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## Structure Reports

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**(Anthracen-9-yl)(piperidin-1-yl)-methanone****Hua-You Hu,<sup>a\*</sup> Yu-Cheng Huang,<sup>b</sup> Hai-Tao Yu<sup>b</sup> and Yan Zhang<sup>b</sup>**

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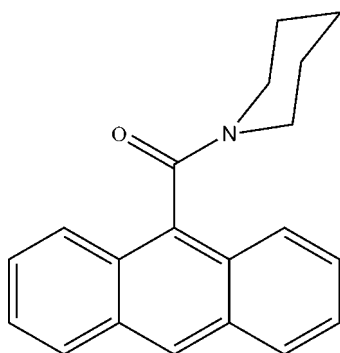
Received 5 September 2008; accepted 13 October 2008

Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å;  $R$  factor = 0.081;  $wR$  factor = 0.278; data-to-parameter ratio = 13.9.

The title compound,  $\text{C}_{20}\text{H}_{19}\text{NO}$ , is a substructure of CP-640186, a potent inhibitor of mammalian acetyl-coenzyme A carboxylases. In the crystal structure, the amide group forms a dihedral angle of  $87.0(1)^\circ$  with the plane of the anthracene unit and the piperidine ring adopts a chair conformation. Molecules are arranged into layers parallel to (100) and adjacent anthracene units within layers form dihedral angles of  $13.2(1)^\circ$ .  $\text{C}-\text{H}\cdots\text{O}$  interactions from the piperidine rings to the  $\text{C}=\text{O}$  group of the amide are observed between layers.

**Related literature**

For further information regarding CP-640186, see: Harwood *et al.* (2003); Zhang *et al.* (2004).

**Experimental***Crystal data*

$\text{C}_{20}\text{H}_{19}\text{NO}$   
 $M_r = 289.36$   
Monoclinic,  $C2/c$   
 $a = 26.393(5)$  Å  
 $b = 7.3950(15)$  Å  
 $c = 18.213(4)$  Å  
 $\beta = 120.29(3)^\circ$   
 $V = 3069.5(14)$  Å<sup>3</sup>  
 $Z = 8$   
Mo  $K\alpha$  radiation  
 $\mu = 0.08$  mm<sup>-1</sup>  
 $T = 293(2)$  K  
 $0.30 \times 0.10 \times 0.10$  mm

*Data collection*

Bruker SMART APEX CCD diffractometer  
Absorption correction: multi-scan (*SADABS*; Bruker, 2000)  
 $T_{\min} = 0.977$ ,  $T_{\max} = 0.992$   
2828 measured reflections  
2762 independent reflections  
1678 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.025$

*Refinement*

$R[F^2 > 2\sigma(F^2)] = 0.081$   
 $wR(F^2) = 0.278$   
 $S = 1.09$   
2762 reflections  
199 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.32$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.29$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C17}-\text{H17B}\cdots\text{O1}^i$	0.97	2.41	3.342 (5)	162
$\text{C20}-\text{H20A}\cdots\text{O1}^{ii}$	0.97	2.71	3.557 (5)	146

Symmetry codes: (i)  $x, y + 1, z$ ; (ii)  $-x + 1, -y, -z + 1$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: B12300).

**References**

- Bruker (2000). *SADABS*, *SAINT* and *SMART*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Harwood, H. J., Petras, S. F., Shelly, L. D., Zaccaro, L. M., Perry, D. A., Makowski, M. R., Hargrove, D. M., Martin, K. A., Tracey, W. R., Chapman, J. G., Magee, W. P., Dalvie, D. K., Soliman, V. F., Martin, W. H., Mularski, C. J. & Eisenbeis, S. A. (2003). *J. Biol. Chem.* **278**, 37099–37111.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
Zhang, H.-L., Tweel, B., Li, J. & Tong, L. (2004). *Structure*, **12**, 1683–1691.

**supplementary materials**

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## (Anthracen-9-yl)(piperidin-1-yl)methanone

H.-Y. Hu, Y.-C. Huang, H.-T. Yu and Y. Zhang

### Comment

Inhibition of acetyl-CoA carboxylase (ACC), with its resultant inhibition of fatty acid synthesis and stimulation of fatty acid oxidation, has the potential to affect favourably the multitude of cardiovascular risk factors associated with the metabolic syndrome. Recent findings reported by Pfizer researchers show that the isozyme-nonspecific ACC inhibitor CP-640186 inhibits both the lipogenic tissue isozyme (ACC1) and the oxidative tissue isozyme (ACC2) (Harwood *et al.*, 2003). The title compound is a sub-structure of CP-640186 (see Zhang *et al.*, 2004).

### Experimental

A mixture of 9-carbonyl anthracene (1 mmol) and piperidine (1.2 mmol) with 1.5 mmol DCC (DCC = *N,N*-dicyclohexylcarbodiimide) was stirred in 5 ml CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 1 h to yield the title compound. Crystals were obtained from acetone/petroleum ether.

### Refinement

H atoms were positioned geometrically and refined using a riding model with C—H = 0.93–0.97 Å and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

### Figures

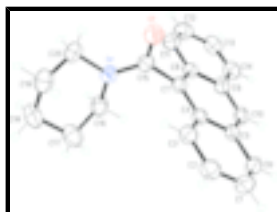


Fig. 1. The molecular structure of the title compound with displacement ellipsoids shown at 30% probability for non-H atoms.

## (Anthracen-9-yl)(piperidin-1-yl)methanone

### Crystal data

C<sub>20</sub>H<sub>19</sub>NO

$M_r = 289.36$

Monoclinic, *C2/c*

Hall symbol: -C 2yc

$a = 26.393(5)$  Å

$F_{000} = 1232$

$D_x = 1.252$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 781 reflections

$\theta = 2.4\text{--}28.0^\circ$

# supplementary materials

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$b = 7.3950 (15) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$c = 18.213 (4) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 120.29 (3)^\circ$	Block, yellow
$V = 3069.5 (14) \text{ \AA}^3$	$0.30 \times 0.10 \times 0.10 \text{ mm}$
$Z = 8$	

## Data collection

Bruker SMART APEX CCD diffractometer	2762 independent reflections
Radiation source: fine-focus sealed tube	1678 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.025$
$T = 293(2) \text{ K}$	$\theta_{\text{max}} = 25.2^\circ$
$\varphi$ and $\omega$ scans	$\theta_{\text{min}} = 1.8^\circ$
Absorption correction: multi-scan (SADABS; Bruker, 2000)	$h = -31 \rightarrow 27$
$T_{\text{min}} = 0.977, T_{\text{max}} = 0.992$	$k = 0 \rightarrow 8$
2828 measured reflections	$l = 0 \rightarrow 21$

## Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.081$	H-atom parameters constrained
$wR(F^2) = 0.278$	$w = 1/[\sigma^2(F_o^2) + (0.1523P)^2 + 2.2222P]$
$S = 1.09$	where $P = (F_o^2 + 2F_c^2)/3$
2762 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
199 parameters	$\Delta\rho_{\text{max}} = 0.32 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$
	Extinction correction: none

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
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C1	0.83354 (17)	0.1537 (5)	0.6866 (3)	0.0733 (11)
H1A	0.8709	0.1589	0.6936	0.088*
C2	0.78453 (17)	0.1309 (5)	0.6048 (3)	0.0680 (10)
H2A	0.7896	0.1254	0.5579	0.082*
C3	0.72989 (15)	0.1169 (5)	0.5938 (2)	0.0577 (9)
H3A	0.6980	0.0993	0.5394	0.069*
C4	0.72079 (13)	0.1286 (4)	0.66384 (19)	0.0472 (8)
C5	0.77079 (14)	0.1544 (4)	0.7464 (2)	0.0504 (8)
C6	0.82684 (15)	0.1679 (5)	0.7541 (2)	0.0611 (9)
H6A	0.8596	0.1871	0.8074	0.073*
C7	0.66500 (13)	0.1159 (4)	0.65502 (19)	0.0480 (8)
C8	0.65705 (14)	0.1315 (4)	0.7257 (2)	0.0478 (8)
C9	0.70746 (15)	0.1503 (4)	0.8081 (2)	0.0534 (8)
C10	0.76293 (14)	0.1644 (4)	0.8162 (2)	0.0538 (9)
H10A	0.7956	0.1809	0.8699	0.065*
C11	0.60162 (16)	0.1229 (5)	0.7191 (3)	0.0626 (10)
H11A	0.5682	0.1098	0.6658	0.075*
C12	0.59630 (19)	0.1335 (5)	0.7889 (3)	0.0751 (12)
H12A	0.5591	0.1308	0.7826	0.090*
C13	0.6455 (2)	0.1485 (5)	0.8706 (3)	0.0718 (11)
H13A	0.6409	0.1535	0.9180	0.086*
C14	0.69916 (19)	0.1556 (5)	0.8799 (2)	0.0677 (10)
H14A	0.7317	0.1642	0.9343	0.081*
C15	0.61267 (14)	0.0678 (5)	0.5689 (2)	0.0524 (8)
C16	0.59033 (17)	0.3956 (5)	0.5392 (2)	0.0693 (11)
H16A	0.6271	0.4108	0.5922	0.083*
H16B	0.5593	0.4463	0.5465	0.083*
C17	0.5933 (2)	0.4948 (6)	0.4705 (3)	0.0870 (13)
H17A	0.6274	0.4549	0.4680	0.104*
H17B	0.5975	0.6232	0.4831	0.104*
C18	0.5372 (2)	0.4617 (6)	0.3833 (3)	0.0848 (13)
H18A	0.5036	0.5164	0.3828	0.102*
H18B	0.5416	0.5166	0.3384	0.102*
C19	0.52736 (19)	0.2577 (7)	0.3679 (2)	0.0861 (13)
H19A	0.4905	0.2372	0.3156	0.103*
H19B	0.5585	0.2072	0.3607	0.103*
C20	0.52607 (17)	0.1630 (6)	0.4382 (2)	0.0798 (12)
H20A	0.4917	0.2008	0.4406	0.096*
H20B	0.5233	0.0337	0.4281	0.096*
N1	0.57930 (13)	0.2033 (4)	0.51987 (17)	0.0652 (9)
O1	0.60244 (11)	-0.0901 (3)	0.54889 (16)	0.0754 (9)

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.058 (2)	0.078 (3)	0.090 (3)	0.0004 (18)	0.042 (2)	0.008 (2)
C2	0.076 (3)	0.071 (2)	0.071 (2)	0.0052 (19)	0.048 (2)	0.0061 (19)
C3	0.059 (2)	0.063 (2)	0.0532 (19)	0.0000 (16)	0.0293 (17)	0.0003 (15)

## supplementary materials

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C4	0.0487 (18)	0.0408 (16)	0.0483 (18)	0.0024 (13)	0.0216 (15)	0.0022 (13)
C5	0.0448 (17)	0.0440 (17)	0.0513 (18)	0.0004 (13)	0.0160 (15)	0.0049 (14)
C6	0.0438 (19)	0.060 (2)	0.068 (2)	0.0006 (15)	0.0189 (17)	-0.0010 (17)
C7	0.0475 (18)	0.0449 (18)	0.0438 (17)	0.0006 (13)	0.0173 (14)	0.0000 (13)
C8	0.0527 (19)	0.0420 (17)	0.0519 (18)	-0.0014 (13)	0.0286 (16)	0.0008 (13)
C9	0.059 (2)	0.0469 (18)	0.0481 (18)	0.0035 (15)	0.0229 (16)	-0.0049 (14)
C10	0.0507 (19)	0.0506 (19)	0.0459 (18)	0.0015 (14)	0.0139 (15)	0.0006 (14)
C11	0.059 (2)	0.062 (2)	0.071 (2)	-0.0058 (16)	0.0350 (19)	-0.0031 (17)
C12	0.078 (3)	0.077 (3)	0.093 (3)	-0.001 (2)	0.060 (3)	-0.002 (2)
C13	0.099 (3)	0.070 (2)	0.068 (2)	0.011 (2)	0.058 (2)	0.0023 (19)
C14	0.084 (3)	0.068 (2)	0.054 (2)	-0.0005 (19)	0.036 (2)	-0.0040 (17)
C15	0.0497 (18)	0.057 (2)	0.0485 (18)	-0.0007 (15)	0.0238 (15)	-0.0044 (16)
C16	0.070 (2)	0.066 (2)	0.057 (2)	0.0051 (18)	0.0207 (19)	-0.0063 (18)
C17	0.112 (4)	0.063 (3)	0.085 (3)	0.003 (2)	0.049 (3)	0.003 (2)
C18	0.110 (3)	0.076 (3)	0.064 (2)	0.013 (2)	0.040 (2)	0.016 (2)
C19	0.086 (3)	0.103 (4)	0.050 (2)	-0.007 (2)	0.020 (2)	-0.003 (2)
C20	0.060 (2)	0.092 (3)	0.058 (2)	-0.012 (2)	0.0075 (19)	0.003 (2)
N1	0.0575 (18)	0.0608 (19)	0.0531 (17)	-0.0090 (14)	0.0100 (14)	0.0008 (14)
O1	0.0812 (19)	0.0499 (15)	0.0646 (16)	-0.0093 (12)	0.0144 (14)	-0.0150 (12)

### *Geometric parameters (Å, °)*

C1—C6	1.333 (5)	C12—H12A	0.930
C1—C2	1.406 (5)	C13—C14	1.339 (6)
C1—H1A	0.930	C13—H13A	0.930
C2—C3	1.356 (5)	C14—H14A	0.930
C2—H2A	0.930	C15—O1	1.213 (4)
C3—C4	1.415 (4)	C15—N1	1.336 (4)
C3—H3A	0.930	C16—N1	1.458 (5)
C4—C7	1.401 (4)	C16—C17	1.486 (6)
C4—C5	1.427 (4)	C16—H16A	0.970
C5—C10	1.389 (5)	C16—H16B	0.970
C5—C6	1.417 (5)	C17—C18	1.549 (6)
C6—H6A	0.930	C17—H17A	0.970
C7—C8	1.409 (4)	C17—H17B	0.970
C7—C15	1.519 (4)	C18—C19	1.532 (6)
C8—C11	1.407 (5)	C18—H18A	0.970
C8—C9	1.424 (5)	C18—H18B	0.970
C9—C10	1.399 (5)	C19—C20	1.476 (6)
C9—C14	1.433 (5)	C19—H19A	0.970
C10—H10A	0.930	C19—H19B	0.970
C11—C12	1.351 (5)	C20—N1	1.472 (4)
C11—H11A	0.930	C20—H20A	0.970
C12—C13	1.401 (6)	C20—H20B	0.970
C6—C1—C2	120.5 (3)	C13—C14—C9	121.5 (4)
C6—C1—H1A	119.7	C13—C14—H14A	119.3
C2—C1—H1A	119.7	C9—C14—H14A	119.3
C3—C2—C1	120.5 (4)	O1—C15—N1	123.3 (3)
C3—C2—H2A	119.8	O1—C15—C7	119.0 (3)

C1—C2—H2A	119.8	N1—C15—C7	117.7 (3)
C2—C3—C4	120.9 (3)	N1—C16—C17	111.5 (3)
C2—C3—H3A	119.5	N1—C16—H16A	109.3
C4—C3—H3A	119.5	C17—C16—H16A	109.3
C7—C4—C3	122.6 (3)	N1—C16—H16B	109.3
C7—C4—C5	119.3 (3)	C17—C16—H16B	109.3
C3—C4—C5	118.1 (3)	H16A—C16—H16B	108.0
C10—C5—C6	122.4 (3)	C16—C17—C18	110.9 (4)
C10—C5—C4	119.0 (3)	C16—C17—H17A	109.5
C6—C5—C4	118.6 (3)	C18—C17—H17A	109.5
C1—C6—C5	121.4 (3)	C16—C17—H17B	109.5
C1—C6—H6A	119.3	C18—C17—H17B	109.5
C5—C6—H6A	119.3	H17A—C17—H17B	108.0
C4—C7—C8	121.4 (3)	C19—C18—C17	109.2 (3)
C4—C7—C15	119.3 (3)	C19—C18—H18A	109.8
C8—C7—C15	119.0 (3)	C17—C18—H18A	109.8
C11—C8—C7	123.0 (3)	C19—C18—H18B	109.8
C11—C8—C9	118.2 (3)	C17—C18—H18B	109.8
C7—C8—C9	118.7 (3)	H18A—C18—H18B	108.3
C10—C9—C8	119.4 (3)	C20—C19—C18	112.6 (4)
C10—C9—C14	122.4 (3)	C20—C19—H19A	109.1
C8—C9—C14	118.2 (3)	C18—C19—H19A	109.1
C5—C10—C9	122.0 (3)	C20—C19—H19B	109.1
C5—C10—H10A	119.0	C18—C19—H19B	109.1
C9—C10—H10A	119.0	H19A—C19—H19B	107.8
C12—C11—C8	120.9 (4)	N1—C20—C19	110.7 (3)
C12—C11—H11A	119.5	N1—C20—H20A	109.5
C8—C11—H11A	119.5	C19—C20—H20A	109.5
C11—C12—C13	121.6 (4)	N1—C20—H20B	109.5
C11—C12—H12A	119.2	C19—C20—H20B	109.5
C13—C12—H12A	119.2	H20A—C20—H20B	108.1
C14—C13—C12	119.5 (4)	C15—N1—C16	125.9 (3)
C14—C13—H13A	120.3	C15—N1—C20	119.7 (3)
C12—C13—H13A	120.3	C16—N1—C20	114.4 (3)
C6—C1—C2—C3	-2.2 (6)	C8—C9—C10—C5	-2.3 (5)
C1—C2—C3—C4	1.3 (5)	C14—C9—C10—C5	178.3 (3)
C2—C3—C4—C7	179.6 (3)	C7—C8—C11—C12	178.1 (3)
C2—C3—C4—C5	-0.4 (5)	C9—C8—C11—C12	0.5 (5)
C7—C4—C5—C10	0.7 (4)	C8—C11—C12—C13	-1.7 (6)
C3—C4—C5—C10	-179.3 (3)	C11—C12—C13—C14	1.1 (6)
C7—C4—C5—C6	-179.7 (3)	C12—C13—C14—C9	0.7 (6)
C3—C4—C5—C6	0.3 (4)	C10—C9—C14—C13	177.5 (3)
C2—C1—C6—C5	2.2 (6)	C8—C9—C14—C13	-1.9 (5)
C10—C5—C6—C1	178.4 (3)	C4—C7—C15—O1	84.2 (4)
C4—C5—C6—C1	-1.2 (5)	C8—C7—C15—O1	-90.4 (4)
C3—C4—C7—C8	-178.8 (3)	C4—C7—C15—N1	-97.4 (4)
C5—C4—C7—C8	1.2 (4)	C8—C7—C15—N1	88.0 (4)
C3—C4—C7—C15	6.7 (5)	N1—C16—C17—C18	54.6 (5)
C5—C4—C7—C15	-173.3 (3)	C16—C17—C18—C19	-53.5 (5)

## supplementary materials

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C4—C7—C8—C11	178.8 (3)	C17—C18—C19—C20	53.9 (5)
C15—C7—C8—C11	-6.7 (5)	C18—C19—C20—N1	-54.1 (5)
C4—C7—C8—C9	-3.6 (4)	O1—C15—N1—C16	-177.9 (4)
C15—C7—C8—C9	170.9 (3)	C7—C15—N1—C16	3.7 (5)
C11—C8—C9—C10	-178.2 (3)	O1—C15—N1—C20	2.1 (5)
C7—C8—C9—C10	4.1 (4)	C7—C15—N1—C20	-176.3 (3)
C11—C8—C9—C14	1.3 (4)	C17—C16—N1—C15	124.0 (4)
C7—C8—C9—C14	-176.5 (3)	C17—C16—N1—C20	-56.0 (5)
C6—C5—C10—C9	-179.8 (3)	C19—C20—N1—C15	-125.0 (4)
C4—C5—C10—C9	-0.1 (5)	C19—C20—N1—C16	55.1 (5)

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C17—H17B $\cdots$ O1 <sup>i</sup>	0.97	2.41	3.342 (5)	162
C20—H20A $\cdots$ O1 <sup>ii</sup>	0.97	2.71	3.557 (5)	146

Symmetry codes: (i)  $x, y+1, z$ ; (ii)  $-x+1, -y, -z+1$ .

