

## (4-*tert*-Butylphenyl)acetic acid

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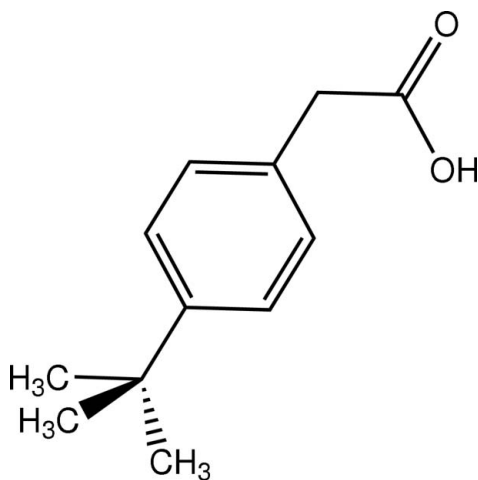
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 Key indicators: single-crystal X-ray study;  $T = 295$  K; mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å; disorder in main residue;  $R$  factor = 0.075;  $wR$  factor = 0.220; data-to-parameter ratio = 15.6.

In the title compound,  $\text{C}_{12}\text{H}_{16}\text{O}_2$ , the plane of the carboxylic acid group is almost perpendicular to the benzene ring [dihedral angle  $80.9(3)^\circ$ ] and the *tert*-butyl unit is disordered over two sets of sites in a 0.503(6):0.497(6) ratio. In the crystal structure, centrosymmetric dimers arise from pairs of  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds involving the carboxylic acid groups.

### Related literature

For general background, see: Liu *et al.* (2006). For a related structure, see: van Koningsveld (1982).



### Experimental

#### Crystal data

$\text{C}_{12}\text{H}_{16}\text{O}_2$	$V = 2327.8(10)$ Å <sup>3</sup>
$M_r = 192.25$	$Z = 8$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 11.209(2)$ Å	$\mu = 0.07$ mm <sup>-1</sup>
$b = 12.442(3)$ Å	$T = 295(2)$ K
$c = 17.250(5)$ Å	$0.30 \times 0.23 \times 0.16$ mm
$\beta = 104.625(12)^\circ$	

#### Data collection

Bruker APEX CCD diffractometer	2047 independent reflections
Absorption correction: none	1375 reflections with $I > 2\sigma(I)$
5829 measured reflections	$R_{\text{int}} = 0.025$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.074$	131 parameters
$wR(F^2) = 0.220$	H-atom parameters constrained
$S = 1.06$	$\Delta\rho_{\text{max}} = 0.30$ e Å <sup>-3</sup>
2047 reflections	$\Delta\rho_{\text{min}} = -0.25$ e Å <sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1}\cdots\text{O2}^i$	0.92	1.74	2.659(3)	176

 Symmetry code: (i)  $-x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$ .

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2766).

### References

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**supplementary materials**

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## (4-*tert*-Butylphenyl)acetic acid

B.-X. Liu, Y.-P. Yu and D.-J. Xu

### Comment

As part of investigation on the nature of aromatic stacking (Liu *et al.*, 2006), the title compound, (I), has recently been prepared in Prof. Liu's laboratory by the hydrolyzation reaction of 2-(4-*tert*-butylphenyl)-1-morpholinoethanethione. Herein we present its X-ray structure (Fig. 1).

The C8—O1 bond distance of 1.309 (3) Å is significantly longer than the C8—O2 bond distance of 1.191 (3) Å. The carboxyl group is nearly perpendicular to the benzene plane, the dihedral angle being 80.9 (3)°. The adjacent molecules are linked together *via* O—H···O hydrogen bonding (Table 1) to form a centrosymmetric supramolecular dimer as shown in Fig. 2, which is comparable to that found in 4-*tert*-butylbenzoic acid (van Koningsveld, 1982).

### Experimental

The title compound was prepared by a hydrolyzation reaction of 2-(4-*tert*-butylphenyl)-1-morpholinoethanethione (55 g) in a solution containing CH<sub>3</sub>COOH (150 ml), H<sub>2</sub>SO<sub>4</sub> (25 ml, 98%) and water (30 ml) at 390 K until the reaction mixture changed colour to dark-green. After cooling to room temperature, the solid product was separated from the reaction mixture, and colourless prisms of (I) were obtained by recrystallization of the solid product from an ethanol–water solution (1:1 v/v) after 2 months.

### Refinement

The carboxyl H atom was located in a difference Fourier map and refined as riding in its as-found relative position with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ . The methyl H atoms were placed in calculated positions with C—H = 0.96 Å and torsion angles were refined to fit the electron density with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ . The other H atoms were placed in calculated positions with C—H = 0.97 Å (methylene) or 0.93 Å (aromatic), and refined in riding mode with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . The *tert*-butyl group is disordered over two positions. Occupancies were initially refined and converged to 0.503 (6) and 0.497 (6), respectively; in the final cycles of refinement, the occupancies were fixed at 0.5, and displacement parameters for the disordered carbon atoms were constrained to be identical.

### Figures

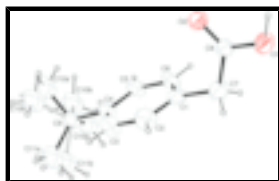


Fig. 1. The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). Double dashed lines indicate one of disordered components.

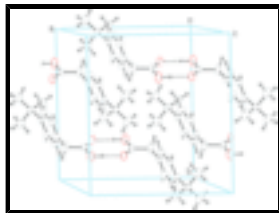


Fig. 2. The unit cell packing diagram showing the supra-molecular dimeric structure linked by the hydrogen bonding (dashed lines).

## (4-*tert*-Butylphenyl)acetic acid

### Crystal data

$C_{12}H_{16}O_2$

$M_r = 192.25$

Monoclinic,  $C2/c$

Hall symbol:  $-C\ 2yc$

$a = 11.209\ (2)\ \text{\AA}$

$b = 12.442\ (3)\ \text{\AA}$

$c = 17.250\ (5)\ \text{\AA}$

$\beta = 104.625\ (12)^\circ$

$V = 2327.8\ (10)\ \text{\AA}^3$

$Z = 8$

$F_{000} = 832$

$D_x = 1.097\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation

$\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2180 reflections

$\theta = 2.8\text{--}25.0^\circ$

$\mu = 0.07\ \text{mm}^{-1}$

$T = 295\ (2)\ \text{K}$

Prism, colourless

$0.30 \times 0.23 \times 0.16\ \text{mm}$

### Data collection

Bruker APEXII CCD  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution:  $10\ \text{pixels mm}^{-1}$

$T = 295\ (2)\ \text{K}$

$\varphi$  and  $\omega$  scans

Absorption correction: none

5829 measured reflections

2047 independent reflections

1375 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.025$

$\theta_{\text{max}} = 25.0^\circ$

$\theta_{\text{min}} = 2.4^\circ$

$h = -13 \rightarrow 12$

$k = -11 \rightarrow 14$

$l = -20 \rightarrow 17$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.075$

$wR(F^2) = 0.220$

$S = 1.06$

2047 reflections

131 parameters

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0943P)^2 + 2.2585P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.003$

$\Delta\rho_{\text{max}} = 0.30\ \text{e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.25\ \text{e \AA}^{-3}$

Primary atom site location: structure-invariant direct methods Extinction correction: none

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
O1	0.3924 (2)	0.7974 (2)	0.56682 (15)	0.1086 (10)	
H1	0.3088	0.7959	0.5607	0.163*	
O2	0.35115 (18)	0.70221 (19)	0.45614 (13)	0.0801 (7)	
C1	0.6042 (2)	0.6780 (3)	0.45859 (19)	0.0733 (9)	
C2	0.6468 (3)	0.5756 (3)	0.4707 (2)	0.0904 (11)	
H2	0.6448	0.5403	0.5179	0.108*	
C3	0.6930 (3)	0.5233 (3)	0.41409 (19)	0.0838 (10)	
H3	0.7216	0.4532	0.4244	0.101*	
C4	0.6984 (2)	0.5702 (2)	0.34325 (15)	0.0582 (7)	
C5	0.6512 (3)	0.6719 (3)	0.33048 (18)	0.0714 (9)	
H5	0.6498	0.7060	0.2823	0.086*	
C6	0.6056 (3)	0.7253 (3)	0.3870 (2)	0.0819 (10)	
H6	0.5753	0.7948	0.3764	0.098*	
C7	0.5608 (3)	0.7390 (4)	0.5219 (2)	0.1051 (14)	
H7A	0.5971	0.7062	0.5735	0.126*	
H7B	0.5917	0.8120	0.5237	0.126*	
C8	0.4244 (3)	0.7432 (3)	0.51024 (17)	0.0653 (8)	
C9	0.7547 (3)	0.5128 (3)	0.28348 (18)	0.0732 (9)	
C10A	0.8271 (11)	0.6028 (9)	0.2430 (6)	0.1125 (15)	0.497 (6)
H10A	0.8575	0.5698	0.2015	0.169*	0.497 (6)
H10B	0.8950	0.6315	0.2833	0.169*	0.497 (6)
H10C	0.7713	0.6598	0.2207	0.169*	0.497 (6)
C11A	0.8538 (13)	0.4343 (11)	0.3166 (9)	0.1125 (15)	0.497 (6)
H11A	0.8187	0.3714	0.3344	0.169*	0.497 (6)
H11B	0.9124	0.4660	0.3611	0.169*	0.497 (6)
H11C	0.8944	0.4147	0.2759	0.169*	0.497 (6)
C12A	0.6605 (9)	0.4711 (9)	0.2153 (6)	0.1125 (15)	0.497 (6)
H12A	0.6206	0.5299	0.1829	0.169*	0.497 (6)
H12B	0.6006	0.4309	0.2344	0.169*	0.497 (6)
H12C	0.6985	0.4251	0.1838	0.169*	0.497 (6)
C10B	0.7561 (11)	0.5733 (9)	0.2101 (6)	0.1125 (15)	0.503 (6)

## supplementary materials

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H10D	0.7923	0.5297	0.1761	0.169*	0.503 (6)
H10E	0.8037	0.6377	0.2243	0.169*	0.503 (6)
H10F	0.6732	0.5917	0.1823	0.169*	0.503 (6)
C11B	0.8802 (13)	0.4692 (11)	0.3269 (8)	0.1125 (15)	0.503 (6)
H11D	0.8717	0.4232	0.3698	0.169*	0.503 (6)
H11E	0.9342	0.5279	0.3482	0.169*	0.503 (6)
H11F	0.9142	0.4289	0.2900	0.169*	0.503 (6)
C12B	0.6679 (9)	0.4091 (9)	0.2531 (6)	0.1125 (15)	0.503 (6)
H12D	0.5836	0.4318	0.2348	0.169*	0.503 (6)
H12E	0.6756	0.3592	0.2965	0.169*	0.503 (6)
H12F	0.6931	0.3750	0.2099	0.169*	0.503 (6)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0599 (14)	0.172 (3)	0.0991 (17)	-0.0057 (14)	0.0303 (12)	-0.0708 (17)
O2	0.0532 (12)	0.1151 (18)	0.0758 (14)	-0.0020 (11)	0.0231 (10)	-0.0310 (12)
C1	0.0408 (15)	0.112 (3)	0.0682 (19)	0.0018 (15)	0.0160 (13)	-0.0220 (18)
C2	0.098 (3)	0.114 (3)	0.068 (2)	0.007 (2)	0.0381 (19)	0.0096 (19)
C3	0.105 (3)	0.078 (2)	0.078 (2)	0.0180 (19)	0.0417 (19)	0.0145 (17)
C4	0.0506 (15)	0.0689 (17)	0.0574 (15)	0.0100 (13)	0.0178 (12)	0.0069 (13)
C5	0.0659 (18)	0.084 (2)	0.0675 (18)	0.0235 (16)	0.0221 (14)	0.0174 (15)
C6	0.064 (2)	0.086 (2)	0.094 (2)	0.0292 (16)	0.0180 (17)	-0.0029 (18)
C7	0.0542 (19)	0.171 (4)	0.091 (2)	0.003 (2)	0.0214 (17)	-0.053 (3)
C8	0.0546 (17)	0.086 (2)	0.0594 (16)	0.0042 (14)	0.0218 (14)	-0.0110 (15)
C9	0.0687 (19)	0.091 (2)	0.0630 (17)	0.0215 (16)	0.0231 (15)	0.0022 (16)
C10A	0.116 (3)	0.136 (4)	0.098 (3)	0.029 (3)	0.050 (3)	-0.008 (2)
C11A	0.116 (3)	0.136 (4)	0.098 (3)	0.029 (3)	0.050 (3)	-0.008 (2)
C12A	0.116 (3)	0.136 (4)	0.098 (3)	0.029 (3)	0.050 (3)	-0.008 (2)
C10B	0.116 (3)	0.136 (4)	0.098 (3)	0.029 (3)	0.050 (3)	-0.008 (2)
C11B	0.116 (3)	0.136 (4)	0.098 (3)	0.029 (3)	0.050 (3)	-0.008 (2)
C12B	0.116 (3)	0.136 (4)	0.098 (3)	0.029 (3)	0.050 (3)	-0.008 (2)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

O1—C8	1.309 (3)	C9—C11B	1.517 (15)
O1—H1	0.9169	C9—C12B	1.621 (11)
O2—C8	1.191 (3)	C9—C10A	1.639 (11)
C1—C2	1.358 (5)	C10A—H10A	0.9600
C1—C6	1.371 (5)	C10A—H10B	0.9600
C1—C7	1.508 (4)	C10A—H10C	0.9600
C2—C3	1.379 (4)	C11A—H11A	0.9600
C2—H2	0.9300	C11A—H11B	0.9600
C3—C4	1.369 (4)	C11A—H11C	0.9600
C3—H3	0.9300	C12A—H12A	0.9600
C4—C5	1.368 (4)	C12A—H12B	0.9600
C4—C9	1.517 (4)	C12A—H12C	0.9600
C5—C6	1.381 (4)	C10B—H10D	0.9600
C5—H5	0.9300	C10B—H10E	0.9600

C6—H6	0.9300	C10B—H10F	0.9600
C7—C8	1.491 (4)	C11B—H11D	0.9600
C7—H7A	0.9700	C11B—H11E	0.9600
C7—H7B	0.9700	C11B—H11F	0.9600
C9—C12A	1.463 (10)	C12B—H12D	0.9600
C9—C10B	1.476 (10)	C12B—H12E	0.9600
C9—C11A	1.480 (15)	C12B—H12F	0.9600
C8—O1—H1	111.6	C10B—C9—C12B	105.3 (5)
C2—C1—C6	117.3 (3)	C4—C9—C12B	106.0 (4)
C2—C1—C7	121.7 (3)	C11B—C9—C12B	106.3 (6)
C6—C1—C7	121.0 (4)	C12A—C9—C10A	103.5 (6)
C1—C2—C3	120.9 (3)	C11A—C9—C10A	102.2 (7)
C1—C2—H2	119.5	C4—C9—C10A	107.7 (4)
C3—C2—H2	119.5	C9—C10A—H10A	109.5
C4—C3—C2	122.7 (3)	C9—C10A—H10B	109.5
C4—C3—H3	118.7	C9—C10A—H10C	109.5
C2—C3—H3	118.7	C9—C11A—H11A	109.5
C5—C4—C3	115.8 (3)	C9—C11A—H11B	109.5
C5—C4—C9	122.4 (3)	C9—C11A—H11C	109.5
C3—C4—C9	121.7 (3)	C9—C12A—H12A	109.5
C4—C5—C6	121.9 (3)	C9—C12A—H12B	109.5
C4—C5—H5	119.0	C9—C12A—H12C	109.5
C6—C5—H5	119.0	C9—C10B—H10D	109.5
C1—C6—C5	121.3 (3)	C9—C10B—H10E	109.5
C1—C6—H6	119.4	H10D—C10B—H10E	109.5
C5—C6—H6	119.4	C9—C10B—H10F	109.5
C8—C7—C1	115.3 (3)	H10D—C10B—H10F	109.5
C8—C7—H7A	108.5	H10E—C10B—H10F	109.5
C1—C7—H7A	108.5	C9—C11B—H11D	109.5
C8—C7—H7B	108.5	C9—C11B—H11E	109.5
C1—C7—H7B	108.5	H11D—C11B—H11E	109.5
H7A—C7—H7B	107.5	C9—C11B—H11F	109.5
O2—C8—O1	122.7 (3)	H11D—C11B—H11F	109.5
O2—C8—C7	124.9 (3)	H11E—C11B—H11F	109.5
O1—C8—C7	112.4 (3)	C9—C12B—H12D	109.5
C12A—C9—C11A	113.3 (6)	C9—C12B—H12E	109.5
C12A—C9—C4	112.0 (4)	H12D—C12B—H12E	109.5
C10B—C9—C4	116.0 (4)	C9—C12B—H12F	109.5
C11A—C9—C4	116.6 (6)	H12D—C12B—H12F	109.5
C10B—C9—C11B	113.3 (7)	H12E—C12B—H12F	109.5
C4—C9—C11B	109.0 (5)		

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O1-H1\cdots O2^i$	0.92	1.74	2.659 (3)	176

Symmetry codes: (i)  $-x+1/2, -y+3/2, -z+1$ .

Fig. 1

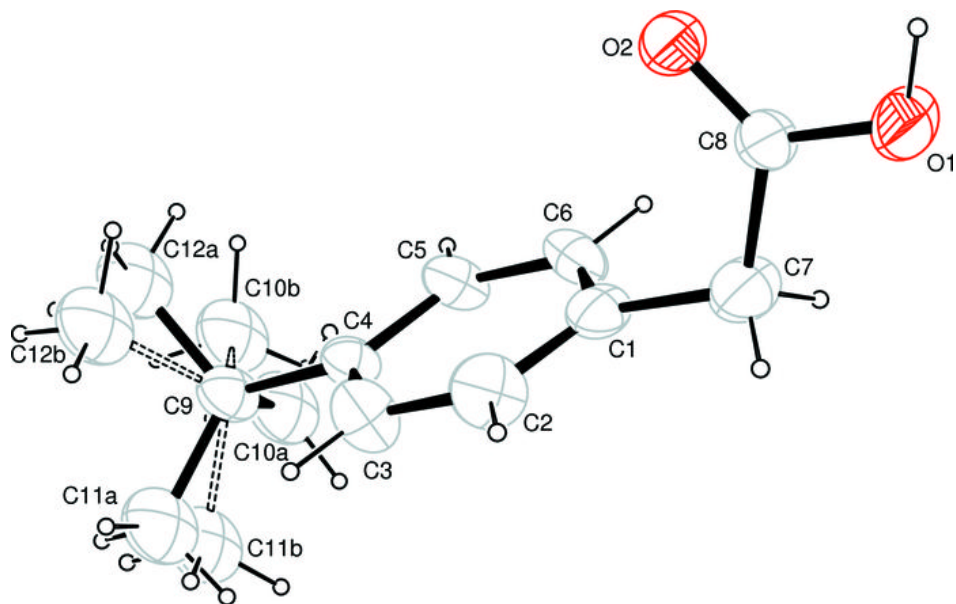


Fig. 2

