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# Sodium 2-nitrocinnamate dihydrate: a one-dimensional hydrogen-bonded coordination polymer

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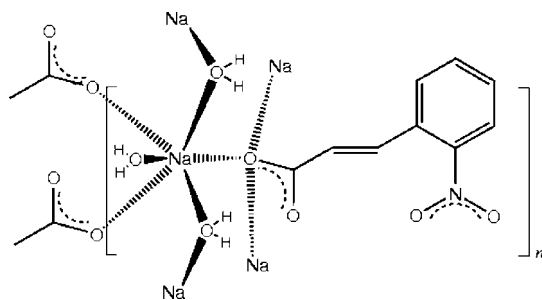
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Key indicators: single-crystal X-ray study;  $T = 297$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.038;  $wR$  factor = 0.113; data-to-parameter ratio = 12.4.

The title compound *catena*-poly[aquasodium- $\mu_2$ -aqua- $\mu_3$ -2-nitrocinnamato],  $[\text{Na}(\text{C}_9\text{H}_6\text{NO}_4)(\text{H}_2\text{O})_2]_n$ , the sodium salt of *trans*-2-nitrocinnamic acid, is a one-dimensional coordination polymer based on six-coordinate octahedral  $\text{NaO}_6$  centres, comprising three facially related monodentate carboxylate O-atom donors from separate ligands (all bridging)  $[\text{Na}-\text{O} = 2.4370(13)-2.5046(13)$  Å], and three water molecules (two bridging and one monodentate)  $[\text{Na}-\text{O} = 2.3782(13)-2.4404(17)$  Å]. The structure is also stabilized by intra-chain water-carboxylate and water-nitro  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds.

## Related literature

For literature on similar compounds, see: Crowther *et al.* (2008); Kariuki *et al.* (1995); Kula *et al.* (2007); Schmidt (1964); Smith *et al.* (2006); Trividi *et al.* (2005).



## Experimental

### Crystal data

$[\text{Na}(\text{C}_9\text{H}_6\text{NO}_4)(\text{H}_2\text{O})_2]$   
 $M_r = 251.17$

Monoclinic,  $P2_1/c$   
 $a = 19.4179(7)$  Å

$b = 3.6899(2)$  Å  
 $c = 14.8738(7)$  Å  
 $\beta = 92.239(4)^\circ$   
 $V = 1064.90(9)$  Å<sup>3</sup>  
 $Z = 4$

Mo  $K\alpha$  radiation  
 $\mu = 0.17$  mm<sup>-1</sup>  
 $T = 297$  K  
 $0.40 \times 0.30 \times 0.13$  mm

### Data collection

Oxford Diffraction Gemini-S CCD-detector diffractometer  
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.93$ ,  $T_{\max} = 0.98$

6531 measured reflections  
2100 independent reflections  
1626 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.019$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$   
 $wR(F^2) = 0.113$   
 $S = 1.09$   
2100 reflections  
170 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\max} = 0.30$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.19$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1W—H11W $\cdots$ O32 <sup>i</sup>	0.78 (3)	2.14 (3)	2.8871 (17)	162 (2)
O1W—H12W $\cdots$ O32 <sup>ii</sup>	0.89 (2)	1.90 (2)	2.7852 (17)	171 (2)
O2W—H21W $\cdots$ O21 <sup>iii</sup>	0.77 (3)	2.49 (3)	3.050 (2)	131 (3)
O2W—H22W $\cdots$ O32 <sup>i</sup>	0.91 (4)	2.04 (5)	2.882 (2)	153 (4)

Symmetry codes: (i)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (ii)  $-x, -y, -z$ ; (iii)  $x, -y + \frac{3}{2}, z - \frac{1}{2}$ .

Data collection: *CrysAlis Pro* (Oxford Diffraction, 2009); cell refinement: *CrysAlis Pro*; data reduction: *CrysAlis Pro*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2131).

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**supplementary materials**

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## Sodium 2-nitrocinnamate dihydrate: a one-dimensional hydrogen-bonded coordination polymer

G. Smith and U. D. Wermuth

### Comment

Although the structures of two polymorphs of *trans*-cinnamic acid have been determined (Schmidt, 1964; Smith *et al.*, 2006), the structures of neither *trans*-2-nitrocinnamic acid [(*E*)-3-(2-nitrophenyl)propenoic acid] nor any of its alkali metal salts are known, although the dicyclohexylammonium salt has been reported (Trividi *et al.*, 2005). The only structures of alkali metal compounds of analogous ring-substituted *trans*-cinnamic acids are the sodium complexes with 2-chlorocinnamic acid (Kariuki *et al.*, 1995), 3-chlorocinnamic acid (Crowther *et al.*, 2008), and 4-hydroxy-2-methoxycinnamic acid (Kula *et al.*, 2007). We have now prepared the sodium salt of *trans*-2-nitrocinnamic acid, a dihydrate [Na(C<sub>9</sub>H<sub>6</sub>NO<sub>4</sub>)(H<sub>2</sub>O)<sub>2</sub>]<sub>n</sub> and its structure is reported here.

The molecular structure of the title compound is illustrated in Fig. 1. The polymeric structure is based on octahedral six-coordinate NaO<sub>6</sub> centres comprising three facially related monodentate carboxylate O-donors from separate ligands (all bridging) [Na–O, 2.4370 (13)–2.5046 (13) Å] and three water molecules (two bridging, one monodentate) [Na–O, 2.3782 (13)–2.4404 (17) Å]. These units are linked into one-dimensional coordination polymer chains which extend along direction [010] (Fig. 1). The structure is similar to that of the sodium 2-chlorocinnamate complex (Kariuki *et al.*, 1995). The polymer chains are stabilized by intra-chain water *O*–*H*···*O*<sub>carboxylate</sub> and *O*–*H*···*O*<sub>nitro</sub> hydrogen bonds (Table 1).

In the substituted cinnamate ligand molecule, the nitro group is rotated out of the plane of the benzene ring [torsion angle C1–C2–N21–O22, 144.65 (17)°], while the carboxylate group is similarly non-coplanar [C11–C21–C31–O31, -169.51 (17)°].

### Experimental

The title compound was synthesized by heating together for 10 minutes under reflux 1 mmol quantities of *trans*-cinnamic acid [(*E*)-3-(2-nitrophenyl)propenoic acid] and sodium carbonate in 50 ml of 50% ethanol-water. After concentration to *ca* 30 ml, partial *rt* evaporation of the hot-filtered solution gave thin colourless plate-like crystals, suitable for X-ray analysis.

### Refinement

The H-atoms of the water molecules were located in difference electron-density maps and were freely refined: O–H = 0.77 (3) – 0.91 (4) Å. The C-bound H-atoms were included in calculated positions and treated as riding atoms: C–H = 0.93 Å with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

## Figures

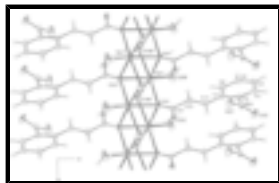


Fig. 1. Molecular configuration and atom naming scheme for the title compound, showing the one-dimensional chain polymer structure extending along direction [010]. Displacement ellipsoids are drawn at the 50% probability level [Symmetry codes: (i)  $x, y + 1, z$ ; (ii)  $-x, -y + 1, -z$ ].

## catena-poly[aquasodium- $\mu_2$ -aqua- $\mu_3$ -2-nitrocinnamato]

### Crystal data

[Na(C<sub>9</sub>H<sub>6</sub>NO<sub>4</sub>)(H<sub>2</sub>O)<sub>2</sub>]

$M_r = 251.17$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 19.4179$  (7) Å

$b = 3.6899$  (2) Å

$c = 14.8738$  (7) Å

$\beta = 92.239$  (4)°

$V = 1064.90$  (9) Å<sup>3</sup>

$Z = 4$

$F_{000} = 520$

$D_x = 1.567$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 2943 reflections

$\theta = 3.0$ – $28.7^\circ$

$\mu = 0.17$  mm<sup>-1</sup>

$T = 297$  K

Plate, colourless

$0.40 \times 0.30 \times 0.13$  mm

### Data collection

Oxford Diffraction Gemini-S CCD-detector diffractometer

Radiation source: Enhance (Mo) X-ray source

Monochromator: graphite

$T = 297$  K

$\omega$  scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

$T_{\min} = 0.93$ ,  $T_{\max} = 0.98$

6531 measured reflections

2100 independent reflections

1626 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

$\theta_{\text{max}} = 26.0^\circ$

$\theta_{\text{min}} = 3.0^\circ$

$h = -23 \rightarrow 21$

$k = -4 \rightarrow 4$

$l = -18 \rightarrow 17$

### Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.038$

$wR(F^2) = 0.113$

$S = 1.09$

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0708P)^2]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

2100 reflections  $\Delta\rho_{\max} = 0.30 \text{ e } \text{\AA}^{-3}$   
 170 parameters  $\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$   
 Primary atom site location: structure-invariant direct methods Extinction correction: none

*Special details*

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ - factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Na1	0.05494 (3)	0.73072 (17)	-0.06704 (4)	0.0272 (2)
O1W	0.02635 (6)	0.2374 (3)	-0.16440 (8)	0.0319 (4)
O2W	0.16105 (8)	0.6812 (5)	-0.14886 (12)	0.0691 (7)
O21	0.29871 (7)	0.4625 (6)	0.31366 (9)	0.0637 (6)
O22	0.39979 (7)	0.2280 (5)	0.32066 (10)	0.0541 (6)
O31	0.06940 (6)	0.2430 (3)	0.04255 (8)	0.0282 (4)
O32	0.10417 (6)	0.0743 (4)	0.18086 (8)	0.0336 (4)
N21	0.35246 (7)	0.3769 (4)	0.27927 (10)	0.0339 (5)
C1	0.30837 (8)	0.4562 (5)	0.12070 (11)	0.0267 (5)
C2	0.36252 (8)	0.4742 (4)	0.18535 (11)	0.0261 (5)
C3	0.42813 (9)	0.5880 (5)	0.16617 (12)	0.0327 (6)
C4	0.44154 (10)	0.6987 (5)	0.08063 (14)	0.0384 (6)
C5	0.38972 (10)	0.6860 (5)	0.01494 (13)	0.0368 (6)
C6	0.32514 (9)	0.5634 (5)	0.03431 (12)	0.0346 (6)
C11	0.23935 (9)	0.3141 (5)	0.13955 (12)	0.0297 (5)
C21	0.18383 (9)	0.3638 (5)	0.08772 (13)	0.0345 (6)
C31	0.11421 (8)	0.2155 (4)	0.10637 (11)	0.0256 (5)
H3	0.46290	0.58950	0.21090	0.0390*
H4	0.48520	0.78150	0.06720	0.0460*
H5	0.39850	0.76120	-0.04320	0.0440*
H6	0.29140	0.55150	-0.01180	0.0420*
H11	0.23490	0.17970	0.19190	0.0360*
H11W	0.0460 (11)	0.241 (6)	-0.209 (2)	0.055 (8)*
H12W	-0.0168 (12)	0.157 (7)	-0.1733 (17)	0.044 (8)*
H21	0.18810	0.50100	0.03580	0.0410*
H21W	0.1828 (16)	0.851 (9)	-0.138 (2)	0.093 (13)*
H22W	0.1500 (18)	0.666 (12)	-0.209 (2)	0.101 (14)*

## supplementary materials

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### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Na1	0.0279 (4)	0.0275 (4)	0.0260 (4)	-0.0019 (3)	-0.0005 (3)	0.0007 (3)
O1W	0.0363 (7)	0.0350 (7)	0.0244 (7)	-0.0082 (5)	0.0026 (5)	-0.0002 (5)
O2W	0.0399 (9)	0.1143 (15)	0.0532 (10)	-0.0214 (9)	0.0046 (7)	0.0098 (9)
O21	0.0357 (8)	0.1222 (15)	0.0337 (8)	0.0081 (9)	0.0067 (6)	-0.0065 (9)
O22	0.0475 (9)	0.0775 (12)	0.0364 (9)	0.0164 (7)	-0.0096 (7)	0.0112 (7)
O31	0.0220 (6)	0.0370 (7)	0.0253 (6)	-0.0013 (5)	-0.0039 (5)	0.0011 (5)
O32	0.0290 (7)	0.0461 (8)	0.0255 (7)	0.0007 (5)	0.0005 (5)	0.0056 (6)
N21	0.0266 (8)	0.0471 (9)	0.0276 (8)	-0.0020 (7)	-0.0027 (6)	-0.0022 (7)
C1	0.0235 (8)	0.0281 (9)	0.0283 (9)	0.0035 (7)	0.0004 (7)	-0.0017 (7)
C2	0.0264 (8)	0.0257 (8)	0.0260 (9)	0.0040 (7)	0.0003 (7)	-0.0020 (7)
C3	0.0255 (9)	0.0331 (10)	0.0393 (11)	-0.0010 (8)	-0.0021 (7)	-0.0033 (8)
C4	0.0299 (10)	0.0362 (10)	0.0499 (13)	-0.0071 (8)	0.0106 (9)	-0.0017 (9)
C5	0.0405 (11)	0.0374 (10)	0.0331 (11)	-0.0045 (8)	0.0097 (8)	0.0035 (8)
C6	0.0340 (10)	0.0415 (11)	0.0281 (9)	0.0024 (8)	-0.0026 (7)	0.0020 (8)
C11	0.0261 (9)	0.0353 (10)	0.0276 (9)	-0.0015 (7)	0.0001 (7)	-0.0008 (7)
C21	0.0266 (9)	0.0435 (11)	0.0334 (10)	-0.0035 (8)	-0.0004 (7)	0.0093 (8)
C31	0.0231 (8)	0.0288 (9)	0.0248 (9)	0.0039 (7)	0.0008 (7)	-0.0028 (7)

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

Na1—O1W	2.3782 (13)	C1—C2	1.399 (2)
Na1—O2W	2.4404 (17)	C1—C6	1.395 (2)
Na1—O31	2.4370 (13)	C1—C11	1.476 (2)
Na1—O1W <sup>i</sup>	2.4162 (13)	C2—C3	1.382 (2)
Na1—O31 <sup>i</sup>	2.5046 (13)	C3—C4	1.371 (3)
Na1—O31 <sup>ii</sup>	2.4577 (13)	C4—C5	1.376 (3)
O21—N21	1.222 (2)	C5—C6	1.374 (3)
O22—N21	1.217 (2)	C11—C21	1.314 (3)
O31—C31	1.267 (2)	C21—C31	1.494 (2)
O32—C31	1.247 (2)	C3—H3	0.9300
O1W—H11W	0.78 (3)	C4—H4	0.9300
O1W—H12W	0.89 (2)	C5—H5	0.9300
O2W—H21W	0.77 (3)	C6—H6	0.9300
O2W—H22W	0.91 (4)	C11—H11	0.9300
N21—C2	1.463 (2)	C21—H21	0.9300
O1W—Na1—O2W	79.67 (5)	C2—C1—C6	115.06 (15)
O1W—Na1—O31	81.96 (4)	C2—C1—C11	123.37 (15)
O1W—Na1—O1W <sup>i</sup>	100.64 (4)	C6—C1—C11	121.46 (15)
O1W—Na1—O31 <sup>i</sup>	172.53 (5)	N21—C2—C1	121.34 (14)
O1W—Na1—O31 <sup>ii</sup>	85.02 (4)	N21—C2—C3	115.49 (14)
O2W—Na1—O31	101.58 (6)	C1—C2—C3	123.16 (15)
O1W <sup>i</sup> —Na1—O2W	86.43 (5)	C2—C3—C4	119.49 (17)
O2W—Na1—O31 <sup>i</sup>	107.80 (5)	C3—C4—C5	119.26 (18)

O2W—Na1—O31 <sup>ii</sup>	158.48 (6)	C4—C5—C6	120.69 (18)
O1W <sup>i</sup> —Na1—O31	171.93 (5)	C1—C6—C5	122.29 (16)
O31—Na1—O31 <sup>i</sup>	96.60 (4)	C1—C11—C21	124.71 (17)
O31—Na1—O31 <sup>ii</sup>	91.04 (4)	C11—C21—C31	124.59 (17)
O1W <sup>i</sup> —Na1—O31 <sup>i</sup>	79.83 (4)	O31—C31—C21	115.57 (14)
O1W <sup>i</sup> —Na1—O31 <sup>ii</sup>	81.62 (4)	O32—C31—C21	119.48 (15)
O31 <sup>i</sup> —Na1—O31 <sup>ii</sup>	87.68 (4)	O31—C31—O32	124.96 (15)
Na1—O1W—Na1 <sup>iii</sup>	100.64 (5)	C2—C3—H3	120.00
Na1—O31—C31	128.20 (10)	C4—C3—H3	120.00
Na1—O31—Na1 <sup>iii</sup>	96.60 (5)	C3—C4—H4	120.00
Na1—O31—Na1 <sup>ii</sup>	88.96 (4)	C5—C4—H4	120.00
Na1 <sup>iii</sup> —O31—C31	118.92 (10)	C4—C5—H5	120.00
Na1 <sup>ii</sup> —O31—C31	122.84 (10)	C6—C5—H5	120.00
Na1 <sup>iii</sup> —O31—Na1 <sup>ii</sup>	92.32 (4)	C1—C6—H6	119.00
H11W—O1W—H12W	112 (2)	C5—C6—H6	119.00
H21W—O2W—H22W	111 (4)	C1—C11—H11	118.00
O21—N21—C2	118.98 (14)	C21—C11—H11	118.00
O22—N21—C2	117.87 (14)	C11—C21—H21	118.00
O21—N21—O22	123.06 (16)	C31—C21—H21	118.00
O2W—Na1—O1W—Na1 <sup>iii</sup>	95.66 (6)	O31—Na1—O31 <sup>ii</sup> —C31 <sup>ii</sup>	-136.41 (11)
O31—Na1—O1W—Na1 <sup>iii</sup>	-7.75 (5)	Na1—O31—C31—O32	145.52 (13)
O1W <sup>i</sup> —Na1—O1W—Na1 <sup>iii</sup>	180.00 (6)	Na1—O31—C31—C21	-35.11 (19)
O31 <sup>ii</sup> —Na1—O1W—Na1 <sup>iii</sup>	-99.51 (5)	Na1 <sup>iii</sup> —O31—C31—O32	-87.48 (18)
O1W—Na1—O31—C31	142.67 (13)	Na1 <sup>iii</sup> —O31—C31—C21	91.89 (14)
O1W—Na1—O31—Na1 <sup>iii</sup>	7.39 (4)	Na1 <sup>ii</sup> —O31—C31—O32	26.8 (2)
O1W—Na1—O31—Na1 <sup>ii</sup>	-84.82 (4)	Na1 <sup>ii</sup> —O31—C31—C21	-153.81 (11)
O2W—Na1—O31—C31	65.02 (14)	O21—N21—C2—C1	-38.6 (2)
O2W—Na1—O31—Na1 <sup>iii</sup>	-70.26 (6)	O21—N21—C2—C3	140.47 (18)
O2W—Na1—O31—Na1 <sup>ii</sup>	-162.47 (5)	O22—N21—C2—C1	144.65 (17)
O31 <sup>i</sup> —Na1—O31—C31	-44.72 (13)	O22—N21—C2—C3	-36.2 (2)
O31 <sup>i</sup> —Na1—O31—Na1 <sup>iii</sup>	180.00 (4)	C6—C1—C2—N21	178.85 (15)
O31 <sup>i</sup> —Na1—O31—Na1 <sup>ii</sup>	87.79 (4)	C6—C1—C2—C3	-0.2 (3)
O31 <sup>ii</sup> —Na1—O31—C31	-132.51 (13)	C11—C1—C2—N21	-4.8 (3)
O31 <sup>ii</sup> —Na1—O31—Na1 <sup>iii</sup>	92.21 (5)	C11—C1—C2—C3	176.18 (17)
O31 <sup>ii</sup> —Na1—O31—Na1 <sup>ii</sup>	0.00 (3)	C2—C1—C6—C5	-1.5 (3)
O1W—Na1—O1W <sup>i</sup> —Na1 <sup>i</sup>	-180.00 (6)	C11—C1—C6—C5	-177.99 (17)
O2W—Na1—O1W <sup>i</sup> —Na1 <sup>i</sup>	-101.22 (6)	C2—C1—C11—C21	164.51 (18)
O2W—Na1—O31 <sup>i</sup> —Na1 <sup>i</sup>	75.56 (6)	C6—C1—C11—C21	-19.4 (3)
O2W—Na1—O31 <sup>i</sup> —C31 <sup>i</sup>	-65.26 (12)	N21—C2—C3—C4	-177.36 (16)
O31—Na1—O31 <sup>i</sup> —Na1 <sup>i</sup>	180.00 (3)	C1—C2—C3—C4	1.7 (3)
O31—Na1—O31 <sup>i</sup> —C31 <sup>i</sup>	39.19 (12)	C2—C3—C4—C5	-1.6 (3)
O1W—Na1—O31 <sup>ii</sup> —Na1 <sup>ii</sup>	81.83 (4)	C3—C4—C5—C6	-0.1 (3)

## supplementary materials

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O1W—Na1—O31 <sup>ii</sup> —C31 <sup>ii</sup>	-54.58 (11)	C4—C5—C6—C1	1.7 (3)
O2W—Na1—O31 <sup>ii</sup> —Na1 <sup>ii</sup>	126.42 (15)	C1—C11—C21—C31	179.16 (16)
O2W—Na1—O31 <sup>ii</sup> —C31 <sup>ii</sup>	-10.0 (2)	C11—C21—C31—O31	-169.51 (17)
O31—Na1—O31 <sup>ii</sup> —Na1 <sup>ii</sup>	0.00 (5)	C11—C21—C31—O32	9.9 (3)

Symmetry codes: (i)  $x, y+1, z$ ; (ii)  $-x, -y+1, -z$ ; (iii)  $x, y-1, z$ .

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1W—H11W $\cdots$ O32 <sup>iv</sup>	0.78 (3)	2.14 (3)	2.8871 (17)	162 (2)
O1W—H12W $\cdots$ O32 <sup>v</sup>	0.89 (2)	1.90 (2)	2.7852 (17)	171 (2)
O2W—H21W $\cdots$ O21 <sup>vi</sup>	0.77 (3)	2.49 (3)	3.050 (2)	131 (3)
O2W—H22W $\cdots$ O32 <sup>iv</sup>	0.91 (4)	2.04 (5)	2.882 (2)	153 (4)
C11—H11 $\cdots$ O21	0.93	2.39	2.846 (2)	110

Symmetry codes: (iv)  $x, -y+1/2, z-1/2$ ; (v)  $-x, -y, -z$ ; (vi)  $x, -y+3/2, z-1/2$ .

Fig. 1

