\text{C}30$ rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}2-\text{H}2\cdots\text{Cl}2^{\text{i}}$	0.93	2.68	3.5955 (13)	167
$\text{C}6-\text{H}6\cdots\text{Cl}2$	0.93	2.80	3.3665 (13)	121
$\text{C}10-\text{H}10\cdots\text{Cl}1$	0.93	2.78	3.4430 (13)	129
$\text{C}20-\text{H}20\cdots\text{Cl}1$	0.93	2.64	3.3522 (13)	134
$\text{C}25-\text{H}25\cdots\text{Cl}4$	0.93	2.71	3.2928 (13)	121
$\text{C}26-\text{H}26\cdots\text{N}9$	0.93	2.62	3.0473 (17)	108
$\text{C}35-\text{H}35\cdots\text{Cl}4$	0.93	2.67	3.3947 (14)	136
$\text{N}4-\text{H}4\text{A}\cdots\text{O}1\text{W}^{\text{ii}}$	0.86	2.38	3.2094 (17)	163
$\text{N}4-\text{H}4\text{B}\cdots\text{Cl}1^{\text{iii}}$	0.86	2.43	3.2893 (13)	175
$\text{N}6-\text{H}6\text{A}\cdots\text{Cl}4^{\text{iv}}$	0.86	2.82	3.4066 (13)	127
$\text{N}8-\text{H}8\text{B}\cdots\text{Cl}2^{\text{iv}}$	0.86	2.56	3.4021 (13)	166
$\text{N}10-\text{H}10\text{B}\cdots\text{Cl}3^{\text{v}}$	0.86	2.41	3.2596 (11)	171
$\text{N}12-\text{H}12\text{A}\cdots\text{O}1\text{W}$	0.86	2.06	2.8908 (16)	162
$\text{N}14-\text{H}14\text{B}\cdots\text{O}1\text{W}^{\text{vi}}$	0.86	2.25	3.0103 (18)	147
$\text{N}16-\text{H}16\text{B}\cdots\text{N}16^{\text{vii}}$	0.86	2.52	3.2036 (17)	137
$\text{O}1\text{W}-\text{H}1\text{W}1\cdots\text{Cl}1^{\text{iii}}$	0.83 (1)	2.26 (1)	3.0614 (12)	163 (2)
$\text{O}1\text{W}-\text{H}2\text{W}1\cdots\text{Cl}4^{\text{viii}}$	0.83 (1)	2.26 (1)	3.0694 (12)	164 (2)
$\text{C}37-\text{H}37\cdots\text{Cg}1^{\text{ix}}$	0.93	2.81	3.5492 (14)	137
$\text{C}39-\text{H}39\cdots\text{Cg}1^{\text{x}}$	0.93	2.95	3.7306 (14)	142
$\text{N}2-\text{H}2\text{B}\cdots\text{Cg}2^{\text{xi}}$	0.86	2.75	3.3359 (13)	126
$\text{C}12-\text{H}12\cdots\text{Cg}3$	0.93	2.79	3.6488 (14)	154
$\text{C}14-\text{H}14\cdots\text{Cg}3^{\text{iv}}$	0.93	2.83	3.6193 (13)	144

Symmetry codes: (i) $-x + 1, -y, -z$; (ii) $-x + 2, -y, -z + 1$; (iii) $-x + 1, -y, -z + 1$; (iv) $x - 1, y, z$; (v) $-x + 2, -y + 1, -z$; (vi) $-x + 1, -y + 1, -z + 1$; (vii) $-x + 2, -y + 2, -z$; (viii) $-x + 2, -y + 1, -z + 1$; (ix) $x, y + 1, z$; (x) $x + 1, y + 1, z$; (xi) $x - 1, y - 1, z$.

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2603).

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supplementary materials

Acta Cryst. (2008). E64, m853-m854 [doi:10.1107/S1600536808015778]

Tetrakis(4-aminopyridine- κN^1)dichloridocopper(II) monohydrate

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Comment

4-Aminopyridine (Fampridine) is used clinically in Lambert–Eaton myasthenic syndrome and multiple sclerosis because by blocking potassium channels it prolongs action potentials thereby increasing transmitter release at the neuromuscular junctions (Judge & Bever, 2006; Schwid *et al.*, 1997; Strupp *et al.*, 2004).

The asymmetric unit of the title compound contains two crystallographically independent complex molecules and two water molecules (Fig. 1). The Cu^{II} ion in each molecule is six-coordinated in an elongated octahedral geometry formed by four pyridine N atoms from four aminopyridine ligands and two Cl atoms. The two Cl atoms are located at apical positions and the four N atoms form the basal plane. The Jahn–Teller elongation observed in Cu1—Cl1 [3.2185 (4) Å] and Cu2—Cl3 [3.1884 (4) Å] distances are consistent with those reported earlier (Moncol *et al.*, 2004). The average Cu—N bond length of 2.0205 (11) Å agree well with that reported for a copper complex (Zaleski *et al.*, 2005). The bond lengths and angles in the 4-aminopyridine units agree well with those reported earlier (Anderson *et al.*, 2005).

The crystal packing is consolidated by intermolecular O—H \cdots Cl, N—H \cdots O, N—H \cdots N, N—H \cdots Cl and C—H \cdots Cl hydrogen bonds, and and C/N—H \cdots π interactions (Table 2) involving the pyridine rings to form a three-dimensional framework.

Experimental

A solution of 4-aminopyridine (0.376 g) in water (20 ml) was added to a solution of CuCl₂·2H₂O (0.170 g) in water (20 ml) and the mixture was stirred at 303 K for 6 h. The clear blue solution obtained was filtered and allowed to evaporate slowly. Blue crystals of the title compound were obtained after a month.

Refinement

H atoms of the water molecules were located in a difference map and refined with O—H and H \cdots H distance restraints of 0.84 (1) and 1.37 (2) Å, respectively. The remaining H atoms were positioned geometrically [C—H = 0.93 Å and N—H = 0.86 Å] and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$.

Figures

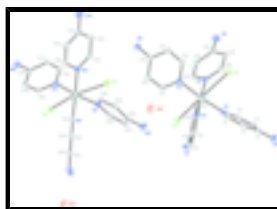


Fig. 1. The asymmetric unit of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme. Hydrogen atoms are omitted for clarity.

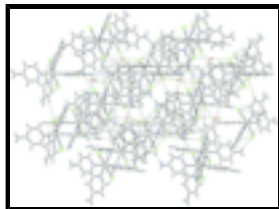


Fig. 2. The crystal packing of the title compound, viewed along the *c* axis. Hydrogen bonds are shown as dashed lines.

Tetrakis(4-aminopyridine- κ N¹)dichloridocopper(II) monohydrate

Crystal data

[CuCl₂(C₅H₆N₂)₄]·H₂O

M_r = 528.93

Triclinic, *P* $\bar{1}$

Hall symbol: -P 1

a = 9.5430 (2) Å

b = 14.1606 (2) Å

c = 17.4662 (3) Å

α = 88.463 (1)°

β = 86.075 (1)°

γ = 85.781 (1)°

V = 2347.81 (7) Å³

Z = 4

*F*₀₀₀ = 1092

D_x = 1.496 Mg m⁻³

Mo *K* α radiation

λ = 0.71073 Å

Cell parameters from 9885 reflections

θ = 2.5–34.6°

μ = 1.19 mm⁻¹

T = 100.0 (1) K

Plate, purple

0.51 × 0.40 × 0.12 mm

Data collection

Bruker SMART APEXII CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

T = 100.0(1) K

φ and ω scans

Absorption correction: multi-scan (SADABS; Bruker, 2005)

*T*_{min} = 0.583, *T*_{max} = 0.871

106511 measured reflections

24484 independent reflections

17417 reflections with *I* > 2 σ (*I*)

*R*_{int} = 0.047

θ _{max} = 37.5°

θ _{min} = 1.2°

h = -16→15

k = -23→24

l = -29→29

Refinement

Refinement on *F*²

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.041$

wR(*F*²) = 0.104

S = 1.07

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0446P)^2 + 0.3682P]$$

where $P = (F_o^2 + 2F_c^2)/3$

(Δ/σ)_{max} = 0.001

24484 reflections $\Delta\rho_{\max} = 0.92 \text{ e } \text{\AA}^{-3}$
 593 parameters $\Delta\rho_{\min} = -1.11 \text{ e } \text{\AA}^{-3}$
 6 restraints Extinction correction: none
 Primary atom site location: structure-invariant direct methods

Special details

Experimental. The data was collected with the Oxford Cyrosystem Cobra low-temperature attachment.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cu1	0.438175 (17)	0.078184 (11)	0.230903 (9)	0.01546 (4)
N1	0.42808 (11)	-0.05177 (8)	0.18839 (6)	0.01482 (18)
N2	0.37359 (14)	-0.31516 (8)	0.09807 (7)	0.0227 (2)
H2A	0.3669	-0.3204	0.0495	0.027*
H2B	0.3679	-0.3641	0.1281	0.027*
N3	0.58224 (11)	0.02793 (7)	0.30393 (6)	0.01460 (18)
N4	0.88631 (12)	-0.05810 (9)	0.45634 (7)	0.0204 (2)
H4A	0.9675	-0.0825	0.4396	0.025*
H4B	0.8678	-0.0508	0.5048	0.025*
N5	0.42913 (11)	0.20525 (7)	0.28145 (6)	0.01477 (18)
N6	0.40879 (14)	0.45962 (9)	0.39775 (8)	0.0280 (3)
H6A	0.3293	0.4829	0.4178	0.034*
H6B	0.4840	0.4882	0.4026	0.034*
N7	0.27731 (11)	0.12641 (8)	0.16710 (6)	0.01593 (19)
N8	-0.06992 (13)	0.23308 (10)	0.05134 (8)	0.0276 (3)
H8A	-0.0617	0.2769	0.0167	0.033*
H8B	-0.1504	0.2106	0.0623	0.033*
C1	0.41925 (13)	-0.06352 (9)	0.11226 (7)	0.0155 (2)
H1	0.4246	-0.0105	0.0799	0.019*
C2	0.40285 (13)	-0.14884 (9)	0.07986 (7)	0.0154 (2)
H2	0.3983	-0.1527	0.0270	0.018*
C3	0.39299 (13)	-0.23054 (9)	0.12703 (7)	0.0160 (2)
C4	0.40297 (14)	-0.21872 (9)	0.20635 (7)	0.0172 (2)
H4	0.3979	-0.2704	0.2401	0.021*
C5	0.42019 (13)	-0.13030 (9)	0.23353 (7)	0.0162 (2)
H5	0.4268	-0.1243	0.2860	0.019*

supplementary materials

C6	0.70980 (14)	-0.01180 (9)	0.27928 (7)	0.0173 (2)
H6	0.7287	-0.0193	0.2268	0.021*
C7	0.81305 (14)	-0.04176 (10)	0.32694 (7)	0.0180 (2)
H7	0.8989	-0.0688	0.3067	0.022*
C8	0.78850 (13)	-0.03145 (9)	0.40679 (7)	0.0144 (2)
C9	0.65482 (13)	0.00904 (9)	0.43265 (7)	0.0158 (2)
H9	0.6324	0.0169	0.4849	0.019*
C10	0.55757 (13)	0.03684 (9)	0.38032 (7)	0.0159 (2)
H10	0.4700	0.0632	0.3987	0.019*
C11	0.54513 (13)	0.25191 (9)	0.28928 (8)	0.0175 (2)
H11	0.6310	0.2255	0.2685	0.021*
C12	0.54381 (14)	0.33629 (9)	0.32624 (8)	0.0190 (2)
H12	0.6270	0.3657	0.3300	0.023*
C13	0.41553 (14)	0.37788 (9)	0.35848 (8)	0.0176 (2)
C14	0.29433 (13)	0.32995 (9)	0.34973 (8)	0.0168 (2)
H14	0.2069	0.3549	0.3696	0.020*
C15	0.30574 (13)	0.24571 (9)	0.31143 (7)	0.0159 (2)
H15	0.2242	0.2151	0.3059	0.019*
C16	0.28786 (14)	0.19626 (9)	0.11362 (8)	0.0180 (2)
H16	0.3750	0.2209	0.1035	0.022*
C17	0.17711 (14)	0.23325 (9)	0.07308 (8)	0.0190 (2)
H17	0.1910	0.2803	0.0357	0.023*
C18	0.04303 (13)	0.19955 (10)	0.08841 (8)	0.0179 (2)
C19	0.03170 (14)	0.12759 (10)	0.14505 (7)	0.0180 (2)
H19	-0.0549	0.1034	0.1581	0.022*
C20	0.14907 (14)	0.09318 (9)	0.18106 (7)	0.0175 (2)
H20	0.1395	0.0442	0.2171	0.021*
Cu2	0.926147 (17)	0.588239 (11)	0.265503 (9)	0.01796 (4)
N9	1.07098 (11)	0.53919 (8)	0.18340 (6)	0.01519 (19)
N10	1.36478 (12)	0.42921 (8)	0.01531 (6)	0.0183 (2)
H10A	1.4401	0.3975	0.0283	0.022*
H10B	1.3490	0.4372	-0.0324	0.022*
N11	0.91724 (11)	0.45781 (8)	0.31403 (6)	0.01547 (19)
N12	0.91541 (14)	0.18712 (8)	0.41402 (7)	0.0248 (3)
H12A	0.9061	0.1801	0.4631	0.030*
H12B	0.9251	0.1382	0.3855	0.030*
N13	0.77568 (12)	0.64322 (8)	0.34276 (6)	0.0170 (2)
N14	0.45970 (14)	0.77011 (10)	0.49076 (7)	0.0262 (3)
H14A	0.4812	0.8030	0.5284	0.031*
H14B	0.3728	0.7633	0.4835	0.031*
N15	0.91270 (11)	0.71282 (8)	0.20637 (6)	0.01664 (19)
N16	0.89456 (13)	0.95381 (8)	0.06323 (7)	0.0204 (2)
H16A	0.8173	0.9718	0.0428	0.024*
H16B	0.9680	0.9850	0.0537	0.024*
C21	1.05091 (13)	0.55075 (9)	0.10788 (7)	0.0157 (2)
H21	0.9680	0.5832	0.0939	0.019*
C22	1.14589 (13)	0.51732 (9)	0.05055 (7)	0.0156 (2)
H22	1.1273	0.5283	-0.0006	0.019*
C23	1.27166 (12)	0.46630 (9)	0.06954 (7)	0.0141 (2)

C24	1.29340 (13)	0.45489 (9)	0.14834 (7)	0.0164 (2)
H24	1.3753	0.4228	0.1642	0.020*
C25	1.19284 (14)	0.49147 (9)	0.20156 (7)	0.0167 (2)
H25	1.2094	0.4830	0.2532	0.020*
C26	0.92843 (13)	0.37962 (9)	0.27115 (7)	0.0165 (2)
H26	0.9361	0.3873	0.2180	0.020*
C27	0.92913 (14)	0.28925 (9)	0.30160 (7)	0.0171 (2)
H27	0.9389	0.2377	0.2693	0.021*
C28	0.91497 (13)	0.27475 (9)	0.38181 (7)	0.0164 (2)
C29	0.89945 (13)	0.35662 (9)	0.42649 (7)	0.0164 (2)
H29	0.8873	0.3515	0.4797	0.020*
C30	0.90246 (13)	0.44415 (9)	0.39074 (7)	0.0164 (2)
H30	0.8938	0.4973	0.4214	0.020*
C31	0.63836 (14)	0.63335 (10)	0.33431 (8)	0.0185 (2)
H31	0.6150	0.5965	0.2944	0.022*
C32	0.53041 (14)	0.67467 (10)	0.38134 (7)	0.0182 (2)
H32	0.4372	0.6666	0.3723	0.022*
C33	0.56243 (14)	0.72915 (9)	0.44304 (7)	0.0177 (2)
C34	0.70551 (15)	0.73799 (10)	0.45233 (8)	0.0210 (2)
H34	0.7323	0.7723	0.4929	0.025*
C35	0.80628 (14)	0.69599 (10)	0.40169 (8)	0.0196 (2)
H35	0.9004	0.7044	0.4084	0.024*
C36	0.79327 (13)	0.74421 (9)	0.17431 (8)	0.0180 (2)
H36	0.7138	0.7105	0.1845	0.022*
C37	0.78247 (13)	0.82367 (9)	0.12713 (8)	0.0174 (2)
H37	0.6975	0.8426	0.1063	0.021*
C38	0.90066 (13)	0.87553 (9)	0.11090 (7)	0.0160 (2)
C39	1.02509 (14)	0.84304 (9)	0.14474 (8)	0.0181 (2)
H39	1.1061	0.8755	0.1359	0.022*
C40	1.02618 (13)	0.76286 (9)	0.19099 (8)	0.0179 (2)
H40	1.1096	0.7422	0.2127	0.021*
Cl1	0.20362 (3)	0.01980 (2)	0.361362 (18)	0.01910 (6)
Cl2	0.64240 (3)	0.12291 (2)	0.122477 (17)	0.01617 (5)
Cl3	0.69715 (4)	0.51337 (2)	0.161801 (18)	0.02111 (6)
Cl4	1.14798 (3)	0.62545 (2)	0.353735 (19)	0.01962 (6)
O1W	0.82107 (13)	0.18216 (8)	0.57474 (6)	0.0285 (2)
O2W	1.62468 (11)	0.33858 (7)	0.07154 (6)	0.02102 (19)
H1W1	0.810 (2)	0.1333 (8)	0.6007 (10)	0.037 (6)*
H2W1	0.834 (2)	0.2276 (9)	0.6020 (10)	0.039 (6)*
H1W2	1.628 (3)	0.2884 (9)	0.0979 (11)	0.052 (7)*
H2W2	1.648 (2)	0.3826 (10)	0.0992 (11)	0.048 (7)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cu1	0.01684 (7)	0.01299 (7)	0.01708 (7)	0.00168 (5)	-0.00697 (5)	-0.00320 (5)
N1	0.0164 (5)	0.0146 (5)	0.0136 (4)	0.0001 (4)	-0.0033 (3)	-0.0015 (3)
N2	0.0330 (6)	0.0165 (5)	0.0193 (5)	-0.0048 (5)	-0.0016 (5)	-0.0046 (4)

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N3	0.0160 (4)	0.0138 (4)	0.0141 (4)	0.0008 (4)	-0.0033 (3)	-0.0021 (3)
N4	0.0158 (5)	0.0302 (6)	0.0149 (5)	0.0041 (4)	-0.0036 (4)	-0.0019 (4)
N5	0.0137 (4)	0.0139 (5)	0.0170 (5)	0.0003 (4)	-0.0035 (3)	-0.0029 (3)
N6	0.0220 (6)	0.0183 (6)	0.0442 (8)	-0.0008 (5)	-0.0034 (5)	-0.0124 (5)
N7	0.0162 (5)	0.0144 (5)	0.0177 (5)	-0.0003 (4)	-0.0053 (4)	-0.0017 (3)
N8	0.0167 (5)	0.0331 (7)	0.0325 (7)	0.0028 (5)	-0.0063 (5)	0.0052 (5)
C1	0.0161 (5)	0.0158 (5)	0.0147 (5)	-0.0011 (4)	-0.0026 (4)	-0.0002 (4)
C2	0.0153 (5)	0.0182 (5)	0.0128 (5)	-0.0008 (4)	-0.0019 (4)	-0.0026 (4)
C3	0.0152 (5)	0.0155 (5)	0.0174 (5)	-0.0013 (4)	-0.0004 (4)	-0.0027 (4)
C4	0.0210 (6)	0.0149 (5)	0.0156 (5)	-0.0019 (4)	-0.0006 (4)	0.0003 (4)
C5	0.0178 (5)	0.0168 (5)	0.0140 (5)	-0.0011 (4)	-0.0024 (4)	-0.0006 (4)
C6	0.0202 (6)	0.0185 (6)	0.0126 (5)	0.0040 (4)	-0.0020 (4)	-0.0032 (4)
C7	0.0173 (5)	0.0215 (6)	0.0146 (5)	0.0044 (5)	-0.0013 (4)	-0.0035 (4)
C8	0.0145 (5)	0.0150 (5)	0.0139 (5)	-0.0009 (4)	-0.0027 (4)	-0.0010 (4)
C9	0.0151 (5)	0.0194 (6)	0.0129 (5)	-0.0009 (4)	-0.0004 (4)	-0.0016 (4)
C10	0.0142 (5)	0.0170 (5)	0.0164 (5)	0.0001 (4)	-0.0014 (4)	-0.0017 (4)
C11	0.0137 (5)	0.0180 (6)	0.0208 (6)	0.0003 (4)	-0.0023 (4)	-0.0026 (4)
C12	0.0149 (5)	0.0165 (6)	0.0262 (6)	-0.0019 (4)	-0.0033 (4)	-0.0036 (4)
C13	0.0188 (5)	0.0138 (5)	0.0202 (6)	0.0003 (4)	-0.0030 (4)	-0.0021 (4)
C14	0.0146 (5)	0.0153 (5)	0.0203 (6)	0.0006 (4)	-0.0007 (4)	-0.0012 (4)
C15	0.0140 (5)	0.0150 (5)	0.0191 (6)	-0.0007 (4)	-0.0032 (4)	-0.0013 (4)
C16	0.0153 (5)	0.0155 (5)	0.0238 (6)	-0.0032 (4)	-0.0044 (4)	0.0006 (4)
C17	0.0177 (5)	0.0158 (6)	0.0239 (6)	-0.0009 (4)	-0.0051 (5)	0.0024 (4)
C18	0.0144 (5)	0.0191 (6)	0.0201 (6)	0.0029 (4)	-0.0045 (4)	-0.0034 (4)
C19	0.0156 (5)	0.0224 (6)	0.0163 (6)	-0.0037 (4)	-0.0010 (4)	-0.0027 (4)
C20	0.0193 (6)	0.0189 (6)	0.0149 (5)	-0.0032 (4)	-0.0023 (4)	-0.0010 (4)
Cu2	0.01974 (8)	0.01387 (7)	0.01829 (8)	0.00360 (6)	0.00663 (6)	0.00303 (5)
N9	0.0154 (4)	0.0139 (4)	0.0155 (5)	0.0015 (4)	0.0009 (3)	0.0007 (3)
N10	0.0168 (5)	0.0225 (5)	0.0146 (5)	0.0031 (4)	0.0012 (4)	-0.0009 (4)
N11	0.0175 (5)	0.0137 (4)	0.0149 (5)	-0.0004 (4)	0.0000 (4)	0.0007 (3)
N12	0.0378 (7)	0.0138 (5)	0.0215 (6)	0.0015 (5)	0.0031 (5)	0.0021 (4)
N13	0.0177 (5)	0.0149 (5)	0.0174 (5)	0.0012 (4)	0.0034 (4)	0.0004 (4)
N14	0.0234 (6)	0.0300 (7)	0.0239 (6)	0.0050 (5)	0.0040 (5)	-0.0088 (5)
N15	0.0157 (5)	0.0145 (5)	0.0186 (5)	0.0015 (4)	0.0030 (4)	0.0015 (4)
N16	0.0218 (5)	0.0163 (5)	0.0226 (6)	-0.0012 (4)	0.0002 (4)	0.0039 (4)
C21	0.0144 (5)	0.0153 (5)	0.0172 (5)	0.0003 (4)	-0.0006 (4)	0.0002 (4)
C22	0.0155 (5)	0.0172 (5)	0.0144 (5)	-0.0011 (4)	-0.0020 (4)	-0.0008 (4)
C23	0.0137 (5)	0.0129 (5)	0.0156 (5)	-0.0017 (4)	0.0002 (4)	-0.0013 (4)
C24	0.0161 (5)	0.0157 (5)	0.0169 (5)	0.0027 (4)	-0.0018 (4)	-0.0002 (4)
C25	0.0192 (5)	0.0156 (5)	0.0147 (5)	0.0021 (4)	0.0000 (4)	0.0006 (4)
C26	0.0176 (5)	0.0172 (5)	0.0144 (5)	-0.0005 (4)	0.0001 (4)	-0.0015 (4)
C27	0.0185 (5)	0.0143 (5)	0.0184 (6)	-0.0003 (4)	0.0003 (4)	-0.0030 (4)
C28	0.0153 (5)	0.0150 (5)	0.0186 (6)	-0.0003 (4)	-0.0005 (4)	0.0009 (4)
C29	0.0185 (5)	0.0159 (5)	0.0143 (5)	0.0004 (4)	0.0000 (4)	0.0005 (4)
C30	0.0189 (5)	0.0148 (5)	0.0153 (5)	0.0002 (4)	-0.0003 (4)	-0.0013 (4)
C31	0.0201 (6)	0.0193 (6)	0.0156 (6)	-0.0007 (5)	0.0018 (4)	-0.0017 (4)
C32	0.0173 (5)	0.0207 (6)	0.0159 (6)	0.0006 (5)	0.0007 (4)	-0.0008 (4)
C33	0.0207 (6)	0.0152 (5)	0.0161 (5)	0.0018 (4)	0.0029 (4)	-0.0001 (4)
C34	0.0226 (6)	0.0208 (6)	0.0199 (6)	-0.0021 (5)	-0.0011 (5)	-0.0051 (5)

C35	0.0180 (6)	0.0182 (6)	0.0226 (6)	-0.0016 (5)	0.0003 (5)	-0.0008 (4)
C36	0.0145 (5)	0.0154 (5)	0.0237 (6)	-0.0011 (4)	0.0011 (4)	0.0000 (4)
C37	0.0146 (5)	0.0154 (5)	0.0219 (6)	0.0000 (4)	-0.0011 (4)	0.0011 (4)
C38	0.0179 (5)	0.0138 (5)	0.0156 (5)	0.0003 (4)	0.0017 (4)	-0.0004 (4)
C39	0.0160 (5)	0.0172 (6)	0.0210 (6)	-0.0026 (4)	0.0010 (4)	0.0003 (4)
C40	0.0143 (5)	0.0174 (6)	0.0215 (6)	0.0008 (4)	0.0000 (4)	0.0006 (4)
Cl1	0.02220 (14)	0.01924 (14)	0.01585 (13)	-0.00149 (11)	-0.00145 (10)	0.00041 (10)
Cl2	0.01525 (12)	0.01728 (13)	0.01575 (13)	-0.00059 (10)	-0.00079 (9)	0.00145 (9)
Cl3	0.02782 (16)	0.02046 (14)	0.01555 (13)	-0.00381 (12)	-0.00212 (11)	-0.00224 (10)
Cl4	0.01772 (13)	0.02048 (14)	0.02104 (14)	-0.00009 (11)	-0.00386 (10)	-0.00442 (11)
O1W	0.0419 (6)	0.0194 (5)	0.0242 (5)	-0.0066 (5)	0.0045 (5)	-0.0017 (4)
O2W	0.0255 (5)	0.0154 (4)	0.0226 (5)	0.0002 (4)	-0.0064 (4)	-0.0015 (4)

Geometric parameters (Å, °)

Cu1—N1	2.0140 (11)	Cu2—N9	2.0224 (10)
Cu1—N3	2.0177 (10)	Cu2—N13	2.0286 (11)
Cu1—N5	2.0190 (11)	Cu2—Cl4	2.7907 (4)
Cu1—N7	2.0258 (11)	Cu2—Cl3	3.1884 (4)
Cu1—Cl2	2.7199 (3)	N9—C21	1.3498 (16)
Cu1—Cl1	3.2185 (4)	N9—C25	1.3539 (16)
N1—C5	1.3494 (16)	N10—C23	1.3435 (16)
N1—C1	1.3536 (16)	N10—H10A	0.86
N2—C3	1.3431 (17)	N10—H10B	0.86
N2—H2A	0.86	N11—C26	1.3469 (17)
N2—H2B	0.86	N11—C30	1.3478 (16)
N3—C10	1.3466 (16)	N12—C28	1.3485 (17)
N3—C6	1.3506 (16)	N12—H12A	0.86
N4—C8	1.3431 (16)	N12—H12B	0.86
N4—H4A	0.86	N13—C31	1.3465 (18)
N4—H4B	0.86	N13—C35	1.3482 (18)
N5—C11	1.3468 (17)	N14—C33	1.3502 (17)
N5—C15	1.3496 (16)	N14—H14A	0.86
N6—C13	1.3564 (18)	N14—H14B	0.86
N6—H6A	0.86	N15—C36	1.3451 (17)
N6—H6B	0.86	N15—C40	1.3453 (17)
N7—C16	1.3466 (17)	N16—C38	1.3685 (17)
N7—C20	1.3477 (17)	N16—H16A	0.86
N8—C18	1.3455 (17)	N16—H16B	0.86
N8—H8A	0.86	C21—C22	1.3735 (17)
N8—H8B	0.86	C21—H21	0.93
C1—C2	1.3716 (18)	C22—C23	1.4095 (17)
C1—H1	0.93	C22—H22	0.93
C2—C3	1.4073 (18)	C23—C24	1.4092 (18)
C2—H2	0.93	C24—C25	1.3726 (17)
C3—C4	1.4104 (18)	C24—H24	0.93
C4—C5	1.3757 (18)	C25—H25	0.93
C4—H4	0.93	C26—C27	1.3723 (18)
C5—H5	0.93	C26—H26	0.93

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C6—C7	1.3695 (18)	C27—C28	1.4091 (18)
C6—H6	0.93	C27—H27	0.93
C7—C8	1.4080 (17)	C28—C29	1.4080 (18)
C7—H7	0.93	C29—C30	1.3742 (18)
C8—C9	1.4098 (17)	C29—H29	0.93
C9—C10	1.3765 (18)	C30—H30	0.93
C9—H9	0.93	C31—C32	1.3783 (18)
C10—H10	0.93	C31—H31	0.93
C11—C12	1.3721 (19)	C32—C33	1.4050 (19)
C11—H11	0.93	C32—H32	0.93
C12—C13	1.4071 (18)	C33—C34	1.401 (2)
C12—H12	0.93	C34—C35	1.3742 (19)
C13—C14	1.4025 (18)	C34—H34	0.93
C14—C15	1.3770 (18)	C35—H35	0.93
C14—H14	0.93	C36—C37	1.3786 (18)
C15—H15	0.93	C36—H36	0.93
C16—C17	1.3762 (18)	C37—C38	1.3996 (18)
C16—H16	0.93	C37—H37	0.93
C17—C18	1.4044 (19)	C38—C39	1.4043 (19)
C17—H17	0.93	C39—C40	1.3755 (18)
C18—C19	1.4057 (19)	C39—H39	0.93
C19—C20	1.3738 (19)	C40—H40	0.93
C19—H19	0.93	O1W—H1W1	0.827 (9)
C20—H20	0.93	O1W—H2W1	0.833 (9)
Cu2—N11	2.0165 (11)	O2W—H1W2	0.835 (9)
Cu2—N15	2.0206 (11)	O2W—H2W2	0.849 (9)
N1—Cu1—N3	91.10 (4)	N11—Cu2—N9	91.26 (4)
N1—Cu1—N5	173.52 (4)	N15—Cu2—N9	88.17 (4)
N3—Cu1—N5	89.52 (4)	N11—Cu2—N13	91.71 (4)
N1—Cu1—N7	89.32 (4)	N15—Cu2—N13	88.57 (4)
N3—Cu1—N7	173.70 (4)	N9—Cu2—N13	176.35 (4)
N5—Cu1—N7	89.35 (4)	N11—Cu2—Cl4	90.90 (3)
N1—Cu1—Cl2	92.32 (3)	N15—Cu2—Cl4	97.84 (3)
N3—Cu1—Cl2	91.83 (3)	N9—Cu2—Cl4	88.13 (3)
N5—Cu1—Cl2	94.11 (3)	N13—Cu2—Cl4	93.94 (3)
N7—Cu1—Cl2	94.43 (3)	N11—Cu2—Cl3	82.30 (3)
N1—Cu1—Cl1	86.84 (3)	N15—Cu2—Cl3	88.92 (3)
N3—Cu1—Cl1	86.48 (3)	N9—Cu2—Cl3	86.13 (3)
N5—Cu1—Cl1	86.76 (3)	N13—Cu2—Cl3	92.17 (3)
N7—Cu1—Cl1	87.27 (3)	Cl4—Cu2—Cl3	170.991 (11)
Cl2—Cu1—Cl1	178.098 (10)	C21—N9—C25	116.39 (11)
C5—N1—C1	116.35 (11)	C21—N9—Cu2	122.13 (8)
C5—N1—Cu1	122.70 (9)	C25—N9—Cu2	121.47 (9)
C1—N1—Cu1	120.80 (9)	C23—N10—H10A	120.0
C3—N2—H2A	120.0	C23—N10—H10B	120.0
C3—N2—H2B	120.0	H10A—N10—H10B	120.0
H2A—N2—H2B	120.0	C26—N11—C30	116.62 (11)
C10—N3—C6	116.61 (11)	C26—N11—Cu2	121.42 (9)
C10—N3—Cu1	120.95 (8)	C30—N11—Cu2	121.96 (9)

C6—N3—Cu1	122.37 (9)	C28—N12—H12A	120.0
C8—N4—H4A	120.0	C28—N12—H12B	120.0
C8—N4—H4B	120.0	H12A—N12—H12B	120.0
H4A—N4—H4B	120.0	C31—N13—C35	116.72 (11)
C11—N5—C15	116.96 (11)	C31—N13—Cu2	120.80 (9)
C11—N5—Cu1	122.09 (8)	C35—N13—Cu2	122.35 (9)
C15—N5—Cu1	120.91 (9)	C33—N14—H14A	120.0
C13—N6—H6A	120.0	C33—N14—H14B	120.0
C13—N6—H6B	120.0	H14A—N14—H14B	120.0
H6A—N6—H6B	120.0	C36—N15—C40	117.21 (11)
C16—N7—C20	116.54 (11)	C36—N15—Cu2	121.08 (9)
C16—N7—Cu1	123.33 (9)	C40—N15—Cu2	121.39 (9)
C20—N7—Cu1	119.96 (9)	C38—N16—H16A	120.0
C18—N8—H8A	120.0	C38—N16—H16B	120.0
C18—N8—H8B	120.0	H16A—N16—H16B	120.0
H8A—N8—H8B	120.0	N9—C21—C22	123.79 (11)
N1—C1—C2	124.04 (12)	N9—C21—H21	118.1
N1—C1—H1	118.0	C22—C21—H21	118.1
C2—C1—H1	118.0	C21—C22—C23	119.77 (11)
C1—C2—C3	119.61 (11)	C21—C22—H22	120.1
C1—C2—H2	120.2	C23—C22—H22	120.1
C3—C2—H2	120.2	N10—C23—C24	121.81 (11)
N2—C3—C2	121.70 (12)	N10—C23—C22	121.67 (11)
N2—C3—C4	121.76 (12)	C24—C23—C22	116.51 (11)
C2—C3—C4	116.53 (11)	C25—C24—C23	119.55 (11)
C5—C4—C3	119.69 (12)	C25—C24—H24	120.2
C5—C4—H4	120.2	C23—C24—H24	120.2
C3—C4—H4	120.2	N9—C25—C24	123.97 (12)
N1—C5—C4	123.77 (12)	N9—C25—H25	118.0
N1—C5—H5	118.1	C24—C25—H25	118.0
C4—C5—H5	118.1	N11—C26—C27	123.57 (12)
N3—C6—C7	123.97 (12)	N11—C26—H26	118.2
N3—C6—H6	118.0	C27—C26—H26	118.2
C7—C6—H6	118.0	C26—C27—C28	119.91 (12)
C6—C7—C8	119.66 (12)	C26—C27—H27	120.0
C6—C7—H7	120.2	C28—C27—H27	120.0
C8—C7—H7	120.2	N12—C28—C29	121.79 (12)
N4—C8—C7	122.44 (11)	N12—C28—C27	121.76 (12)
N4—C8—C9	121.15 (11)	C29—C28—C27	116.45 (11)
C7—C8—C9	116.41 (11)	C30—C29—C28	119.36 (12)
C10—C9—C8	119.74 (11)	C30—C29—H29	120.3
C10—C9—H9	120.1	C28—C29—H29	120.3
C8—C9—H9	120.1	N11—C30—C29	124.06 (12)
N3—C10—C9	123.60 (11)	N11—C30—H30	118.0
N3—C10—H10	118.2	C29—C30—H30	118.0
C9—C10—H10	118.2	N13—C31—C32	123.80 (13)
N5—C11—C12	123.69 (12)	N13—C31—H31	118.1
N5—C11—H11	118.2	C32—C31—H31	118.1
C12—C11—H11	118.2	C31—C32—C33	119.43 (13)

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C11—C12—C13	119.42 (12)	C31—C32—H32	120.3
C11—C12—H12	120.3	C33—C32—H32	120.3
C13—C12—H12	120.3	N14—C33—C34	122.24 (13)
N6—C13—C14	121.09 (12)	N14—C33—C32	121.20 (13)
N6—C13—C12	121.88 (12)	C34—C33—C32	116.56 (12)
C14—C13—C12	117.02 (12)	C35—C34—C33	120.14 (13)
C15—C14—C13	119.50 (11)	C35—C34—H34	119.9
C15—C14—H14	120.2	C33—C34—H34	119.9
C13—C14—H14	120.2	N13—C35—C34	123.33 (13)
N5—C15—C14	123.41 (12)	N13—C35—H35	118.3
N5—C15—H15	118.3	C34—C35—H35	118.3
C14—C15—H15	118.3	N15—C36—C37	123.47 (12)
N7—C16—C17	123.80 (12)	N15—C36—H36	118.3
N7—C16—H16	118.1	C37—C36—H36	118.3
C17—C16—H16	118.1	C36—C37—C38	119.33 (12)
C16—C17—C18	119.65 (12)	C36—C37—H37	120.3
C16—C17—H17	120.2	C38—C37—H37	120.3
C18—C17—H17	120.2	N16—C38—C37	120.96 (12)
N8—C18—C17	122.71 (13)	N16—C38—C39	121.87 (12)
N8—C18—C19	120.80 (13)	C37—C38—C39	117.17 (11)
C17—C18—C19	116.49 (12)	C40—C39—C38	119.49 (12)
C20—C19—C18	119.76 (12)	C40—C39—H39	120.3
C20—C19—H19	120.1	C38—C39—H39	120.3
C18—C19—H19	120.1	N15—C40—C39	123.33 (12)
N7—C20—C19	123.73 (12)	N15—C40—H40	118.3
N7—C20—H20	118.1	C39—C40—H40	118.3
C19—C20—H20	118.1	H1W1—O1W—H2W1	111.7 (14)
N11—Cu2—N15	171.22 (5)	H1W2—O2W—H2W2	107.8 (13)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C2—H2...Cl2 ⁱ	0.93	2.68	3.5955 (13)	167
C6—H6...Cl2	0.93	2.80	3.3665 (13)	121
C10—H10...C11	0.93	2.78	3.4430 (13)	129
C20—H20...C11	0.93	2.64	3.3522 (13)	134
C25—H25...C14	0.93	2.71	3.2928 (13)	121
C26—H26...N9	0.93	2.62	3.0473 (17)	108
C35—H35...C14	0.93	2.67	3.3947 (14)	136
N4—H4A...O1W ⁱⁱ	0.86	2.38	3.2094 (17)	163
N4—H4B...C11 ⁱⁱⁱ	0.86	2.43	3.2893 (13)	175
N6—H6A...C14 ^{iv}	0.86	2.82	3.4066 (13)	127
N8—H8B...Cl2 ^{iv}	0.86	2.56	3.4021 (13)	166
N10—H10B...Cl3 ^v	0.86	2.41	3.2596 (11)	171
N12—H12A...O1W	0.86	2.06	2.8908 (16)	162
N14—H14B...O1W ^{vi}	0.86	2.25	3.0103 (18)	147
N16—H16B...N16 ^{vii}	0.86	2.52	3.2036 (17)	137

O1W—H1W1...C11 ⁱⁱⁱ	0.83 (1)	2.26 (1)	3.0614 (12)	163 (2)
O1W—H2W1...C14 ^{viii}	0.83 (1)	2.26 (1)	3.0694 (12)	164 (2)
C37—H37...Cg1 ^{ix}	0.93	2.81	3.5492 (14)	137
C39—H39...Cg1 ^x	0.93	2.95	3.7306 (14)	142
N2—H2B...Cg2 ^{xi}	0.86	2.75	3.3359 (13)	126
C12—H12...Cg3	0.93	2.79	3.6488 (14)	154
C14—H14...Cg3 ^{iv}	0.93	2.83	3.6193 (13)	144

Symmetry codes: (i) $-x+1, -y, -z$; (ii) $-x+2, -y, -z+1$; (iii) $-x+1, -y, -z+1$; (iv) $x-1, y, z$; (v) $-x+2, -y+1, -z$; (vi) $-x+1, -y+1, -z+1$; (vii) $-x+2, -y+2, -z$; (viii) $-x+2, -y+1, -z+1$; (ix) $x, y+1, z$; (x) $x+1, y+1, z$; (xi) $x-1, y-1, z$.

Fig. 1

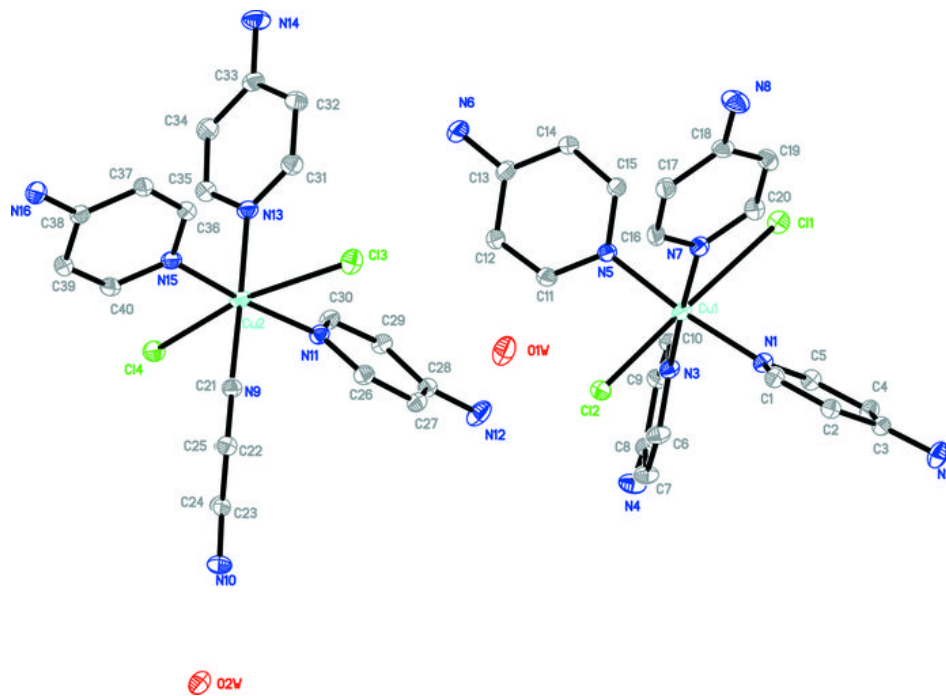


Fig. 2

