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## Structure Reports

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# (3*a*S,7*S*,9*a*S,9*b*R)-3*a*,6,6,9*a*-Tetramethyl-2-oxoperhydronaphtho[2,1-*b*]furan-7-yl acetate

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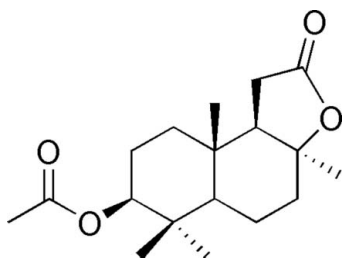
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Key indicators: single-crystal X-ray study;  $T = 153$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.032;  $wR$  factor = 0.087; data-to-parameter ratio = 10.0.

The title compound (common name: 3β-acetoxy-8-*epi*-sclareolide), C<sub>18</sub>H<sub>28</sub>O<sub>4</sub>, is a sclareolide derivative, which was synthesized from 9(11)-*en*-3β-acetoxy-8-*epi*-sclareolide. In the molecular structure, the two six-membered rings display chair conformations and the five-membered ring displays an envelope conformation. Weak intermolecular C—H...O hydrogen bonding is present in the crystal structure.

## Related literature

For general background, see: Choudhary *et al.* (2004); Quideau *et al.* (2002). For related structures, see: Devi *et al.* (2004); Bhattacharyya *et al.* (2006). For synthesis, see: Yang *et al.* (2006).



## Experimental

## Crystal data

C<sub>18</sub>H<sub>28</sub>O<sub>4</sub>  
 $M_r = 308.40$   
 Monoclinic,  $P2_1$   
 $a = 10.1935$  (5) Å  
 $b = 7.3226$  (3) Å  
 $c = 11.3056$  (4) Å  
 $\beta = 99.2940$  (1)°  
 $V = 832.81$  (6) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 153$  (2) K  
 $0.60 \times 0.54 \times 0.47$  mm

## Data collection

Rigaku R-AXIS RAPID IP diffractometer  
 Absorption correction: none  
 8195 measured reflections  
 2046 independent reflections  
 1943 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.018$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.032$   
 $wR(F^2) = 0.086$   
 $S = 0.99$   
 2046 reflections  
 205 parameters  
 1 restraint  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.20$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.14$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C16}-\text{H16B}\cdots\text{O2}^i$	0.98	2.55	3.383 (3)	143

 Symmetry code: (i)  $-x + 1, y + \frac{1}{2}, -z + 1$ .

Data collection: *RAPID-AUTO* (Rigaku, 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2447).

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**supplementary materials**

*Acta Cryst.* (2008). E64, o1765 [ doi:10.1107/S1600536808025737 ]

**(3a*S*,7*S*,9a*S*,9b*R*)-3a,6,6,9a-Tetramethyl-2-oxoperhydronaphtho[2,1-*b*]furan-7-yl acetate**

**Q.-C. Huang, B.-G. Li, Y.-P. Xie, K.-B. Yu and G.-L. Zhang**

**Comment**

Sclareolide has exhibited phytotoxic activity and cytotoxicity against human cancer cell lines (Choudhary *et al.*, 2004). It also be important pharmaceutical intermediates (Quideau *et al.*, 2002). Here we report the synthesis and crystal structure of the title compound which is a 8-epi-sclareolide type compound.

The molecular structure is shown in Fig. 1. The molecule contains two six-membered rings, A (atoms C1–C5/C10) and B (atoms C5–C10), and one five-membered lactone rings C (C8/C9/O1/C11/C12). The cyclohexane ring A and the cyclohexane ring B exist both in chair conformation. The  $\gamma$ -lactone rings C adopt envelope conformations with C9 at the flap (Devi *et al.*, 2004; Bhattacharyya *et al.*, 2006). The rings A/B are *trans* fused and the rings B/C are *cis* fused. The C1/C2/C4/C5, C6/C7/C9/C10, C8/C11/C12/O1 form least square plane D, E and F, respectively. The dihedral angles between planes D and E is 15.50 (8)°, between planes E and F is 59.03 (7)°, between planes D and F is 43.54 (7)°. The C3 and C10 deviate from plane D by 0.634 (2) and 0.644 (2) Å, respectively. The C5 and C8 deviate from plane E by 0.765 (2) and 0.382 (3) Å, respectively. The C9 deviates from plane F by 0.566 (3) Å.

The intermolecular weak C—H $\cdots$ O hydrogen bonding presents in the crystal structure (Table 1).

**Experimental**

To a methanol solution (10 ml) of 9(11)-en-3 $\beta$ -Acetoxy-8-epi-sclareolide (1 mmol) (Yang *et al.*, 2006) was added NiCl<sub>2</sub>.6H<sub>2</sub>O (1 mmol), and the mixture was cooled to 273 K with an ice bath, then NaBH<sub>4</sub> (4 mmol) was added in small portions over 30 min. The ice bath was removed and the reaction mixture was left stirred for 4 h at room temperature. Then the suspension was filtered, and after usual workup, the residue was purified by flash chromatography eluted with petroleum ether–ethyl acetate (10:1) to afford the title compound. Yield (97%).

**Refinement**

H atoms were placed in calculated positions with C—H = 0.98–1.00 Å, and refined in riding mode with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . The absolute configuration has been assigned by reference to an unchanging chiral centre in the synthetic procedure; Friedel pairs were merged.

**Figures**

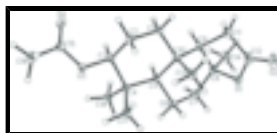


Fig. 1. The molecular structure of the title compound, displacement ellipsoids are drawn at the 30% probability level.

## (3aS,7S,9aS,9bR)-3a,6,6,9a-Tetramethyl-2-oxoperhydronaphtho[2,1-b]furan-7-yl acetate

### Crystal data

$C_{18}H_{28}O_4$	$F_{000} = 336$
$M_r = 308.40$	$D_x = 1.230 \text{ Mg m}^{-3}$
Monoclinic, $P2_1$	Mo $K\alpha$ radiation
Hall symbol: P 2yb	$\lambda = 0.71073 \text{ \AA}$
$a = 10.1935 (5) \text{ \AA}$	Cell parameters from 7613 reflections
$b = 7.3226 (3) \text{ \AA}$	$\theta = 3.3\text{--}27.5^\circ$
$c = 11.3056 (4) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 99.2940 (1)^\circ$	$T = 153 (2) \text{ K}$
$V = 832.81 (6) \text{ \AA}^3$	Block, colourless
$Z = 2$	$0.60 \times 0.54 \times 0.47 \text{ mm}$

### Data collection

Rigaku R-Axis RAPID IP diffractometer	1943 reflections with $I > 2\sigma(I)$
Radiation source: Rotating Anode	$R_{\text{int}} = 0.018$
Monochromator: graphite	$\theta_{\text{max}} = 27.5^\circ$
$T = 153(2) \text{ K}$	$\theta_{\text{min}} = 3.3^\circ$
$\omega$ scans	$h = -13 \rightarrow 13$
Absorption correction: none	$k = -9 \rightarrow 9$
8195 measured reflections	$l = -14 \rightarrow 12$
2046 independent reflections	

### Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.032$	H-atom parameters constrained
$wR(F^2) = 0.086$	$w = 1/[\sigma^2(F_o^2) + (0.0526P)^2 + 0.1209P]$
$S = 0.99$	where $P = (F_o^2 + 2F_c^2)/3$
2046 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
205 parameters	$\Delta\rho_{\text{max}} = 0.20 \text{ e \AA}^{-3}$
1 restraint	$\Delta\rho_{\text{min}} = -0.14 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: SHELXTL (Sheldrick, 2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
	Extinction coefficient: 0.065 (6)

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.15453 (13)	0.2174 (2)	0.55280 (13)	0.0477 (4)
O2	0.10361 (18)	0.3434 (4)	0.37252 (14)	0.0809 (6)
O3	0.78754 (11)	0.58869 (17)	0.86839 (11)	0.0377 (3)
O4	0.77677 (15)	0.8581 (2)	0.96045 (15)	0.0599 (4)
C1	0.44238 (16)	0.6709 (2)	0.70097 (15)	0.0344 (3)
H1A	0.4104	0.7464	0.6295	0.041*
H1B	0.4025	0.7199	0.7687	0.041*
C2	0.59341 (16)	0.6859 (2)	0.73086 (15)	0.0370 (4)
H2A	0.6341	0.6395	0.6630	0.044*
H2B	0.6194	0.8154	0.7443	0.044*
C3	0.64282 (15)	0.5757 (2)	0.84229 (14)	0.0301 (3)
H3	0.6051	0.6290	0.9110	0.036*
C4	0.60784 (14)	0.3708 (2)	0.83267 (13)	0.0280 (3)
C5	0.45489 (14)	0.3533 (2)	0.78539 (13)	0.0259 (3)
H5	0.4102	0.3959	0.8529	0.031*
C6	0.40994 (16)	0.1550 (2)	0.76243 (16)	0.0360 (4)
H6A	0.4350	0.1121	0.6861	0.043*
H6B	0.4551	0.0762	0.8277	0.043*
C7	0.25934 (17)	0.1404 (3)	0.75616 (17)	0.0427 (4)
H7A	0.2310	0.0175	0.7257	0.051*
H7B	0.2388	0.1502	0.8386	0.051*
C8	0.17711 (15)	0.2812 (3)	0.67859 (15)	0.0386 (4)
C9	0.24142 (15)	0.4709 (3)	0.66974 (15)	0.0345 (3)
H9	0.2175	0.5507	0.7348	0.041*
C10	0.39595 (14)	0.4729 (2)	0.67643 (13)	0.0275 (3)
C11	0.1657 (2)	0.5373 (3)	0.54912 (19)	0.0482 (5)
H11A	0.0824	0.6003	0.5595	0.058*
H11B	0.2210	0.6213	0.5093	0.058*
C12	0.13711 (18)	0.3637 (4)	0.47870 (18)	0.0523 (5)
C13	0.43749 (17)	0.4084 (3)	0.55821 (14)	0.0384 (4)
H13A	0.5347	0.4090	0.5663	0.046*
H13B	0.4000	0.4909	0.4932	0.046*
H13C	0.4043	0.2844	0.5399	0.046*

## supplementary materials

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C14	0.03976 (18)	0.2954 (4)	0.7163 (2)	0.0566 (6)
H14A	-0.0028	0.1750	0.7102	0.068*
H14B	-0.0153	0.3816	0.6635	0.068*
H14C	0.0494	0.3388	0.7993	0.068*
C15	0.84029 (17)	0.7424 (3)	0.92062 (16)	0.0399 (4)
C16	0.98747 (19)	0.7482 (3)	0.9221 (2)	0.0523 (5)
H16A	1.0279	0.8355	0.9830	0.063*
H16B	1.0056	0.7864	0.8432	0.063*
H16C	1.0252	0.6266	0.9411	0.063*
C17	0.69680 (16)	0.2695 (3)	0.75667 (16)	0.0395 (4)
H17A	0.6594	0.1485	0.7353	0.047*
H17B	0.7863	0.2562	0.8028	0.047*
H17C	0.7014	0.3393	0.6835	0.047*
C18	0.63524 (17)	0.2918 (3)	0.96055 (16)	0.0410 (4)
H18A	0.7252	0.3255	0.9987	0.049*
H18B	0.6274	0.1585	0.9569	0.049*
H18C	0.5705	0.3412	1.0074	0.049*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0391 (7)	0.0550 (9)	0.0468 (7)	-0.0076 (6)	0.0003 (5)	-0.0144 (6)
O2	0.0773 (11)	0.1122 (17)	0.0453 (8)	-0.0063 (12)	-0.0140 (7)	-0.0085 (11)
O3	0.0267 (6)	0.0346 (6)	0.0499 (7)	-0.0043 (5)	0.0006 (5)	-0.0051 (5)
O4	0.0517 (8)	0.0469 (8)	0.0773 (10)	-0.0032 (7)	-0.0014 (7)	-0.0222 (8)
C1	0.0347 (8)	0.0271 (7)	0.0390 (8)	0.0021 (7)	-0.0014 (6)	0.0043 (6)
C2	0.0352 (8)	0.0283 (8)	0.0455 (9)	-0.0064 (7)	0.0008 (6)	0.0055 (7)
C3	0.0243 (7)	0.0283 (8)	0.0365 (8)	-0.0006 (6)	0.0017 (5)	-0.0027 (6)
C4	0.0223 (6)	0.0265 (7)	0.0347 (7)	0.0003 (6)	0.0027 (5)	-0.0008 (6)
C5	0.0233 (6)	0.0255 (7)	0.0291 (7)	0.0001 (5)	0.0050 (5)	-0.0019 (6)
C6	0.0320 (8)	0.0273 (8)	0.0477 (9)	-0.0029 (7)	0.0039 (7)	0.0012 (7)
C7	0.0347 (9)	0.0406 (10)	0.0529 (10)	-0.0109 (8)	0.0072 (7)	0.0021 (8)
C8	0.0267 (7)	0.0466 (10)	0.0418 (9)	-0.0057 (7)	0.0039 (6)	-0.0079 (8)
C9	0.0268 (7)	0.0371 (9)	0.0382 (8)	0.0044 (7)	0.0006 (6)	-0.0045 (7)
C10	0.0267 (6)	0.0274 (7)	0.0278 (7)	0.0006 (6)	0.0026 (5)	-0.0010 (6)
C11	0.0353 (9)	0.0544 (12)	0.0501 (11)	0.0060 (8)	-0.0077 (8)	0.0034 (9)
C12	0.0358 (9)	0.0719 (14)	0.0454 (10)	-0.0014 (10)	-0.0050 (7)	-0.0045 (11)
C13	0.0367 (8)	0.0486 (10)	0.0301 (7)	-0.0023 (7)	0.0060 (6)	-0.0045 (7)
C14	0.0300 (8)	0.0750 (15)	0.0665 (12)	-0.0064 (10)	0.0132 (8)	-0.0086 (12)
C15	0.0383 (8)	0.0364 (9)	0.0415 (8)	-0.0063 (7)	-0.0041 (7)	0.0003 (8)
C16	0.0373 (9)	0.0492 (11)	0.0660 (12)	-0.0129 (9)	-0.0051 (8)	-0.0014 (10)
C17	0.0290 (7)	0.0357 (9)	0.0551 (10)	0.0030 (7)	0.0107 (7)	-0.0073 (8)
C18	0.0372 (8)	0.0409 (9)	0.0419 (9)	0.0002 (7)	-0.0021 (6)	0.0117 (7)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

O1—C12	1.354 (3)	C7—H7B	0.9900
O1—C8	1.479 (2)	C8—C14	1.532 (2)
O2—C12	1.203 (2)	C8—C9	1.546 (3)

O3—C15	1.343 (2)	C9—C11	1.534 (2)
O3—C3	1.4601 (18)	C9—C10	1.565 (2)
O4—C15	1.197 (3)	C9—H9	1.0000
C1—C2	1.526 (2)	C10—C13	1.540 (2)
C1—C10	1.536 (2)	C11—C12	1.504 (3)
C1—H1A	0.9900	C11—H11A	0.9900
C1—H1B	0.9900	C11—H11B	0.9900
C2—C3	1.512 (2)	C13—H13A	0.9800
C2—H2A	0.9900	C13—H13B	0.9800
C2—H2B	0.9900	C13—H13C	0.9800
C3—C4	1.542 (2)	C14—H14A	0.9800
C3—H3	1.0000	C14—H14B	0.9800
C4—C17	1.537 (2)	C14—H14C	0.9800
C4—C18	1.540 (2)	C15—C16	1.498 (2)
C4—C5	1.5696 (18)	C16—H16A	0.9800
C5—C6	1.532 (2)	C16—H16B	0.9800
C5—C10	1.551 (2)	C16—H16C	0.9800
C5—H5	1.0000	C17—H17A	0.9800
C6—C7	1.529 (2)	C17—H17B	0.9800
C6—H6A	0.9900	C17—H17C	0.9800
C6—H6B	0.9900	C18—H18A	0.9800
C7—C8	1.516 (3)	C18—H18B	0.9800
C7—H7A	0.9900	C18—H18C	0.9800
C12—O1—C8	109.23 (16)	C8—C9—C10	116.08 (13)
C15—O3—C3	117.60 (13)	C11—C9—H9	108.8
C2—C1—C10	112.21 (13)	C8—C9—H9	108.8
C2—C1—H1A	109.2	C10—C9—H9	108.8
C10—C1—H1A	109.2	C1—C10—C13	109.04 (14)
C2—C1—H1B	109.2	C1—C10—C5	108.73 (11)
C10—C1—H1B	109.2	C13—C10—C5	112.98 (13)
H1A—C1—H1B	107.9	C1—C10—C9	107.23 (13)
C3—C2—C1	109.55 (13)	C13—C10—C9	111.77 (12)
C3—C2—H2A	109.8	C5—C10—C9	106.90 (12)
C1—C2—H2A	109.8	C12—C11—C9	103.26 (17)
C3—C2—H2B	109.8	C12—C11—H11A	111.1
C1—C2—H2B	109.8	C9—C11—H11A	111.1
H2A—C2—H2B	108.2	C12—C11—H11B	111.1
O3—C3—C2	108.86 (13)	C9—C11—H11B	111.1
O3—C3—C4	106.99 (12)	H11A—C11—H11B	109.1
C2—C3—C4	114.73 (13)	O2—C12—O1	120.5 (2)
O3—C3—H3	108.7	O2—C12—C11	129.2 (3)
C2—C3—H3	108.7	O1—C12—C11	110.28 (15)
C4—C3—H3	108.7	C10—C13—H13A	109.5
C17—C4—C18	108.03 (14)	C10—C13—H13B	109.5
C17—C4—C3	111.00 (13)	H13A—C13—H13B	109.5
C18—C4—C3	107.13 (13)	C10—C13—H13C	109.5
C17—C4—C5	114.41 (13)	H13A—C13—H13C	109.5
C18—C4—C5	107.97 (12)	H13B—C13—H13C	109.5
C3—C4—C5	108.02 (12)	C8—C14—H14A	109.5

## supplementary materials

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C6—C5—C10	109.50 (12)	C8—C14—H14B	109.5
C6—C5—C4	112.88 (12)	H14A—C14—H14B	109.5
C10—C5—C4	117.40 (12)	C8—C14—H14C	109.5
C6—C5—H5	105.3	H14A—C14—H14C	109.5
C10—C5—H5	105.3	H14B—C14—H14C	109.5
C4—C5—H5	105.3	O4—C15—O3	123.78 (16)
C7—C6—C5	110.17 (14)	O4—C15—C16	125.24 (18)
C7—C6—H6A	109.6	O3—C15—C16	110.98 (17)
C5—C6—H6A	109.6	C15—C16—H16A	109.5
C7—C6—H6B	109.6	C15—C16—H16B	109.5
C5—C6—H6B	109.6	H16A—C16—H16B	109.5
H6A—C6—H6B	108.1	C15—C16—H16C	109.5
C8—C7—C6	115.88 (15)	H16A—C16—H16C	109.5
C8—C7—H7A	108.3	H16B—C16—H16C	109.5
C6—C7—H7A	108.3	C4—C17—H17A	109.5
C8—C7—H7B	108.3	C4—C17—H17B	109.5
C6—C7—H7B	108.3	H17A—C17—H17B	109.5
H7A—C7—H7B	107.4	C4—C17—H17C	109.5
O1—C8—C7	109.08 (16)	H17A—C17—H17C	109.5
O1—C8—C14	106.37 (14)	H17B—C17—H17C	109.5
C7—C8—C14	109.27 (17)	C4—C18—H18A	109.5
O1—C8—C9	102.92 (14)	C4—C18—H18B	109.5
C7—C8—C9	116.62 (13)	H18A—C18—H18B	109.5
C14—C8—C9	111.92 (17)	C4—C18—H18C	109.5
C11—C9—C8	100.65 (14)	H18A—C18—H18C	109.5
C11—C9—C10	113.45 (15)	H18B—C18—H18C	109.5
C10—C1—C2—C3	-60.61 (18)	C14—C8—C9—C11	-77.70 (18)
C15—O3—C3—C2	77.96 (18)	O1—C8—C9—C10	-86.75 (15)
C15—O3—C3—C4	-157.52 (14)	C7—C8—C9—C10	32.6 (2)
C1—C2—C3—O3	178.97 (13)	C14—C8—C9—C10	159.43 (15)
C1—C2—C3—C4	59.15 (18)	C2—C1—C10—C13	-68.69 (16)
O3—C3—C4—C17	-45.26 (17)	C2—C1—C10—C5	54.89 (17)
C2—C3—C4—C17	75.59 (16)	C2—C1—C10—C9	170.13 (13)
O3—C3—C4—C18	72.47 (15)	C6—C5—C10—C1	-179.85 (13)
C2—C3—C4—C18	-166.68 (13)	C4—C5—C10—C1	-49.46 (17)
O3—C3—C4—C5	-171.45 (11)	C6—C5—C10—C13	-58.66 (16)
C2—C3—C4—C5	-50.60 (17)	C4—C5—C10—C13	71.73 (17)
C17—C4—C5—C6	51.22 (18)	C6—C5—C10—C9	64.69 (15)
C18—C4—C5—C6	-69.07 (17)	C4—C5—C10—C9	-164.91 (13)
C3—C4—C5—C6	175.39 (13)	C11—C9—C10—C1	79.44 (17)
C17—C4—C5—C10	-77.59 (18)	C8—C9—C10—C1	-164.68 (14)
C18—C4—C5—C10	162.11 (14)	C11—C9—C10—C13	-40.0 (2)
C3—C4—C5—C10	46.57 (17)	C8—C9—C10—C13	75.88 (18)
C10—C5—C6—C7	-65.31 (17)	C11—C9—C10—C5	-164.10 (15)
C4—C5—C6—C7	161.90 (13)	C8—C9—C10—C5	-48.22 (17)
C5—C6—C7—C8	47.1 (2)	C8—C9—C11—C12	-31.47 (18)
C12—O1—C8—C7	-152.81 (14)	C10—C9—C11—C12	93.22 (18)
C12—O1—C8—C14	89.45 (19)	C8—O1—C12—O2	-172.27 (18)
C12—O1—C8—C9	-28.36 (17)	C8—O1—C12—C11	7.91 (19)

C6—C7—C8—O1	84.82 (18)	C9—C11—C12—O2	-163.8 (2)
C6—C7—C8—C14	-159.28 (17)	C9—C11—C12—O1	16.0 (2)
C6—C7—C8—C9	-31.1 (2)	C3—O3—C15—O4	9.9 (3)
O1—C8—C9—C11	36.12 (16)	C3—O3—C15—C16	-170.25 (14)
C7—C8—C9—C11	155.46 (16)		

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C16—H16B $\cdots$ O2 <sup>i</sup>	0.98	2.55	3.383 (3)	143

Symmetry codes: (i)  $-x+1, y+1/2, -z+1$ .

Fig. 1

