

r-2,c-6-Bis(4-methoxyphenyl)-c-3,t-3-dimethyl-1-nitrosopiperidin-4-one

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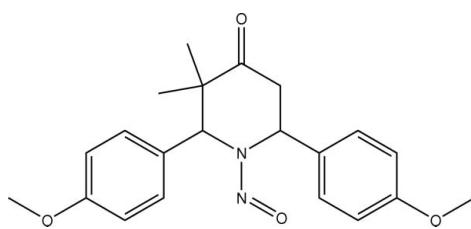
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Key indicators: single-crystal X-ray study; $T = 293\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$;
 R factor = 0.039; wR factor = 0.109; data-to-parameter ratio = 13.2.

In the title compound, $\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4$, the piperidine ring adopts a distorted boat conformation. The crystal structure is stabilized by $\text{C}-\text{H}\cdots\pi$ interactions involving one of the methoxyphenyl rings.

Related literature

For the biological activity of piperidones, see: Dimmock *et al.* (1990); Mutus *et al.* (1989); Perumal *et al.* (2001). For ring conformations, see: Cremer & Pople (1975); Nardelli (1983).



Experimental

Crystal data

$\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4$	$V = 1863.64 (13)\text{ \AA}^3$
$M_r = 368.42$	$Z = 4$
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
$a = 7.2540 (3)\text{ \AA}$	$\mu = 0.09\text{ mm}^{-1}$
$b = 15.0469 (6)\text{ \AA}$	$T = 293\text{ K}$
$c = 17.0741 (7)\text{ \AA}$	$0.30 \times 0.25 \times 0.20\text{ mm}$

Data collection

Bruker Kappa APEXII CCD area-detector diffractometer	24656 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 2001)	3211 independent reflections
$R_{\text{int}} = 0.026$	2595 reflections with $I > 2\sigma(I)$
$T_{\text{min}} = 0.973$, $T_{\text{max}} = 0.982$	

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$	244 parameters
$wR(F^2) = 0.109$	H-atom parameters constrained
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.20\text{ e \AA}^{-3}$
3211 reflections	$\Delta\rho_{\text{min}} = -0.15\text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$ is the centroid of the C16–C21 ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
C15–H15C \cdots CG1 ⁱ	0.96	2.97	3.9108 (26)	167
C23–H23C \cdots CG1 ⁱⁱ	0.96	2.86	3.7201 (27)	149

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x - \frac{1}{2}, -y + \frac{1}{2}, -z$.

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1983).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2805).

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supporting information

Acta Cryst. (2009). E65, o1420 [doi:10.1107/S1600536809019357]

r-2,c-6-Bis(4-methoxyphenyl)-c-3,t-3-dimethyl-1-nitrosopiperidin-4-one

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S1. Comment

2,6-Disubstituted 4-piperidones are found to have various biological and pharmacological activities (Dimmock *et al.*, 1990; Mutus *et al.*, 1989). Piperidones are also reported to possess analgesic, anti-inflammatory, central nervous system (CNS), local anaesthetic, anticancer and antimicrobial activities (Perumal *et al.*, 2001).

The piperidine ring adopts a distorted boat conformation, with puckering parameters (Cremer & Pople, 1975) $q_2 = 0.613$ (2) Å, $q_3 = -0.123$ (2) Å and $\varphi_2 = 260.6$ (2)° and the asymmetry parameters $\Delta C_s(C2) = \Delta C_s(C5) = 21.7$ (2)° (Nardelli, 1983). The C8—C13 and C16—C21 phenyl rings are oriented at angles of 88.04 (6)° and 82.38 (7)°, respectively, with the best plane (N1/C3/C4/C6) through the piperidine ring. The C14 methyl group is oriented axially [N1—C2—C3—C14 = 53.8 (2)°] while the C15 methyl group is oriented equatorially [N1—C2—C3—C15 = 173.3 (2)°] to the piperidinone ring. The sum of bond angles around N1 [358.1°] shows that the atom N1 is in sp^2 hybridized state. There is a delocalization between the lone pair of electrons and the hetero π -electrons of the nitroso group.

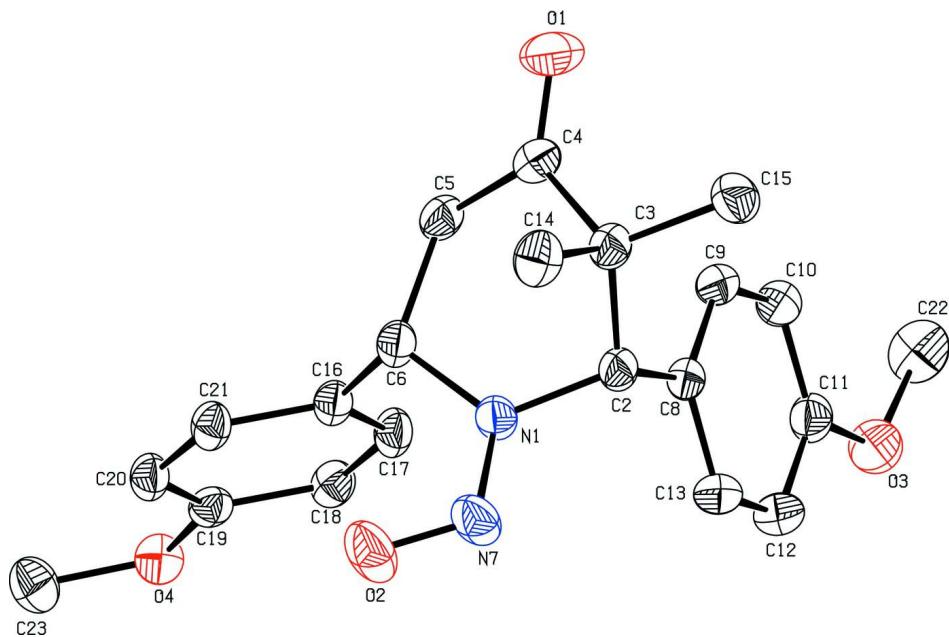
The packing of the molecules in the crystal is through C—H \cdots π interactions.

S2. Experimental

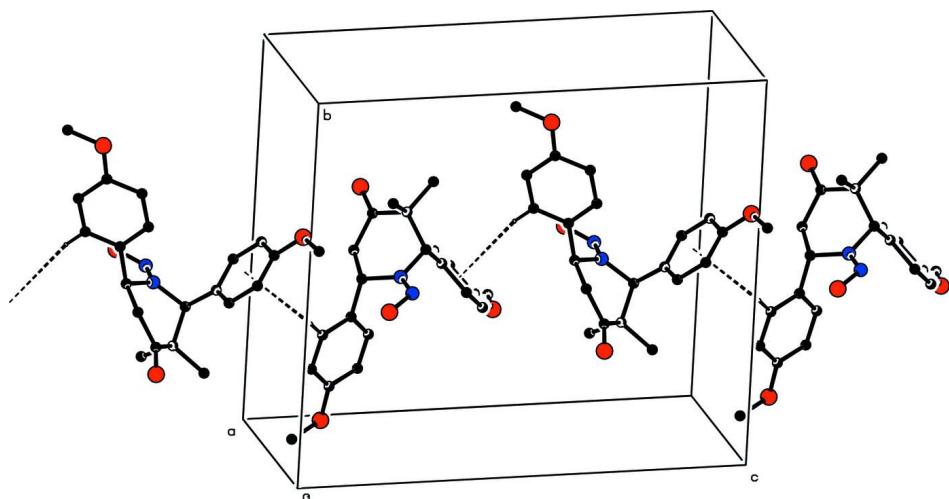
To a solution of r-2,c-6-bis(4-methoxyphenyl)-c-3,t-3-dimethylpiperidin-4-one (1.69 g, 5 mmol) in chloroform (10 ml) was added with conc. HCl (1.5 ml) and water (1.5 ml) and while stirring, solid NaNO₂ (0.84 g, 12 mmol) was added in portions over the period of 0.5 h. The solution was stirred at room temperature for another 0.5 h. The organic layer was washed with water, saturated with aqueous NaHCO₃ and dried over anhydrous Na₂SO₄. The resulting solution was concentrated and the residue was crystallized from ethanol.

S3. Refinement

H atoms were positioned geometrically (C—H = 0.93–0.98 Å) and allowed to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H and $1.2U_{\text{eq}}(\text{C})$ for other H atoms. In the absence of significant anomalous scattering effects, Friedel pairs were averaged.

**Figure 1**

Molecular structure of the title compound. Displacement ellipsoids are drawn at the 30% probability level. H atoms have been omitted for clarity.

**Figure 2**

Part of the crystal packing of the title compound, viewed approximately along the a axis. Dashed lines indicate $\text{C}—\text{H}\cdots\pi$ interactions. H atoms not involved in the interactions have been omitted.

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Crystal data

$\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4$

$M_r = 368.42$

Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 7.2540 (3)$ Å

$b = 15.0469 (6)$ Å

$c = 17.0741 (7)$ Å

$V = 1863.64 (13)$ Å³

$Z = 4$

$F(000) = 784$

$D_x = 1.313 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
 Cell parameters from 3211 reflections
 $\theta = 2.4\text{--}30.5^\circ$

$\mu = 0.09 \text{ mm}^{-1}$
 $T = 293 \text{ K}$
 Block, colourless
 $0.30 \times 0.25 \times 0.20 \text{ mm}$

Data collection

Bruker Kappa-APEXII CCD area-detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 ω scans
 Absorption correction: multi-scan
 (*SADABS*; Sheldrick, 2001)
 $T_{\min} = 0.973$, $T_{\max} = 0.982$

24656 measured reflections
 3211 independent reflections
 2595 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$
 $\theta_{\max} = 30.5^\circ$, $\theta_{\min} = 2.4^\circ$
 $h = -10 \rightarrow 10$
 $k = -21 \rightarrow 21$
 $l = -24 \rightarrow 24$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.109$
 $S = 1.03$
 3211 reflections
 244 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0532P)^2 + 0.2754P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.20 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.15 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.0483 (2)	0.52808 (12)	0.21105 (9)	0.0433 (4)
C2	0.1305 (2)	0.58054 (13)	0.27489 (10)	0.0417 (4)
H2	0.0276	0.6015	0.3072	0.050*
C3	0.2206 (3)	0.66398 (14)	0.23982 (11)	0.0453 (4)
C4	0.3709 (3)	0.63719 (14)	0.18259 (11)	0.0478 (4)
C5	0.3450 (3)	0.54846 (14)	0.14164 (11)	0.0464 (4)
H5A	0.4235	0.5054	0.1675	0.056*
H5B	0.3898	0.5547	0.0884	0.056*
C6	0.1494 (3)	0.50917 (13)	0.13767 (10)	0.0419 (4)
H6	0.0840	0.5392	0.0950	0.050*
N7	-0.1331 (2)	0.51769 (14)	0.21570 (12)	0.0589 (5)
C8	0.2461 (3)	0.52025 (13)	0.32719 (10)	0.0413 (4)

C9	0.4369 (3)	0.51945 (15)	0.33242 (11)	0.0461 (4)
H9	0.5048	0.5589	0.3020	0.055*
C10	0.5290 (3)	0.46138 (15)	0.38177 (12)	0.0500 (5)
H10	0.6571	0.4620	0.3837	0.060*
C11	0.4317 (3)	0.40259 (13)	0.42807 (12)	0.0484 (4)
C12	0.2405 (3)	0.40122 (16)	0.42284 (13)	0.0562 (5)
H12	0.1730	0.3611	0.4528	0.067*
C13	0.1513 (3)	0.45897 (15)	0.37357 (12)	0.0517 (5)
H13	0.0233	0.4573	0.3710	0.062*
C15	0.2886 (4)	0.72672 (16)	0.30428 (13)	0.0581 (5)
H15A	0.3445	0.7781	0.2809	0.087*
H15B	0.3777	0.6965	0.3362	0.087*
H15C	0.1863	0.7449	0.3361	0.087*
C14	0.0774 (3)	0.71461 (16)	0.18933 (14)	0.0578 (5)
H14A	0.1337	0.7666	0.1673	0.087*
H14B	-0.0251	0.7319	0.2215	0.087*
H14C	0.0347	0.6767	0.1479	0.087*
C16	0.1535 (2)	0.41093 (12)	0.11881 (10)	0.0393 (4)
C17	0.2309 (3)	0.34891 (14)	0.16965 (10)	0.0449 (4)
H17	0.2793	0.3677	0.2173	0.054*
C18	0.2366 (3)	0.26016 (14)	0.15030 (11)	0.0443 (4)
H18	0.2889	0.2195	0.1848	0.053*
C19	0.1647 (2)	0.23100 (12)	0.07945 (10)	0.0402 (4)
C20	0.0887 (3)	0.29189 (13)	0.02797 (11)	0.0430 (4)
H20	0.0414	0.2731	-0.0199	0.052*
C21	0.0838 (3)	0.38045 (14)	0.04814 (10)	0.0431 (4)
H21	0.0322	0.4210	0.0134	0.052*
C22	0.7010 (4)	0.34767 (19)	0.49303 (17)	0.0719 (7)
H22A	0.7349	0.3045	0.5318	0.108*
H22B	0.7358	0.4059	0.5107	0.108*
H22C	0.7632	0.3346	0.4448	0.108*
C23	0.1004 (4)	0.10995 (15)	-0.00595 (14)	0.0605 (6)
H23A	0.1152	0.0466	-0.0089	0.091*
H23B	0.1638	0.1374	-0.0490	0.091*
H23C	-0.0283	0.1245	-0.0085	0.091*
O1	0.5033 (3)	0.68301 (12)	0.16938 (12)	0.0732 (5)
O2	-0.2050 (2)	0.48520 (14)	0.15739 (11)	0.0753 (5)
O3	0.5071 (3)	0.34479 (11)	0.48087 (10)	0.0640 (4)
O4	0.1751 (2)	0.14144 (9)	0.06582 (8)	0.0509 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0311 (6)	0.0576 (9)	0.0412 (7)	0.0002 (7)	0.0008 (6)	-0.0093 (7)
C2	0.0358 (8)	0.0543 (10)	0.0348 (8)	-0.0009 (8)	0.0034 (7)	-0.0084 (7)
C3	0.0446 (9)	0.0519 (10)	0.0393 (8)	0.0004 (8)	0.0037 (8)	-0.0051 (8)
C4	0.0461 (10)	0.0531 (11)	0.0442 (9)	-0.0003 (9)	0.0077 (8)	0.0035 (8)
C5	0.0432 (9)	0.0573 (11)	0.0387 (8)	0.0012 (9)	0.0088 (8)	-0.0034 (8)

C6	0.0403 (8)	0.0522 (10)	0.0331 (8)	0.0041 (8)	-0.0015 (7)	-0.0029 (7)
N7	0.0347 (8)	0.0767 (12)	0.0654 (11)	-0.0027 (8)	-0.0038 (8)	-0.0179 (10)
C8	0.0390 (8)	0.0525 (11)	0.0323 (7)	-0.0055 (8)	0.0015 (7)	-0.0049 (8)
C9	0.0401 (9)	0.0592 (11)	0.0389 (9)	-0.0079 (9)	0.0024 (7)	0.0025 (9)
C10	0.0432 (9)	0.0621 (12)	0.0449 (10)	-0.0046 (9)	-0.0008 (9)	-0.0002 (9)
C11	0.0600 (12)	0.0453 (10)	0.0400 (9)	-0.0028 (9)	0.0003 (9)	-0.0036 (8)
C12	0.0600 (12)	0.0569 (12)	0.0516 (11)	-0.0158 (10)	0.0075 (10)	0.0059 (10)
C13	0.0410 (9)	0.0638 (13)	0.0504 (10)	-0.0121 (10)	0.0056 (9)	0.0009 (10)
C15	0.0579 (12)	0.0581 (12)	0.0582 (12)	-0.0077 (10)	0.0029 (11)	-0.0143 (10)
C14	0.0607 (13)	0.0600 (12)	0.0527 (11)	0.0116 (11)	-0.0001 (11)	-0.0026 (10)
C16	0.0356 (8)	0.0494 (9)	0.0329 (7)	0.0046 (8)	-0.0010 (7)	-0.0024 (7)
C17	0.0425 (9)	0.0604 (11)	0.0319 (8)	0.0043 (9)	-0.0063 (7)	-0.0014 (8)
C18	0.0390 (9)	0.0555 (11)	0.0383 (8)	0.0070 (8)	-0.0033 (7)	0.0074 (8)
C19	0.0324 (8)	0.0480 (9)	0.0403 (8)	0.0009 (7)	0.0050 (7)	0.0008 (7)
C20	0.0420 (9)	0.0533 (10)	0.0336 (8)	0.0022 (8)	-0.0047 (7)	-0.0021 (8)
C21	0.0432 (9)	0.0523 (10)	0.0337 (8)	0.0060 (8)	-0.0065 (7)	0.0020 (8)
C22	0.0729 (16)	0.0707 (16)	0.0723 (15)	0.0205 (14)	-0.0010 (14)	0.0133 (13)
C23	0.0649 (13)	0.0527 (12)	0.0638 (13)	-0.0002 (11)	-0.0052 (12)	-0.0134 (11)
O1	0.0684 (10)	0.0652 (10)	0.0858 (13)	-0.0185 (9)	0.0299 (10)	-0.0047 (9)
O2	0.0420 (8)	0.1026 (14)	0.0812 (11)	0.0025 (9)	-0.0132 (8)	-0.0327 (11)
O3	0.0740 (11)	0.0592 (9)	0.0589 (9)	-0.0050 (9)	-0.0057 (9)	0.0118 (8)
O4	0.0543 (8)	0.0467 (7)	0.0516 (7)	0.0036 (6)	-0.0011 (7)	0.0001 (6)

Geometric parameters (Å, °)

N1—N7	1.328 (2)	C13—H13	0.93
N1—C2	1.472 (2)	C15—H15A	0.96
N1—C6	1.479 (2)	C15—H15B	0.96
C2—C8	1.524 (3)	C15—H15C	0.96
C2—C3	1.537 (3)	C14—H14A	0.96
C2—H2	0.98	C14—H14B	0.96
C3—C4	1.519 (3)	C14—H14C	0.96
C3—C15	1.532 (3)	C16—C21	1.386 (2)
C3—C14	1.550 (3)	C16—C17	1.393 (3)
C4—O1	1.204 (2)	C17—C18	1.376 (3)
C4—C5	1.519 (3)	C17—H17	0.93
C5—C6	1.539 (3)	C18—C19	1.389 (3)
C5—H5A	0.97	C18—H18	0.93
C5—H5B	0.97	C19—O4	1.370 (2)
C6—C16	1.513 (3)	C19—C20	1.384 (3)
C6—H6	0.98	C20—C21	1.377 (3)
N7—O2	1.225 (2)	C20—H20	0.93
C8—C9	1.387 (3)	C21—H21	0.93
C8—C13	1.396 (3)	C22—O3	1.423 (3)
C9—C10	1.385 (3)	C22—H22A	0.96
C9—H9	0.93	C22—H22B	0.96
C10—C11	1.380 (3)	C22—H22C	0.96
C10—H10	0.93	C23—O4	1.421 (3)

C11—O3	1.367 (3)	C23—H23A	0.96
C11—C12	1.390 (3)	C23—H23B	0.96
C12—C13	1.372 (3)	C23—H23C	0.96
C12—H12	0.93		
N7—N1—C2	114.87 (16)	C12—C13—H13	118.9
N7—N1—C6	121.28 (17)	C8—C13—H13	118.9
C2—N1—C6	121.97 (14)	C3—C15—H15A	109.5
N1—C2—C8	109.72 (15)	C3—C15—H15B	109.5
N1—C2—C3	108.78 (15)	H15A—C15—H15B	109.5
C8—C2—C3	118.74 (16)	C3—C15—H15C	109.5
N1—C2—H2	106.3	H15A—C15—H15C	109.5
C8—C2—H2	106.3	H15B—C15—H15C	109.5
C3—C2—H2	106.3	C3—C14—H14A	109.5
C4—C3—C15	113.24 (18)	C3—C14—H14B	109.5
C4—C3—C2	109.81 (16)	H14A—C14—H14B	109.5
C15—C3—C2	111.14 (16)	C3—C14—H14C	109.5
C4—C3—C14	104.71 (16)	H14A—C14—H14C	109.5
C15—C3—C14	108.22 (18)	H14B—C14—H14C	109.5
C2—C3—C14	109.47 (17)	C21—C16—C17	117.90 (17)
O1—C4—C5	121.09 (19)	C21—C16—C6	120.07 (16)
O1—C4—C3	122.78 (19)	C17—C16—C6	122.00 (16)
C5—C4—C3	116.13 (17)	C18—C17—C16	120.84 (17)
C4—C5—C6	118.19 (17)	C18—C17—H17	119.6
C4—C5—H5A	107.8	C16—C17—H17	119.6
C6—C5—H5A	107.8	C17—C18—C19	120.29 (18)
C4—C5—H5B	107.8	C17—C18—H18	119.9
C6—C5—H5B	107.8	C19—C18—H18	119.9
H5A—C5—H5B	107.1	O4—C19—C20	124.43 (17)
N1—C6—C16	112.22 (16)	O4—C19—C18	115.98 (17)
N1—C6—C5	110.23 (15)	C20—C19—C18	119.58 (18)
C16—C6—C5	111.48 (16)	C21—C20—C19	119.48 (17)
N1—C6—H6	107.6	C21—C20—H20	120.3
C16—C6—H6	107.6	C19—C20—H20	120.3
C5—C6—H6	107.6	C20—C21—C16	121.90 (17)
O2—N7—N1	114.82 (19)	C20—C21—H21	119.1
C9—C8—C13	116.69 (19)	C16—C21—H21	119.1
C9—C8—C2	126.27 (18)	O3—C22—H22A	109.5
C13—C8—C2	117.03 (17)	O3—C22—H22B	109.5
C10—C9—C8	121.70 (19)	H22A—C22—H22B	109.5
C10—C9—H9	119.2	O3—C22—H22C	109.5
C8—C9—H9	119.2	H22A—C22—H22C	109.5
C11—C10—C9	120.41 (19)	H22B—C22—H22C	109.5
C11—C10—H10	119.8	O4—C23—H23A	109.5
C9—C10—H10	119.8	O4—C23—H23B	109.5
O3—C11—C10	125.5 (2)	H23A—C23—H23B	109.5
O3—C11—C12	115.6 (2)	O4—C23—H23C	109.5
C10—C11—C12	118.9 (2)	H23A—C23—H23C	109.5

C13—C12—C11	120.0 (2)	H23B—C23—H23C	109.5
C13—C12—H12	120.0	C11—O3—C22	118.20 (19)
C11—C12—H12	120.0	C19—O4—C23	116.96 (16)
C12—C13—C8	122.26 (19)		
N7—N1—C2—C8	110.4 (2)	C3—C2—C8—C13	164.11 (17)
C6—N1—C2—C8	−85.0 (2)	C13—C8—C9—C10	−0.6 (3)
N7—N1—C2—C3	−118.2 (2)	C2—C8—C9—C10	−179.66 (17)
C6—N1—C2—C3	46.4 (2)	C8—C9—C10—C11	−0.6 (3)
N1—C2—C3—C4	−60.6 (2)	C9—C10—C11—O3	−177.43 (19)
C8—C2—C3—C4	65.7 (2)	C9—C10—C11—C12	1.6 (3)
N1—C2—C3—C15	173.29 (16)	O3—C11—C12—C13	177.69 (18)
C8—C2—C3—C15	−60.3 (2)	C10—C11—C12—C13	−1.4 (3)
N1—C2—C3—C14	53.80 (19)	C11—C12—C13—C8	0.2 (3)
C8—C2—C3—C14	−179.83 (16)	C9—C8—C13—C12	0.8 (3)
C15—C3—C4—O1	−26.9 (3)	C2—C8—C13—C12	179.92 (19)
C2—C3—C4—O1	−151.7 (2)	N1—C6—C16—C21	122.51 (18)
C14—C3—C4—O1	90.8 (3)	C5—C6—C16—C21	−113.29 (19)
C15—C3—C4—C5	153.57 (19)	N1—C6—C16—C17	−59.4 (2)
C2—C3—C4—C5	28.7 (2)	C5—C6—C16—C17	64.8 (2)
C14—C3—C4—C5	−88.7 (2)	C21—C16—C17—C18	−0.4 (3)
O1—C4—C5—C6	−158.8 (2)	C6—C16—C17—C18	−178.54 (18)
C3—C4—C5—C6	20.8 (3)	C16—C17—C18—C19	−0.1 (3)
N7—N1—C6—C16	−69.4 (2)	C17—C18—C19—O4	−179.38 (18)
C2—N1—C6—C16	127.05 (19)	C17—C18—C19—C20	0.6 (3)
N7—N1—C6—C5	165.7 (2)	O4—C19—C20—C21	179.33 (18)
C2—N1—C6—C5	2.1 (2)	C18—C19—C20—C21	−0.7 (3)
C4—C5—C6—N1	−37.0 (2)	C19—C20—C21—C16	0.2 (3)
C4—C5—C6—C16	−162.27 (16)	C17—C16—C21—C20	0.3 (3)
C2—N1—N7—O2	170.12 (19)	C6—C16—C21—C20	178.52 (18)
C6—N1—N7—O2	5.4 (3)	C10—C11—O3—C22	3.6 (3)
N1—C2—C8—C9	109.0 (2)	C12—C11—O3—C22	−175.5 (2)
C3—C2—C8—C9	−16.9 (3)	C20—C19—O4—C23	−0.7 (3)
N1—C2—C8—C13	−70.0 (2)	C18—C19—O4—C23	179.28 (18)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A
C15—H15C···Cg1 ⁱ	0.96	2.97	3.911 (3)	167
C23—H23C···Cg1 ⁱⁱ	0.96	2.86	3.720 (3)	149

Symmetry codes: (i) $-x, y+1/2, -z+1/2$; (ii) $x-1/2, -y+1/2, -z$.