Cylinder, $40 \times 1 \text{ mm}$

T = 293 K



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Alaptide from synchrotron powder diffraction data

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Key indicators: powder synchrotron study; T = 293 K; mean $\sigma(C-C) = 0.005 \text{ Å}$; R factor = 0.058; wR factor = 0.089; data-to-parameter ratio = 295.7.

The title compound [systematic name: (8S)-8-methyl-6,9diazaspiro[4.5]decane-7,10-dione], C₉H₁₄N₂O₂, consists of two connected rings, viz. a piperazine-2,5-dione (DKP) ring and a five-membered ring. The DKP ring adopts a slight boat conformation and the bonded methyl group is in an equatorial position. The five-membered ring is in an envelope conformation. In the crystal structure, intermolecular N-H···O hydrogen bonds link molecules into chains running parallel to the c axis.

Related literature

For background to alaptide and its biological activity, see: Kasafírek et al. (1992); Hliňák et al. (1996). For a related structure, see: Symerský et al. (1987). For the original powder diffraction data, see: Maixner et al. (2009). For the synthetic procedure, see: Sturc & Kacafirek (1992). For a description of the Cambridge Structural Database, see: Allen (2002). For the March-Dollase orientation correction, see: (Dollase, 1986).

Experimental

Crystal data

 $C_9H_{14}N_2O_2$ $M_r = 182.22$ Orthorhombic, P2₁2₁2₁ a = 21.14118 (7) Åb = 7.22207 (2) Å c = 6.14610 (3) Å

 $V = 938.41 (1) \text{ Å}^3$ 7 - 4

Synchrotron radiation, $\lambda = 0.79984 \text{ Å}$

Data collection

ID31 ESRF Grenoble Scan method: step

 $2\theta_{\min}=1.00^{\circ},\,2\theta_{\max}=48.01^{\circ},\,2\theta_{\mathrm{step}}=$ diffractometer Specimen mounting: capilary

Data collection mode: transmission

Refinement

 $R_{\rm p} = 0.058$ 15671 data points $R_{\rm wp} = 0.089$ 53 parameters $R_{\rm exp} = 0.023$ 37 restraints

 $R_{\text{Bragg}} = 0.102$ $\chi^2 = 15.210$ H-atom parameters not refined

Table 1 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
N4—H41···O8 ⁱ	0.86	2.10	2.929 (3)	164
N7—H71···O13 ⁱⁱ	0.86	2.01	2.826 (3)	159

Symmetry codes: (i) x, y, z + 1; (ii) x, y, z - 1.

Data collection: ESRF SPEC (Certified Scientific Software, 2003); cell refinement: EXPO2004 (Altomare et al., 1999); data reduction: CRYSFIRE2004 (Shirley, 2000); program(s) used to solve structure: EXPO2004; program(s) used to refine structure: GSAS (Larson & Von Dreele, 1994); molecular graphics: Mercury (Macrae et al., 2006) and PLATON (Spek, 2009); software used to prepare material for publication: enCIFer (Allen et al., 2004).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2977).

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S1. Comment

Alaptide is a small molecule belonging to the group of spirocyclic dipeptides (Kasafirek *et al.*, 1992). The systematic research during the last twenty years has shown a positive effect of alaptide and its derivatives on the memory of animals and on healing of burns (Hliňák *et al.*, 1996).

The molecular structure of the title compound is shown in Fig. 1. The crystal structure contains two types of intermolecular N—H···O hydrogen bonds between DKP rings. The DKP ring adopts a slight boat conformation and is connected via the spiro junction to a five-membered carbon ring which is in an envelope conformation. The methyl group bonded to the dipeptide ring is in an equatorial position. A search in the Cambridge Structural Database (Allen, 2002) found the crystal structure of a similar type of molecule, namely: (8S)-8-Hydroxymethyl-6,9-diazaspiro[4.5]decane-7,10-dione (CSD refcode FEPFOV; Symerský *et al.*, 1987). This structure has the same spacegroup and comparable unit-cell parameters as the reported structure of the title copmound. Two similar hydrogen bonds N—H···O connecting DKP rings of neighboring molecules occur in both crystal structures. In both structures, the hydrogen bonding connects molecules to form one-dimensional chains. The third hydrogen bond O—H···O is missing in the structure of alaptide, which causes a different formation of extended chains in these structures, see Fig. 2.

S2. Experimental

The title compound was synthesized according to the procedure of Sturc & Kacafirek (1992). Alaptide was crystallized from various solvents in order to check polymorphism, but only one solid form was found (Maixner *et al.*, 2009). The sample for measurement was recrystallized from methanol by slow evaporation technique.

S3. Refinement

X-Ray diffraction data were collected on the high resolution diffractometer ID31 of the European Synchrotron Radiation Facility. The monochromatic wavelength was fixed at 0.79984 (4) Å. Si (111) crystal multi-analyzer combined with Si (111) monochromator was used (beam offset angle $\alpha = 2^{\circ}$). A rotating 1-mm-diameter borosilicate glass capillary with alaptide powder was used for the experiment. Data were measured from 1.002° 2θ to 48.012° 2θ at the room temperature, steps scans were set to 0.003° 2θ .

Indexation was done in CRYSFIRE 2004 (Shirley, 2000) package. It confirmed previously presented unit-cell parameters and space group (Maixner *et al.*, 2009): a = 21.136 (4), b = 7.212 (4), c = 6.126 (3) Å, $P2_12_12_1$, V = 933.8 (8) Å³, and Z = 4. The structure was solved by using direct space methods implemented in EXPO2004 package (Altomare *et al.*,1999). All non-hydrogen atoms were found in the structure solution process. Hydrogen atoms were placed in their theoretical positions and structure was refined by Rietveld method as implemented in *GSAS* (Larson & Von Dreele, 1994). Bonds, angles and planar group restraints were used during refinement. At final stages atomic coordinates and U_{iso}

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parameters of non-hydrogen atoms were refined to the final agreement factors $R_p = 0.059$ and $R_{wp} = 0.089$. The diffraction profiles and differences between the measured and calculated profiles are shown in Fig. 3.

The isotropic displacement parameters of atoms C10, C11 and C12 are large compared to those of the other atoms. A disorder model was attempted but this did not improve the refinement and therefore was not used.

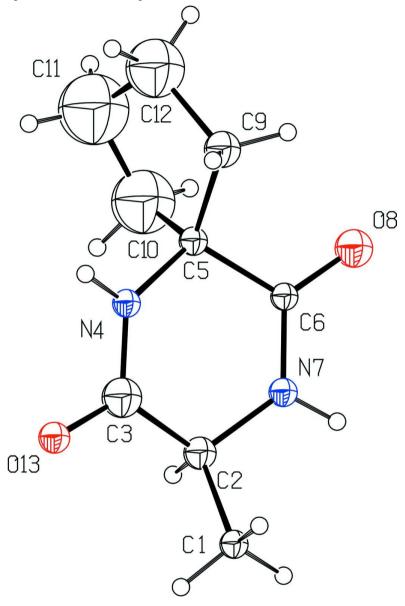


Figure 1The molecular structure of alaptide showing the atomic numbering. Displacement spheres are drawn at 30% probability level.

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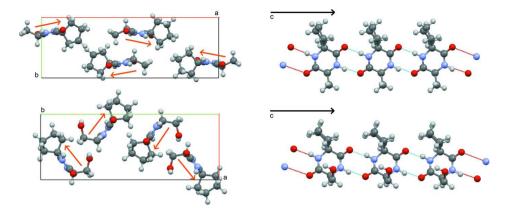


Figure 2Comparison of molecular packing (left - arrows show directions of dipeptide rings) and hydrogen bonding system (right) of two structures. Top: Structure of alaptide, bottom: Structure of (8*S*)-8-Hydroxymethyl-6,9-diazaspiro[4.5]decane-7,10-dione.

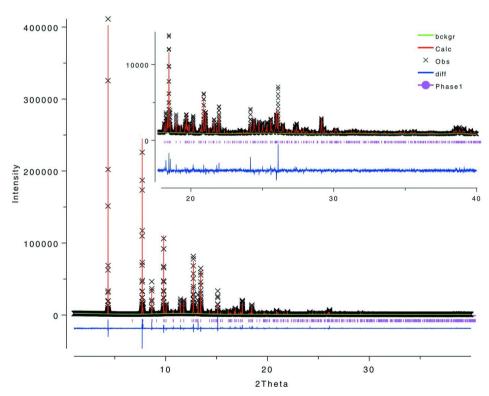


Figure 3

The final Rietveld plot showing the measured data (black thin-plus), calculated data (red line) and difference curve (blue line). Calculated positions of the reflection are shown by vertical bars.

(8S)-8-methyl-6,9-diazaspiro[4.5]decane-7,10-dione

Crystal data $C_9H_{14}N_2O_2$ $M_r = 182.22$ Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab a = 21.14118 (7) Å b = 7.22207 (2) Å

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c = 6.14610 (3) Å $V = 938.41 (1) \text{ Å}^3$ Z = 4F(000) = 392 $D_{\rm x} = 1.290 {\rm \ Mg \ m^{-3}}$ Synchrotron radiation, $\lambda = 0.79984 \text{ Å}$

Data collection

ID31 ESRF Grenoble diffractometer Radiation source: synchrotron

Si(111) monochromator

Refinement

37 restraints

Least-squares matrix: full

 $R_{\rm p} = 0.058$ $R_{\rm wp} = 0.089$ $R_{\rm exp} = 0.023$ $R_{\rm Bragg} = 0.102$ $\chi^2 = 15.210$ 15671 data points Excluded region(s): no 53 parameters

T = 293 K

Particle morphology: no specific habit

white

cylinder, 40 × 1 mm

Specimen preparation: Prepared at 293 K and

101 kPa

Specimen mounting: capilary Data collection mode: transmission

Scan method: step

 $2\theta_{\min} = 1.001^{\circ}, 2\theta_{\max} = 48.011^{\circ}, 2\theta_{\text{step}} = 0.003^{\circ}$

0 constraints

H-atom parameters not refined

Weighting scheme based on measured s.u.'s w =

 $1/\sigma(Y_{obs})^2$

 $(\Delta/\sigma)_{\text{max}} = 0.06$

Background function: Shifted Chebyschev Preferred orientation correction: March-Dollase (Dollase, 1986); direction of preferred

orientation is 101; MD = 0.93

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

	x	y	Z	$U_{ m iso}$ */ $U_{ m eq}$
C1	-0.08373 (10)	0.2121 (8)	-0.0450(5)	0.027 (3)*
C2	-0.01869 (8)	0.3035 (3)	-0.0158 (4)	0.035 (3)*
C3	0.01033 (9)	0.27000 (17)	0.2032 (3)	0.052 (3)*
N4	0.07052 (9)	0.2345 (4)	0.2137 (3)	0.026 (2)*
C5	0.11718 (7)	0.2470(3)	0.0365 (3)	0.027 (3)*
C6	0.08529 (8)	0.23092 (17)	-0.1846(3)	0.025 (3)*
N7	0.02321 (9)	0.2505 (4)	-0.1937(3)	0.027 (2)*
O8	0.11977 (11)	0.2014 (4)	-0.3425(4)	0.047 (2)*
C9	0.16843 (14)	0.0925 (6)	0.0590 (5)	0.043 (4)*
C10	0.15361 (16)	0.4304 (5)	0.0487 (5)	0.133 (6)*
C11	0.20873 (16)	0.3907 (9)	0.1995 (5)	0.165 (5)*
C12	0.22476 (14)	0.1870 (9)	0.1704 (8)	0.114 (4)*
O13	-0.02052 (12)	0.2734 (4)	0.3727 (4)	0.0319 (18)*
H11	-0.0992	0.2386	-0.1875	0.0346*
H12	-0.1121	0.2595	0.0598	0.0346*
H13	-0.0795	0.0821	-0.0282	0.0346*
H21	-0.025	0.4337	-0.0286	0.0516*
H91	0.181	0.0494	-0.0796	0.063*
H92	0.1531	-0.0061	0.1445	0.063*
H101	0.1686	0.4665	-0.0916	0.18*
H102	0.1276	0.5269	0.1053	0.18*
H111	0.2441	0.4664	0.1615	0.2445*
H112	0.197	0.4181	0.3464	0.2445*

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H121	0.2617	0.1746	0.089	0.168*			
H122	0.2309	0.1335	0.313	0.168*			
H71	0.006	0.2305	-0.3181	0.0296*			
H41	0.0848	0.2013	0.3384	0.0271*			
Geometric	Geometric parameters (Å, °)						
O8—C6		1.232 (3)	N4—H41	0.86			
O13—C3		1.229 (3)	N7—H71	0.86			
N4—C3		1.300(3)	C1—H11	0.95			
N4—C5		1.472 (3)	C1—H12	0.94			
N7—C2		1.458 (3)	C1—H13	0.95			
N7—C6		1.321 (3)	C2—H21	0.95			
C1—C2		1.536 (4)	C9—H91	0.95			
C2—C3		1.499 (3)	C9—H92	0.94			
C5—C6		1.521 (3)	C10—H101	0.95			
C5—C9		1.561 (4)	C10—H102	0.95			
C5—C10		1.534 (4)	C11—H111	0.96			
C9—C12		1.534 (5)	C11—H112	0.96			
C10—C11		1.516 (5)	C12—H121	0.93			
C11—C12		1.520 (9)	C12—H122	0.97			
C3—N4—		127.39 (18)	C2—C1—H13	109			
C2—N7—		126.85 (18)	H11—C1—H12	110			
N7—C2—		110.1 (2)	H11—C1—H13	109			
N7—C2—		112.48 (16)	H12—C1—H13	110			
C1—C2—		113.7 (2)	N7—C2—H21	106			
O13—C3-		118.8 (2)	C1—C2—H21	107			
O13—C3-		122.7 (2)	C3—C2—H21	107			
N4—C3—		118.50 (17)	C5—C9—H91	111			
N4—C5—		111.05 (14)	C5—C9—H92	111			
N4—C5—		110.8 (2)	C12—C9—H91	109			
N4—C5—		110.7 (2)	C12—C9—H92	111			
C6—C5—		109.39 (18)	H91—C9—H92	111			
C6—C5—		109.39 (18)	C5—C10—H101	111			
C9—C5—		105.3 (2)	C5—C10—H102	111			
O8—C6—		124.94 (19)	C11—C10—H101	110			
O8—C6—		117.03 (17)	C11—C10—H102	111			
N7—C6—		118.02 (16)	H101—C10—H102	109			
C5—C9—		105.1 (3)	C10—C11—H111	110			
C5—C10–		104.6 (3)	C10—C11—H112	110			
C10—C11		106.4 (4)	C12—C11—H111	111			
C9—C12-		108.1 (3)	C12—C11—H112	112			
C3—N4—		116	H111—C11—H112	108			
C5—N4—		116	C9—C12—H121	112			
C2—N7—		117	C9—C12—H122	109			
C6—N7—		116	C11—C12—H121	110			
C2—C1—	·H11	109	C11—C12—H122	108			

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<u>C2</u> —C1—H12	109	H121—C12—H122		110	
Hydrogen-bond geometry (Å, °)					
<i>D</i> —H··· <i>A</i>		<i>D</i> —Н	H···A	D···A	<i>D</i> —H··· <i>A</i>
N4—H41···O8 ⁱ		0.86	2.10	2.929 (3)	164
N7—H71···O13 ⁱⁱ		0.86	2.01	2.826 (3)	159

Symmetry codes: (i) x, y, z+1; (ii) x, y, z-1.

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