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Ethyl 3-(4-hydroxyphenoxy)-2-(4-methoxyphenoxy)acrylate

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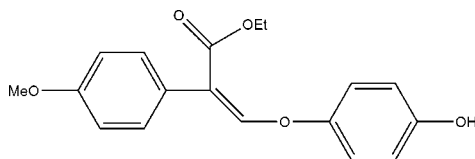
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 Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.051; wR factor = 0.134; data-to-parameter ratio = 8.6.

In the title compound, $\text{C}_{18}\text{H}_{18}\text{O}_5$, the dihedral angle between the two benzene rings is $55.2(3)^\circ$. The ethyl acrylate linkage is planar and forms dihedral angles of $21.3(3)$ and $41.0(3)^\circ$, respectively, with the hydroxyphenyl and methoxyphenyl rings. In the crystal structure, molecules are linked into zigzag chains along the b axis by $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For general background, see: Huang *et al.* (2008); Li *et al.* (2008); Liu *et al.* (2008); Shi *et al.* (2008); Xiao *et al.* (2008). For bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

 $\text{C}_{18}\text{H}_{18}\text{O}_5$
 $M_r = 314.33$

 Orthorhombic, $P2_12_12_1$
 $a = 7.4773(16)$ Å

 $b = 11.661(2)$ Å

 $c = 18.417(4)$ Å

 $V = 1605.8(6)$ Å³
 $Z = 4$

 Mo $K\alpha$ radiation

 $\mu = 0.10$ mm⁻¹
 $T = 298(2)$ K

 $0.40 \times 0.30 \times 0.20$ mm

Data collection

Bruker SMART CCD area-detector diffractometer

Absorption correction: multi-scan

 (*SADABS*; Bruker, 2001)

 $T_{\min} = 0.963$, $T_{\max} = 0.981$

5527 measured reflections

1822 independent reflections

 1486 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.032$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.134$
 $S = 1.05$

1822 reflections

211 parameters

H-atom parameters constrained

 $\Delta\rho_{\max} = 0.26$ e Å⁻³
 $\Delta\rho_{\min} = -0.34$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O4}-\text{H4}\cdots\text{O1}^i$	0.82	2.00	2.812 (3)	169

 Symmetry code: (i) $-x, y + \frac{1}{2}, -z + \frac{3}{2}$.

Data collection: *SMART* (Bruker, 2007); cell refinement: *SAINTE* (Bruker, 2007); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2693).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bruker (2001). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2007). *SMART* and *SAINTE*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Huang, X.-F., Li, H.-Q., Shi, L., Xue, J.-Y. & Zhu, H.-L. (2008). *Chem. Biodiver.* **5**, 636–642.
- Li, H.-Q., Xue, J.-Y., Shi, L., Gui, S.-Y. & Zhu, H.-L. (2008). *Eur. J. Med. Chem.* **43**, 662–667.
- Liu, X.-H., Lv, P.-C., Li, B. & Zhu, H.-L. (2008). *Aust. J. Chem.* **61**, 223–230.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Shi, L., Huang, X.-F., Zhu, Z.-W., Li, H.-Q., Xue, J.-Y. & Zhu, H.-L. (2008). *Aust. J. Chem.* **61**, 472–475.
- Xiao, Z.-P., Li, H.-Q., Xue, J.-Y., Shi, L. & Zhu, H.-L. (2008). *Synth. Commun.* **38**, 525–529.

supplementary materials

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Ethyl 3-(4-hydroxyphenoxy)-2-(4-methoxyphenyl)acrylate

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Comment

Phenylacetate and styrene derivatives are important for their extensive biological activities. Recently a great deal of such kinds of compounds were synthesized, which were found to exhibit good activities (Huang *et al.*, 2008; Li *et al.*, 2008; Liu *et al.*, 2008; Shi *et al.*, 2008; Xiao *et al.*, 2008)

Bond lengths in the title compound (Fig.1) are within normal ranges (Allen *et al.*, 1987). The dihedral angle between the C1—C6 and C7—C12 benzene rings is 55.2 (3)°. The O1/O2/C13—C17 plane forms dihedral angles of 21.3 (3)° and 41.0 (3)°, respectively, with C1—C6 and C7—C12 benzene rings. In the crystal structure, O—H···O hydrogen bonds (Table 1) link the molecules into zigzag chains along the *b* axis (Fig. 2).

Experimental

Equimolar ethyl 3-bromo-2-(4-methoxyphenyl)acrylate and hydroquinone reacted in chloroform overnight, gave the title compound in high yield (88%). Colourless crystals of the title compound were grown by slow evaporation of a methanol solution.

Refinement

H atoms were positioned geometrically (O—H = 0.82 Å and C—H = 0.93 Å) and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and $1.5U_{\text{eq}}(\text{O})$. In the absence of significant anomalous scattering, Friedel pairs were merged prior to the final refinement.

Figures

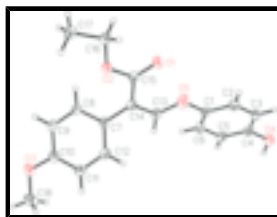


Fig. 1. The molecular structure of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme.



Fig. 2. The crystal structure of the title compound, viewed along the *c* axis. Hydrogen bonds are shown as dashed lines.

Ethyl 3-(4-hydroxyphenoxy)-2-(4-methoxyphenyl)acrylate

Crystal data

$C_{18}H_{18}O_5$	$F_{000} = 664$
$M_r = 314.33$	$D_x = 1.304 \text{ Mg m}^{-3}$
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
Hall symbol: P 2ac 2ab	$\lambda = 0.71073 \text{ \AA}$
$a = 7.4773 (16) \text{ \AA}$	Cell parameters from 1213 reflections
$b = 11.661 (2) \text{ \AA}$	$\theta = 3.2\text{--}26.1^\circ$
$c = 18.417 (4) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$V = 1605.8 (6) \text{ \AA}^3$	$T = 298 (2) \text{ K}$
$Z = 4$	Prism, colourless
	$0.40 \times 0.30 \times 0.20 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	1822 independent reflections
Radiation source: fine-focus sealed tube	1486 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.032$
$T = 298(2) \text{ K}$	$\theta_{\text{max}} = 26.0^\circ$
φ and ω scans	$\theta_{\text{min}} = 2.1^\circ$
Absorption correction: multi-scan (SADABS; Bruker, 2001)	$h = -9 \rightarrow 9$
$T_{\text{min}} = 0.963$, $T_{\text{max}} = 0.981$	$k = -10 \rightarrow 14$
5527 measured reflections	$l = -22 \rightarrow 22$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.051$	H-atom parameters constrained
$wR(F^2) = 0.134$	$w = 1/[\sigma^2(F_o^2) + (0.0689P)^2 + 0.4887P]$
$S = 1.05$	where $P = (F_o^2 + 2F_c^2)/3$
1822 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
211 parameters	$\Delta\rho_{\text{max}} = 0.26 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\text{min}} = -0.34 \text{ e \AA}^{-3}$
	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations

between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.0040 (5)	0.4964 (3)	0.73124 (16)	0.0401 (8)
C2	0.0773 (6)	0.4306 (3)	0.78656 (17)	0.0485 (9)
H2	0.1463	0.3669	0.7750	0.058*
C3	0.0490 (6)	0.4589 (3)	0.85909 (18)	0.0489 (10)
H3	0.0990	0.4144	0.8957	0.059*
C4	-0.0543 (6)	0.5537 (3)	0.87635 (17)	0.0444 (9)
C5	-0.1311 (5)	0.6181 (3)	0.82193 (17)	0.0454 (9)
H5	-0.2019	0.6810	0.8335	0.054*
C6	-0.1024 (5)	0.5887 (3)	0.74921 (17)	0.0471 (9)
H6	-0.1556	0.6319	0.7126	0.057*
C7	0.0419 (5)	0.6087 (3)	0.47594 (16)	0.0370 (7)
C8	-0.0503 (5)	0.5959 (3)	0.41071 (17)	0.0407 (8)
H8	-0.0970	0.5246	0.3984	0.049*
C9	-0.0734 (5)	0.6872 (3)	0.36402 (17)	0.0462 (9)
H9	-0.1336	0.6762	0.3204	0.055*
C10	-0.0083 (5)	0.7946 (3)	0.38124 (18)	0.0441 (9)
C11	0.0842 (5)	0.8107 (3)	0.44550 (19)	0.0463 (9)
H11	0.1287	0.8826	0.4578	0.056*
C12	0.1095 (5)	0.7176 (3)	0.49137 (19)	0.0440 (9)
H12	0.1741	0.7282	0.5340	0.053*
C13	0.0301 (5)	0.5385 (3)	0.60130 (17)	0.0415 (8)
H13	0.0000	0.6138	0.6123	0.050*
C14	0.0590 (5)	0.5141 (3)	0.52943 (17)	0.0394 (8)
C15	0.1049 (5)	0.3973 (3)	0.50927 (17)	0.0397 (8)
C16	0.1644 (7)	0.2682 (3)	0.41167 (18)	0.0513 (10)
H16A	0.0775	0.2121	0.4281	0.062*
H16B	0.2814	0.2462	0.4297	0.062*
C17	0.1658 (6)	0.2743 (3)	0.33027 (17)	0.0549 (10)
H17A	0.0455	0.2826	0.3128	0.082*
H17B	0.2170	0.2053	0.3109	0.082*
H17C	0.2357	0.3390	0.3150	0.082*
C18	-0.0024 (7)	0.9941 (3)	0.3518 (3)	0.0668 (13)
H18A	-0.0656	1.0139	0.3954	0.100*
H18B	-0.0374	1.0450	0.3134	0.100*
H18C	0.1239	1.0008	0.3600	0.100*
O5	0.0407 (4)	0.4650 (3)	0.65807 (14)	0.0662 (8)
O1	0.1315 (4)	0.3188 (2)	0.55214 (12)	0.0541 (7)
O2	0.1174 (4)	0.3810 (2)	0.43764 (12)	0.0468 (7)

supplementary materials

O3	-0.0441 (5)	0.8795 (2)	0.33197 (14)	0.0618 (8)
O4	-0.0756 (5)	0.5803 (2)	0.94804 (12)	0.0601 (8)
H4	-0.1019	0.6483	0.9520	0.090*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.047 (2)	0.0446 (17)	0.0288 (16)	-0.0058 (17)	-0.0002 (15)	-0.0054 (14)
C2	0.062 (2)	0.0465 (19)	0.0373 (17)	-0.003 (2)	0.0026 (18)	-0.0020 (15)
C3	0.071 (3)	0.0439 (18)	0.0320 (16)	-0.008 (2)	-0.0065 (18)	0.0043 (15)
C4	0.060 (2)	0.0450 (19)	0.0281 (15)	-0.011 (2)	0.0040 (16)	-0.0038 (14)
C5	0.053 (2)	0.0458 (19)	0.0379 (18)	0.0008 (19)	0.0010 (17)	-0.0091 (16)
C6	0.054 (2)	0.056 (2)	0.0313 (16)	0.002 (2)	-0.0039 (16)	0.0043 (16)
C7	0.0403 (17)	0.0406 (17)	0.0302 (15)	-0.0005 (17)	0.0009 (15)	-0.0029 (14)
C8	0.046 (2)	0.0405 (18)	0.0352 (17)	-0.0062 (17)	-0.0021 (16)	-0.0060 (14)
C9	0.058 (2)	0.052 (2)	0.0282 (15)	-0.004 (2)	-0.0067 (16)	-0.0010 (15)
C10	0.048 (2)	0.0461 (19)	0.0376 (18)	-0.0014 (18)	0.0041 (17)	-0.0001 (15)
C11	0.053 (2)	0.0395 (17)	0.0464 (19)	-0.0077 (18)	0.0007 (18)	-0.0043 (16)
C12	0.045 (2)	0.049 (2)	0.0385 (18)	-0.0055 (18)	-0.0048 (17)	-0.0065 (16)
C13	0.048 (2)	0.0437 (18)	0.0329 (16)	0.0006 (18)	-0.0030 (15)	-0.0030 (15)
C14	0.0419 (19)	0.0441 (18)	0.0321 (16)	-0.0035 (17)	-0.0041 (16)	-0.0026 (14)
C15	0.048 (2)	0.0404 (17)	0.0303 (16)	-0.0023 (18)	0.0012 (16)	-0.0017 (14)
C16	0.081 (3)	0.0400 (19)	0.0332 (17)	0.001 (2)	0.002 (2)	-0.0028 (15)
C17	0.075 (3)	0.055 (2)	0.0348 (18)	-0.005 (2)	0.0052 (19)	-0.0094 (17)
C18	0.068 (3)	0.047 (2)	0.086 (3)	0.000 (2)	0.000 (3)	0.014 (2)
O5	0.080 (2)	0.0751 (18)	0.0436 (14)	0.0007 (19)	0.0009 (15)	-0.0010 (14)
O1	0.087 (2)	0.0428 (13)	0.0328 (12)	0.0010 (15)	0.0023 (14)	0.0029 (11)
O2	0.0716 (17)	0.0392 (12)	0.0296 (11)	0.0028 (13)	0.0012 (13)	-0.0037 (10)
O3	0.092 (2)	0.0451 (14)	0.0488 (14)	-0.0026 (16)	-0.0042 (16)	0.0095 (12)
O4	0.098 (2)	0.0527 (15)	0.0292 (12)	-0.0039 (18)	0.0073 (14)	-0.0048 (11)

Geometric parameters (\AA , $^\circ$)

C1—C6	1.379 (5)	C11—C12	1.388 (5)
C1—C2	1.388 (5)	C11—H11	0.93
C1—O5	1.423 (4)	C12—H12	0.93
C2—C3	1.392 (5)	C13—O5	1.354 (4)
C2—H2	0.93	C13—C14	1.371 (4)
C3—C4	1.386 (5)	C13—H13	0.93
C3—H3	0.93	C14—C15	1.454 (4)
C4—O4	1.366 (4)	C15—O1	1.225 (4)
C4—C5	1.377 (5)	C15—O2	1.336 (4)
C5—C6	1.399 (5)	C16—O2	1.444 (4)
C5—H5	0.93	C16—C17	1.501 (5)
C6—H6	0.93	C16—H16A	0.97
C7—C8	1.393 (5)	C16—H16B	0.97
C7—C12	1.396 (4)	C17—H17A	0.96
C7—C14	1.484 (4)	C17—H17B	0.96
C8—C9	1.379 (5)	C17—H17C	0.96

C8—H8	0.93	C18—O3	1.420 (5)
C9—C10	1.381 (5)	C18—H18A	0.96
C9—H9	0.93	C18—H18B	0.96
C10—O3	1.369 (4)	C18—H18C	0.96
C10—C11	1.384 (5)	O4—H4	0.82
C6—C1—C2	118.9 (3)	C11—C12—H12	118.7
C6—C1—O5	122.7 (3)	C7—C12—H12	118.7
C2—C1—O5	118.5 (3)	O5—C13—C14	127.3 (3)
C1—C2—C3	120.9 (4)	O5—C13—H13	116.4
C1—C2—H2	119.6	C14—C13—H13	116.4
C3—C2—H2	119.6	C13—C14—C15	118.5 (3)
C4—C3—C2	119.6 (3)	C13—C14—C7	118.3 (3)
C4—C3—H3	120.2	C15—C14—C7	123.2 (3)
C2—C3—H3	120.2	O1—C15—O2	121.3 (3)
O4—C4—C5	122.1 (3)	O1—C15—C14	125.0 (3)
O4—C4—C3	117.9 (3)	O2—C15—C14	113.7 (3)
C5—C4—C3	120.0 (3)	O2—C16—C17	106.8 (3)
C4—C5—C6	120.0 (3)	O2—C16—H16A	110.4
C4—C5—H5	120.0	C17—C16—H16A	110.4
C6—C5—H5	120.0	O2—C16—H16B	110.4
C1—C6—C5	120.6 (3)	C17—C16—H16B	110.4
C1—C6—H6	119.7	H16A—C16—H16B	108.6
C5—C6—H6	119.7	C16—C17—H17A	109.5
C8—C7—C12	116.9 (3)	C16—C17—H17B	109.5
C8—C7—C14	122.3 (3)	H17A—C17—H17B	109.5
C12—C7—C14	120.7 (3)	C16—C17—H17C	109.5
C9—C8—C7	121.1 (3)	H17A—C17—H17C	109.5
C9—C8—H8	119.4	H17B—C17—H17C	109.5
C7—C8—H8	119.4	O3—C18—H18A	109.5
C8—C9—C10	120.9 (3)	O3—C18—H18B	109.5
C8—C9—H9	119.6	H18A—C18—H18B	109.5
C10—C9—H9	119.6	O3—C18—H18C	109.5
O3—C10—C9	115.8 (3)	H18A—C18—H18C	109.5
O3—C10—C11	124.5 (3)	H18B—C18—H18C	109.5
C9—C10—C11	119.7 (3)	C13—O5—C1	123.8 (3)
C10—C11—C12	118.9 (3)	C15—O2—C16	118.3 (3)
C10—C11—H11	120.6	C10—O3—C18	117.8 (3)
C12—C11—H11	120.6	C4—O4—H4	109.5
C11—C12—C7	122.5 (3)		
C6—C1—C2—C3	-1.9 (6)	C14—C7—C12—C11	-174.8 (3)
O5—C1—C2—C3	178.6 (3)	O5—C13—C14—C15	0.5 (6)
C1—C2—C3—C4	0.2 (6)	O5—C13—C14—C7	-179.4 (3)
C2—C3—C4—O4	-178.4 (4)	C8—C7—C14—C13	-134.7 (4)
C2—C3—C4—C5	1.2 (6)	C12—C7—C14—C13	41.3 (5)
O4—C4—C5—C6	178.7 (4)	C8—C7—C14—C15	45.3 (5)
C3—C4—C5—C6	-1.0 (6)	C12—C7—C14—C15	-138.6 (4)
C2—C1—C6—C5	2.2 (6)	C13—C14—C15—O1	-4.9 (6)
O5—C1—C6—C5	-178.4 (3)	C7—C14—C15—O1	175.0 (4)

supplementary materials

C4—C5—C6—C1	-0.7 (6)	C13—C14—C15—O2	175.4 (3)
C12—C7—C8—C9	-0.2 (5)	C7—C14—C15—O2	-4.7 (5)
C14—C7—C8—C9	176.0 (3)	C14—C13—O5—C1	-178.3 (4)
C7—C8—C9—C10	-1.1 (6)	C6—C1—O5—C13	23.2 (6)
C8—C9—C10—O3	-178.0 (4)	C2—C1—O5—C13	-157.3 (4)
C8—C9—C10—C11	1.2 (6)	O1—C15—O2—C16	-0.4 (6)
O3—C10—C11—C12	179.1 (4)	C14—C15—O2—C16	179.3 (3)
C9—C10—C11—C12	0.0 (6)	C17—C16—O2—C15	179.1 (3)
C10—C11—C12—C7	-1.4 (6)	C9—C10—O3—C18	171.3 (4)
C8—C7—C12—C11	1.5 (6)	C11—C10—O3—C18	-7.8 (6)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O4—H4 \cdots O1 ⁱ	0.82	2.00	2.812 (3)	169

Symmetry codes: (i) $-x, y+1/2, -z+3/2$.

Fig. 1

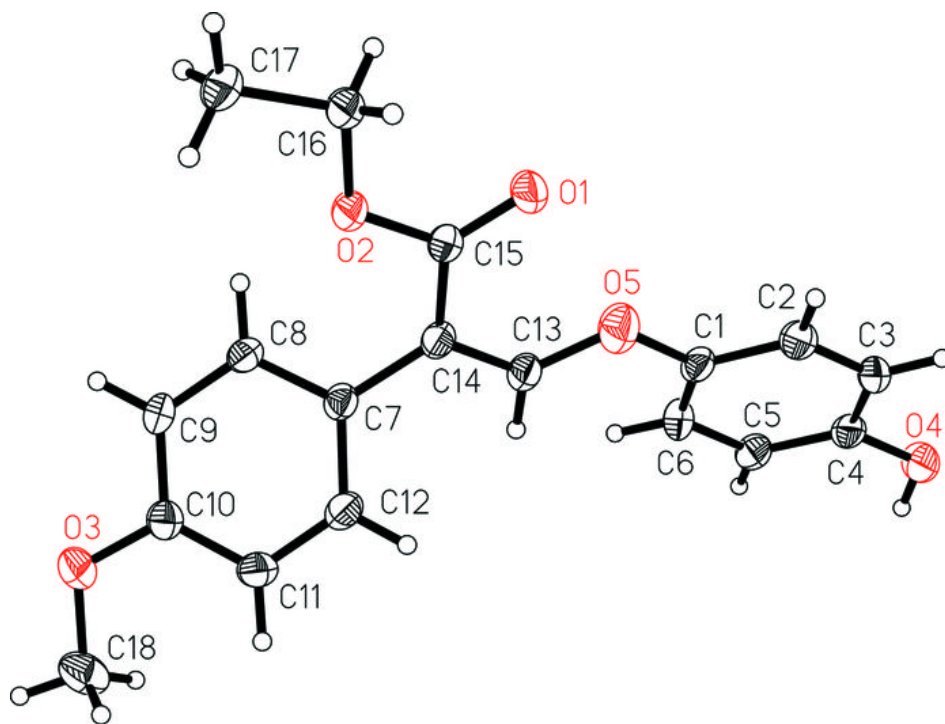


Fig. 2

