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2,2,4-Trimethyl-5-(4-tolylsulfonyl)-2,3,4,5-tetrahydro-1H-1,5-benzodiazepine

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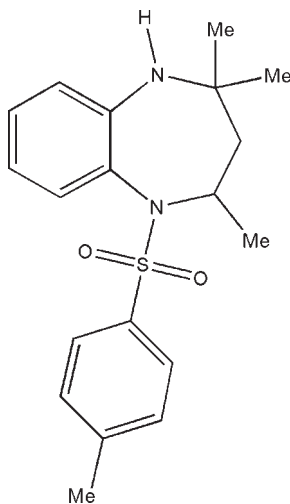
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.047; wR factor = 0.145; data-to-parameter ratio = 23.0.

In the title compound, $\text{C}_{19}\text{H}_{24}\text{N}_2\text{O}_2\text{S}$, the benzodiazepine ring adopts a distorted boat conformation. The S atom shows a distorted tetrahedral geometry, with the $\text{O}-\text{S}-\text{O}$ [$119.16(14)^\circ$] and $\text{N}-\text{S}-\text{C}$ [$107.48(10)^\circ$] angles deviating significantly from ideal values. The crystal packing is controlled by $\text{C}-\text{H}\cdots\text{O}$, $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For the use of benzodiazepines in the treatment of gastrointestinal and central nervous system disorders, see: Rahbaek *et al.* (1999). For hydrogen-bond motifs, see: Bernstein *et al.* (1995). For puckering and asymmetry parameters, see: Cremer & Pople (1975); Nardelli (1983). For the Thorpe-Ingold effect, see: Bassindale (1984). For details of the preparation, see: Ponnuswamy *et al.* (2006).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{24}\text{N}_2\text{O}_2\text{S}$
 $M_r = 344.46$
 Orthorhombic, $P2_12_12_1$
 $a = 7.3658(3)$ Å
 $b = 14.8013(8)$ Å
 $c = 17.4556(10)$ Å
 $V = 1903.07(17)$ Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.18$ mm⁻¹
 $T = 293$ K
 $0.25 \times 0.20 \times 0.20$ mm

Data collection

Bruker Kappa APEXII area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 2001)
 $T_{\min} = 0.957$, $T_{\max} = 0.964$
 13544 measured reflections
 5154 independent reflections
 3278 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.145$
 $S = 1.04$
 5154 reflections
 224 parameters
 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.19$ e Å⁻³
 Absolute structure: Flack (1983), 2171 Friedel pairs
 Flack parameter: $-0.12(9)$

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}20-\text{H}20\cdots\text{O}2$	0.93	2.51	2.885 (4)	105
$\text{N}1-\text{H}1\cdots\text{O}2^i$	0.75 (3)	2.52 (3)	3.268 (3)	176 (3)
$\text{C}14-\text{H}14\text{B}\cdots\text{O}1^{\text{ii}}$	0.96	2.58	3.436 (4)	149
$\text{C}13-\text{H}13\text{A}\cdots\text{C}g2$	0.96	2.90	3.7592 (31)	150
$\text{C}19-\text{H}19\cdots\text{C}g2^i$	0.93	2.90	3.5192 (34)	126

Symmetry codes: (i) $-x, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x - 1, y, z$. $\text{C}g2$ is the centroid of the = C15-C20 ring.

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5042).

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supplementary materials

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2,2,4-Trimethyl-5-(4-tolylsulfonyl)-2,3,4,5-tetrahydro-1H-1,5-benzodiazepine

K. Ravichandran, K. Sathiyaraj, S. S. Ilango, S. Ponnuswamy and M. N. Ponnuswamy

Comment

Benzodiazepines are known for their natural occurrence in filamentous fungi and actinomycetes of the *genera penicillium, aspergillus and streptomyces*. Benzodiazepines from *aspergillus* include *asperlicin*, which is used for treatment of gastrointestinal and central nervous system (CNS) disorders (Rahbaek *et al.*, 1999). In view of these importance and to ascertain the molecular conformation, a crystallographic study of the title compound has been carried out.

The *ORTEP* diagram of the title compound is shown in Fig.1. The benzodiazepine ring adopts a distorted boat conformation. The puckering parameters (Cremer & Pople, 1975) and the asymmetry parameters (Nardelli, 1983) for this ring are $q_2 = 0.976(3) \text{ \AA}$, $q_3 = 0.068(3) \text{ \AA}$, $\varphi_2 = 152.5(2)^\circ$, $\varphi_3 = 30.0(1)^\circ$ and $\Delta 2(C4) = 7.8(1)^\circ$. Atom S takes up a distorted tetrahedral geometry, with the O—S—O and N—S—C angles deviating significantly from ideal values, and this may be attributed to the Thorpe-Ingold effect (Bassindale, 1984). The sum of the bond angles at N1 (342.9°) of the benzodiazepine ring is in accordance with sp^3 hybridization, whereas the one at N5 (359.7°) is in sp^2 hybridization, respectively.

The crystal packing is controlled by C—H \cdots O, N—H \cdots O and C—H \cdots π types of intra and intermolecular interactions. Atom N1 at (x, y, z) donates a proton to O1 ($-x, y + 1/2, -z + 1/2 + 1$), which forms a one-dimensional C7 chain (Bernstein *et al.*, 1995) running along *b*-axis. The intermolecular hydrogen bond C14—H14B \cdots O1 connects the molecule into a C6 one dimensional chain running along *a*-axis as shown in Fig 2. The methyl group H atom in the benzodiazepine ring interacts with the centroid atom of the toluenesulfonyl ring through an intramolecular C—H \cdots π interaction involving atom C13, the separation between H13A and the centroid (*Cg*2) of the ring (C15/C16/C17/C18/C19/C20) being 2.899 \AA .

Experimental

To a solution of 2,2,4-Trimethyl-1H-tetrahydro-1,5-benzodiazepine (0.59 g) in anhydrous benzene (16 ml), triethylamine (0.7 ml) and 4-toluenesulfonyl chloride (1 gm) were added. The reaction mixture was allowed to reflux for 16 hrs. The benzene solution was washed with water, dried over anhydrous Na₂SO₄ and concentrated. The resulting mass was purified by crystallization from benzene (Ponnuswamy *et al.*, 2006).

Refinement

The amino H atom was freely refined. The other H atoms were positioned geometrically (C—H=0.93–0.98 \AA) and allowed to ride on their parent atoms, with $1.5U_{eq}(C_{methyl})$ or $1.2U_{eq}(C)$. *PLATON* (Spek, 2009) detected a solvent accessible void of approximately 31 \AA^3 . This void could have initially contained solvent molecules but these molecules have since evaporated from the structure without degradation of the crystal.

Figures

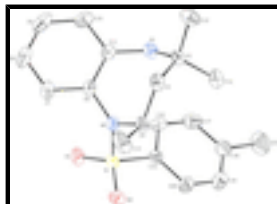


Fig. 1. Perspective view of the molecule showing the displacement ellipsoids at 30% probability level. H atoms have been omitted for clarity.

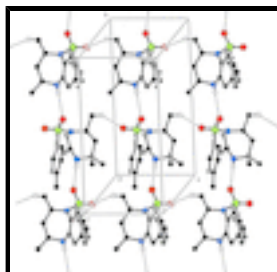


Fig. 2. The crystal packing of the molecules viewed down *c*-axis. H atoms not involved in hydrogen bonding have been omitted for clarity.

2,2,4-Trimethyl-5-(4-tolylsulfonyl)-2,3,4,5-tetrahydro-1*H*-1,5-benzodiazepine

Crystal data

$C_{19}H_{24}N_2O_2S$

$M_r = 344.46$

Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 7.3658$ (3) Å

$b = 14.8013$ (8) Å

$c = 17.4556$ (10) Å

$V = 1903.07$ (17) Å³

$Z = 4$

$F_{000} = 736$

$D_x = 1.202$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2546 reflections

$\theta = 1.8$ – 29.4°

$\mu = 0.18$ mm⁻¹

$T = 293$ K

Block, colourless

$0.25 \times 0.20 \times 0.20$ mm

Data collection

Bruker Kappa APEXII area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 293$ K

ω and φ scans

Absorption correction: Multi-scan (SADABS; Sheldrick, 2001)

$T_{\min} = 0.957$, $T_{\max} = 0.964$

13544 measured reflections

5154 independent reflections

3278 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.034$

$\theta_{\text{max}} = 29.4^\circ$

$\theta_{\text{min}} = 1.8^\circ$

$h = -5 \rightarrow 10$

$k = -20 \rightarrow 18$

$l = -23 \rightarrow 23$

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.047$	$w = 1/[\sigma^2(F_o^2) + (0.0742P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.145$	$(\Delta/\sigma)_{\max} = 0.009$
$S = 1.04$	$\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$
5154 reflections	$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$
224 parameters	Extinction correction: none
Primary atom site location: structure-invariant direct methods	Absolute structure: Flack (1983), 2171 Friedel pairs
Secondary atom site location: difference Fourier map	Flack parameter: -0.12 (9)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.08799 (9)	0.63541 (4)	0.80960 (4)	0.0620 (2)
O1	0.2717 (3)	0.63113 (17)	0.78264 (12)	0.0849 (6)
O2	0.0096 (3)	0.55749 (13)	0.84498 (12)	0.0870 (7)
N1	-0.0887 (3)	0.84795 (14)	0.70370 (13)	0.0566 (5)
H1	-0.072 (4)	0.897 (2)	0.6945 (18)	0.073 (9)*
C2	-0.2833 (3)	0.83013 (16)	0.71790 (15)	0.0520 (6)
C3	-0.3323 (3)	0.73221 (16)	0.70025 (16)	0.0577 (6)
H3A	-0.4623	0.7257	0.7074	0.069*
H3B	-0.3078	0.7218	0.6464	0.069*
C4	-0.2402 (3)	0.65759 (15)	0.74511 (17)	0.0589 (7)
H4	-0.2719	0.6642	0.7993	0.071*
N5	-0.0387 (3)	0.66325 (13)	0.73728 (11)	0.0524 (5)
C6	0.0396 (3)	0.70241 (16)	0.66986 (13)	0.0497 (6)
C7	0.1455 (3)	0.6493 (2)	0.62250 (16)	0.0659 (7)
H7	0.1565	0.5877	0.6320	0.079*
C8	0.2354 (4)	0.6879 (3)	0.56082 (18)	0.0855 (10)

supplementary materials

H8	0.3070	0.6524	0.5289	0.103*
C9	0.2182 (4)	0.7779 (3)	0.54737 (17)	0.0862 (11)
H9	0.2819	0.8041	0.5071	0.103*
C10	0.1077 (4)	0.8310 (2)	0.59246 (15)	0.0676 (7)
H10	0.0954	0.8922	0.5814	0.081*
C11	0.0141 (3)	0.79417 (17)	0.65442 (13)	0.0501 (6)
C12	-0.3967 (5)	0.8894 (2)	0.6654 (2)	0.0813 (9)
H12A	-0.3735	0.9518	0.6767	0.122*
H12B	-0.5231	0.8766	0.6732	0.122*
H12C	-0.3652	0.8773	0.6130	0.122*
C13	-0.3238 (4)	0.8554 (2)	0.80092 (17)	0.0704 (7)
H13A	-0.2526	0.8182	0.8345	0.106*
H13B	-0.4504	0.8460	0.8113	0.106*
H13C	-0.2939	0.9177	0.8092	0.106*
C14	-0.3082 (5)	0.5663 (2)	0.7171 (3)	0.0999 (13)
H14A	-0.2821	0.5600	0.6635	0.150*
H14B	-0.4369	0.5623	0.7251	0.150*
H14C	-0.2485	0.5190	0.7451	0.150*
C15	0.0750 (3)	0.72366 (16)	0.87704 (14)	0.0516 (5)
C16	0.1482 (4)	0.80712 (19)	0.85876 (15)	0.0640 (7)
H16	0.2035	0.8165	0.8115	0.077*
C17	0.1378 (4)	0.8753 (2)	0.91126 (17)	0.0759 (8)
H17	0.1873	0.9313	0.8988	0.091*
C18	0.0569 (4)	0.8647 (2)	0.98198 (17)	0.0722 (7)
C19	-0.0125 (4)	0.7808 (2)	0.99895 (18)	0.0796 (9)
H19	-0.0666	0.7712	1.0464	0.095*
C20	-0.0035 (4)	0.7106 (2)	0.94719 (17)	0.0700 (8)
H20	-0.0508	0.6543	0.9600	0.084*
C21	0.0455 (7)	0.9425 (3)	1.0377 (2)	0.1170 (14)
H21A	-0.0147	0.9229	1.0836	0.176*
H21B	-0.0217	0.9911	1.0150	0.176*
H21C	0.1657	0.9628	1.0501	0.176*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0663 (4)	0.0509 (3)	0.0687 (4)	0.0183 (3)	0.0157 (3)	0.0095 (3)
O1	0.0646 (11)	0.1087 (16)	0.0812 (13)	0.0422 (12)	0.0166 (10)	0.0088 (13)
O2	0.1298 (19)	0.0453 (10)	0.0858 (13)	0.0116 (11)	0.0213 (13)	0.0147 (10)
N1	0.0579 (12)	0.0403 (11)	0.0715 (13)	-0.0081 (9)	0.0017 (11)	-0.0036 (10)
C2	0.0474 (12)	0.0439 (12)	0.0648 (15)	0.0013 (9)	-0.0032 (11)	-0.0035 (11)
C3	0.0432 (11)	0.0530 (13)	0.0770 (17)	-0.0057 (10)	0.0066 (12)	-0.0146 (13)
C4	0.0530 (13)	0.0434 (13)	0.0803 (17)	-0.0027 (11)	0.0239 (13)	-0.0049 (12)
N5	0.0497 (11)	0.0486 (11)	0.0591 (11)	0.0025 (8)	0.0130 (9)	-0.0015 (9)
C6	0.0443 (12)	0.0558 (13)	0.0491 (12)	-0.0020 (10)	0.0070 (10)	-0.0089 (10)
C7	0.0542 (14)	0.0773 (19)	0.0663 (15)	0.0042 (13)	0.0106 (12)	-0.0189 (15)
C8	0.0705 (19)	0.128 (3)	0.0583 (18)	-0.001 (2)	0.0171 (16)	-0.0234 (19)
C9	0.0677 (19)	0.140 (4)	0.0504 (16)	-0.031 (2)	0.0072 (14)	0.005 (2)

C10	0.0623 (16)	0.087 (2)	0.0541 (14)	-0.0195 (15)	-0.0074 (13)	0.0098 (14)
C11	0.0442 (11)	0.0583 (14)	0.0478 (12)	-0.0100 (10)	-0.0031 (10)	-0.0029 (11)
C12	0.081 (2)	0.0666 (17)	0.097 (2)	0.0052 (16)	-0.0252 (18)	0.0046 (16)
C13	0.0620 (15)	0.0673 (17)	0.0819 (18)	0.0051 (13)	0.0077 (14)	-0.0183 (16)
C14	0.077 (2)	0.0482 (15)	0.174 (4)	-0.0146 (14)	0.024 (2)	-0.018 (2)
C15	0.0412 (11)	0.0547 (13)	0.0590 (13)	0.0080 (10)	0.0036 (11)	0.0126 (11)
C16	0.0731 (17)	0.0651 (17)	0.0538 (14)	-0.0065 (14)	-0.0021 (13)	0.0170 (13)
C17	0.091 (2)	0.0607 (17)	0.0764 (19)	-0.0149 (15)	-0.0173 (16)	0.0148 (15)
C18	0.0755 (18)	0.0681 (18)	0.0730 (17)	0.0079 (16)	-0.0034 (15)	-0.0024 (16)
C19	0.079 (2)	0.092 (2)	0.0674 (17)	0.0025 (17)	0.0194 (16)	-0.0025 (17)
C20	0.0712 (17)	0.0671 (18)	0.0717 (17)	-0.0057 (14)	0.0213 (15)	0.0072 (14)
C21	0.144 (4)	0.103 (3)	0.104 (3)	0.019 (3)	-0.017 (3)	-0.032 (2)

Geometric parameters (Å, °)

S1—O2	1.430 (2)	C10—C11	1.394 (3)
S1—O1	1.434 (2)	C10—H10	0.9300
S1—N5	1.623 (2)	C12—H12A	0.9600
S1—C15	1.761 (3)	C12—H12B	0.9600
N1—C11	1.395 (3)	C12—H12C	0.9600
N1—C2	1.478 (3)	C13—H13A	0.9600
N1—H1	0.75 (3)	C13—H13B	0.9600
C2—C12	1.519 (4)	C13—H13C	0.9600
C2—C13	1.526 (4)	C14—H14A	0.9600
C2—C3	1.525 (3)	C14—H14B	0.9600
C3—C4	1.515 (4)	C14—H14C	0.9600
C3—H3A	0.9700	C15—C20	1.368 (4)
C3—H3B	0.9700	C15—C16	1.385 (4)
C4—N5	1.493 (3)	C16—C17	1.366 (4)
C4—C14	1.522 (4)	C16—H16	0.9300
C4—H4	0.9800	C17—C18	1.380 (4)
N5—C6	1.433 (3)	C17—H17	0.9300
C6—C7	1.382 (3)	C18—C19	1.375 (4)
C6—C11	1.397 (3)	C18—C21	1.510 (5)
C7—C8	1.387 (5)	C19—C20	1.379 (4)
C7—H7	0.9300	C19—H19	0.9300
C8—C9	1.359 (5)	C20—H20	0.9300
C8—H8	0.9300	C21—H21A	0.9600
C9—C10	1.378 (5)	C21—H21B	0.9600
C9—H9	0.9300	C21—H21C	0.9600
O2—S1—O1	119.16 (14)	N1—C11—C10	121.6 (2)
O2—S1—N5	107.96 (13)	N1—C11—C6	120.6 (2)
O1—S1—N5	107.36 (11)	C10—C11—C6	117.6 (2)
O2—S1—C15	106.71 (12)	C2—C12—H12A	109.5
O1—S1—C15	107.67 (14)	C2—C12—H12B	109.5
N5—S1—C15	107.48 (10)	H12A—C12—H12B	109.5
C11—N1—C2	121.86 (19)	C2—C12—H12C	109.5
C11—N1—H1	109 (2)	H12A—C12—H12C	109.5
C2—N1—H1	112 (3)	H12B—C12—H12C	109.5

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N1—C2—C12	109.2 (2)	C2—C13—H13A	109.5
N1—C2—C13	107.8 (2)	C2—C13—H13B	109.5
C12—C2—C13	108.9 (2)	H13A—C13—H13B	109.5
N1—C2—C3	111.42 (19)	C2—C13—H13C	109.5
C12—C2—C3	107.3 (2)	H13A—C13—H13C	109.5
C13—C2—C3	112.2 (2)	H13B—C13—H13C	109.5
C4—C3—C2	118.8 (2)	C4—C14—H14A	109.5
C4—C3—H3A	107.6	C4—C14—H14B	109.5
C2—C3—H3A	107.6	H14A—C14—H14B	109.5
C4—C3—H3B	107.6	C4—C14—H14C	109.5
C2—C3—H3B	107.6	H14A—C14—H14C	109.5
H3A—C3—H3B	107.0	H14B—C14—H14C	109.5
N5—C4—C3	110.93 (19)	C20—C15—C16	119.8 (3)
N5—C4—C14	110.3 (2)	C20—C15—S1	121.1 (2)
C3—C4—C14	109.5 (3)	C16—C15—S1	119.09 (19)
N5—C4—H4	108.7	C17—C16—C15	118.9 (3)
C3—C4—H4	108.7	C17—C16—H16	120.6
C14—C4—H4	108.7	C15—C16—H16	120.6
C6—N5—C4	119.9 (2)	C16—C17—C18	122.7 (3)
C6—N5—S1	120.68 (16)	C16—C17—H17	118.7
C4—N5—S1	119.08 (17)	C18—C17—H17	118.7
C7—C6—C11	120.9 (2)	C19—C18—C17	117.2 (3)
C7—C6—N5	119.2 (2)	C19—C18—C21	121.9 (3)
C11—C6—N5	119.8 (2)	C17—C18—C21	120.9 (3)
C6—C7—C8	120.0 (3)	C18—C19—C20	121.5 (3)
C6—C7—H7	120.0	C18—C19—H19	119.3
C8—C7—H7	120.0	C20—C19—H19	119.3
C9—C8—C7	119.6 (3)	C15—C20—C19	120.0 (3)
C9—C8—H8	120.2	C15—C20—H20	120.0
C7—C8—H8	120.2	C19—C20—H20	120.0
C8—C9—C10	121.0 (3)	C18—C21—H21A	109.5
C8—C9—H9	119.5	C18—C21—H21B	109.5
C10—C9—H9	119.5	H21A—C21—H21B	109.5
C9—C10—C11	120.8 (3)	C18—C21—H21C	109.5
C9—C10—H10	119.6	H21A—C21—H21C	109.5
C11—C10—H10	119.6	H21B—C21—H21C	109.5
C11—N1—C2—C12	-95.9 (3)	C8—C9—C10—C11	1.7 (4)
C11—N1—C2—C13	145.9 (2)	C2—N1—C11—C10	123.8 (3)
C11—N1—C2—C3	22.4 (3)	C2—N1—C11—C6	-62.0 (3)
N1—C2—C3—C4	61.1 (3)	C9—C10—C11—N1	175.7 (2)
C12—C2—C3—C4	-179.4 (2)	C9—C10—C11—C6	1.3 (4)
C13—C2—C3—C4	-59.8 (3)	C7—C6—C11—N1	-178.1 (2)
C2—C3—C4—N5	-57.2 (3)	N5—C6—C11—N1	-0.7 (3)
C2—C3—C4—C14	-179.3 (2)	C7—C6—C11—C10	-3.7 (3)
C3—C4—N5—C6	-27.7 (3)	N5—C6—C11—C10	173.7 (2)
C14—C4—N5—C6	93.9 (3)	O2—S1—C15—C20	-1.4 (3)
C3—C4—N5—S1	145.32 (18)	O1—S1—C15—C20	-130.5 (2)
C14—C4—N5—S1	-93.1 (3)	N5—S1—C15—C20	114.2 (2)
O2—S1—N5—C6	-146.99 (18)	O2—S1—C15—C16	177.6 (2)

O1—S1—N5—C6	-17.3 (2)	O1—S1—C15—C16	48.5 (2)
C15—S1—N5—C6	98.24 (18)	N5—S1—C15—C16	-66.8 (2)
O2—S1—N5—C4	40.06 (19)	C20—C15—C16—C17	-1.0 (4)
O1—S1—N5—C4	169.70 (18)	S1—C15—C16—C17	-180.0 (2)
C15—S1—N5—C4	-74.71 (19)	C15—C16—C17—C18	-0.1 (4)
C4—N5—C6—C7	-116.8 (2)	C16—C17—C18—C19	1.0 (4)
S1—N5—C6—C7	70.3 (3)	C16—C17—C18—C21	-178.9 (3)
C4—N5—C6—C11	65.7 (3)	C17—C18—C19—C20	-0.8 (5)
S1—N5—C6—C11	-107.2 (2)	C21—C18—C19—C20	179.1 (3)
C11—C6—C7—C8	3.2 (4)	C16—C15—C20—C19	1.2 (4)
N5—C6—C7—C8	-174.3 (2)	S1—C15—C20—C19	-179.8 (2)
C6—C7—C8—C9	-0.1 (4)	C18—C19—C20—C15	-0.3 (5)
C7—C8—C9—C10	-2.3 (5)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C20—H20 \cdots O2	0.93	2.51	2.885 (4)	105
N1—H1 \cdots O2 ⁱ	0.75 (3)	2.52 (3)	3.268 (3)	176 (3)
C14—H14B \cdots O1 ⁱⁱ	0.96	2.58	3.436 (4)	149
C13—H13A \cdots Cg2	0.96	2.90	3.7592 (31)	150
C19—H19 \cdots Cg2 ⁱ	0.93	2.90	3.5192 (34)	126

Symmetry codes: (i) $-x, y+1/2, -z+3/2$; (ii) $x-1, y, z$.

Fig. 1

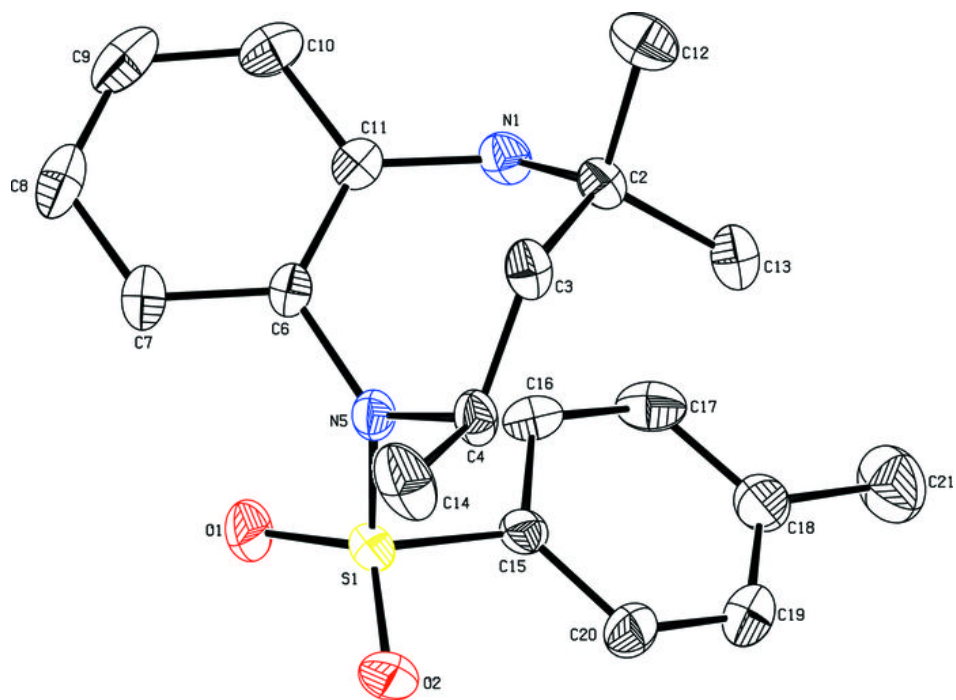


Fig. 2

