

## Poly[pentaqua<sup>tetrakis(μ<sub>2</sub>-nicotinato-κ<sup>2</sup>N:O)(perchlorato-κO)lanthanum(III)-disilver(I)]</sup>

Biao Guan,<sup>a</sup> Chao-Hua Zhang<sup>b</sup> and Wen-Dong Song<sup>c\*</sup>

<sup>a</sup>Laboratory and Facility Management Division, Guang Dong Ocean University, Zhanjiang 524088, People's Republic of China, <sup>b</sup>School of Food Science and Technology, Guang Dong Ocean University, Zhanjiang 524088, People's Republic of China, and <sup>c</sup>College of Science, Guang Dong Ocean University, Zhanjiang 524088, People's Republic of China  
Correspondence e-mail: songwd60@126.com

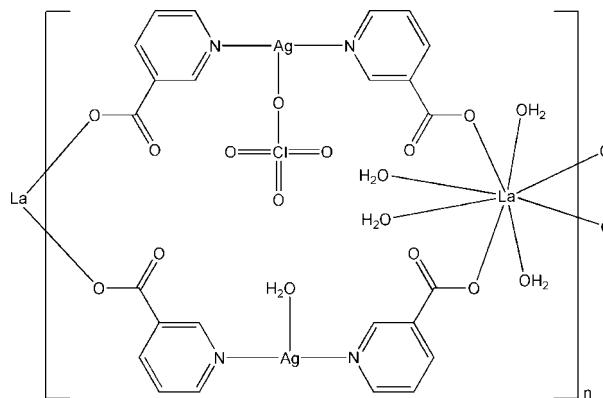
Received 13 May 2009; accepted 5 July 2009

Key indicators: single-crystal X-ray study;  $T = 298\text{ K}$ ; mean  $\sigma(\text{C}-\text{C}) = 0.011\text{ \AA}$ ; disorder in main residue;  $R$  factor = 0.049;  $wR$  factor = 0.090; data-to-parameter ratio = 14.1.

In the title complex,  $[\text{Ag}_2\text{La}(\text{C}_6\text{H}_4\text{NO}_2)_4(\text{ClO}_4)(\text{H}_2\text{O})_5]_n$ , the  $\text{La}^{\text{III}}$  atom, lying on a twofold rotation axis, is eight-coordinated by four O atoms from four nicotinate (nic) ligands and four water molecules in a distorted square-antiprismatic coordination geometry. The  $\text{Ag}^{\text{I}}$  atom is coordinated in an almost linear fashion by two pyridyl N atoms of two nic ligands. The linear coordination is augmented by weak interactions with one O atom from a half-occupied  $\text{ClO}_4^-$  anion and a water molecule lying on a twofold axis. Two  $\text{Ag}(\text{nic})_2$  units connect two La atoms, forming a cyclic unit. These units are further extended into an infinite zigzag chain. The chains are bridged by the disordered perchlorate ions via weak  $\text{Ag}-\text{O}$  [2.678 (2) Å] interactions.  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds, weak  $\text{Ag}\cdots\text{Ag}$  [3.3340 (15) Å] interactions and  $\pi-\pi$  interactions between the pyridyl rings [centroid–centroid distance = 3.656 (2) Å] lead to a three-dimensional network.

### Related literature

For related structures see: Evans & Lin (2001); Luo *et al.* (2004).



### Experimental

#### Crystal data

$[\text{Ag}_2\text{La}(\text{C}_6\text{H}_4\text{NO}_2)_4(\text{ClO}_4)(\text{H}_2\text{O})_5]$	$V = 6522.8 (15)\text{ \AA}^3$
$M_r = 1032.59$	$Z = 8$
Orthorhombic, $Cmca$	Mo $K\alpha$ radiation
$a = 35.140 (5)\text{ \AA}$	$\mu = 2.64\text{ mm}^{-1}$
$b = 12.3371 (16)\text{ \AA}$	$T = 298\text{ K}$
$c = 15.046 (2)\text{ \AA}$	$0.30 \times 0.25 \times 0.22\text{ mm}$

#### Data collection

Bruker APEXII CCD diffractometer	15911 measured reflections
Absorption correction: multi-scan ( <i>SADABS</i> ; Sheldrick, 1996)	2999 independent reflections
$T_{\min} = 0.465$ , $T_{\max} = 0.567$	2251 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.067$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$	48 restraints
$wR(F^2) = 0.090$	H-atom parameters constrained
$S = 1.95$	$\Delta\rho_{\text{max}} = 1.90\text{ e \AA}^{-3}$
2999 reflections	$\Delta\rho_{\text{min}} = -0.97\text{ e \AA}^{-3}$
212 parameters	

**Table 1**  
Selected bond lengths (Å).

La1—O1	2.511 (5)	Ag1—N2	2.161 (6)
La1—O3 <sup>i</sup>	2.401 (4)	Ag1—O6	2.681 (2)
La1—O1W	2.498 (5)	Ag1—O3W	2.877 (6)
La1—O2W	2.494 (4)	Ag1—Ag1 <sup>ii</sup>	3.3352 (14)
Ag1—N1	2.175 (6)		

Symmetry codes: (i)  $x, -y + \frac{3}{2}, z - \frac{1}{2}$ ; (ii)  $x, -y + 1, -z + 1$ .

**Table 2**  
Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
O1W—H1W <sup>iii</sup> ···O2 <sup>iii</sup>	0.86	1.85	2.667 (6)	159
O1W—H2W <sup>iv</sup> ···O4 <sup>ii</sup>	0.84	1.80	2.611 (7)	161
O2W—H3W <sup>v</sup> ···O2 <sup>iv</sup>	0.84	1.92	2.738 (7)	165
O2W—H4W <sup>v</sup> ···O2 <sup>y</sup>	0.84	1.89	2.693 (7)	161
O3W—H5W <sup>v</sup> ···O1W <sup>i</sup>	0.82	2.11	2.883 (5)	157

Symmetry codes: (ii)  $x, -y + 1, -z + 1$ ; (iii)  $-x + \frac{1}{2}, y, -z + \frac{1}{2}$ ; (iv)  $-x + \frac{1}{2}, y + \frac{1}{2}, z$ ; (v)  $x, -y + 2, -z + 1$ ; (vi)  $x, -y + \frac{3}{2}, z + \frac{1}{2}$ .

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge Guang Dong Ocean University for support of this work.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HY2200).

## References

- Brandenburg, K. (1999). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.  
Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
Evans, O. R. & Lin, W. B. (2001). *Chem. Mater.* **13**, 3009–3017.  
Luo, J. H., Jiang, F. L., Wang, R. H., Han, L., Lin, Z. Z., Cao, R. & Hong, M. C. (2004). *J. Mol. Struct.* **707**, 211–216.  
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.  
Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.

# supporting information

*Acta Cryst.* (2009). E65, m896–m897 [doi:10.1107/S1600536809026130]

## Poly[pentaaquatetrakis( $\mu_2$ -nicotinato- $\kappa^2N:O$ )(perchlorato- $\kappa O$ )lanthanum(III)disilver(I)]

Biao Guan, Chao-Hua Zhang and Wen-Dong Song

### S1. Comment

In the structural investigation of nictinate complexes, it has been found that nictinate functions as a multidentate ligand with versatile binding and coordination modes (Evans & Lin, 2001; Luo *et al.*, 2004). In this paper, we report the crystal structure of the title compound, a new La<sup>III</sup> complex, resulted from the hydrothermal treatment of La<sub>2</sub>O<sub>3</sub>, AgNO<sub>3</sub>, perchloric acid and nicotinic acid in water.

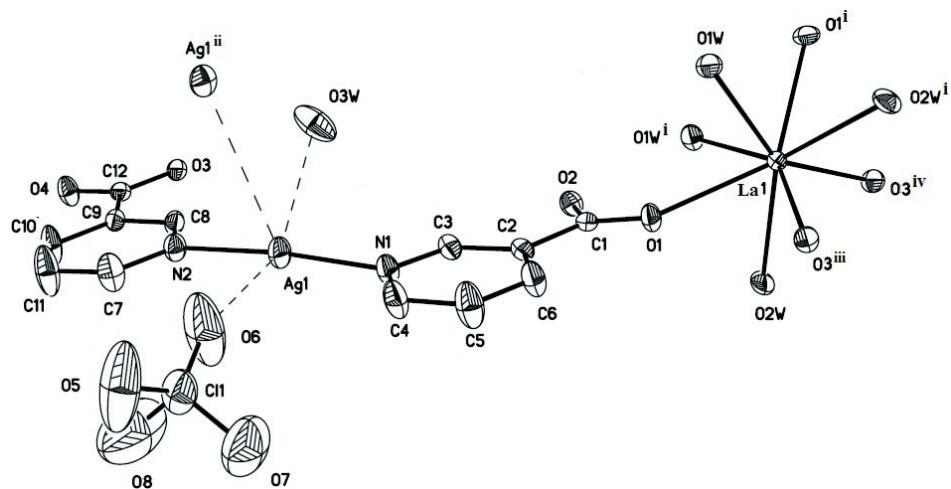
As depicted in Fig. 1, the La<sup>III</sup> atom, lying on a twofold rotation axis, is surrounded by four O atoms from four nic ligands and four water molecules in a distorted square-antiprismatic coordination geometry. The Ag<sup>I</sup> atom is coordinated in an almost linear fashion by two pyridyl N atoms of two nic ligands. The linear coordination is augmented by weak interactions with one O atom from a half-occupied ClO<sub>4</sub><sup>-</sup> anion and a water molecule lying on a twofold rotation axis. The two pyridyl rings of the nic ligands coordinating to the Ag atom are almost coplanar and have a dihedral angle of 1.74 (2) $^\circ$ . Two Ag(nic)<sub>2</sub> units connect two La atoms, forming a cyclic unit. These cycles are further extended into an infinite zigzag chain. The chains are bridged by disordered perchlorate ions *via* the weak Ag—O [2.678 (2) Å] interactions into a two-dimensional wavelike layer in the *b* axis direction (Fig. 2). Finally, the layers are further self-assembled into a three-dimensional supramolecular network (Fig. 3) *via* O—H···O hydrogen bonds involving the coordinated water molecules and carboxylate O atoms from the nic ligands (Table 1), weak Ag···Ag [3.3340 (15) Å] interactions and  $\pi$ — $\pi$  stacking interactions between the pyridyl rings [centroid–centroid distance = 3.656 (2) Å].

### S2. Experimental

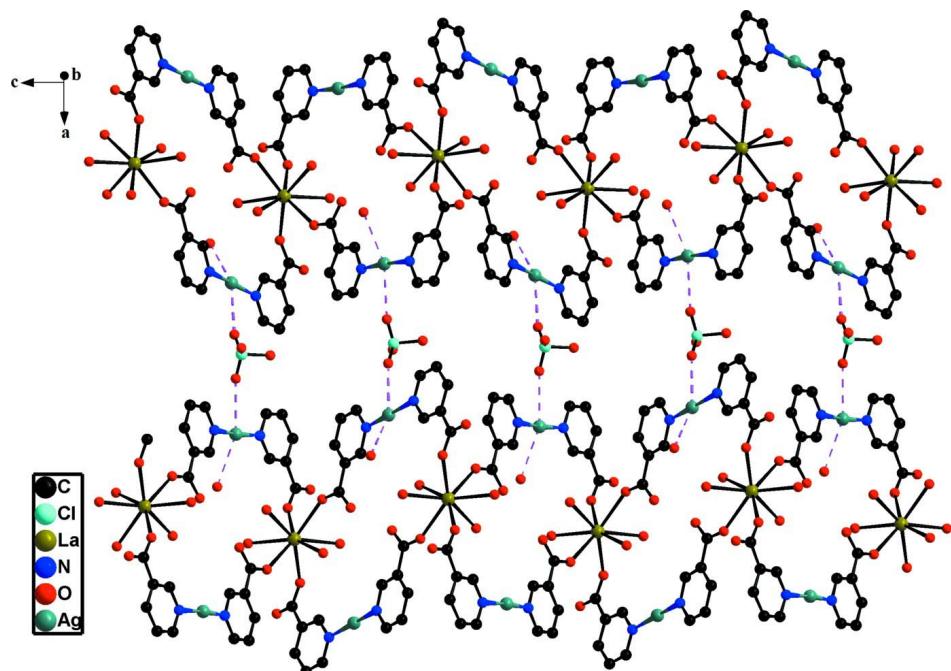
A mixture of La<sub>2</sub>O<sub>3</sub> (0.162 g, 0.5 mmol), AgNO<sub>3</sub> (0.169 g, 1 mmol), nicotinic acid (0.123 g, 1 mmol), HClO<sub>4</sub> (0.12 ml) and H<sub>2</sub>O (10 ml) was placed in a 23 ml Teflon-lined reactor, which was heated to 433 K for 3 d, and then cooled to room temperature at a rate of 10 K h<sup>-1</sup>. The pale-purple crystals obtained were washed with water and dried in air (yield 46% based on La).

### S3. Refinement

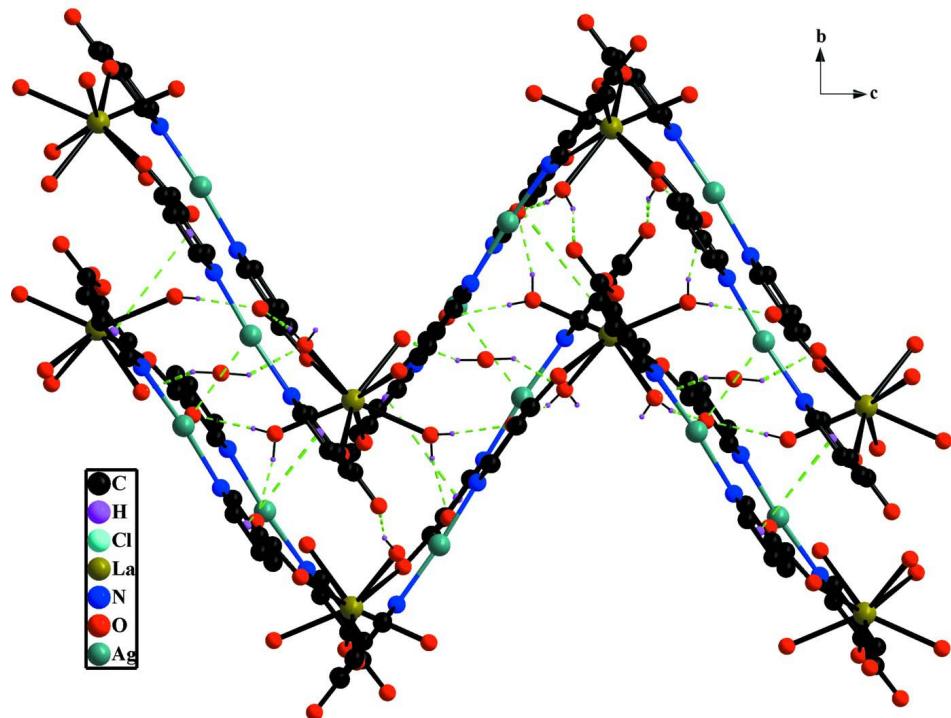
H atoms on C atoms were positioned geometrically and treated as riding on the parent C atoms, with C—H = 0.93 Å, and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . H atoms of water molecules were located in difference Fourier maps and refined as riding atoms, with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ . The perchlorate anion is disordered with an occupancy factor of 0.5. The highest peak in final difference map is located 1.00 Å from La1 and the deepest hole is located 0.94 Å from La1.

**Figure 1**

The asymmetric unit of the title compound. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry codes: (i)  $1/2 - x, y, 1/2 - z$ ; (ii)  $x, 1 - y, 1 - z$ ; (iii)  $x, 3/2 - y, -1/2 + z$ ; (iv)  $1/2 - x, 3/2 - y, 1 - z$ .]

**Figure 2**

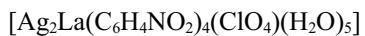
View of the two-dimensional wavelike layer of the title compound. Dashed lines denote weak  $\text{Ag}\cdots\text{O}$  interactions.

**Figure 3**

View of the three-dimensional network *via* hydrogen bonds, weak Ag···O, Ag···Ag, and  $\pi$ - $\pi$  interactions (dashed lines).

### Poly[pentaqua<sup>4</sup>tetrakis( $\mu_2$ -nicotinato- $\kappa^2$ N:O)(perchlorato- $\kappa$ O)lanthanum(III)disilver(I)]

#### Crystal data



$M_r = 1032.59$

Orthorhombic,  $Cmca$

Hall symbol: -C 2bc 2

$a = 35.140 (5)$  Å

$b = 12.3371 (16)$  Å

$c = 15.046 (2)$  Å

$V = 6522.8 (15)$  Å<sup>3</sup>

$Z = 8$

$F(000) = 4016$

$D_x = 2.103 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 3600 reflections

$\theta = 1.4\text{--}28^\circ$

$\mu = 2.64 \text{ mm}^{-1}$

$T = 298$  K

Block, colorless

$0.30 \times 0.25 \times 0.22$  mm

#### Data collection

Bruker APEXII CCD  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan  
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.465$ ,  $T_{\max} = 0.567$

15911 measured reflections

2999 independent reflections

2251 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.067$

$\theta_{\max} = 25.2^\circ$ ,  $\theta_{\min} = 2.2^\circ$

$h = -42 \rightarrow 41$

$k = -14 \rightarrow 11$

$l = -18 \rightarrow 15$

*Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.049$  $wR(F^2) = 0.090$  $S = 1.95$ 

2999 reflections

212 parameters

48 restraints

Primary atom site location: structure-invariant  
direct methodsSecondary atom site location: difference Fourier  
mapHydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2)]$   
where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\text{max}} = 0.008$  $\Delta\rho_{\text{max}} = 1.90 \text{ e \AA}^{-3}$  $\Delta\rho_{\text{min}} = -0.97 \text{ e \AA}^{-3}$ *Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
La1	0.2500	1.08273 (5)	0.2500	0.02358 (16)	
Ag1	0.389767 (18)	0.60965 (6)	0.56481 (4)	0.0465 (2)	
C1	0.3039 (2)	0.9196 (6)	0.3991 (4)	0.0299 (18)	
C2	0.34312 (19)	0.8866 (6)	0.4214 (5)	0.0318 (18)	
C3	0.3495 (2)	0.7964 (6)	0.4729 (4)	0.0331 (19)	
H3	0.3287	0.7591	0.4956	0.040*	
C4	0.4138 (2)	0.8140 (8)	0.4627 (5)	0.053 (3)	
H4	0.4380	0.7896	0.4770	0.064*	
C5	0.4097 (2)	0.9060 (8)	0.4114 (7)	0.071 (3)	
H5	0.4311	0.9430	0.3913	0.085*	
C6	0.3744 (2)	0.9422 (7)	0.3905 (5)	0.048 (2)	
H6	0.3714	1.0039	0.3557	0.057*	
C7	0.4322 (2)	0.4349 (7)	0.6693 (6)	0.054 (3)	
H7	0.4534	0.4729	0.6489	0.065*	
C8	0.3685 (2)	0.4135 (6)	0.6768 (5)	0.0313 (18)	
H8	0.3444	0.4376	0.6605	0.038*	
C9	0.37074 (19)	0.3248 (6)	0.7314 (4)	0.0288 (18)	
C10	0.4064 (2)	0.2925 (7)	0.7546 (6)	0.060 (3)	
H10	0.4098	0.2331	0.7919	0.072*	
C11	0.4374 (2)	0.3480 (8)	0.7226 (7)	0.078 (4)	
H11	0.4619	0.3257	0.7375	0.094*	
C12	0.3359 (2)	0.2677 (6)	0.7620 (5)	0.0292 (17)	
N1	0.38410 (17)	0.7606 (5)	0.4913 (4)	0.0369 (16)	
N2	0.39810 (17)	0.4670 (5)	0.6458 (4)	0.0375 (16)	
O1	0.30034 (13)	0.9960 (4)	0.3448 (3)	0.0366 (13)	
O2	0.27688 (13)	0.8707 (4)	0.4351 (3)	0.0331 (13)	
O3	0.30462 (12)	0.3020 (4)	0.7338 (3)	0.0328 (12)	
O4	0.33989 (13)	0.1890 (5)	0.8112 (3)	0.0432 (15)	
O1W	0.28251 (13)	0.9406 (4)	0.1593 (3)	0.0373 (14)	
H1W	0.2679	0.9115	0.1204	0.056*	
H2W	0.2983	0.8959	0.1795	0.056*	
O2W	0.24825 (13)	1.1647 (4)	0.4015 (3)	0.0478 (14)	
H3W	0.2385	1.2269	0.4029	0.072*	
H4W	0.2620	1.1535	0.4461	0.072*	

O3W	0.32305 (14)	0.5000	0.5000	0.074 (3)	
H5W	0.3086	0.5297	0.5356	0.111*	
C11	0.5030 (4)	0.6862 (4)	0.5923 (4)	0.0816 (15)	0.50
O5	0.5071 (4)	0.8007 (6)	0.5862 (9)	0.118 (3)	0.50
O6	0.4640 (3)	0.6544 (14)	0.5835 (10)	0.118 (3)	0.50
O7	0.5133 (4)	0.6545 (12)	0.6849 (7)	0.118 (3)	0.50
O8	0.5275 (4)	0.6292 (12)	0.5351 (9)	0.118 (3)	0.50

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
La1	0.0267 (3)	0.0230 (3)	0.0210 (3)	0.000	-0.0014 (3)	0.000
Ag1	0.0542 (4)	0.0386 (4)	0.0466 (4)	0.0050 (4)	-0.0029 (3)	0.0150 (3)
C1	0.043 (4)	0.027 (5)	0.020 (4)	0.006 (4)	-0.007 (4)	-0.007 (3)
C2	0.034 (4)	0.032 (5)	0.028 (4)	-0.004 (4)	-0.004 (4)	-0.004 (4)
C3	0.032 (4)	0.036 (5)	0.030 (4)	0.005 (4)	0.000 (4)	0.010 (4)
C4	0.038 (5)	0.061 (8)	0.060 (6)	-0.003 (5)	-0.004 (4)	0.024 (5)
C5	0.033 (5)	0.075 (9)	0.105 (9)	-0.004 (5)	-0.001 (5)	0.049 (7)
C6	0.044 (5)	0.052 (7)	0.047 (5)	-0.006 (4)	-0.011 (4)	0.029 (4)
C7	0.033 (5)	0.051 (7)	0.078 (7)	-0.004 (4)	-0.002 (5)	0.026 (5)
C8	0.029 (4)	0.037 (5)	0.028 (4)	0.002 (4)	-0.001 (3)	-0.004 (4)
C9	0.033 (4)	0.028 (5)	0.025 (5)	0.003 (4)	-0.006 (3)	-0.001 (4)
C10	0.039 (5)	0.055 (6)	0.087 (7)	-0.002 (4)	-0.007 (5)	0.040 (6)
C11	0.027 (5)	0.069 (8)	0.139 (10)	-0.003 (5)	-0.007 (5)	0.059 (7)
C12	0.035 (4)	0.025 (5)	0.027 (5)	-0.003 (3)	0.004 (4)	-0.010 (4)
N1	0.037 (4)	0.037 (5)	0.037 (4)	0.002 (4)	-0.004 (3)	0.011 (3)
N2	0.040 (4)	0.033 (4)	0.040 (4)	0.004 (3)	0.004 (3)	0.008 (3)
O1	0.041 (3)	0.031 (4)	0.038 (3)	0.006 (2)	-0.005 (3)	0.013 (3)
O2	0.037 (3)	0.036 (4)	0.027 (3)	-0.002 (2)	0.001 (2)	0.005 (2)
O3	0.027 (3)	0.033 (3)	0.038 (3)	0.007 (2)	-0.004 (2)	-0.002 (2)
O4	0.037 (3)	0.038 (4)	0.054 (4)	-0.004 (3)	-0.007 (3)	0.021 (3)
O1W	0.039 (3)	0.034 (4)	0.039 (3)	0.007 (2)	-0.010 (2)	-0.009 (2)
O2W	0.073 (3)	0.046 (4)	0.025 (3)	0.026 (3)	-0.017 (3)	-0.009 (2)
O3W	0.050 (5)	0.123 (10)	0.048 (6)	0.000	0.000	-0.011 (5)
C11	0.044 (3)	0.069 (3)	0.132 (4)	0.015 (5)	0.044 (5)	0.014 (3)
O5	0.075 (5)	0.108 (7)	0.172 (8)	0.003 (5)	0.005 (5)	0.035 (6)
O6	0.075 (5)	0.108 (7)	0.172 (8)	0.003 (5)	0.005 (5)	0.035 (6)
O7	0.075 (5)	0.108 (7)	0.172 (8)	0.003 (5)	0.005 (5)	0.035 (6)
O8	0.075 (5)	0.108 (7)	0.172 (8)	0.003 (5)	0.005 (5)	0.035 (6)

*Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )*

La1—O1	2.511 (5)	C7—C11	1.352 (11)
La1—O3 <sup>i</sup>	2.401 (4)	C7—H7	0.9300
La1—O1W	2.498 (5)	C8—N2	1.317 (8)
La1—O2W	2.494 (4)	C8—C9	1.371 (10)
Ag1—N1	2.175 (6)	C8—H8	0.9300
Ag1—N2	2.161 (6)	C9—C10	1.360 (9)

Ag1—O6	2.681 (2)	C9—C12	1.484 (10)
Ag1—O3W	2.877 (6)	C10—C11	1.375 (11)
Ag1—Ag1 <sup>ii</sup>	3.3352 (14)	C10—H10	0.9300
C1—O1	1.254 (8)	C11—H11	0.9300
C1—O2	1.249 (8)	C12—O4	1.230 (8)
C1—C2	1.475 (9)	C12—O3	1.253 (8)
C2—C6	1.378 (10)	O3—La1 <sup>iii</sup>	2.401 (4)
C2—C3	1.373 (10)	O1W—H1W	0.8564
C3—N1	1.324 (8)	O1W—H2W	0.8388
C3—H3	0.9300	O2W—H3W	0.8404
C4—N1	1.306 (9)	O2W—H4W	0.8395
C4—C5	1.379 (11)	O3W—H5W	0.8241
C4—H4	0.9300	C11—O8	1.4076
C5—C6	1.354 (10)	C11—O5	1.4226
C5—H5	0.9300	C11—O6	1.4309
C6—H6	0.9300	C11—O7	1.4925
C7—N2	1.312 (9)		
O3 <sup>iii</sup> —La1—O3 <sup>i</sup>	107.4 (2)	N1—C4—H4	119.5
O3 <sup>iii</sup> —La1—O2W <sup>iv</sup>	82.65 (16)	C5—C4—H4	119.5
O3 <sup>i</sup> —La1—O2W <sup>iv</sup>	69.35 (15)	C4—C5—C6	119.8 (8)
O3 <sup>iii</sup> —La1—O2W	69.35 (15)	C4—C5—H5	120.1
O3 <sup>i</sup> —La1—O2W	82.65 (16)	C6—C5—H5	120.1
O2W <sup>iv</sup> —La1—O2W	132.1 (2)	C2—C6—C5	119.2 (8)
O3 <sup>iii</sup> —La1—O1W	146.78 (15)	C2—C6—H6	120.4
O3 <sup>i</sup> —La1—O1W	89.71 (15)	C5—C6—H6	120.4
O2W <sup>iv</sup> —La1—O1W	76.96 (16)	N2—C7—C11	121.5 (8)
O2W—La1—O1W	142.66 (15)	N2—C7—H7	119.2
O3 <sup>iii</sup> —La1—O1W <sup>iv</sup>	89.71 (15)	C11—C7—H7	119.2
O3 <sup>i</sup> —La1—O1W <sup>iv</sup>	146.78 (15)	N2—C8—C9	124.5 (7)
O2W <sup>iv</sup> —La1—O1W <sup>iv</sup>	142.66 (15)	N2—C8—H8	117.8
O2W—La1—O1W <sup>iv</sup>	76.96 (16)	C9—C8—H8	117.8
O1W—La1—O1W <sup>iv</sup>	90.8 (2)	C10—C9—C8	116.2 (7)
O3 <sup>iii</sup> —La1—O1 <sup>iv</sup>	75.34 (16)	C10—C9—C12	122.7 (7)
O3 <sup>i</sup> —La1—O1 <sup>iv</sup>	139.26 (15)	C8—C9—C12	121.1 (6)
O2W <sup>iv</sup> —La1—O1 <sup>iv</sup>	70.81 (16)	C9—C10—C11	119.7 (8)
O2W—La1—O1 <sup>iv</sup>	132.40 (16)	C9—C10—H10	120.2
O1W—La1—O1 <sup>iv</sup>	73.32 (16)	C11—C10—H10	120.2
O1W <sup>iv</sup> —La1—O1 <sup>iv</sup>	71.89 (16)	C7—C11—C10	119.7 (8)
O3 <sup>iii</sup> —La1—O1	139.26 (15)	C7—C11—H11	120.1
O3 <sup>i</sup> —La1—O1	75.34 (16)	C10—C11—H11	120.1
O2W <sup>iv</sup> —La1—O1	132.40 (16)	O4—C12—O3	124.7 (7)
O2W—La1—O1	70.81 (16)	O4—C12—C9	117.9 (7)
O1W—La1—O1	71.89 (16)	O3—C12—C9	117.4 (7)
O1W <sup>iv</sup> —La1—O1	73.32 (16)	C4—N1—C3	119.7 (7)
O1 <sup>iv</sup> —La1—O1	129.6 (2)	C4—N1—Ag1	121.8 (5)
N2—Ag1—N1	175.3 (2)	C3—N1—Ag1	118.5 (5)
N2—Ag1—Ag1 <sup>ii</sup>	70.66 (16)	C7—N2—C8	118.4 (7)

N1—Ag1—Ag1 <sup>ii</sup>	113.42 (17)	C7—N2—Ag1	121.4 (5)
O6—Ag1—N2	88.67 (6)	C8—N2—Ag1	120.0 (5)
O6—Ag1—N1	88.06 (7)	C1—O1—La1	139.5 (5)
O3W—Ag1—N1	98.95 (17)	C12—O3—La1 <sup>iii</sup>	150.2 (5)
O3W—Ag1—N2	85.33 (16)	La1—O1W—H1W	113.1
O3W—Ag1—O6	157.70 (7)	La1—O1W—H2W	124.4
O1—C1—O2	124.7 (7)	H1W—O1W—H2W	111.6
O1—C1—C2	116.7 (7)	La1—O2W—H3W	113.8
O2—C1—C2	118.6 (7)	La1—O2W—H4W	130.5
C6—C2—C3	117.6 (7)	H3W—O2W—H4W	111.4
C6—C2—C1	122.1 (7)	O8—Cl1—O5	113.3
C3—C2—C1	120.3 (7)	O8—Cl1—O6	113.1
N1—C3—C2	122.5 (7)	O5—Cl1—O6	111.3
N1—C3—H3	118.7	O8—Cl1—O7	106.8
C2—C3—H3	118.7	O5—Cl1—O7	107.2
N1—C4—C5	121.1 (8)	O6—Cl1—O7	104.5

Symmetry codes: (i)  $x, -y+3/2, z-1/2$ ; (ii)  $x, -y+1, -z+1$ ; (iii)  $-x+1/2, -y+3/2, -z+1$ ; (iv)  $-x+1/2, y, -z+1/2$ .

#### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D\cdots H$	$D—H$	$H\cdots A$	$D\cdots A$	$D—H\cdots A$
O1W—H1W <sup>iv</sup> ···O2 <sup>iv</sup>	0.86	1.85	2.667 (6)	159
O1W—H2W <sup>vii</sup> ···O4 <sup>ii</sup>	0.84	1.80	2.611 (7)	161
O2W—H3W <sup>vii</sup> ···O2 <sup>v</sup>	0.84	1.92	2.738 (7)	165
O2W—H4W <sup>vii</sup> ···O2 <sup>vi</sup>	0.84	1.89	2.693 (7)	161
O3W—H5W <sup>vii</sup> ···O1W <sup>vii</sup>	0.82	2.11	2.883 (5)	157

Symmetry codes: (ii)  $x, -y+1, -z+1$ ; (iv)  $-x+1/2, y, -z+1/2$ ; (v)  $-x+1/2, y+1/2, z$ ; (vi)  $x, -y+2, -z+1$ ; (vii)  $x, -y+3/2, z+1/2$ .