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(Z)-Methyl 4-({3-[(2,5-dioxoimidazolidin-4-ylidene)methyl]-1H-indol-1-yl}methyl)-benzoate

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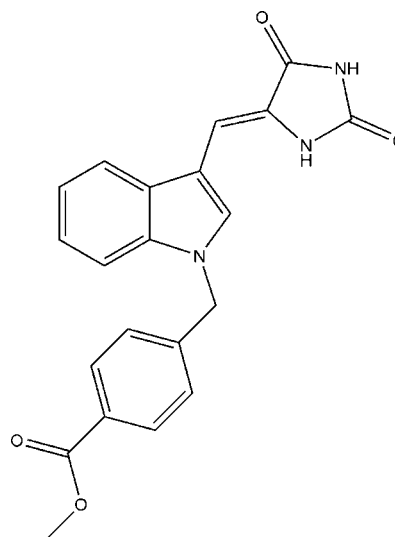
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Key indicators: single-crystal X-ray study; $T = 90$ K; mean $\sigma(C-C) = 0.002$ Å; R factor = 0.043; wR factor = 0.127; data-to-parameter ratio = 15.7.

In the title compound, $C_{21}H_{17}N_3O_4$, pairs of molecules form a planar [maximum deviation 0.0566 (9) Å] centrosymmetric imidazole dimer *via* two N—H...O hydrogen bonds. These dimeric units are linked by further N—H...O hydrogen bonds between the ester carbonyl group and the imidazolidine ring, forming chains parallel to the c -axis direction. In addition, there are π – π stacking interactions between the planar imidazole pairs, with an interplanar spacing of 3.301 (2) Å. There is a double bond with Z geometry connecting the imidazolidine and indole units.

Related literature

For general background to the radiosensitization activity of (Z)-2-(*N*-benzylindol-3-ylmethylene)quinuclidin-3-one and (Z)-(\pm)-2-(*N*-benzylindol-3-ylmethylene)quinuclidin-3-ol derivatives, see: Sekhar *et al.* (2003); Sonar *et al.*, (2007). For related structures, see: Mason *et al.* (2003); Zarza *et al.* (1988).



Experimental

Crystal data

$C_{21}H_{17}N_3O_4$	$\gamma = 104.4242$ (9) ^o
$M_r = 375.38$	$V = 883.25$ (2) Å ³
Triclinic, $P\bar{1}$	$Z = 2$
$a = 7.6390$ (1) Å	Mo $K\alpha$ radiation
$b = 8.0013$ (1) Å	$\mu = 0.10$ mm ⁻¹
$c = 15.0405$ (3) Å	$T = 90.0$ (2) K
$\alpha = 91.9853$ (9) ^o	$0.25 \times 0.22 \times 0.15$ mm
$\beta = 96.2291$ (9) ^o	

Data collection

Nonius KappaCCD diffractometer	19513 measured reflections
Absorption correction: multi-scan (SCALEPACK; Otwinowski & Minor, 1997)	3997 independent reflections
$T_{\min} = 0.976$, $T_{\max} = 0.985$	3595 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.015$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$	255 parameters
$wR(F^2) = 0.127$	H-atom parameters constrained
$S = 1.07$	$\Delta\rho_{\text{max}} = 0.48$ e Å ⁻³
3997 reflections	$\Delta\rho_{\text{min}} = -0.34$ e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N11—H11...O12 ⁱ	0.88	2.11	2.9658 (15)	163
N13—H13...O22 ⁱⁱ	0.88	2.29	2.9699 (15)	134

Symmetry codes: (i) $-x + 2, -y + 1, -z + 1$; (ii) $x, y, z - 1$.

Data collection: COLLECT (Nonius, 1998); cell refinement: SCALEPACK (Otwinowski & Minor, 1997); data reduction: DENZO-SMN (Otwinowski & Minor, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XP in SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97 and local procedures.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FJ2166).

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supplementary materials

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(Z)-Methyl 4-({3-[(2,5-dioxoimidazolidin-4-ylidene)methyl]-1*H*-indol-1-yl}methyl)benzoate

N. R. Penthala, T. R. Y. Reddy, S. Parkin and P. A. Crooks

Comment

In continuation of our work on the radiosensitization activity of (*Z*)-2-(*N*-benzylindol-3-ylmethylene)quinuclidin-3-one and (*Z*)-(\pm)-2-(*N*-benzylindol-3-ylmethylene) quinuclidin-3-ol derivatives (Sekhar *et al.*, 2003; Sonar *et al.*, 2007), we have undertaken the design, synthesis and structural analysis of a series of (*N*-benzylindol-3-ylmethylene)imidazolidine-2,4-dione analogs with different substituents on both indole moiety and on the benzene ring of the *N*-benzyl group. The primary goal for X-ray analysis of the title compound is to confirm the double-bond geometry and to obtain detailed information on the structural conformation of the molecule. This information will be useful in structure-activity relationship (SAR) analysis. The title compound was prepared by the reaction of methyl 4-((3-formyl-1*H*-indol-1-yl)methyl)benzoate with imidazolidine-2,4-dione in the presence of ammonium acetate in acetic acid at 391 K. The compound was crystallized from a mixture of methanol and ethylacetate. The molecular structure and the atom-numbering scheme are shown in Fig.1. The indole ring is planar with bond distances and angles comparable with those previously reported for other indole derivatives (Mason *et al.*, 2003; Zarza, *et al.*, 1988). The X-ray studies revealed that the title compound is the *Z* isomer. The C8—C9 bond is in a *transoid* geometry with respect to the C10—C14 bond. The olefinic bond (C9=C10) has a planar atomic arrangement, since the r.m.s. deviation from the mean plane passing through atoms C1, C8, C9, N11 is 0.0349 (6) Å. Deviations from ideal geometry are observed in the bond angles around atoms C9, C10 and N11 (130.48 (12)°) due to repulsion between the indole ring C1 hydrogen and imidazolidine ring N11 hydrogen. The imidazolidine ring, which makes a dihedral angle of 10.03 (7)° with the adjacent aromatic ring, presents very small distortions around atoms N11, C12, N13 and C14.

Significant intermolecular hydrogen-bonding interactions are found between N(11)—H(11)⋯O(12) and N(13)—H(13)⋯O(22), and molecules are linked into chains by N—H⋯O hydrogen bonding.

Experimental

A mixture of methyl 4-((3-formyl-1*H*-indol-1-yl)methyl)benzoate (0.5 g, 1.70 mmol), imidazolidine-2,4-dione (0.18 g, 1.80 mmol) and ammonium acetate (0.132 g, 1.71 mmol) was stirred in acetic acid (5 ml) at 391 K for 8 hrs. The reaction mixture was cooled to room temperature and the yellow solid that separated was collected by filtration, washed with cold water and dried to afford the crude product. Crystallization from methanol and ethyl acetate (1:1) afforded a yellow crystalline product of (*Z*)Methyl-4-((3-((2,5- dioxoimidazolidin-4-ylidene)methyl)-1*H*-indol-1-yl)methyl)benzoate that was suitable for X-ray analysis. ¹H NMR (DMSO *d*₆): δ 3.81 (*s*, 3H), 5.54 (*s*, 2H), 6.74 (*s*, 1H), 7.15–7.23 (*m*, 2H), 7.40–7.42 (*d*, 2H), 7.50–7.52 (*d*, 1H), 7.79–7.81 (*d*, 1H), 7.91–7.93 (*d*, 2H), 8.32 (*s*, 1H), 10.15 (*bs*, 1H), 11.06 (*bs*, 1H); ¹³C NMR (DMSO *d*₆): δ 49.52, 52.2, 101.44, 109.26, 111.26, 119.22, 121.35, 123.44, 124.01, 127.29, 128.95, 129.35, 131.23, 136.28, 142.64, 155.93, 165.85, 167.12.

Refinement

H atoms were found in difference Fourier maps and subsequently placed in idealized positions with constrained distances of 0.98 Å (RCH₃), 0.99 Å (R₂CH₂), 0.95 Å (C_{Ar}H), 0.88 Å (N—H), and with $U_{\text{iso}}(\text{H})$ values set to either $1.2U_{\text{eq}}$ or $1.5U_{\text{eq}}$ (RCH₃) of the attached atom.

Figures

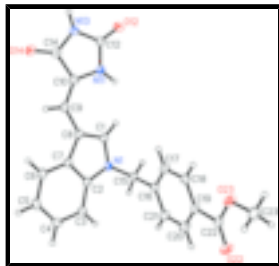


Fig. 1. A view of the molecule with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

(Z)-Methyl 4-({3-[(2,5-dioxoimidazolidin-4-ylidene)methyl]-1H-indol-1-yl}methyl)benzoate

Crystal data

C₂₁H₁₇N₃O₄

$M_r = 375.38$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 7.6390$ (1) Å

$b = 8.0013$ (1) Å

$c = 15.0405$ (3) Å

$\alpha = 91.9853$ (9)°

$\beta = 96.2291$ (9)°

$\gamma = 104.4242$ (9)°

$V = 883.25$ (2) Å³

$Z = 2$

$F_{000} = 392$

$D_x = 1.411$ Mg m⁻³

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 3966 reflections

$\theta = 1.0\text{--}27.5^\circ$

$\mu = 0.10$ mm⁻¹

$T = 90.0$ (2) K

Block, colourless

$0.25 \times 0.22 \times 0.15$ mm

Data collection

Nonius KappaCCD
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

Detector resolution: 18 pixels mm⁻¹

$T = 90.0$ (2) K

ω scans at fixed $\chi = 55^\circ$

Absorption correction: multi-scan
(SCALEPACK; Otwinowski & Minor, 1997)

$T_{\text{min}} = 0.976$, $T_{\text{max}} = 0.985$

3997 independent reflections

3595 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.015$

$\theta_{\text{max}} = 27.4^\circ$

$\theta_{\text{min}} = 1.4^\circ$

$h = -9 \rightarrow 9$

$k = -10 \rightarrow 10$

$l = -19 \rightarrow 19$

19513 measured reflections

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.043$	$w = 1/[\sigma^2(F_o^2) + (0.0697P)^2 + 0.3949P]$
$wR(F^2) = 0.127$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.07$	$(\Delta/\sigma)_{\max} < 0.001$
3997 reflections	$\Delta\rho_{\max} = 0.48 \text{ e } \text{\AA}^{-3}$
255 parameters	$\Delta\rho_{\min} = -0.34 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: SHELXL97 (Sheldrick, 2008), $F_c^* = kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Secondary atom site location: difference Fourier map	Extinction coefficient: 0.090 (15)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.62738 (14)	0.06683 (14)	0.69424 (7)	0.0179 (2)
C1	0.64933 (17)	0.16626 (17)	0.62219 (8)	0.0189 (3)
H1	0.7632	0.2298	0.6066	0.023*
C2	0.44342 (17)	-0.00489 (16)	0.69679 (8)	0.0173 (3)
C3	0.35580 (18)	-0.11990 (17)	0.75529 (9)	0.0196 (3)
H3	0.4221	-0.1562	0.8048	0.024*
C4	0.16793 (19)	-0.17896 (17)	0.73816 (9)	0.0221 (3)
H4	0.1044	-0.2593	0.7760	0.027*
C5	0.06985 (18)	-0.12239 (18)	0.66605 (9)	0.0226 (3)
H5	-0.0589	-0.1643	0.6562	0.027*
C6	0.15795 (18)	-0.00650 (18)	0.60918 (9)	0.0209 (3)
H6	0.0904	0.0331	0.5613	0.025*
C7	0.34801 (17)	0.05157 (16)	0.62314 (8)	0.0177 (3)
C8	0.48188 (17)	0.16104 (17)	0.57534 (8)	0.0181 (3)
C9	0.44195 (17)	0.23455 (16)	0.49230 (8)	0.0187 (3)

supplementary materials

H9	0.3177	0.2033	0.4673	0.022*
C10	0.55628 (18)	0.34163 (16)	0.44489 (8)	0.0186 (3)
N11	0.74423 (15)	0.41748 (15)	0.46348 (7)	0.0197 (3)
H11	0.8114	0.4057	0.5131	0.024*
O12	0.96568 (13)	0.59079 (14)	0.38998 (7)	0.0276 (3)
C12	0.80892 (18)	0.51160 (16)	0.39495 (9)	0.0198 (3)
N13	0.66123 (15)	0.50316 (14)	0.33104 (7)	0.0205 (3)
H13	0.6677	0.5574	0.2811	0.025*
O14	0.35435 (13)	0.36140 (13)	0.31095 (7)	0.0247 (2)
C14	0.50269 (18)	0.39906 (16)	0.35539 (8)	0.0191 (3)
C15	0.77114 (17)	0.06679 (17)	0.76616 (8)	0.0197 (3)
H15A	0.8908	0.1013	0.7432	0.024*
H15B	0.7557	-0.0516	0.7868	0.024*
C16	0.76755 (17)	0.19009 (17)	0.84440 (8)	0.0186 (3)
C17	0.77627 (19)	0.36332 (18)	0.83052 (9)	0.0230 (3)
H17	0.7891	0.4045	0.7725	0.028*
C18	0.76628 (19)	0.47605 (17)	0.90102 (9)	0.0226 (3)
H18	0.7718	0.5937	0.8910	0.027*
C19	0.74814 (17)	0.41658 (17)	0.98642 (8)	0.0190 (3)
C20	0.7411 (2)	0.24417 (18)	1.00064 (9)	0.0244 (3)
H20	0.7298	0.2033	1.0588	0.029*
C21	0.7508 (2)	0.13161 (18)	0.92977 (9)	0.0242 (3)
H21	0.7459	0.0140	0.9398	0.029*
O22	0.70603 (16)	0.48178 (14)	1.13762 (7)	0.0298 (3)
C22	0.73352 (17)	0.53113 (17)	1.06416 (9)	0.0196 (3)
O23	0.75282 (15)	0.69475 (13)	1.04391 (6)	0.0274 (3)
C23	0.7376 (2)	0.8120 (2)	1.11646 (10)	0.0316 (4)
H23A	0.6119	0.7839	1.1309	0.047*
H23B	0.7709	0.9311	1.0982	0.047*
H23C	0.8197	0.8005	1.1694	0.047*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0158 (5)	0.0211 (5)	0.0150 (5)	0.0023 (4)	0.0008 (4)	-0.0007 (4)
C1	0.0186 (6)	0.0219 (6)	0.0150 (6)	0.0023 (5)	0.0034 (4)	-0.0005 (5)
C2	0.0163 (6)	0.0184 (6)	0.0162 (6)	0.0033 (5)	0.0020 (4)	-0.0038 (5)
C3	0.0216 (6)	0.0186 (6)	0.0180 (6)	0.0039 (5)	0.0034 (5)	-0.0005 (5)
C4	0.0230 (7)	0.0205 (6)	0.0219 (6)	0.0018 (5)	0.0077 (5)	-0.0007 (5)
C5	0.0168 (6)	0.0256 (7)	0.0236 (7)	0.0016 (5)	0.0044 (5)	-0.0046 (5)
C6	0.0196 (6)	0.0248 (7)	0.0173 (6)	0.0052 (5)	0.0003 (5)	-0.0022 (5)
C7	0.0189 (6)	0.0196 (6)	0.0141 (6)	0.0040 (5)	0.0029 (4)	-0.0025 (4)
C8	0.0185 (6)	0.0198 (6)	0.0152 (6)	0.0031 (5)	0.0034 (4)	-0.0011 (4)
C9	0.0194 (6)	0.0184 (6)	0.0176 (6)	0.0040 (5)	0.0025 (5)	-0.0010 (5)
C10	0.0202 (6)	0.0176 (6)	0.0176 (6)	0.0047 (5)	0.0018 (5)	-0.0010 (5)
N11	0.0199 (5)	0.0237 (6)	0.0156 (5)	0.0054 (4)	0.0020 (4)	0.0028 (4)
O12	0.0211 (5)	0.0310 (6)	0.0260 (5)	-0.0024 (4)	0.0022 (4)	0.0074 (4)
C12	0.0234 (6)	0.0173 (6)	0.0177 (6)	0.0032 (5)	0.0021 (5)	0.0006 (5)

N13	0.0232 (6)	0.0196 (5)	0.0172 (5)	0.0025 (4)	0.0013 (4)	0.0039 (4)
O14	0.0234 (5)	0.0281 (5)	0.0219 (5)	0.0058 (4)	-0.0005 (4)	0.0062 (4)
C14	0.0222 (6)	0.0180 (6)	0.0181 (6)	0.0066 (5)	0.0036 (5)	0.0011 (5)
C15	0.0170 (6)	0.0242 (6)	0.0174 (6)	0.0056 (5)	-0.0011 (5)	-0.0013 (5)
C16	0.0151 (6)	0.0222 (6)	0.0174 (6)	0.0041 (5)	-0.0008 (4)	-0.0009 (5)
C17	0.0283 (7)	0.0230 (7)	0.0158 (6)	0.0032 (5)	0.0021 (5)	0.0024 (5)
C18	0.0300 (7)	0.0185 (6)	0.0180 (6)	0.0037 (5)	0.0018 (5)	0.0021 (5)
C19	0.0187 (6)	0.0206 (6)	0.0166 (6)	0.0038 (5)	0.0000 (5)	-0.0001 (5)
C20	0.0336 (8)	0.0239 (7)	0.0167 (6)	0.0089 (6)	0.0033 (5)	0.0037 (5)
C21	0.0330 (7)	0.0202 (6)	0.0205 (6)	0.0091 (5)	0.0023 (5)	0.0024 (5)
O22	0.0460 (7)	0.0272 (5)	0.0174 (5)	0.0101 (5)	0.0075 (4)	0.0024 (4)
C22	0.0182 (6)	0.0214 (6)	0.0180 (6)	0.0039 (5)	-0.0010 (5)	0.0012 (5)
O23	0.0444 (6)	0.0197 (5)	0.0185 (5)	0.0090 (4)	0.0046 (4)	-0.0005 (4)
C23	0.0490 (10)	0.0245 (7)	0.0224 (7)	0.0130 (7)	0.0023 (6)	-0.0047 (6)

Geometric parameters (Å, °)

N1—C1	1.3668 (17)	C12—N13	1.3869 (17)
N1—C2	1.3846 (16)	N13—C14	1.3819 (17)
N1—C15	1.4554 (16)	N13—H13	0.8800
C1—C8	1.3821 (18)	O14—C14	1.2137 (16)
C1—H1	0.9500	C15—C16	1.5162 (17)
C2—C3	1.3951 (18)	C15—H15A	0.9900
C2—C7	1.4093 (18)	C15—H15B	0.9900
C3—C4	1.3866 (19)	C16—C21	1.3898 (18)
C3—H3	0.9500	C16—C17	1.3951 (19)
C4—C5	1.403 (2)	C17—C18	1.3890 (19)
C4—H4	0.9500	C17—H17	0.9500
C5—C6	1.3833 (19)	C18—C19	1.3940 (18)
C5—H5	0.9500	C18—H18	0.9500
C6—C7	1.3996 (18)	C19—C20	1.3921 (19)
C6—H6	0.9500	C19—C22	1.4916 (18)
C7—C8	1.4445 (17)	C20—C21	1.3915 (19)
C8—C9	1.4340 (18)	C20—H20	0.9500
C9—C10	1.3454 (18)	C21—H21	0.9500
C9—H9	0.9500	O22—C22	1.2083 (16)
C10—N11	1.4053 (17)	C22—O23	1.3303 (16)
C10—C14	1.4858 (18)	O23—C23	1.4475 (16)
N11—C12	1.3612 (17)	C23—H23A	0.9800
N11—H11	0.8800	C23—H23B	0.9800
O12—C12	1.2189 (17)	C23—H23C	0.9800
C1—N1—C2	109.12 (11)	C14—N13—H13	124.2
C1—N1—C15	124.49 (11)	C12—N13—H13	124.2
C2—N1—C15	125.24 (11)	O14—C14—N13	126.35 (12)
N1—C1—C8	110.17 (11)	O14—C14—C10	128.62 (12)
N1—C1—H1	124.9	N13—C14—C10	105.00 (11)
C8—C1—H1	124.9	N1—C15—C16	111.22 (10)
N1—C2—C3	129.95 (12)	N1—C15—H15A	109.4
N1—C2—C7	107.54 (11)	C16—C15—H15A	109.4

supplementary materials

C3—C2—C7	122.43 (12)	N1—C15—H15B	109.4
C4—C3—C2	117.24 (12)	C16—C15—H15B	109.4
C4—C3—H3	121.4	H15A—C15—H15B	108.0
C2—C3—H3	121.4	C21—C16—C17	119.35 (12)
C3—C4—C5	121.30 (12)	C21—C16—C15	120.72 (12)
C3—C4—H4	119.3	C17—C16—C15	119.91 (11)
C5—C4—H4	119.3	C18—C17—C16	120.43 (12)
C6—C5—C4	120.95 (12)	C18—C17—H17	119.8
C6—C5—H5	119.5	C16—C17—H17	119.8
C4—C5—H5	119.5	C17—C18—C19	120.04 (12)
C5—C6—C7	119.10 (12)	C17—C18—H18	120.0
C5—C6—H6	120.5	C19—C18—H18	120.0
C7—C6—H6	120.5	C20—C19—C18	119.66 (12)
C6—C7—C2	118.93 (12)	C20—C19—C22	118.03 (12)
C6—C7—C8	133.86 (12)	C18—C19—C22	122.31 (12)
C2—C7—C8	107.19 (11)	C21—C20—C19	120.11 (12)
C1—C8—C9	128.94 (12)	C21—C20—H20	119.9
C1—C8—C7	105.98 (11)	C19—C20—H20	119.9
C9—C8—C7	124.90 (12)	C16—C21—C20	120.42 (13)
C10—C9—C8	128.91 (12)	C16—C21—H21	119.8
C10—C9—H9	115.5	C20—C21—H21	119.8
C8—C9—H9	115.5	O22—C22—O23	123.20 (12)
C9—C10—N11	130.48 (12)	O22—C22—C19	124.14 (12)
C9—C10—C14	124.45 (12)	O23—C22—C19	112.66 (11)
N11—C10—C14	105.04 (11)	C22—O23—C23	115.26 (11)
C12—N11—C10	111.28 (11)	O23—C23—H23A	109.5
C12—N11—H11	124.4	O23—C23—H23B	109.5
C10—N11—H11	124.4	H23A—C23—H23B	109.5
O12—C12—N11	127.32 (12)	O23—C23—H23C	109.5
O12—C12—N13	125.72 (12)	H23A—C23—H23C	109.5
N11—C12—N13	106.95 (11)	H23B—C23—H23C	109.5
C14—N13—C12	111.67 (11)		
C2—N1—C1—C8	-0.64 (14)	C10—N11—C12—N13	1.94 (14)
C15—N1—C1—C8	-168.87 (11)	O12—C12—N13—C14	178.35 (13)
C1—N1—C2—C3	177.11 (13)	N11—C12—N13—C14	-2.48 (15)
C15—N1—C2—C3	-14.8 (2)	C12—N13—C14—O14	-176.40 (13)
C1—N1—C2—C7	0.36 (14)	C12—N13—C14—C10	1.99 (14)
C15—N1—C2—C7	168.48 (11)	C9—C10—C14—O14	-0.3 (2)
N1—C2—C3—C4	-176.00 (12)	N11—C10—C14—O14	177.59 (13)
C7—C2—C3—C4	0.32 (18)	C9—C10—C14—N13	-178.67 (12)
C2—C3—C4—C5	-1.33 (19)	N11—C10—C14—N13	-0.75 (13)
C3—C4—C5—C6	0.5 (2)	C1—N1—C15—C16	94.54 (14)
C4—C5—C6—C7	1.38 (19)	C2—N1—C15—C16	-71.81 (15)
C5—C6—C7—C2	-2.33 (18)	N1—C15—C16—C21	122.96 (13)
C5—C6—C7—C8	175.72 (13)	N1—C15—C16—C17	-55.33 (16)
N1—C2—C7—C6	178.56 (11)	C21—C16—C17—C18	-0.7 (2)
C3—C2—C7—C6	1.52 (18)	C15—C16—C17—C18	177.60 (12)
N1—C2—C7—C8	0.03 (14)	C16—C17—C18—C19	0.3 (2)
C3—C2—C7—C8	-177.01 (11)	C17—C18—C19—C20	0.4 (2)

N1—C1—C8—C9	-174.63 (12)	C17—C18—C19—C22	-178.91 (12)
N1—C1—C8—C7	0.63 (14)	C18—C19—C20—C21	-0.5 (2)
C6—C7—C8—C1	-178.62 (14)	C22—C19—C20—C21	178.79 (12)
C2—C7—C8—C1	-0.40 (14)	C17—C16—C21—C20	0.6 (2)
C6—C7—C8—C9	-3.1 (2)	C15—C16—C21—C20	-177.73 (12)
C2—C7—C8—C9	175.11 (12)	C19—C20—C21—C16	0.0 (2)
C1—C8—C9—C10	-7.9 (2)	C20—C19—C22—O22	-4.2 (2)
C7—C8—C9—C10	177.63 (13)	C18—C19—C22—O22	175.06 (13)
C8—C9—C10—N11	-2.6 (2)	C20—C19—C22—O23	176.23 (12)
C8—C9—C10—C14	174.74 (12)	C18—C19—C22—O23	-4.50 (18)
C9—C10—N11—C12	177.01 (13)	O22—C22—O23—C23	-0.3 (2)
C14—C10—N11—C12	-0.74 (14)	C19—C22—O23—C23	179.24 (11)
C10—N11—C12—O12	-178.91 (13)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N11—H11 \cdots O12 ⁱ	0.88	2.11	2.9658 (15)	163
N13—H13 \cdots O22 ⁱⁱ	0.88	2.29	2.9699 (15)	134

Symmetry codes: (i) $-x+2, -y+1, -z+1$; (ii) $x, y, z-1$.

Fig. 1

