

# Hexaaquacadmium(II) 2,2'-(azino-dimethylidene)dibenzenesulfonate dihydrate

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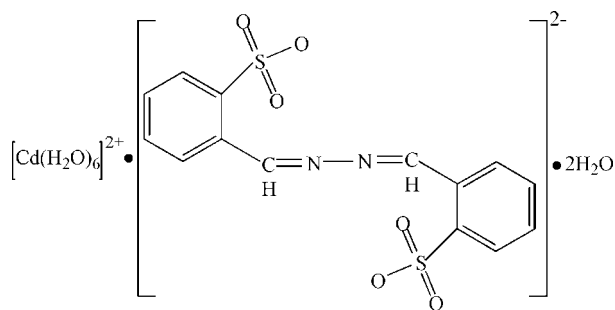
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 Key indicators: single-crystal X-ray study;  $T = 298$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.023;  $wR$  factor = 0.061; data-to-parameter ratio = 13.4.

In the title compound,  $[\text{Cd}(\text{H}_2\text{O})_6](\text{C}_{14}\text{H}_{10}\text{O}_6\text{N}_2\text{S}_2) \cdot 2\text{H}_2\text{O}$ , the complete cation and anion are each generated by crystallographic inversion symmetry. In the crystal structure, the components form a three-dimensional network by way of  $\text{O}-\text{H} \cdots \text{O}$  and  $\text{O}-\text{H} \cdots \text{N}$  hydrogen bonds.

## Related literature

For background to the properties and potential applications of organic-inorganic hybrid materials, see: Hagrman *et al.* (1998); Ranford *et al.* (1998).



## Experimental

### Crystal data

 $[\text{Cd}(\text{H}_2\text{O})_6](\text{C}_{14}\text{H}_{10}\text{O}_6\text{N}_2\text{S}_2) \cdot 2\text{H}_2\text{O}$ 
 $M_r = 622.89$ 

 Triclinic,  $P\bar{1}$ 
 $a = 7.8329$  (11) Å

 $b = 7.9824$  (12) Å

 $c = 10.1010$  (15) Å

 $\alpha = 92.723$  (1)°

 $\beta = 102.076$  (2)°

 $\gamma = 105.924$  (2)°

 $V = 590.19$  (15) Å<sup>3</sup>
 $Z = 1$ 

 Mo  $K\alpha$  radiation

 $\mu = 1.17$  mm<sup>-1</sup>
 $T = 298$  (2) K

 $0.45 \times 0.40 \times 0.28$  mm

### Data collection

Bruker SMART CCD

diffractometer

Absorption correction: multi-scan

(SADABS; Bruker, 2000)

 $T_{\min} = 0.621$ ,  $T_{\max} = 0.735$ 

3081 measured reflections

2041 independent reflections

 1929 reflections with  $I > 2\sigma(I)$ 
 $R_{\text{int}} = 0.018$ 

### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.023$ 
 $wR(F^2) = 0.061$ 
 $S = 1.06$ 

2041 reflections

152 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.33$  e Å<sup>-3</sup>
 $\Delta\rho_{\text{min}} = -0.42$  e Å<sup>-3</sup>
**Table 1**

Selected bond lengths (Å).

Cd1—O5	2.2555 (18)	Cd1—O6	2.2947 (18)
Cd1—O4	2.2589 (17)		

**Table 2**

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O4—H4A $\cdots$ O2 <sup>i</sup>	0.85	1.94	2.783 (3)	171
O4—H4B $\cdots$ O1 <sup>ii</sup>	0.85	2.03	2.872 (2)	174
O5—H5A $\cdots$ O1 <sup>iii</sup>	0.85	1.99	2.831 (3)	173
O5—H5B $\cdots$ O7 <sup>iv</sup>	0.85	2.00	2.843 (3)	173
O6—H6A $\cdots$ O7	0.85	2.08	2.881 (3)	157
O6—H6B $\cdots$ N1 <sup>v</sup>	0.85	2.15	2.993 (3)	169
O7—H7A $\cdots$ O3 <sup>vi</sup>	0.85	1.85	2.692 (3)	172
O7—H7B $\cdots$ O2 <sup>iii</sup>	0.85	2.21	3.002 (3)	156

Symmetry codes: (i)  $x-1, y-1, z$ ; (ii)  $-x+1, -y+1, -z+1$ ; (iii)  $-x+2, -y+1, -z+1$ ; (iv)  $-x+1, -y, -z+1$ ; (v)  $-x+1, -y+1, -z$ ; (vi)  $x, y-1, z$ .

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

The authors thank the Science and Technology Foundation of Weifang (2008–19) for a research grant.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2858).

## References

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- Ranford, J. D., Vittal, J. J. & Wang, Y. M. (1998). *Inorg. Chem.* **37**, 1226–1231.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

**supplementary materials**

*Acta Cryst.* (2008). E64, m1625 [ doi:10.1107/S1600536808039032 ]

## Hexaaquacadmium(II) 2,2'-(azinodimethylidyne)dibenzenesulfonate dihydrate

L.-C. Du

### Comment

The design and synthesis of organic/inorganic hybrid materials have attracted intense attention in recent years owing to their potential practical applications, such as antitumor, antidiabetic, antitubercular activities, magnetism and catalysis [Ranford, *et al.*, 1998; Hagrman, *et al.*, 1998]. As part of our studies in this area, we now report the synthesis and crystal structure of the title compound, (I).

The Cd(II) centre is six-coordinate with six O donors of H<sub>2</sub>O, and adopts distorted octahedral coordination (Table 1, Fig. 1). In the crystal, the molecules form a three-dimensional network by way of O—H...O and O—H...N hydrogen bonds (Table 2).

### Experimental

A solution of 1.0 mmol 2-formyl-benzenesulfonic acid-hydrazine and 1.0 mmol NaOH in 5 ml 95% ethanol was added to a solution of 0.5 mmol Cd(CH<sub>3</sub>COO)<sub>2</sub>·4H<sub>2</sub>O in 5 ml ethanol at room temperature. The mixture was refluxed for 4 h with stirring, then the resulting precipitate was filtered, washed, and dried *in vacuo* over P<sub>4</sub>O<sub>10</sub> for 48 h. Colourless blocks of (I) were obtained by slowly evaporating from methanol at room temperature.

### Refinement

H atom treatment??

### Figures

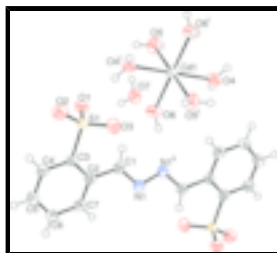


Fig. 1. The molecular structure of (I) showing 50% displacement ellipsoids for the non-hydrogen atoms. Symmetry codes: (i) 1-x, 1-y, 1-z; (ii) 1-x, 1-y, -z.

## Hexaaquacadmium(II) 2,2'-(azinodimethylidyne)dibenzenesulfonate dihydrate

### Crystal data

[Cd(H<sub>2</sub>O)<sub>6</sub>](C<sub>14</sub>H<sub>10</sub>O<sub>6</sub>N<sub>2</sub>S<sub>2</sub>)·2H<sub>2</sub>O  
*M<sub>r</sub>* = 622.89

*Z* = 1  
*F*<sub>000</sub> = 316

# supplementary materials

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Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 7.8329$  (11) Å

$b = 7.9824$  (12) Å

$c = 10.1010$  (15) Å

$\alpha = 92.723$  (1)°

$\beta = 102.076$  (2)°

$\gamma = 105.924$  (2)°

$V = 590.19$  (15) Å<sup>3</sup>

$D_x = 1.753$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 2719 reflections

$\theta = 2.7$ – $28.3$ °

$\mu = 1.17$  mm<sup>-1</sup>

$T = 298$  (2) K

Block, colourless

$0.45 \times 0.40 \times 0.28$  mm

## Data collection

Bruker SMART CCD  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 298$ (2) K

$\omega$  scans

Absorption correction: multi-scan  
(SADABS; Bruker, 2000)

$T_{\min} = 0.621$ ,  $T_{\max} = 0.735$

3081 measured reflections

2041 independent reflections

1929 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.018$

$\theta_{\text{max}} = 25.0$ °

$\theta_{\text{min}} = 2.1$ °

$h = -9 \rightarrow 9$

$k = -5 \rightarrow 9$

$l = -11 \rightarrow 12$

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.023$

$wR(F^2) = 0.061$

$S = 1.06$

2041 reflections

152 parameters

Primary atom site location: structure-invariant direct  
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring  
sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0313P)^2 + 0.2007P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.33$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.42$  e Å<sup>-3</sup>

Extinction correction: SHELXL97 (Sheldrick, 2008),

$$F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$$

Extinction coefficient: 0.067 (3)

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
Cd1	0.5000	0.5000	0.5000	0.02922 (13)
N1	0.5732 (3)	0.5473 (3)	-0.0282 (2)	0.0336 (5)
O1	1.0298 (2)	0.6986 (2)	0.35433 (17)	0.0384 (4)
O2	1.1817 (3)	1.0036 (3)	0.35043 (19)	0.0465 (5)
O3	0.8508 (3)	0.8917 (3)	0.2790 (2)	0.0477 (5)
O4	0.2044 (2)	0.3385 (2)	0.45818 (18)	0.0406 (4)
H4A	0.1847	0.2342	0.4229	0.049*
H4B	0.1378	0.3362	0.5150	0.049*
O5	0.6085 (3)	0.3128 (3)	0.6297 (2)	0.0471 (5)
H5A	0.7201	0.3195	0.6358	0.057*
H5B	0.5490	0.2060	0.6283	0.057*
O6	0.5347 (3)	0.3561 (3)	0.30977 (19)	0.0448 (5)
H6A	0.5293	0.2485	0.3123	0.054*
H6B	0.4908	0.3730	0.2289	0.054*
O7	0.6167 (3)	0.0337 (3)	0.3703 (2)	0.0468 (5)
H7A	0.6908	-0.0043	0.3358	0.056*
H7B	0.6594	0.0466	0.4560	0.056*
S1	1.02370 (7)	0.85670 (8)	0.28844 (6)	0.02847 (16)
C1	0.7223 (3)	0.5998 (3)	0.0615 (2)	0.0303 (5)
H1	0.7234	0.5806	0.1517	0.036*
C2	0.8948 (3)	0.6913 (3)	0.0237 (2)	0.0265 (5)
C3	1.0408 (3)	0.8087 (3)	0.1183 (2)	0.0258 (5)
C4	1.2027 (3)	0.8863 (3)	0.0822 (3)	0.0348 (6)
H4	1.2984	0.9650	0.1451	0.042*
C5	1.2224 (4)	0.8469 (4)	-0.0476 (3)	0.0423 (6)
H5	1.3316	0.8989	-0.0717	0.051*
C6	1.0809 (3)	0.7311 (4)	-0.1414 (3)	0.0386 (6)
H6	1.0948	0.7044	-0.2284	0.046*
C7	0.9177 (3)	0.6542 (3)	-0.1061 (2)	0.0337 (5)
H7	0.8223	0.5768	-0.1701	0.040*

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cd1	0.03192 (17)	0.02725 (17)	0.02920 (17)	0.01001 (10)	0.00609 (10)	0.00611 (10)
N1	0.0251 (10)	0.0408 (12)	0.0293 (11)	0.0001 (9)	0.0060 (8)	0.0066 (9)
O1	0.0449 (10)	0.0429 (10)	0.0318 (9)	0.0163 (8)	0.0123 (8)	0.0108 (8)
O2	0.0477 (11)	0.0433 (11)	0.0375 (10)	-0.0004 (9)	0.0074 (8)	-0.0093 (9)
O3	0.0429 (10)	0.0629 (13)	0.0486 (12)	0.0291 (10)	0.0168 (9)	0.0054 (10)

## supplementary materials

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O4	0.0398 (10)	0.0394 (10)	0.0389 (10)	0.0022 (8)	0.0151 (8)	0.0000 (8)
O5	0.0416 (10)	0.0394 (11)	0.0614 (13)	0.0147 (9)	0.0069 (9)	0.0211 (9)
O6	0.0639 (12)	0.0415 (11)	0.0329 (10)	0.0200 (10)	0.0132 (9)	0.0038 (8)
O7	0.0447 (11)	0.0518 (12)	0.0501 (12)	0.0243 (9)	0.0123 (9)	-0.0007 (9)
S1	0.0288 (3)	0.0306 (3)	0.0257 (3)	0.0085 (2)	0.0063 (2)	0.0008 (2)
C1	0.0290 (12)	0.0316 (13)	0.0265 (12)	0.0036 (10)	0.0048 (10)	0.0033 (10)
C2	0.0251 (11)	0.0282 (12)	0.0260 (12)	0.0080 (9)	0.0044 (9)	0.0069 (9)
C3	0.0253 (11)	0.0268 (12)	0.0258 (12)	0.0086 (9)	0.0054 (9)	0.0055 (9)
C4	0.0259 (12)	0.0389 (14)	0.0329 (13)	0.0008 (10)	0.0040 (10)	0.0036 (11)
C5	0.0325 (13)	0.0565 (17)	0.0374 (15)	0.0052 (12)	0.0158 (11)	0.0127 (13)
C6	0.0401 (14)	0.0504 (16)	0.0287 (13)	0.0143 (12)	0.0131 (11)	0.0086 (12)
C7	0.0335 (13)	0.0382 (14)	0.0265 (12)	0.0087 (11)	0.0029 (10)	0.0021 (10)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

Cd1—O5 <sup>i</sup>	2.2555 (18)	O6—H6B	0.8500
Cd1—O5	2.2555 (18)	O7—H7A	0.8499
Cd1—O4 <sup>i</sup>	2.2589 (17)	O7—H7B	0.8500
Cd1—O4	2.2589 (17)	S1—C3	1.783 (2)
Cd1—O6 <sup>i</sup>	2.2947 (18)	C1—C2	1.485 (3)
Cd1—O6	2.2947 (18)	C1—H1	0.9300
N1—C1	1.270 (3)	C2—C7	1.390 (3)
N1—N1 <sup>ii</sup>	1.431 (4)	C2—C3	1.403 (3)
O1—S1	1.4621 (19)	C3—C4	1.382 (3)
O2—S1	1.4523 (19)	C4—C5	1.384 (4)
O3—S1	1.4414 (18)	C4—H4	0.9300
O4—H4A	0.8500	C5—C6	1.377 (4)
O4—H4B	0.8500	C5—H5	0.9300
O5—H5A	0.8500	C6—C7	1.385 (4)
O5—H5B	0.8500	C6—H6	0.9300
O6—H6A	0.8499	C7—H7	0.9300
O5 <sup>i</sup> —Cd1—O5	180.0	O3—S1—O1	111.68 (12)
O5 <sup>i</sup> —Cd1—O4 <sup>i</sup>	95.45 (7)	O2—S1—O1	111.24 (12)
O5—Cd1—O4 <sup>i</sup>	84.55 (7)	O3—S1—C3	106.73 (11)
O5 <sup>i</sup> —Cd1—O4	84.55 (7)	O2—S1—C3	106.56 (11)
O5—Cd1—O4	95.45 (7)	O1—S1—C3	105.47 (10)
O4 <sup>i</sup> —Cd1—O4	180.0	N1—C1—C2	120.7 (2)
O5 <sup>i</sup> —Cd1—O6 <sup>i</sup>	89.86 (7)	N1—C1—H1	119.7
O5—Cd1—O6 <sup>i</sup>	90.14 (7)	C2—C1—H1	119.7
O4 <sup>i</sup> —Cd1—O6 <sup>i</sup>	90.19 (7)	C7—C2—C3	118.4 (2)
O4—Cd1—O6 <sup>i</sup>	89.81 (7)	C7—C2—C1	119.8 (2)
O5 <sup>i</sup> —Cd1—O6	90.14 (7)	C3—C2—C1	121.7 (2)
O5—Cd1—O6	89.86 (7)	C4—C3—C2	120.5 (2)
O4 <sup>i</sup> —Cd1—O6	89.81 (7)	C4—C3—S1	118.63 (17)
O4—Cd1—O6	90.19 (7)	C2—C3—S1	120.88 (17)
O6 <sup>i</sup> —Cd1—O6	180.0	C3—C4—C5	120.0 (2)

C1—N1—N1 <sup>ii</sup>	111.5 (2)	C3—C4—H4	120.0
Cd1—O4—H4A	113.7	C5—C4—H4	120.0
Cd1—O4—H4B	123.5	C6—C5—C4	120.3 (2)
H4A—O4—H4B	108.3	C6—C5—H5	119.9
Cd1—O5—H5A	116.0	C4—C5—H5	119.9
Cd1—O5—H5B	123.2	C5—C6—C7	120.0 (2)
H5A—O5—H5B	108.7	C5—C6—H6	120.0
Cd1—O6—H6A	116.9	C7—C6—H6	120.0
Cd1—O6—H6B	123.6	C6—C7—C2	120.8 (2)
H6A—O6—H6B	109.5	C6—C7—H7	119.6
H7A—O7—H7B	105.6	C2—C7—H7	119.6
O3—S1—O2	114.51 (12)		
N1 <sup>ii</sup> —N1—C1—C2	176.2 (2)	O3—S1—C3—C2	-47.1 (2)
N1—C1—C2—C7	-29.2 (4)	O2—S1—C3—C2	-169.90 (19)
N1—C1—C2—C3	154.3 (2)	O1—S1—C3—C2	71.8 (2)
C7—C2—C3—C4	0.5 (3)	C2—C3—C4—C5	-0.7 (4)
C1—C2—C3—C4	177.1 (2)	S1—C3—C4—C5	177.7 (2)
C7—C2—C3—S1	-177.85 (18)	C3—C4—C5—C6	0.3 (4)
C1—C2—C3—S1	-1.2 (3)	C4—C5—C6—C7	0.3 (4)
O3—S1—C3—C4	134.5 (2)	C5—C6—C7—C2	-0.5 (4)
O2—S1—C3—C4	11.7 (2)	C3—C2—C7—C6	0.1 (4)
O1—S1—C3—C4	-106.6 (2)	C1—C2—C7—C6	-176.6 (2)

Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x+1, -y+1, -z$ .

*Hydrogen-bond geometry* ( $\text{\AA}, ^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O4—H4A $\cdots$ O2 <sup>iii</sup>	0.85	1.94	2.783 (3)	171
O4—H4B $\cdots$ O1 <sup>i</sup>	0.85	2.03	2.872 (2)	174
O5—H5A $\cdots$ O1 <sup>iv</sup>	0.85	1.99	2.831 (3)	173
O5—H5B $\cdots$ O7 <sup>v</sup>	0.85	2.00	2.843 (3)	173
O6—H6A $\cdots$ O7	0.85	2.08	2.881 (3)	157
O6—H6B $\cdots$ N1 <sup>ii</sup>	0.85	2.15	2.993 (3)	169
O7—H7A $\cdots$ O3 <sup>vi</sup>	0.85	1.85	2.692 (3)	172
O7—H7B $\cdots$ O2 <sup>iv</sup>	0.85	2.21	3.002 (3)	156

Symmetry codes: (iii)  $x-1, y-1, z$ ; (i)  $-x+1, -y+1, -z+1$ ; (iv)  $-x+2, -y+1, -z+1$ ; (v)  $-x+1, -y, -z+1$ ; (ii)  $-x+1, -y+1, -z$ ; (vi)  $x, y-1, z$ .

Fig. 1

