

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

(Z)-Ethyl 3-(4-chlorophenyl)-2-cyano-3-(2,6-difluorobenzamido)acrylateZhang Dehua^a and Zhang Xiaoyan^{b*}

^aDepartment of Chemistry and Environmental Engineering, Hubei Normal University, Huangshi 435002, People's Republic of China, and ^bSchool of Mathematics and Physics, Huangshi Institute of Technology, Huangshi 435003, People's Republic of China

Correspondence e-mail: zhangdehua200@163.com

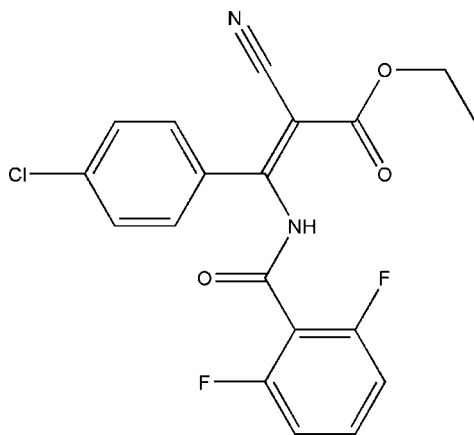
Received 16 September 2008; accepted 24 October 2008

Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; some non-H atoms missing; disorder in main residue; R factor = 0.052; wR factor = 0.154; data-to-parameter ratio = 13.4.

The title compound, $\text{C}_{19}\text{H}_{13}\text{ClF}_2\text{N}_2\text{O}_3$, was prepared by the reaction of (Z)-ethyl 3-amino-3-(4-chlorophenyl)-2-cyanoacrylate and 2,6-difluorobenzoyl chloride. The dihedral angle between the chlorobenzene and fluorobenzene rings is $37.0(1)^\circ$. The ethyl group is disordered over two positions [occupancies = 0.52 (2):0.48 (2)]. In addition to intramolecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{N}-\text{H}\cdots\text{F}$ hydrogen bonds, the crystal packing shows the molecules to be connected by intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds.

Related literature

The title compound is useful as an inhibitor of *Pyricularia oryzae*, *Rhizoctonia solani*, *Botrytis cinerea* and *Gibberella zeae*, see: Heller *et al.* (2004); Creagh & Hubbell (1992); Ibers & Hamilton (1964).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{13}\text{ClF}_2\text{N}_2\text{O}_3$	$\gamma = 91.4490(10)^\circ$
$M_r = 390.76$	$V = 919.9(5) \text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 8.919(5) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 9.7560(6) \text{ \AA}$	$\mu = 0.25 \text{ mm}^{-1}$
$c = 11.2717(7) \text{ \AA}$	$T = 298(2) \text{ K}$
$\alpha = 91.9710(10)^\circ$	$0.23 \times 0.20 \times 0.10 \text{ mm}$
$\beta = 110.0940(10)^\circ$	

Data collection

Bruker SMART CCD area-detector diffractometer	3556 independent reflections
Absorption correction: none	2524 reflections with $I > 2\sigma(I)$
7196 measured reflections	$R_{\text{int}} = 0.053$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.052$	6 restraints
$wR(F^2) = 0.154$	H-atom parameters constrained
$S = 1.10$	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
3556 reflections	$\Delta\rho_{\text{min}} = -0.31 \text{ e \AA}^{-3}$
265 parameters	

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O2}$	0.86	2.05	2.674 (2)	129
$\text{N1}-\text{H1}\cdots\text{F1}$	0.86	2.36	2.827 (2)	115
$\text{C18}-\text{H18B}\cdots\text{O1}^{\text{i}}$	0.97	2.58	2.990 (7)	106
$\text{C10}-\text{H10}\cdots\text{N2}^{\text{ii}}$	0.93	2.62	3.302 (3)	131
$\text{C5}-\text{H5}\cdots\text{N2}^{\text{iii}}$	0.93	2.59	3.432 (3)	150

Symmetry codes: (i) $x, y + 1, z$; (ii) $-x, -y + 2, -z + 1$; (iii) $x + 1, y, z + 1$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FL2223).

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supplementary materials

Acta Cryst. (2008). E64, o2254 [doi:10.1107/S1600536808034818]

(Z)-Ethyl 3-(4-chlorophenyl)-2-cyano-3-(2,6-difluorobenzamido)acrylate

Z. Dehua and Z. Xiaoyan

Comment

Recently, 2-cyanoacrylates have been in widespread used as agrochemicals because of their unique mechanism of action and good environmental profiles. The title compound is useful as an inhibitor of *Pyricularia oryzae*, *Rhizoctonia solani*, *Botrytis cinerea* and *Gibberella zeae* (Heller *et al.*, 2004; Creagh & Hubbell, 1992; Ibers & Hamilton, 1964).

In the title compound (Fig. 1), all bond lengths and angles are unexceptional. The planar chlorobenzene ring is approximately perpendicular to the fluorobenzene ring with a dihedral angle of 37.0 (1)°. The ethyl group is disordered over two positions occupancies (0.52 (2):0.48 (2)). The molecular conformation is stabilized by C—H···O and N—H···O hydrogen bonds (Table 1). The crystal packing is governed by additional N—H···O and N—H···F interactions (Fig. 2).

Experimental

To a solution of (Z)-ethyl 3-amino-3-(4-chlorophenyl)-2-cyanoacrylate (1.25 g, 0.0050 mol) in CH₂Cl₂ (18 ml), 2,6-difluorobenzoyl chloride (2.65 g, 0.015 mol) was added. Subsequently, Et₃N (1.52 g, 0.015 mol) was dropped into the solution under stirring. Then, the reaction mixture was heated to reflux and stirred for 4 h and then cooled to room temperature. The reaction solution was filtered off and some white solid was separated. The organic phase was washed with water and then dried over Na₂SO₄. After removal of the solvent, a brown dope was obtained. After column chromatography using ethylacetate/light petroleum (1:6) as the eluent. Small single crystals were grown from a solution of ethyl acetate/petroleum ether (3:1) after 45 days, at room temperature.

Refinement

Methyl H atoms were placed in calculated positions with C—H = 0.96 Å and the torsion angle was refined to fit the electron density, with $U_{iso}(H) = 1.5U_{eq}(C)$. Other H atoms were placed in calculated positions with C—H = 0.96 Å (methylene) and 0.93 Å (aromatic C—H), and refined in riding mode, with $U_{iso}(H) = 1.2U_{eq}(C)$. In the absence of significant anomalous scattering, Friedel pairs were merged.

Figures

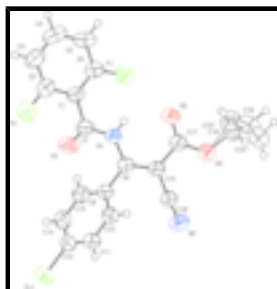


Fig. 1. The molecular structure of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme.

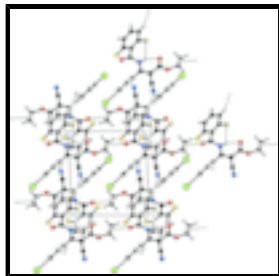


Fig. 2. The packing of the title compound, viewed down the *c* axis.

(Z)-Ethyl 3-(4-chlorophenyl)-2-cyano-3-(2,6-difluorobenzamido)acrylate

Crystal data

$C_{19}H_{13}ClF_2N_2O_3$

$M_r = 390.76$

Triclinic, $P\bar{1}$

$a = 8.919 (5) \text{ \AA}$

$b = 9.7560 (6) \text{ \AA}$

$c = 11.2717 (7) \text{ \AA}$

$\alpha = 91.9710 (10)^\circ$

$\beta = 110.0940 (10)^\circ$

$\gamma = 91.4490 (10)^\circ$

$V = 919.9 (5) \text{ \AA}^3$

$Z = 2$

$F_{000} = 400$

$D_x = 1.411 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71073 \text{ \AA}$

Cell parameters from 2672 reflections

$\theta = 2.4\text{--}26.8^\circ$

$\mu = 0.25 \text{ mm}^{-1}$

$T = 298 (2) \text{ K}$

Block, colorless

$0.23 \times 0.20 \times 0.10 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 298(2) \text{ K}$

φ and ω scans

Absorption correction: none

7196 measured reflections

3556 independent reflections

2524 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.053$

$\theta_{\text{max}} = 26.0^\circ$

$\theta_{\text{min}} = 1.9^\circ$

$h = -10 \rightarrow 9$

$k = -12 \rightarrow 12$

$l = -10 \rightarrow 13$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.052$

$wR(F^2) = 0.154$

$S = 1.10$

3556 reflections

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0794P)^2 + 0.0184P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$

265 parameters

$$\Delta\rho_{\min} = -0.31 \text{ e } \text{\AA}^{-3}$$

6 restraints

Extinction correction: none

Primary atom site location: structure-invariant direct methods

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	0.7937 (2)	0.8595 (2)	0.98955 (19)	0.0535 (5)	
C2	0.9102 (3)	0.7653 (3)	1.0005 (2)	0.0732 (7)	
C3	1.0531 (3)	0.7682 (3)	1.0992 (3)	0.0838 (8)	
H3	1.1291	0.7036	1.1024	0.101*	
C4	1.0803 (3)	0.8678 (3)	1.1917 (3)	0.0803 (8)	
H4	1.1756	0.8699	1.2601	0.096*	
C5	0.9711 (3)	0.9653 (3)	1.1869 (2)	0.0751 (7)	
H5	0.9911	1.0336	1.2506	0.090*	
C6	0.8309 (3)	0.9595 (2)	1.0850 (2)	0.0595 (6)	
C7	0.6359 (3)	0.8440 (2)	0.8831 (2)	0.0559 (5)	
C8	0.4458 (2)	0.97457 (19)	0.71787 (19)	0.0488 (5)	
C9	0.3884 (2)	0.85359 (19)	0.62766 (19)	0.0502 (5)	
C10	0.2296 (3)	0.8088 (2)	0.5863 (2)	0.0622 (6)	
H10	0.1586	0.8521	0.6180	0.075*	
C11	0.1759 (3)	0.7012 (2)	0.4989 (2)	0.0750 (8)	
H11	0.0695	0.6703	0.4725	0.090*	
C12	0.2800 (4)	0.6397 (2)	0.4511 (2)	0.0769 (8)	
C13	0.4383 (3)	0.6827 (2)	0.4905 (2)	0.0755 (7)	
H13	0.5082	0.6400	0.4573	0.091*	
C14	0.4923 (3)	0.7894 (2)	0.5793 (2)	0.0621 (6)	
H14	0.5994	0.8184	0.6069	0.074*	
C15	0.3749 (2)	1.09761 (19)	0.6912 (2)	0.0513 (5)	
C16	0.2483 (2)	1.1126 (2)	0.5730 (2)	0.0560 (5)	
C17	0.4278 (3)	1.2223 (2)	0.7746 (2)	0.0620 (6)	
C18	0.3811 (12)	1.4663 (7)	0.7837 (10)	0.068 (2)	0.523 (18)
H18A	0.3563	1.5287	0.7150	0.081*	0.523 (18)
H18B	0.4946	1.4746	0.8316	0.081*	0.523 (18)
C19	0.2853 (15)	1.4949 (11)	0.8666 (11)	0.092 (3)	0.523 (18)
H19A	0.1736	1.4875	0.8169	0.138*	0.523 (18)

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H19B	0.3120	1.5860	0.9044	0.138*	0.523 (18)
H19C	0.3087	1.4296	0.9317	0.138*	0.523 (18)
H18C	0.3968	1.4029	0.9188	0.105*	0.477 (18)
H18D	0.4638	1.4940	0.8343	0.105*	0.477 (18)
H19D	0.1927	1.5423	0.7192	0.136*	0.477 (18)
H19E	0.2613	1.6103	0.8561	0.136*	0.477 (18)
H19F	0.1498	1.4776	0.8294	0.136*	0.477 (18)
Cl1	0.21456 (12)	0.50444 (8)	0.33944 (8)	0.1331 (5)	
F1	0.72205 (17)	1.05606 (16)	1.07956 (13)	0.0850 (5)	
F2	0.8855 (2)	0.66854 (19)	0.90845 (19)	0.1238 (7)	
N1	0.57876 (19)	0.96329 (16)	0.82386 (16)	0.0544 (5)	
H1	0.6319	1.0382	0.8567	0.065*	
N2	0.1517 (2)	1.1285 (2)	0.4780 (2)	0.0741 (6)	
O1	0.5673 (2)	0.73410 (15)	0.85157 (17)	0.0825 (6)	
O2	0.5482 (2)	1.23045 (16)	0.86745 (17)	0.0837 (6)	
O3	0.3324 (2)	1.32363 (15)	0.73535 (19)	0.0887 (6)	

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0522 (12)	0.0567 (11)	0.0446 (11)	-0.0027 (9)	0.0076 (9)	0.0076 (9)
C2	0.0721 (16)	0.0698 (14)	0.0661 (16)	0.0128 (12)	0.0083 (13)	0.0047 (12)
C3	0.0652 (16)	0.0906 (18)	0.082 (2)	0.0167 (13)	0.0062 (15)	0.0179 (16)
C4	0.0527 (14)	0.111 (2)	0.0633 (16)	-0.0062 (14)	0.0010 (12)	0.0252 (15)
C5	0.0707 (16)	0.0961 (18)	0.0465 (14)	-0.0138 (14)	0.0067 (12)	-0.0015 (12)
C6	0.0531 (13)	0.0744 (14)	0.0479 (12)	-0.0006 (10)	0.0138 (10)	0.0029 (10)
C7	0.0562 (12)	0.0551 (11)	0.0467 (12)	-0.0040 (9)	0.0056 (10)	0.0024 (9)
C8	0.0412 (10)	0.0534 (10)	0.0477 (12)	-0.0037 (8)	0.0106 (9)	-0.0004 (9)
C9	0.0500 (11)	0.0491 (10)	0.0445 (11)	0.0017 (8)	0.0076 (9)	0.0003 (8)
C10	0.0521 (12)	0.0533 (11)	0.0711 (15)	0.0009 (9)	0.0098 (11)	-0.0098 (10)
C11	0.0645 (15)	0.0569 (13)	0.0773 (17)	0.0016 (11)	-0.0080 (13)	-0.0112 (12)
C12	0.0947 (19)	0.0522 (12)	0.0568 (15)	0.0208 (12)	-0.0091 (13)	-0.0072 (11)
C13	0.095 (2)	0.0717 (15)	0.0563 (15)	0.0297 (14)	0.0200 (14)	-0.0033 (12)
C14	0.0616 (14)	0.0673 (13)	0.0572 (14)	0.0130 (11)	0.0197 (11)	0.0023 (11)
C15	0.0438 (11)	0.0514 (11)	0.0498 (12)	-0.0033 (8)	0.0056 (9)	-0.0014 (9)
C16	0.0464 (12)	0.0516 (11)	0.0612 (14)	0.0009 (9)	0.0076 (11)	-0.0018 (10)
C17	0.0554 (13)	0.0512 (11)	0.0671 (15)	-0.0036 (10)	0.0063 (12)	-0.0033 (10)
C18	0.084 (4)	0.037 (3)	0.067 (5)	-0.004 (3)	0.007 (4)	0.001 (3)
C19	0.102 (8)	0.082 (6)	0.082 (6)	-0.018 (5)	0.022 (5)	-0.025 (5)
Cl1	0.1606 (9)	0.0804 (5)	0.0965 (6)	0.0416 (5)	-0.0328 (6)	-0.0442 (4)
F1	0.0830 (10)	0.0996 (10)	0.0646 (9)	0.0167 (8)	0.0164 (8)	-0.0169 (8)
F2	0.1283 (15)	0.1036 (12)	0.1083 (14)	0.0450 (11)	0.0014 (11)	-0.0327 (11)
N1	0.0465 (10)	0.0488 (9)	0.0534 (11)	-0.0040 (7)	-0.0005 (8)	-0.0004 (8)
N2	0.0581 (12)	0.0760 (13)	0.0698 (14)	0.0017 (10)	-0.0014 (11)	0.0026 (10)
O1	0.0875 (12)	0.0556 (9)	0.0744 (12)	-0.0176 (8)	-0.0097 (9)	0.0096 (8)
O2	0.0777 (11)	0.0612 (9)	0.0783 (12)	-0.0010 (8)	-0.0145 (10)	-0.0155 (8)
O3	0.0693 (11)	0.0497 (9)	0.1138 (15)	0.0055 (7)	-0.0093 (10)	-0.0193 (9)

Geometric parameters (Å, °)

C1—C6	1.372 (3)	C11—C12	1.365 (4)
C1—C2	1.382 (3)	C11—H11	0.9300
C1—C7	1.504 (3)	C12—C13	1.376 (4)
C2—F2	1.335 (3)	C12—Cl1	1.734 (2)
C2—C3	1.373 (3)	C13—C14	1.374 (3)
C3—C4	1.355 (4)	C13—H13	0.9300
C3—H3	0.9300	C14—H14	0.9300
C4—C5	1.368 (4)	C15—C16	1.435 (3)
C4—H4	0.9300	C15—C17	1.474 (3)
C5—C6	1.375 (3)	C16—N2	1.139 (3)
C5—H5	0.9300	C17—O2	1.214 (3)
C6—F1	1.358 (3)	C17—O3	1.308 (3)
C7—O1	1.200 (2)	C18—O3	1.475 (6)
C7—N1	1.382 (3)	C18—C19	1.491 (8)
C8—C15	1.365 (3)	C18—H18A	0.9700
C8—N1	1.374 (2)	C18—H18B	0.9700
C8—C9	1.489 (3)	C19—H19A	0.9600
C9—C14	1.379 (3)	C19—H19B	0.9600
C9—C10	1.384 (3)	C19—H19C	0.9600
C10—C11	1.372 (3)	N1—H1	0.8600
C10—H10	0.9300		
C6—C1—C2	115.1 (2)	C12—C11—H11	120.3
C6—C1—C7	124.45 (19)	C10—C11—H11	120.3
C2—C1—C7	120.34 (19)	C11—C12—C13	121.0 (2)
F2—C2—C3	117.8 (2)	C11—C12—Cl1	120.3 (2)
F2—C2—C1	118.6 (2)	C13—C12—Cl1	118.7 (2)
C3—C2—C1	123.6 (2)	C14—C13—C12	119.5 (2)
C4—C3—C2	118.2 (3)	C14—C13—H13	120.3
C4—C3—H3	120.9	C12—C13—H13	120.3
C2—C3—H3	120.9	C13—C14—C9	120.3 (2)
C3—C4—C5	121.6 (2)	C13—C14—H14	119.9
C3—C4—H4	119.2	C9—C14—H14	119.9
C5—C4—H4	119.2	C8—C15—C16	119.88 (17)
C4—C5—C6	118.1 (2)	C8—C15—C17	123.52 (18)
C4—C5—H5	121.0	C16—C15—C17	116.48 (17)
C6—C5—H5	121.0	N2—C16—C15	177.2 (2)
F1—C6—C1	118.17 (18)	O2—C17—O3	124.01 (19)
F1—C6—C5	118.3 (2)	O2—C17—C15	123.6 (2)
C1—C6—C5	123.5 (2)	O3—C17—C15	112.34 (18)
O1—C7—N1	123.20 (19)	O3—C18—C19	103.8 (6)
O1—C7—C1	121.33 (19)	O3—C18—H18A	111.0
N1—C7—C1	115.46 (16)	C19—C18—H18A	111.0
C15—C8—N1	120.33 (16)	O3—C18—H18B	111.0
C15—C8—C9	120.59 (17)	C19—C18—H18B	111.0
N1—C8—C9	118.91 (17)	H18A—C18—H18B	109.0
C14—C9—C10	119.25 (19)	C8—N1—C7	126.72 (16)

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C14—C9—C8	119.85 (19)	C8—N1—H1	116.6
C10—C9—C8	120.82 (18)	C7—N1—H1	116.6
C11—C10—C9	120.6 (2)	C17—O3—C18	121.6 (5)
C11—C10—H10	119.7	C17—O3—C18'	110.6 (4)
C9—C10—H10	119.7	C18—O3—C18'	26.1 (5)
C12—C11—C10	119.4 (2)		

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1 \cdots O2	0.86	2.05	2.674 (2)	129
N1—H1 \cdots F1	0.86	2.36	2.827 (2)	115
C18—H18B \cdots O1 ⁱ	0.97	2.58	2.990 (7)	106
C10—H10 \cdots N2 ⁱⁱ	0.93	2.62	3.302 (3)	131
C5—H5 \cdots N2 ⁱⁱⁱ	0.93	2.59	3.432 (3)	150

Symmetry codes: (i) $x, y+1, z$; (ii) $-x, -y+2, -z+1$; (iii) $x+1, y, z+1$.

Fig. 1

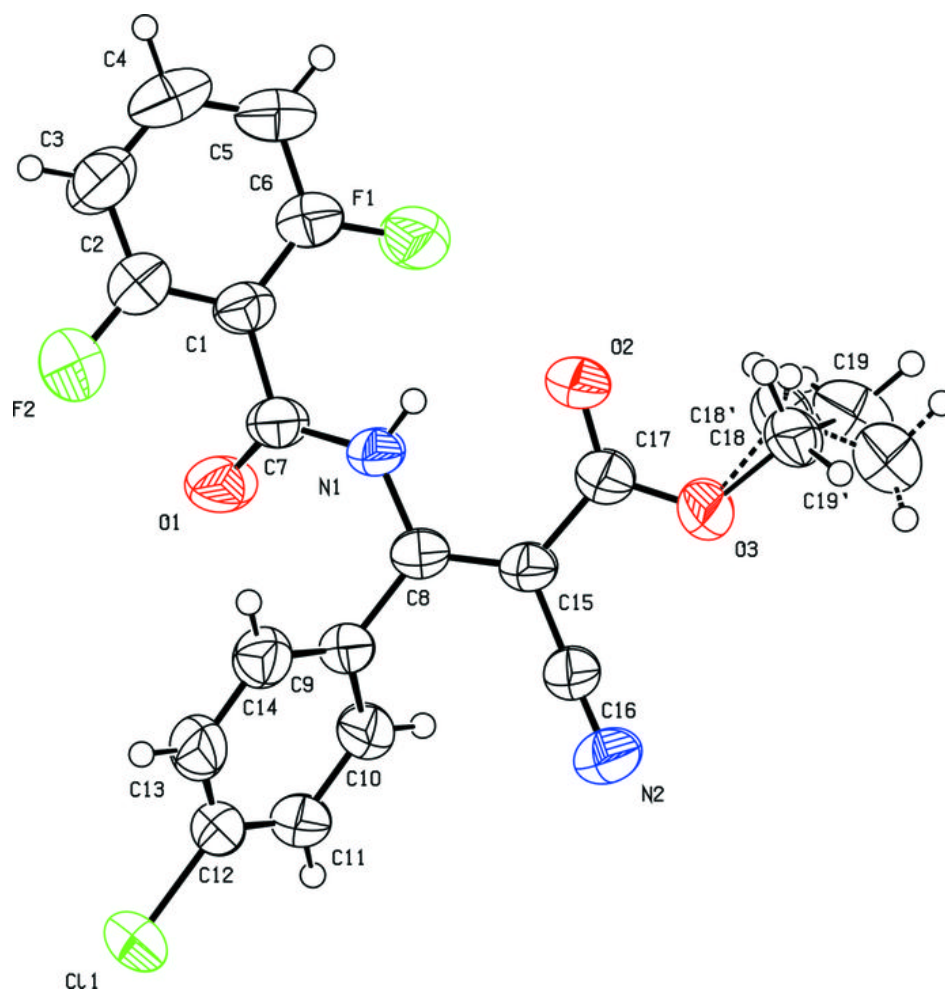


Fig. 2

