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6-Cyclohexylmethyl-2-cyclohexylsulfanyl-5-isopropylpyrimidin-4(3H)-one

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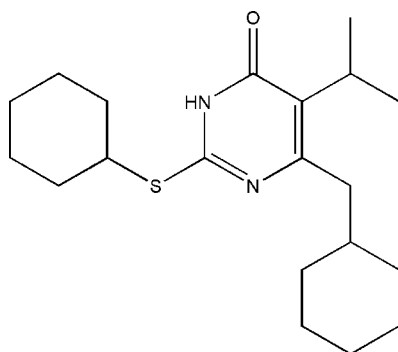
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.066; wR factor = 0.216; data-to-parameter ratio = 21.4.

The title compound, $\text{C}_{20}\text{H}_{32}\text{N}_2\text{O}_2\text{S}$, was obtained during the course of our investigation on 2-alkylsulfanyl-6-benzyl-3,4-dihydropyrimidin-4(3H)-ones (S-DABOs) showing favourable anti-HIV-1 activity. Both cyclohexane rings adopt chair conformations. The angle at the methylene C atom linking the pyrimidine and cyclohexane ring is 113.7 (3)°, which is in the range considered optimal for maximum activity of non-nucleoside reverse transcriptase inhibitors. Intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds link the molecules into dimers and stabilize the crystal structure of the compound. In addition, an intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bond is observed.

Related literature

For related literature, see: He *et al.* (2004); Ettorre *et al.* (1996, 1998); Rao *et al.* (2007).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{32}\text{N}_2\text{O}_2\text{S}$
 $M_r = 348.54$
 Triclinic, $P\bar{1}$
 $a = 9.9549$ (16) Å
 $b = 10.9542$ (17) Å
 $c = 12.1054$ (19) Å
 $\alpha = 63.250$ (2)°
 $\beta = 69.195$ (2)°
 $\gamma = 63.033$ (2)°
 $V = 1031.5$ (3) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.17$ mm⁻¹
 $T = 298$ (2) K
 $0.19 \times 0.14 \times 0.12$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 1998)
 $T_{\min} = 0.969$, $T_{\max} = 0.980$
 9017 measured reflections
 4697 independent reflections
 2103 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.065$
 $wR(F^2) = 0.216$
 $S = 0.88$
 4697 reflections
 219 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.17$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H2}\cdots\text{O1}^1$	0.86	1.91	2.761 (4)	170
$\text{C11}-\text{H11C}\cdots\text{O1}$	0.96	2.53	3.115 (5)	119

Symmetry code: (i) $-x, -y, -z + 1$.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WN2273).

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supplementary materials

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6-Cyclohexylmethyl-2-cyclohexylsulfanyl-5-isopropylpyrimidin-4(3*H*)-one

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Comment

As part of our ongoing investigation of S-DABO analogues which are a potent family of non-nucleoside reverse transcriptase inhibitors (NNRTIs), the title compound was synthesized as a novel inhibitor and shows favourable anti-HIV-1 activity.

The molecular structure is shown in Fig. 1. Both cyclohexane rings adopt the lowest energy chair conformation. C13—C14—C5 is 113.7 (3)°, which is in the range considered optimal for maximum activity of NNRTIs, *viz.* 110°–115° (Ettorre *et al.*, 1996).

A comparison of the crystal structure of the title compound with some reported S-DABOs show that their spatial arrangement is similar (Ettorre *et al.*, 1998; Rao *et al.*, 2007). Although these molecules assume a similar conformation, they show differences in their activities. Thus, further structural investigations are needed.

Intermolecular N—H···O hydrogen bonds link the molecules into dimers and stabilize the crystal structure of the compound. In addition, an intramolecular C—H···O hydrogen bond is observed.

Experimental

With 2-cyclohexylacetonitrile as the starting material, the title compound was synthesized according to the procedure of He *et al.* (2004). Single crystals were obtained from a mixture of ethyl acetate and petroleum ether by slow evaporation at room temperature.

Refinement

Methyl H atoms were placed in calculated positions with C—H = 0.96 Å and the torsion angle was refined to fit the electron density; $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. Other H atoms were placed in calculated positions with C—H = 0.97–0.98 Å and N—H = 0.86 Å, and refined in riding mode; $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$.

Figures

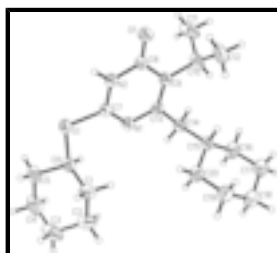


Fig. 1. The molecular structure of the title compound, showing the atom labelling scheme and 30% probability displacement ellipsoids.

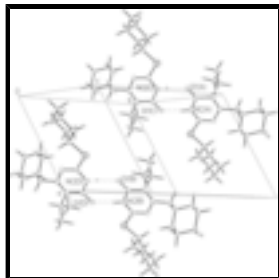


Fig. 2. The crystal packing of title compound showing the intermolecular hydrogen bonding (dashed lines)

6-Cyclohexylmethyl-2-cyclohexylsulfanyl-5-isopropylpyrimidin-4(3H)-one

Crystal data

$C_{20}H_{32}N_2OS$	$Z = 2$
$M_r = 348.54$	$F_{000} = 380$
Triclinic, $P\bar{1}$	$D_x = 1.122 \text{ Mg m}^{-3}$
$a = 9.9549 (16) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 10.9542 (17) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$c = 12.1054 (19) \text{ \AA}$	Cell parameters from 4697 reflections
$\alpha = 63.250 (2)^\circ$	$\theta = 1.9\text{--}28.3^\circ$
$\beta = 69.195 (2)^\circ$	$\mu = 0.17 \text{ mm}^{-1}$
$\gamma = 63.033 (2)^\circ$	$T = 298 (2) \text{ K}$
$V = 1031.5 (3) \text{ \AA}^3$	Block, colourless
	$0.19 \times 0.14 \times 0.12 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	4697 independent reflections
Radiation source: fine-focus sealed tube	2103 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.047$
$T = 298(2) \text{ K}$	$\theta_{\text{max}} = 28.3^\circ$
φ and ω scans	$\theta_{\text{min}} = 1.9^\circ$
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 1998)	$h = -13 \rightarrow 13$
$T_{\text{min}} = 0.969$, $T_{\text{max}} = 0.980$	$k = -13 \rightarrow 14$
9017 measured reflections	$l = -16 \rightarrow 15$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.065$	H-atom parameters constrained
$wR(F^2) = 0.216$	$w = 1/[\sigma^2(F_o^2) + (0.1P)^2 + 0.3399P]$
$S = 0.88$	where $P = (F_o^2 + 2F_c^2)/3$
	$(\Delta/\sigma)_{\text{max}} < 0.001$

4697 reflections $\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$
 219 parameters $\Delta\rho_{\min} = -0.20 \text{ e } \text{\AA}^{-3}$
 Primary atom site location: structure-invariant direct methods Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors (gt) etc, and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.33860 (11)	0.15552 (12)	0.30395 (9)	0.0638 (3)
O1	-0.1114 (3)	0.0731 (3)	0.6134 (2)	0.0669 (8)
N1	0.1985 (3)	0.2729 (3)	0.4863 (2)	0.0473 (7)
N2	0.0978 (3)	0.1230 (3)	0.4780 (2)	0.0514 (7)
H2	0.1069	0.0683	0.4405	0.062*
C8	-0.0217 (4)	0.1378 (4)	0.5792 (3)	0.0492 (8)
C7	0.2013 (4)	0.1900 (4)	0.4349 (3)	0.0465 (8)
C13	0.0816 (3)	0.2939 (3)	0.5876 (3)	0.0439 (8)
C9	-0.0283 (4)	0.2320 (4)	0.6356 (3)	0.0492 (8)
C10	-0.1620 (4)	0.2617 (4)	0.7421 (4)	0.0686 (11)
H10	-0.1477	0.3255	0.7702	0.082*
C14	0.0877 (4)	0.3902 (4)	0.6429 (3)	0.0522 (9)
H14A	-0.0149	0.4592	0.6585	0.063*
H14B	0.1521	0.4456	0.5817	0.063*
C6	0.4652 (4)	0.2481 (4)	0.2780 (3)	0.0551 (9)
H6	0.4021	0.3453	0.2833	0.066*
C15	0.1495 (4)	0.3055 (4)	0.7659 (3)	0.0531 (9)
H15	0.0998	0.2329	0.8188	0.064*
C5	0.5566 (5)	0.2656 (5)	0.1449 (4)	0.0801 (13)
H5A	0.4871	0.3252	0.0857	0.096*
H5B	0.6106	0.1706	0.1356	0.096*
C1	0.5698 (4)	0.1675 (4)	0.3730 (4)	0.0675 (11)
H1A	0.6276	0.0684	0.3734	0.081*
H1B	0.5087	0.1628	0.4566	0.081*
C20	0.3203 (4)	0.2241 (4)	0.7442 (4)	0.0671 (11)
H20A	0.3720	0.2930	0.6876	0.081*
H20B	0.3431	0.1559	0.7039	0.081*
C16	0.1092 (5)	0.4041 (5)	0.8375 (4)	0.0779 (12)

supplementary materials

H16A	0.1537	0.4795	0.7857	0.094*
H16B	-0.0014	0.4514	0.8545	0.094*
C18	0.3361 (5)	0.2395 (5)	0.9399 (4)	0.0860 (13)
H18A	0.3673	0.1805	1.0210	0.103*
H18B	0.3898	0.3084	0.8947	0.103*
C2	0.6803 (5)	0.2449 (5)	0.3405 (4)	0.0806 (13)
H2A	0.6225	0.3418	0.3455	0.097*
H2B	0.7477	0.1910	0.4013	0.097*
C4	0.6718 (6)	0.3368 (6)	0.1148 (4)	0.1012 (17)
H4A	0.7343	0.3389	0.0320	0.121*
H4B	0.6168	0.4369	0.1124	0.121*
C17	0.1662 (5)	0.3218 (6)	0.9600 (4)	0.0909 (15)
H17A	0.1430	0.3900	1.0003	0.109*
H17B	0.1131	0.2538	1.0158	0.109*
C19	0.3812 (5)	0.1413 (5)	0.8657 (4)	0.0842 (13)
H19A	0.4922	0.0985	0.8466	0.101*
H19B	0.3415	0.0624	0.9170	0.101*
C3	0.7749 (5)	0.2569 (6)	0.2104 (6)	0.1060 (18)
H3A	0.8383	0.1600	0.2068	0.127*
H3B	0.8420	0.3090	0.1908	0.127*
C12	-0.1652 (6)	0.1229 (6)	0.8561 (4)	0.0948 (15)
H12A	-0.1853	0.0600	0.8335	0.142*
H12B	-0.2446	0.1483	0.9239	0.142*
H12C	-0.0679	0.0732	0.8823	0.142*
C11	-0.3132 (4)	0.3452 (5)	0.6950 (5)	0.0978 (16)
H11A	-0.3070	0.4320	0.6245	0.147*
H11B	-0.3949	0.3714	0.7613	0.147*
H11C	-0.3330	0.2846	0.6694	0.147*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0637 (6)	0.0911 (8)	0.0556 (6)	-0.0395 (6)	0.0099 (4)	-0.0446 (6)
O1	0.0614 (16)	0.088 (2)	0.0772 (18)	-0.0426 (15)	0.0112 (13)	-0.0516 (16)
N1	0.0487 (16)	0.0526 (17)	0.0457 (16)	-0.0201 (14)	-0.0045 (12)	-0.0222 (14)
N2	0.0519 (16)	0.0645 (19)	0.0499 (16)	-0.0248 (15)	0.0003 (13)	-0.0329 (15)
C8	0.0451 (19)	0.059 (2)	0.051 (2)	-0.0177 (17)	-0.0052 (15)	-0.0282 (17)
C7	0.0478 (19)	0.052 (2)	0.0429 (18)	-0.0194 (16)	-0.0072 (14)	-0.0181 (16)
C13	0.0449 (18)	0.046 (2)	0.0426 (18)	-0.0123 (15)	-0.0075 (14)	-0.0211 (16)
C9	0.0432 (18)	0.058 (2)	0.053 (2)	-0.0192 (16)	-0.0007 (15)	-0.0295 (17)
C10	0.063 (2)	0.082 (3)	0.080 (3)	-0.037 (2)	0.018 (2)	-0.055 (2)
C14	0.052 (2)	0.053 (2)	0.057 (2)	-0.0199 (17)	-0.0023 (16)	-0.0273 (18)
C6	0.061 (2)	0.054 (2)	0.052 (2)	-0.0245 (18)	0.0040 (17)	-0.0273 (18)
C15	0.060 (2)	0.062 (2)	0.050 (2)	-0.0301 (18)	0.0006 (16)	-0.0299 (18)
C5	0.099 (3)	0.096 (3)	0.058 (2)	-0.060 (3)	0.017 (2)	-0.036 (2)
C1	0.058 (2)	0.069 (3)	0.072 (3)	-0.022 (2)	-0.015 (2)	-0.020 (2)
C20	0.064 (2)	0.076 (3)	0.066 (2)	-0.013 (2)	-0.0122 (19)	-0.040 (2)
C16	0.082 (3)	0.091 (3)	0.080 (3)	-0.021 (2)	-0.011 (2)	-0.060 (3)

C18	0.093 (3)	0.109 (4)	0.074 (3)	-0.038 (3)	-0.018 (2)	-0.043 (3)
C2	0.062 (3)	0.075 (3)	0.113 (4)	-0.023 (2)	-0.024 (3)	-0.035 (3)
C4	0.130 (4)	0.115 (4)	0.079 (3)	-0.089 (4)	0.035 (3)	-0.045 (3)
C17	0.091 (3)	0.128 (4)	0.080 (3)	-0.033 (3)	-0.005 (2)	-0.072 (3)
C19	0.080 (3)	0.094 (3)	0.082 (3)	-0.012 (2)	-0.026 (2)	-0.043 (3)
C3	0.067 (3)	0.101 (4)	0.168 (6)	-0.042 (3)	0.016 (3)	-0.078 (4)
C12	0.112 (4)	0.116 (4)	0.066 (3)	-0.066 (3)	0.029 (3)	-0.048 (3)
C11	0.053 (3)	0.095 (4)	0.133 (4)	-0.021 (2)	0.017 (3)	-0.062 (3)

Geometric parameters (Å, °)

S1—C7	1.748 (3)	C1—H1B	0.9700
S1—C6	1.817 (3)	C20—C19	1.513 (5)
O1—C8	1.233 (4)	C20—H20A	0.9700
N1—C7	1.299 (4)	C20—H20B	0.9700
N1—C13	1.381 (4)	C16—C17	1.505 (6)
N2—C7	1.353 (4)	C16—H16A	0.9700
N2—C8	1.383 (4)	C16—H16B	0.9700
N2—H2	0.8600	C18—C17	1.502 (6)
C8—C9	1.441 (4)	C18—C19	1.520 (6)
C13—C9	1.364 (4)	C18—H18A	0.9700
C13—C14	1.512 (4)	C18—H18B	0.9700
C9—C10	1.518 (4)	C2—C3	1.503 (6)
C10—C11	1.523 (6)	C2—H2A	0.9700
C10—C12	1.531 (6)	C2—H2B	0.9700
C10—H10	0.9800	C4—C3	1.509 (7)
C14—C15	1.531 (5)	C4—H4A	0.9700
C14—H14A	0.9700	C4—H4B	0.9700
C14—H14B	0.9700	C17—H17A	0.9700
C6—C1	1.514 (5)	C17—H17B	0.9700
C6—C5	1.516 (5)	C19—H19A	0.9700
C6—H6	0.9800	C19—H19B	0.9700
C15—C20	1.508 (5)	C3—H3A	0.9700
C15—C16	1.511 (5)	C3—H3B	0.9700
C15—H15	0.9800	C12—H12A	0.9600
C5—C4	1.521 (6)	C12—H12B	0.9600
C5—H5A	0.9700	C12—H12C	0.9600
C5—H5B	0.9700	C11—H11A	0.9600
C1—C2	1.523 (5)	C11—H11B	0.9600
C1—H1A	0.9700	C11—H11C	0.9600
C7—S1—C6	103.08 (16)	C19—C20—H20B	109.2
C7—N1—C13	117.1 (3)	H20A—C20—H20B	107.9
C7—N2—C8	123.4 (3)	C17—C16—C15	112.2 (4)
C7—N2—H2	118.3	C17—C16—H16A	109.2
C8—N2—H2	118.3	C15—C16—H16A	109.2
O1—C8—N2	119.8 (3)	C17—C16—H16B	109.2
O1—C8—C9	125.8 (3)	C15—C16—H16B	109.2
N2—C8—C9	114.4 (3)	H16A—C16—H16B	107.9
N1—C7—N2	122.9 (3)	C17—C18—C19	111.7 (4)

supplementary materials

N1—C7—S1	123.0 (3)	C17—C18—H18A	109.3
N2—C7—S1	114.1 (2)	C19—C18—H18A	109.3
C9—C13—N1	123.6 (3)	C17—C18—H18B	109.3
C9—C13—C14	122.5 (3)	C19—C18—H18B	109.3
N1—C13—C14	113.8 (3)	H18A—C18—H18B	107.9
C13—C9—C8	118.6 (3)	C3—C2—C1	111.1 (4)
C13—C9—C10	123.7 (3)	C3—C2—H2A	109.4
C8—C9—C10	117.7 (3)	C1—C2—H2A	109.4
C9—C10—C11	110.7 (3)	C3—C2—H2B	109.4
C9—C10—C12	112.6 (3)	C1—C2—H2B	109.4
C11—C10—C12	112.1 (3)	H2A—C2—H2B	108.0
C9—C10—H10	107.0	C3—C4—C5	112.2 (4)
C11—C10—H10	107.0	C3—C4—H4A	109.2
C12—C10—H10	107.0	C5—C4—H4A	109.2
C13—C14—C15	113.7 (3)	C3—C4—H4B	109.2
C13—C14—H14A	108.8	C5—C4—H4B	109.2
C15—C14—H14A	108.8	H4A—C4—H4B	107.9
C13—C14—H14B	108.8	C18—C17—C16	111.4 (3)
C15—C14—H14B	108.8	C18—C17—H17A	109.3
H14A—C14—H14B	107.7	C16—C17—H17A	109.3
C1—C6—C5	111.3 (3)	C18—C17—H17B	109.3
C1—C6—S1	113.1 (3)	C16—C17—H17B	109.3
C5—C6—S1	106.7 (2)	H17A—C17—H17B	108.0
C1—C6—H6	108.5	C20—C19—C18	111.9 (4)
C5—C6—H6	108.5	C20—C19—H19A	109.2
S1—C6—H6	108.5	C18—C19—H19A	109.2
C20—C15—C16	110.4 (3)	C20—C19—H19B	109.2
C20—C15—C14	112.2 (3)	C18—C19—H19B	109.2
C16—C15—C14	111.6 (3)	H19A—C19—H19B	107.9
C20—C15—H15	107.5	C2—C3—C4	110.2 (4)
C16—C15—H15	107.5	C2—C3—H3A	109.6
C14—C15—H15	107.5	C4—C3—H3A	109.6
C6—C5—C4	111.2 (3)	C2—C3—H3B	109.6
C6—C5—H5A	109.4	C4—C3—H3B	109.6
C4—C5—H5A	109.4	H3A—C3—H3B	108.1
C6—C5—H5B	109.4	C10—C12—H12A	109.5
C4—C5—H5B	109.4	C10—C12—H12B	109.5
H5A—C5—H5B	108.0	H12A—C12—H12B	109.5
C6—C1—C2	110.4 (3)	C10—C12—H12C	109.5
C6—C1—H1A	109.6	H12A—C12—H12C	109.5
C2—C1—H1A	109.6	H12B—C12—H12C	109.5
C6—C1—H1B	109.6	C10—C11—H11A	109.5
C2—C1—H1B	109.6	C10—C11—H11B	109.5
H1A—C1—H1B	108.1	H11A—C11—H11B	109.5
C15—C20—C19	112.1 (3)	C10—C11—H11C	109.5
C15—C20—H20A	109.2	H11A—C11—H11C	109.5
C19—C20—H20A	109.2	H11B—C11—H11C	109.5
C15—C20—H20B	109.2		
C7—N2—C8—O1	-179.8 (3)	C9—C13—C14—C15	75.8 (4)

C7—N2—C8—C9	0.3 (5)	N1—C13—C14—C15	-103.1 (3)
C13—N1—C7—N2	-1.4 (5)	C7—S1—C6—C1	74.8 (3)
C13—N1—C7—S1	178.5 (2)	C7—S1—C6—C5	-162.5 (3)
C8—N2—C7—N1	1.0 (5)	C13—C14—C15—C20	73.0 (4)
C8—N2—C7—S1	-178.9 (2)	C13—C14—C15—C16	-162.6 (3)
C6—S1—C7—N1	4.1 (3)	C1—C6—C5—C4	-53.9 (5)
C6—S1—C7—N2	-176.0 (2)	S1—C6—C5—C4	-177.7 (3)
C7—N1—C13—C9	0.4 (5)	C5—C6—C1—C2	55.9 (4)
C7—N1—C13—C14	179.3 (3)	S1—C6—C1—C2	176.0 (3)
N1—C13—C9—C8	0.9 (5)	C16—C15—C20—C19	54.5 (5)
C14—C13—C9—C8	-177.9 (3)	C14—C15—C20—C19	179.6 (3)
N1—C13—C9—C10	-176.4 (3)	C20—C15—C16—C17	-55.8 (5)
C14—C13—C9—C10	4.8 (5)	C14—C15—C16—C17	178.7 (3)
O1—C8—C9—C13	179.0 (3)	C6—C1—C2—C3	-58.1 (5)
N2—C8—C9—C13	-1.2 (5)	C6—C5—C4—C3	54.0 (5)
O1—C8—C9—C10	-3.6 (5)	C19—C18—C17—C16	-53.6 (5)
N2—C8—C9—C10	176.2 (3)	C15—C16—C17—C18	55.8 (5)
C13—C9—C10—C11	112.6 (4)	C15—C20—C19—C18	-53.5 (5)
C8—C9—C10—C11	-64.7 (4)	C17—C18—C19—C20	52.6 (5)
C13—C9—C10—C12	-121.1 (4)	C1—C2—C3—C4	57.6 (5)
C8—C9—C10—C12	61.7 (5)	C5—C4—C3—C2	-55.7 (5)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2...O1 ⁱ	0.86	1.91	2.761 (4)	170
C11—H11C...O1	0.96	2.53	3.115 (5)	119

Symmetry codes: (i) $-x, -y, -z+1$.

Fig. 1

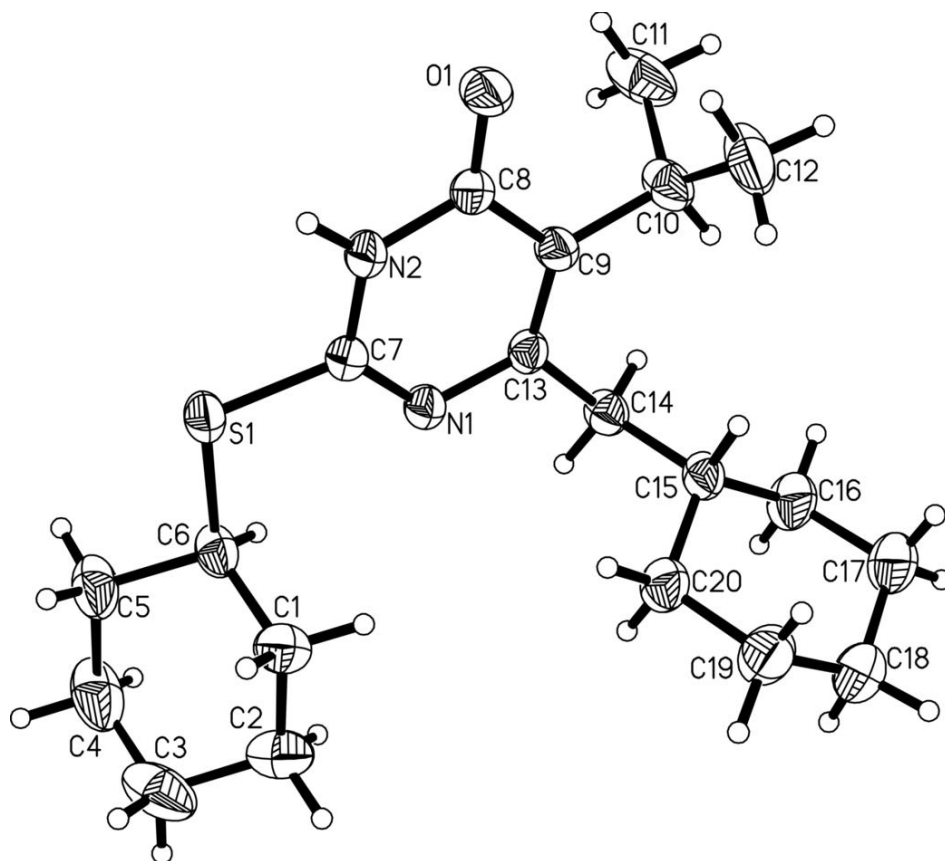


Fig. 2

