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N'-(2-Hydroxy-3-methoxybenzylidene)-1,3-benzodioxole-5-carbohydrazide monohydrate

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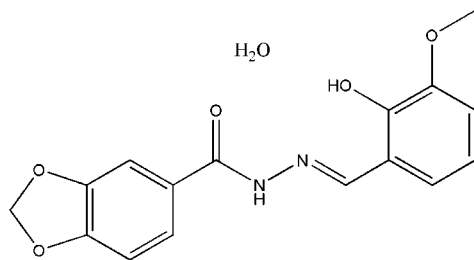
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 Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.032; wR factor = 0.076; data-to-parameter ratio = 8.4.

Single crystals of the title compound, $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_5 \cdot \text{H}_2\text{O}$, were obtained from a condensation reaction of 1,3-benzodioxole-5-carbohydrazide and 3-methoxysalicylaldehyde in a 95% ethanol solution. The asymmetric unit consists of a Schiff base molecule, which assumes an *E* configuration with respect to the $\text{C}=\text{N}$ bond, and a water molecule of crystallization. The dihedral angle between the two substituted benzene rings is $12.7(2)^\circ$. In the crystal structure, molecules are linked through intermolecular $\text{N}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{O}$ hydrogen bonds, forming layers parallel to the *bc* plane.

Related literature

For the biological properties of hydrazones, see: Bedia *et al.* (2006); Rollas *et al.* (2002); Okabe *et al.* (1993). For bond-length data, see: Allen *et al.* (1987). For related structures, see: Fun *et al.* (2008); Qu *et al.* (2008); Shan *et al.* (2008); Yehye *et al.* (2008).



Experimental

Crystal data

 $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_5 \cdot \text{H}_2\text{O}$
 $M_r = 332.31$

 Orthorhombic, $P2_12_12_1$
 $a = 4.792(2)$ Å

 $b = 12.916(3)$ Å
 $c = 24.002(6)$ Å
 $V = 1485.6(7)$ Å³
 $Z = 4$

 Mo $K\alpha$ radiation
 $\mu = 0.12$ mm⁻¹
 $T = 298(2)$ K
 $0.23 \times 0.20 \times 0.20$ mm

Data collection

 Bruker SMART 1K CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.974$, $T_{\max} = 0.977$

 8595 measured reflections
 1907 independent reflections
 1639 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.030$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.032$
 $wR(F^2) = 0.076$
 $S = 1.05$
 1907 reflections
 228 parameters
 4 restraints

 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.13$ e Å⁻³
 $\Delta\rho_{\min} = -0.13$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H} \cdots A$	$D-\text{H}$	$\text{H} \cdots A$	$D \cdots A$	$D-\text{H} \cdots A$
$\text{O1}-\text{H1} \cdots \text{N1}$	0.82	2.04	2.743 (2)	143
$\text{O1}-\text{H1} \cdots \text{O6}$	0.82	2.56	3.001 (2)	115
$\text{N2}-\text{H2} \cdots \text{O6}^i$	0.899 (10)	2.075 (11)	2.962 (2)	168 (3)
$\text{O6}-\text{H6A} \cdots \text{O2}$	0.857 (10)	1.880 (12)	2.728 (2)	170 (3)
$\text{O6}-\text{H6B} \cdots \text{O1}^{ii}$	0.848 (10)	2.269 (16)	3.043 (2)	152 (2)
$\text{O6}-\text{H6B} \cdots \text{O3}^{ii}$	0.848 (10)	2.538 (19)	3.226 (2)	139 (2)

 Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x - 1, y, z$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SJ2559).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bedia, K.-K., Elcin, O., Seda, U., Fatma, K., Nathaly, S., Sevim, R. & Dimoglo, A. (2006). *Eur. J. Med. Chem.* **41**, 1253–1261.
- Bruker (2002). SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Fun, H.-K., Sujith, K. V., Patil, P. S., Kalluraya, B. & Chantrapromma, S. (2008). *Acta Cryst.* **E64**, o1961–o1962.
- Okabe, N., Nakamura, T. & Fukuda, H. (1993). *Acta Cryst.* **C49**, 1678–1680.
- Qu, L.-Z., Yang, T., Cao, G.-B. & Wang, X.-Y. (2008). *Acta Cryst.* **E64**, o2061.
- Rollas, S., Gülerman, N. & Erdeniz, H. (2002). *Farmaco*, **57**, 171–174.
- Shan, S., Tian, Y.-L., Wang, S.-H., Wang, W.-L. & Xu, Y.-L. (2008). *Acta Cryst.* **E64**, o1363.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Yehye, W. A., Ariffin, A. & Ng, S. W. (2008). *Acta Cryst.* **E64**, o960.

supplementary materials

Acta Cryst. (2009). E65, o29 [doi:10.1107/S1600536808040117]

N'-(2-Hydroxy-3-methoxybenzylidene)-1,3-benzodioxole-5-carbohydrazide monohydrate

C.-L. Du

Comment

Hydrazone compounds, derived from the condensation reactions of aldehydes with hydrazides, show interesting biological properties (Okabe *et al.*, 1993; Bedia *et al.*, 2006; Rollas *et al.*, 2002). Recently, a large number of hydrazone derivatives have been reported (Shan *et al.*, 2008; Fun *et al.*, 2008; Qu *et al.*, 2008; Yehye *et al.*, 2008). We report here the structure of a new hydrazone compound, I, Fig. 1, with a Schiff base molecule, which assumes an *E* configuration with respect to the C=N bond and a water molecule in the asymmetric unit. The dihedral angle between the two substituted benzene rings is 12.7 (2)°. All the bond lengths are within normal ranges (Allen *et al.*, 1987).

In the crystal structure, molecules are linked through intermolecular N—H···O and O—H···O hydrogen bonds (Table 1), forming layers parallel to the *bc* direction (Fig. 2).

Experimental

The title compound was prepared by Schiff base condensation reaction of 1,3-benzodioxole-5-carbohydrazide (1.0 mmol) and 3-methoxysalicylaldehyde (1.0 mmol) in a 95% ethanol solution (50 ml). Needle colorless crystals were formed by gradual evaporation of the solution in air for a few days.

Refinement

The imino H atom was located in a difference map and refined with a N—H distance restraint of 0.90 (1) Å. The water H atoms were also located in a difference map and refined with O—H and H···H distances restraints of 0.85 (1) and 1.37 (2) Å, respectively. The other H atoms were positioned geometrically [C—H = 0.93–0.97 Å] and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$. A rotating group model was used for the methyl group. In the absence of significant anomalous scattering effects, 1034 Friedel pairs were merged.

Figures

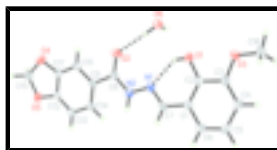


Fig. 1. The structure of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme.

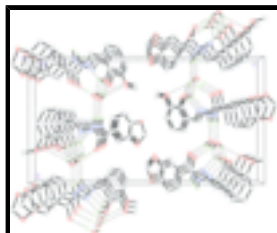


Fig. 2. The crystal packing of the title compound, viewed down the *a* axis.

N'-(2-Hydroxy-3-methoxybenzylidene)-1,3-benzodioxole-5-carbohydrazide monohydrate

Crystal data

C₁₆H₁₄N₂O₅·H₂O

M_r = 332.31

Orthorhombic, *P*2₁2₁2₁

Hall symbol: P 2ac 2ab

a = 4.792 (2) Å

b = 12.916 (3) Å

c = 24.002 (6) Å

V = 1485.6 (7) Å³

Z = 4

*F*₀₀₀ = 696

D_x = 1.486 Mg m⁻³

Mo *K*α radiation

λ = 0.71073 Å

Cell parameters from 3222 reflections

θ = 2.4–25.6°

μ = 0.12 mm⁻¹

T = 298 (2) K

Cut from needle, colorless

0.23 × 0.20 × 0.20 mm

Data collection

Bruker SMART 1K CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

T = 298(2) K

ω scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

T_{min} = 0.974, *T_{max}* = 0.977

8595 measured reflections

1907 independent reflections

1639 reflections with *I* > 2σ(*I*)

R_{int} = 0.030

θ_{max} = 27.0°

θ_{min} = 1.7°

h = -6→6

k = -16→13

l = -30→28

Refinement

Refinement on *F*²

Least-squares matrix: full

R [*F*² > 2σ(*F*²)] = 0.032

wR (*F*²) = 0.076

S = 1.05

1907 reflections

228 parameters

4 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0385P)^2 + 0.1335P]$$

where *P* = (*F_o*² + 2*F_c*²)/3

(Δ/σ)_{max} < 0.001

Δρ_{max} = 0.13 e Å⁻³

Δρ_{min} = -0.13 e Å⁻³

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.9802 (3)	0.80320 (10)	0.83305 (6)	0.0447 (4)
H1	0.8775	0.8121	0.8061	0.067*
O2	0.3794 (4)	0.78499 (11)	0.69933 (6)	0.0563 (5)
O3	1.3470 (3)	0.79317 (11)	0.91162 (6)	0.0489 (4)
O4	-0.3191 (4)	0.81446 (12)	0.53841 (6)	0.0599 (5)
O5	-0.3991 (3)	0.98897 (11)	0.52200 (6)	0.0490 (4)
O6	0.4857 (4)	0.67908 (12)	0.79533 (6)	0.0528 (4)
N1	0.7019 (4)	0.92306 (13)	0.75678 (6)	0.0368 (4)
N2	0.5153 (4)	0.95059 (13)	0.71517 (7)	0.0374 (4)
C1	1.0510 (4)	0.98797 (15)	0.82015 (7)	0.0349 (4)
C2	1.1069 (4)	0.89400 (14)	0.84647 (8)	0.0330 (4)
C3	1.3082 (4)	0.88967 (15)	0.88912 (8)	0.0360 (4)
C4	1.4481 (5)	0.97831 (16)	0.90522 (8)	0.0415 (5)
H4	1.5789	0.9754	0.9338	0.050*
C5	1.3937 (5)	1.07197 (16)	0.87871 (8)	0.0449 (5)
H5	1.4897	1.1314	0.8893	0.054*
C6	1.1978 (5)	1.07680 (15)	0.83680 (8)	0.0418 (5)
H6	1.1624	1.1397	0.8193	0.050*
C7	0.8453 (5)	0.99888 (16)	0.77596 (8)	0.0388 (5)
H7	0.8163	1.0643	0.7608	0.047*
C8	0.3608 (5)	0.87806 (15)	0.68878 (7)	0.0371 (5)
C9	0.1638 (5)	0.91647 (14)	0.64528 (7)	0.0330 (4)
C10	0.0244 (5)	0.84061 (15)	0.61392 (8)	0.0383 (5)
H10	0.0559	0.7705	0.6200	0.046*
C11	-0.1597 (5)	0.87353 (15)	0.57404 (8)	0.0385 (5)
C12	-0.2088 (5)	0.97742 (15)	0.56457 (7)	0.0359 (4)
C13	-0.0799 (5)	1.05292 (15)	0.59480 (8)	0.0412 (5)
H13	-0.1162	1.1227	0.5885	0.049*
C14	0.1088 (5)	1.02087 (15)	0.63555 (8)	0.0386 (5)
H14	0.2004	1.0705	0.6568	0.046*
C15	-0.4686 (6)	0.88603 (18)	0.50433 (8)	0.0497 (6)
H15A	-0.6678	0.8746	0.5080	0.060*
H15B	-0.4179	0.8765	0.4655	0.060*

supplementary materials

C16	1.5697 (5)	0.78022 (19)	0.95022 (9)	0.0522 (6)
H16A	1.5365	0.8226	0.9824	0.078*
H16B	1.7421	0.8005	0.9330	0.078*
H16C	1.5806	0.7089	0.9613	0.078*
H2	0.499 (7)	1.0186 (9)	0.7078 (10)	0.080*
H6A	0.474 (6)	0.7133 (18)	0.7649 (6)	0.080*
H6B	0.366 (5)	0.706 (2)	0.8170 (8)	0.080*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0460 (9)	0.0381 (8)	0.0500 (9)	-0.0062 (7)	-0.0165 (7)	-0.0011 (7)
O2	0.0772 (12)	0.0351 (8)	0.0567 (9)	0.0061 (9)	-0.0217 (9)	0.0055 (7)
O3	0.0506 (10)	0.0414 (8)	0.0548 (8)	-0.0064 (8)	-0.0222 (8)	0.0105 (7)
O4	0.0706 (12)	0.0453 (8)	0.0638 (10)	0.0007 (9)	-0.0276 (10)	-0.0127 (8)
O5	0.0524 (10)	0.0516 (9)	0.0430 (8)	0.0055 (8)	-0.0141 (8)	-0.0009 (7)
O6	0.0684 (12)	0.0394 (8)	0.0505 (9)	0.0050 (9)	-0.0110 (9)	0.0016 (7)
N1	0.0366 (9)	0.0429 (9)	0.0308 (8)	0.0065 (8)	-0.0034 (8)	0.0020 (7)
N2	0.0394 (10)	0.0373 (9)	0.0356 (8)	0.0034 (8)	-0.0070 (8)	0.0044 (7)
C1	0.0326 (11)	0.0389 (10)	0.0333 (9)	0.0042 (9)	0.0022 (8)	-0.0020 (8)
C2	0.0319 (10)	0.0339 (9)	0.0332 (9)	-0.0007 (9)	-0.0005 (8)	-0.0039 (8)
C3	0.0357 (11)	0.0380 (10)	0.0344 (9)	-0.0002 (10)	-0.0025 (9)	0.0003 (8)
C4	0.0389 (12)	0.0462 (11)	0.0393 (10)	-0.0042 (10)	-0.0058 (10)	-0.0047 (9)
C5	0.0442 (13)	0.0380 (11)	0.0525 (12)	-0.0065 (10)	-0.0029 (11)	-0.0063 (9)
C6	0.0450 (13)	0.0335 (10)	0.0469 (11)	0.0031 (10)	0.0012 (11)	0.0001 (9)
C7	0.0418 (12)	0.0381 (10)	0.0365 (10)	0.0068 (11)	-0.0019 (9)	0.0038 (8)
C8	0.0424 (12)	0.0366 (10)	0.0323 (10)	0.0055 (10)	0.0005 (9)	0.0019 (8)
C9	0.0356 (11)	0.0333 (9)	0.0302 (9)	0.0016 (9)	0.0021 (8)	0.0020 (7)
C10	0.0446 (12)	0.0325 (10)	0.0377 (10)	0.0031 (10)	-0.0014 (10)	0.0007 (8)
C11	0.0419 (13)	0.0376 (10)	0.0360 (10)	-0.0013 (10)	-0.0009 (9)	-0.0064 (8)
C12	0.0357 (11)	0.0438 (11)	0.0283 (9)	0.0039 (9)	0.0002 (9)	0.0030 (8)
C13	0.0508 (13)	0.0314 (10)	0.0414 (10)	0.0050 (10)	-0.0056 (10)	0.0021 (9)
C14	0.0442 (12)	0.0345 (10)	0.0372 (10)	-0.0001 (9)	-0.0066 (10)	-0.0010 (8)
C15	0.0480 (14)	0.0602 (14)	0.0410 (11)	-0.0083 (13)	-0.0079 (11)	0.0006 (10)
C16	0.0456 (13)	0.0630 (14)	0.0478 (12)	-0.0021 (12)	-0.0128 (11)	0.0145 (11)

Geometric parameters (\AA , $^\circ$)

O1—C2	1.359 (2)	C4—H4	0.9300
O1—H1	0.8200	C5—C6	1.377 (3)
O2—C8	1.232 (2)	C5—H5	0.9300
O3—C3	1.371 (2)	C6—H6	0.9300
O3—C16	1.423 (2)	C7—H7	0.9300
O4—C11	1.377 (2)	C8—C9	1.492 (3)
O4—C15	1.427 (3)	C9—C14	1.394 (3)
O5—C12	1.378 (2)	C9—C10	1.405 (3)
O5—C15	1.435 (3)	C10—C11	1.369 (3)
O6—H6A	0.857 (10)	C10—H10	0.9300
O6—H6B	0.848 (10)	C11—C12	1.381 (3)

N1—C7	1.282 (3)	C12—C13	1.363 (3)
N1—N2	1.387 (2)	C13—C14	1.395 (3)
N2—C8	1.352 (3)	C13—H13	0.9300
N2—H2	0.899 (10)	C14—H14	0.9300
C1—C2	1.394 (3)	C15—H15A	0.9700
C1—C6	1.404 (3)	C15—H15B	0.9700
C1—C7	1.455 (3)	C16—H16A	0.9600
C2—C3	1.408 (3)	C16—H16B	0.9600
C3—C4	1.382 (3)	C16—H16C	0.9600
C4—C5	1.392 (3)		
C2—O1—H1	109.5	N2—C8—C9	116.35 (16)
C3—O3—C16	117.72 (17)	C14—C9—C10	119.66 (18)
C11—O4—C15	105.99 (16)	C14—C9—C8	123.96 (18)
C12—O5—C15	105.81 (15)	C10—C9—C8	116.36 (17)
H6A—O6—H6B	105.5 (19)	C11—C10—C9	117.68 (18)
C7—N1—N2	114.10 (16)	C11—C10—H10	121.2
C8—N2—N1	120.86 (16)	C9—C10—H10	121.2
C8—N2—H2	122.6 (19)	C10—C11—O4	128.26 (18)
N1—N2—H2	116.6 (19)	C10—C11—C12	121.75 (19)
C2—C1—C6	119.08 (18)	O4—C11—C12	109.99 (18)
C2—C1—C7	123.01 (19)	C13—C12—O5	128.10 (17)
C6—C1—C7	117.90 (18)	C13—C12—C11	122.03 (18)
O1—C2—C1	123.91 (17)	O5—C12—C11	109.86 (17)
O1—C2—C3	116.37 (16)	C12—C13—C14	117.03 (17)
C1—C2—C3	119.72 (17)	C12—C13—H13	121.5
O3—C3—C4	125.24 (18)	C14—C13—H13	121.5
O3—C3—C2	114.56 (17)	C9—C14—C13	121.83 (18)
C4—C3—C2	120.20 (18)	C9—C14—H14	119.1
C3—C4—C5	120.09 (19)	C13—C14—H14	119.1
C3—C4—H4	120.0	O4—C15—O5	108.32 (17)
C5—C4—H4	120.0	O4—C15—H15A	110.0
C6—C5—C4	120.1 (2)	O5—C15—H15A	110.0
C6—C5—H5	120.0	O4—C15—H15B	110.0
C4—C5—H5	120.0	O5—C15—H15B	110.0
C5—C6—C1	120.83 (19)	H15A—C15—H15B	108.4
C5—C6—H6	119.6	O3—C16—H16A	109.5
C1—C6—H6	119.6	O3—C16—H16B	109.5
N1—C7—C1	123.44 (18)	H16A—C16—H16B	109.5
N1—C7—H7	118.3	O3—C16—H16C	109.5
C1—C7—H7	118.3	H16A—C16—H16C	109.5
O2—C8—N2	122.73 (19)	H16B—C16—H16C	109.5
O2—C8—C9	120.92 (19)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O1—H1 \cdots N1	0.82	2.04	2.743 (2)	143
O1—H1 \cdots O6	0.82	2.56	3.001 (2)	115
N2—H2 \cdots O6 ⁱ	0.899 (10)	2.075 (11)	2.962 (2)	168 (3)

supplementary materials

O6—H6A···O2	0.857 (10)	1.880 (12)	2.728 (2)	170 (3)
O6—H6B···O1 ⁱⁱ	0.848 (10)	2.269 (16)	3.043 (2)	152 (2)
O6—H6B···O3 ⁱⁱ	0.848 (10)	2.538 (19)	3.226 (2)	139 (2)

Symmetry codes: (i) $-x+1, y+1/2, -z+3/2$; (ii) $x-1, y, z$.

Fig. 1

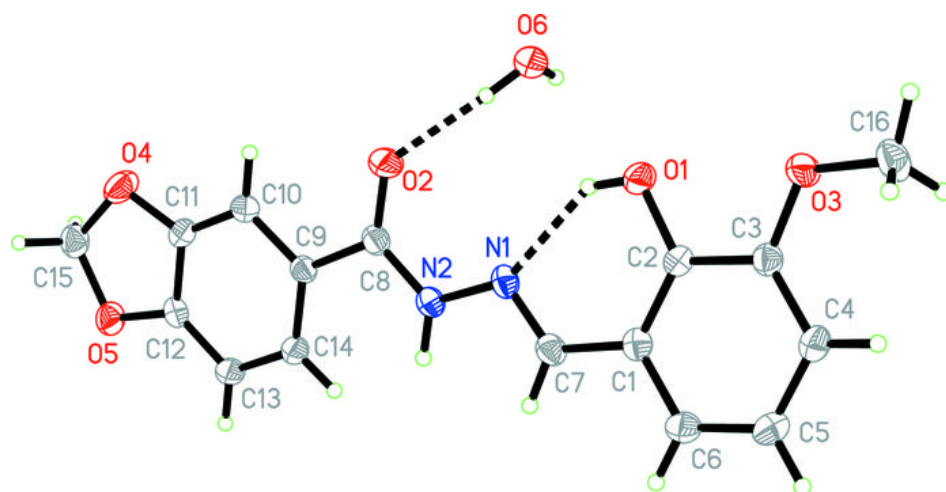


Fig. 2

