

1'-Ethylsulfanyl-1,1'-bicyclohexyl-2-one

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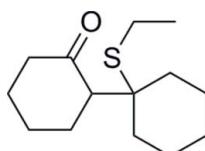
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Key indicators: single-crystal X-ray study; $T = 90\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$; R factor = 0.034; wR factor = 0.092; data-to-parameter ratio = 20.6.

There are two independent molecules in the asymmetric unit of the title cyclohexanone derivative, $\text{C}_{14}\text{H}_{24}\text{OS}$, in which both cyclohexane rings exhibit chair conformations. They are also equatorial to each other, which permits the ethanethiol substituent to be in a *syn* conformation with the α -H atom of the parent attached cyclohexanone.

Related literature

For background literature on the synthesis, see Bach & Klix (1985); Trost *et al.* (1976); Reetz & Giannis (1981). For the preparation of the starting materials, see: Ito *et al.* (1979); Kumar & Dev (1983).

**Experimental***Crystal data*

$\text{C}_{14}\text{H}_{24}\text{OS}$

$M_r = 240.39$

| | |
|------------------------------|--|
| Triclinic, $P\bar{1}$ | $V = 1307.09 (4)\text{ \AA}^3$ |
| $a = 10.3662 (2)\text{ \AA}$ | $Z = 4$ |
| $b = 11.2090 (2)\text{ \AA}$ | Mo $K\alpha$ radiation |
| $c = 11.5026 (2)\text{ \AA}$ | $\mu = 0.23\text{ mm}^{-1}$ |
| $\alpha = 92.5786 (8)^\circ$ | $T = 90\text{ K}$ |
| $\beta = 101.7513 (8)^\circ$ | $0.18 \times 0.15 \times 0.10\text{ mm}$ |
| $\gamma = 90.2145 (8)^\circ$ | |

Data collection

| | |
|--|---|
| Nonius KappaCCD diffractometer | 32350 measured reflections |
| Absorption correction: multi-scan (<i>SCALEPACK</i> ; Otwinowski & Minor, 1997) | 5986 independent reflections |
| | 5000 reflections with $I > 2\sigma(I)$ |
| | $R_{\text{int}} = 0.037$ |
| | $T_{\text{min}} = 0.960$, $T_{\text{max}} = 0.978$ |

Refinement

| | |
|---------------------------------|---|
| $R[F^2 > 2\sigma(F^2)] = 0.034$ | 291 parameters |
| $wR(F^2) = 0.092$ | H-atom parameters constrained |
| $S = 1.05$ | $\Delta\rho_{\text{max}} = 0.40\text{ e \AA}^{-3}$ |
| 5986 reflections | $\Delta\rho_{\text{min}} = -0.34\text{ e \AA}^{-3}$ |

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97* and local procedures.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2746).

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supporting information

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1'-Ethylsulfanyl-1,1'-bicyclohexyl-2-one

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S1. Comment

Self condensation of cyclohexanone followed by *in situ* dehydration provides an isomeric mixture of products, specifically, 2-(1-cyclohexen-1-yl)cyclohexanone and cyclohexylidenecyclohexanone. In our efforts to obtain only cyclohexylidene cyclohexanone, an improved route was developed. The title compound, C₁₄H₂₄OS, was prepared through a Lewis acid mediated alkylation between 1,1-bis(ethylsulfanyl)cyclohexane and 1-trimethylsilyloxy cyclohexene at low temperature. The title compound can be oxidized with NaIO₄ and the corresponding cyclohexylidene cyclohexanone is produced as the only product in moderate yield.

The conformational energy for a cyclohexyl ring is 2.15 kcal/mol while the energy for ethanethiol is approximately 0.7 kcal/mol. From this information it is expected that both ring systems would be in an equatorial position leaving the thiol axial. The x-ray analysis provided agreement to our hypothesis.

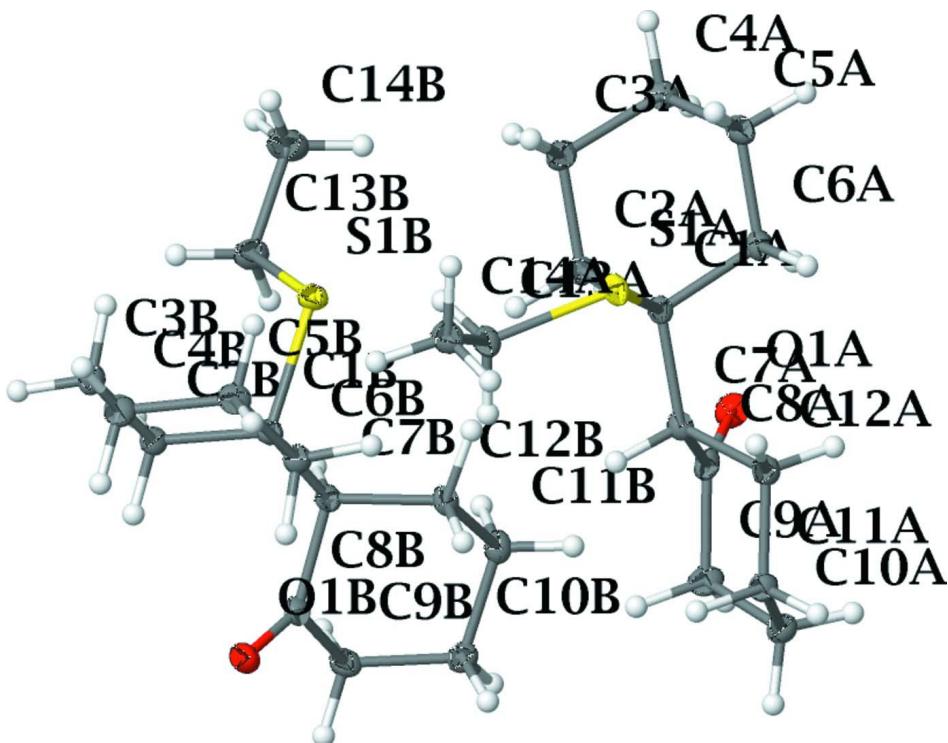
S2. Experimental

SnCl₄(10 ml, 1M in CH₂Cl₂, 10 mmol) was added to 20 ml of anhydrous CH₂Cl₂ at -60°C. A cooled (-60°C) solution of 1,1-bis(ethylsulfanyl)cyclohexane (2.04 g, 10 mmol) in 5 ml anhydrous CH₂Cl₂ was added dropwise. Immediately following the final addition of 1,1-bis(ethylsulfanyl)cyclohexane was slowly added a cooled (-60°C) solution of 1-trimethylsilyloxy cyclohexene (1.70 g, 10 mmol) in 5 ml anhydrous CH₂Cl₂. The solution stirred at -60°C for 45 min and was poured on to 100 ml of ice water. The aqueous phase was extracted with CH₂Cl₂(3 X 50 ml). The combined organic phases were washed with 10% aqueous NaHCO₃ (1 X 100 ml), water (1 X 100 ml) and dried over MgSO₄. The filtrate was concentrated under reduced pressure providing the title compound (2.05 g) as a white solid. Recrystallization from a solution of hexane-CH₂Cl₂ (3:1) provided 1.90 g (79% yield) of 1'-(ethylsulfanyl)-1,1'-bi(cyclohexyl)-2-one. mp = 72°C. ¹H NMR (CDCl₃, 500 MHz) δ : 2.58–2.52 (m, 1H); 2.47 (dd, J=11.7, 5 Hz, 1H); 2.37 (dq, J=7.3, 1 Hz, 2H); 2.34–2.24 (m, 2H); 2.06–1.88 (m, 4H); 1.84–1.74 (m, 1H); 1.74–1.56 (m, 6H); 1.56–1.48 (m, 1H); 1.48–1.40 (m, 2H); 1.32–1.22 (m, 1H); 1.22–1.16 (dt, J=7.5, 1.5 Hz, 3H). ¹³C NMR (CDCl₃, 125 MHz) δ: 212.2, 58.5, 52.2, 44.3, 32.6, 31.4, 29.9, 28.6, 25.9, 25.7, 22.0, 21.9, 21.1, 14.1. IR (ν_{max}): 2931, 1697, 1500, 1310, 1115, 1063, 884 cm⁻¹.

1,1-bis(ethylsulfanyl)cyclohexane was prepared by following the procedure of Kumar & Dev (1983) and 1-trimethylsilyloxy cyclohexene was prepared by following the procedure of Ito *et al.* (1979).

S3. Refinement

H atoms were found in difference Fourier maps and subsequently placed in idealized positions with constrained distances of 0.98 Å (RCH₃), 0.99 Å (R₂CH₂), 1.00 Å (R₃CH), and with U_{iso}(H) values set to either 1.2U_{eq} or 1.5U_{eq} (RCH₃) of the attached atom.

**Figure 1**

Displacement ellipsoids drawn at 50% probability level.

1'-Ethylsulfanyl-1,1'-bicyclohexyl-2-one

Crystal data

$C_{14}H_{24}OS$
 $M_r = 240.39$
Triclinic, $P\bar{1}$
Hall symbol: -P 1
 $a = 10.3662 (2)$ Å
 $b = 11.2090 (2)$ Å
 $c = 11.5026 (2)$ Å
 $\alpha = 92.5786 (8)^\circ$
 $\beta = 101.7513 (8)^\circ$
 $\gamma = 90.2145 (8)^\circ$
 $V = 1307.09 (4)$ Å³

$Z = 4$
 $F(000) = 528$
 $D_x = 1.222 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 5942 reflections
 $\theta = 1.0\text{--}27.4^\circ$
 $\mu = 0.23 \text{ mm}^{-1}$
 $T = 90$ K
Rod, colourless
 $0.18 \times 0.15 \times 0.10$ mm

Data collection

Nonius KappaCCD
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 9.1 pixels mm⁻¹
 ω scans at fixed $\chi = 55^\circ$
Absorption correction: multi-scan
(SCALEPACK; Otwinowski & Minor, 1997)
 $T_{\min} = 0.960$, $T_{\max} = 0.978$

32350 measured reflections
5986 independent reflections
5000 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.037$
 $\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 1.8^\circ$
 $h = -13 \rightarrow 13$
 $k = -14 \rightarrow 14$
 $l = -14 \rightarrow 14$

*Refinement*Refinement on F^2

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.034$$

$$wR(F^2) = 0.092$$

$$S = 1.05$$

5986 reflections

291 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0458P)^2 + 0.4279P]$$
$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.002$$

$$\Delta\rho_{\max} = 0.40 \text{ e \AA}^{-3}$$

$$\Delta\rho_{\min} = -0.34 \text{ e \AA}^{-3}$$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|--------------|---------------|----------------------------------|
| S1A | 0.88224 (3) | 0.94850 (3) | 0.30582 (3) | 0.01526 (9) |
| O1A | 0.53397 (9) | 0.80151 (9) | -0.01038 (8) | 0.0194 (2) |
| C1A | 0.72797 (12) | 0.89244 (11) | 0.20595 (11) | 0.0130 (2) |
| C2A | 0.68219 (13) | 0.77443 (11) | 0.24908 (11) | 0.0152 (3) |
| H2A1 | 0.6038 | 0.7436 | 0.1911 | 0.018* |
| H2A2 | 0.7529 | 0.7150 | 0.2511 | 0.018* |
| C3A | 0.64777 (14) | 0.78678 (12) | 0.37204 (12) | 0.0182 (3) |
| H3A1 | 0.7281 | 0.8086 | 0.4321 | 0.022* |
| H3A2 | 0.6142 | 0.7092 | 0.3924 | 0.022* |
| C4A | 0.54321 (14) | 0.88253 (12) | 0.37498 (12) | 0.0195 (3) |
| H4A1 | 0.4601 | 0.8572 | 0.3203 | 0.023* |
| H4A2 | 0.5255 | 0.8920 | 0.4563 | 0.023* |
| C5A | 0.59008 (13) | 1.00195 (12) | 0.33805 (12) | 0.0172 (3) |
| H5A1 | 0.6682 | 1.0309 | 0.3974 | 0.021* |
| H5A2 | 0.5196 | 1.0616 | 0.3367 | 0.021* |
| C6A | 0.62570 (13) | 0.99037 (12) | 0.21515 (11) | 0.0154 (3) |
| H6A1 | 0.6612 | 1.0679 | 0.1970 | 0.019* |
| H6A2 | 0.5446 | 0.9720 | 0.1547 | 0.019* |
| C7A | 0.75555 (12) | 0.87558 (11) | 0.07876 (11) | 0.0132 (2) |
| H7A | 0.8342 | 0.8230 | 0.0853 | 0.016* |
| C8A | 0.64834 (13) | 0.81697 (11) | -0.01800 (11) | 0.0153 (3) |
| C9A | 0.69592 (14) | 0.78561 (12) | -0.13183 (12) | 0.0188 (3) |
| H9A1 | 0.6224 | 0.7492 | -0.1917 | 0.023* |
| H9A2 | 0.7671 | 0.7263 | -0.1158 | 0.023* |
| C10A | 0.74760 (13) | 0.89752 (12) | -0.18129 (12) | 0.0176 (3) |

| | | | | |
|------|--------------|--------------|---------------|-------------|
| H10A | 0.7887 | 0.8737 | -0.2492 | 0.021* |
| H10B | 0.6731 | 0.9506 | -0.2105 | 0.021* |
| C11A | 0.84846 (13) | 0.96463 (12) | -0.08583 (11) | 0.0162 (3) |
| H11A | 0.8741 | 1.0401 | -0.1171 | 0.019* |
| H11B | 0.9284 | 0.9157 | -0.0647 | 0.019* |
| C12A | 0.79214 (13) | 0.99232 (11) | 0.02493 (12) | 0.0155 (3) |
| H12A | 0.7127 | 1.0419 | 0.0039 | 0.019* |
| H12B | 0.8580 | 1.0384 | 0.0847 | 0.019* |
| C13A | 0.99018 (13) | 0.82077 (12) | 0.31977 (12) | 0.0166 (3) |
| H13A | 0.9536 | 0.7569 | 0.3608 | 0.020* |
| H13B | 0.9986 | 0.7887 | 0.2401 | 0.020* |
| C14A | 1.12463 (13) | 0.86198 (13) | 0.39135 (12) | 0.0199 (3) |
| H14A | 1.1623 | 0.9216 | 0.3477 | 0.030* |
| H14B | 1.1834 | 0.7934 | 0.4040 | 0.030* |
| H14C | 1.1145 | 0.8974 | 0.4684 | 0.030* |
| S1B | 0.75895 (3) | 0.44991 (3) | 0.30868 (3) | 0.01604 (9) |
| O1B | 0.96037 (9) | 0.30177 (9) | -0.00856 (8) | 0.0195 (2) |
| C1B | 0.86605 (12) | 0.39287 (11) | 0.20787 (11) | 0.0134 (2) |
| C2B | 0.92957 (13) | 0.27433 (11) | 0.25006 (11) | 0.0155 (3) |
| H2B1 | 0.8590 | 0.2149 | 0.2513 | 0.019* |
| H2B2 | 0.9818 | 0.2439 | 0.1922 | 0.019* |
| C3B | 1.01917 (14) | 0.28581 (12) | 0.37345 (11) | 0.0181 (3) |
| H3B1 | 1.0607 | 0.2080 | 0.3936 | 0.022* |
| H3B2 | 0.9659 | 0.3077 | 0.4333 | 0.022* |
| C4B | 1.12643 (14) | 0.38092 (12) | 0.37728 (12) | 0.0198 (3) |
| H4B1 | 1.1803 | 0.3900 | 0.4588 | 0.024* |
| H4B2 | 1.1851 | 0.3552 | 0.3231 | 0.024* |
| C5B | 1.06533 (13) | 0.50073 (12) | 0.34018 (12) | 0.0171 (3) |
| H5B1 | 1.1364 | 0.5597 | 0.3390 | 0.020* |
| H5B2 | 1.0144 | 0.5303 | 0.3994 | 0.020* |
| C6B | 0.97415 (13) | 0.49000 (11) | 0.21687 (11) | 0.0153 (3) |
| H6B1 | 1.0276 | 0.4710 | 0.1565 | 0.018* |
| H6B2 | 0.9318 | 0.5679 | 0.1988 | 0.018* |
| C7B | 0.78001 (12) | 0.37651 (11) | 0.08113 (11) | 0.0137 (3) |
| H7B | 0.7038 | 0.3242 | 0.0880 | 0.016* |
| C8B | 0.84277 (13) | 0.31733 (11) | -0.01577 (11) | 0.0157 (3) |
| C9B | 0.74329 (14) | 0.28622 (12) | -0.12940 (12) | 0.0189 (3) |
| H9B1 | 0.6784 | 0.2275 | -0.1132 | 0.023* |
| H9B2 | 0.7891 | 0.2492 | -0.1892 | 0.023* |
| C10B | 0.67096 (13) | 0.39844 (12) | -0.17934 (12) | 0.0181 (3) |
| H10C | 0.7332 | 0.4512 | -0.2084 | 0.022* |
| H10D | 0.5991 | 0.3749 | -0.2474 | 0.022* |
| C11B | 0.61377 (13) | 0.46594 (12) | -0.08390 (12) | 0.0167 (3) |
| H11C | 0.5427 | 0.4173 | -0.0628 | 0.020* |
| H11D | 0.5749 | 0.5415 | -0.1153 | 0.020* |
| C12B | 0.72028 (13) | 0.49357 (11) | 0.02716 (12) | 0.0155 (3) |
| H12C | 0.6817 | 0.5397 | 0.0868 | 0.019* |
| H12D | 0.7908 | 0.5430 | 0.0063 | 0.019* |

| | | | | |
|------|--------------|--------------|--------------|------------|
| C13B | 0.65969 (14) | 0.32216 (12) | 0.32900 (12) | 0.0196 (3) |
| H13C | 0.6113 | 0.2890 | 0.2510 | 0.024* |
| H13D | 0.7167 | 0.2591 | 0.3687 | 0.024* |
| C14B | 0.56281 (15) | 0.36372 (14) | 0.40545 (14) | 0.0266 (3) |
| H14D | 0.6116 | 0.3972 | 0.4820 | 0.040* |
| H14E | 0.5088 | 0.2957 | 0.4188 | 0.040* |
| H14F | 0.5057 | 0.4250 | 0.3647 | 0.040* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|------|--------------|--------------|--------------|---------------|--------------|---------------|
| S1A | 0.01464 (16) | 0.01284 (16) | 0.01680 (16) | 0.00131 (12) | 0.00018 (12) | -0.00173 (12) |
| O1A | 0.0168 (5) | 0.0218 (5) | 0.0189 (5) | -0.0030 (4) | 0.0018 (4) | 0.0022 (4) |
| C1A | 0.0126 (6) | 0.0127 (6) | 0.0129 (6) | 0.0011 (5) | 0.0008 (5) | 0.0005 (5) |
| C2A | 0.0170 (6) | 0.0128 (6) | 0.0156 (6) | 0.0005 (5) | 0.0029 (5) | 0.0005 (5) |
| C3A | 0.0226 (7) | 0.0164 (7) | 0.0162 (6) | -0.0002 (5) | 0.0050 (5) | 0.0020 (5) |
| C4A | 0.0203 (7) | 0.0219 (7) | 0.0175 (7) | 0.0018 (5) | 0.0070 (5) | 0.0008 (5) |
| C5A | 0.0163 (6) | 0.0177 (7) | 0.0172 (6) | 0.0035 (5) | 0.0029 (5) | -0.0024 (5) |
| C6A | 0.0156 (6) | 0.0135 (6) | 0.0168 (6) | 0.0017 (5) | 0.0026 (5) | 0.0008 (5) |
| C7A | 0.0126 (6) | 0.0119 (6) | 0.0149 (6) | 0.0008 (5) | 0.0024 (5) | 0.0004 (5) |
| C8A | 0.0189 (7) | 0.0112 (6) | 0.0153 (6) | -0.0001 (5) | 0.0014 (5) | 0.0028 (5) |
| C9A | 0.0212 (7) | 0.0196 (7) | 0.0148 (6) | -0.0035 (5) | 0.0023 (5) | -0.0020 (5) |
| C10A | 0.0183 (7) | 0.0191 (7) | 0.0155 (6) | 0.0004 (5) | 0.0033 (5) | 0.0024 (5) |
| C11A | 0.0162 (6) | 0.0155 (6) | 0.0177 (6) | -0.0002 (5) | 0.0052 (5) | 0.0021 (5) |
| C12A | 0.0164 (6) | 0.0130 (6) | 0.0174 (6) | -0.0002 (5) | 0.0040 (5) | 0.0008 (5) |
| C13A | 0.0161 (6) | 0.0156 (6) | 0.0171 (6) | 0.0033 (5) | 0.0012 (5) | 0.0000 (5) |
| C14A | 0.0169 (7) | 0.0240 (7) | 0.0180 (7) | 0.0032 (5) | 0.0018 (5) | -0.0006 (5) |
| S1B | 0.02009 (17) | 0.01311 (16) | 0.01656 (17) | -0.00010 (12) | 0.00800 (13) | -0.00127 (12) |
| O1B | 0.0182 (5) | 0.0221 (5) | 0.0192 (5) | 0.0057 (4) | 0.0060 (4) | 0.0018 (4) |
| C1B | 0.0152 (6) | 0.0118 (6) | 0.0137 (6) | 0.0008 (5) | 0.0047 (5) | -0.0004 (5) |
| C2B | 0.0192 (6) | 0.0124 (6) | 0.0151 (6) | 0.0011 (5) | 0.0038 (5) | 0.0001 (5) |
| C3B | 0.0235 (7) | 0.0156 (7) | 0.0143 (6) | 0.0038 (5) | 0.0018 (5) | 0.0014 (5) |
| C4B | 0.0202 (7) | 0.0211 (7) | 0.0168 (6) | 0.0026 (5) | 0.0011 (5) | -0.0012 (5) |
| C5B | 0.0183 (7) | 0.0156 (6) | 0.0168 (6) | -0.0022 (5) | 0.0031 (5) | -0.0021 (5) |
| C6B | 0.0174 (6) | 0.0128 (6) | 0.0162 (6) | -0.0008 (5) | 0.0046 (5) | 0.0006 (5) |
| C7B | 0.0151 (6) | 0.0118 (6) | 0.0147 (6) | 0.0006 (5) | 0.0040 (5) | 0.0002 (5) |
| C8B | 0.0192 (7) | 0.0118 (6) | 0.0168 (6) | 0.0032 (5) | 0.0054 (5) | 0.0017 (5) |
| C9B | 0.0210 (7) | 0.0203 (7) | 0.0151 (6) | 0.0044 (5) | 0.0037 (5) | -0.0028 (5) |
| C10B | 0.0187 (7) | 0.0207 (7) | 0.0145 (6) | 0.0010 (5) | 0.0023 (5) | 0.0022 (5) |
| C11B | 0.0158 (6) | 0.0154 (6) | 0.0184 (6) | 0.0018 (5) | 0.0018 (5) | 0.0025 (5) |
| C12B | 0.0165 (6) | 0.0122 (6) | 0.0178 (6) | 0.0020 (5) | 0.0034 (5) | 0.0010 (5) |
| C13B | 0.0213 (7) | 0.0183 (7) | 0.0210 (7) | -0.0022 (5) | 0.0083 (6) | 0.0015 (5) |
| C14B | 0.0274 (8) | 0.0280 (8) | 0.0282 (8) | -0.0014 (6) | 0.0146 (6) | 0.0004 (6) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|----------|-------------|----------|-------------|
| S1A—C13A | 1.8130 (13) | S1B—C13B | 1.8113 (14) |
| S1A—C1A | 1.8577 (13) | S1B—C1B | 1.8571 (13) |

| | | | |
|--------------|-------------|--------------|-------------|
| O1A—C8A | 1.2189 (16) | O1B—C8B | 1.2186 (16) |
| C1A—C2A | 1.5399 (17) | C1B—C2B | 1.5392 (17) |
| C1A—C6A | 1.5440 (17) | C1B—C6B | 1.5444 (17) |
| C1A—C7A | 1.5504 (17) | C1B—C7B | 1.5490 (17) |
| C2A—C3A | 1.5281 (17) | C2B—C3B | 1.5295 (18) |
| C2A—H2A1 | 0.9900 | C2B—H2B1 | 0.9900 |
| C2A—H2A2 | 0.9900 | C2B—H2B2 | 0.9900 |
| C3A—C4A | 1.5324 (19) | C3B—C4B | 1.5303 (19) |
| C3A—H3A1 | 0.9900 | C3B—H3B1 | 0.9900 |
| C3A—H3A2 | 0.9900 | C3B—H3B2 | 0.9900 |
| C4A—C5A | 1.5293 (19) | C4B—C5B | 1.5279 (19) |
| C4A—H4A1 | 0.9900 | C4B—H4B1 | 0.9900 |
| C4A—H4A2 | 0.9900 | C4B—H4B2 | 0.9900 |
| C5A—C6A | 1.5323 (18) | C5B—C6B | 1.5354 (18) |
| C5A—H5A1 | 0.9900 | C5B—H5B1 | 0.9900 |
| C5A—H5A2 | 0.9900 | C5B—H5B2 | 0.9900 |
| C6A—H6A1 | 0.9900 | C6B—H6B1 | 0.9900 |
| C6A—H6A2 | 0.9900 | C6B—H6B2 | 0.9900 |
| C7A—C8A | 1.5265 (17) | C7B—C8B | 1.5290 (17) |
| C7A—C12A | 1.5507 (17) | C7B—C12B | 1.5535 (17) |
| C7A—H7A | 1.0000 | C7B—H7B | 1.0000 |
| C8A—C9A | 1.5179 (18) | C8B—C9B | 1.5168 (18) |
| C9A—C10A | 1.5387 (18) | C9B—C10B | 1.5398 (18) |
| C9A—H9A1 | 0.9900 | C9B—H9B1 | 0.9900 |
| C9A—H9A2 | 0.9900 | C9B—H9B2 | 0.9900 |
| C10A—C11A | 1.5221 (18) | C10B—C11B | 1.5242 (18) |
| C10A—H10A | 0.9900 | C10B—H10C | 0.9900 |
| C10A—H10B | 0.9900 | C10B—H10D | 0.9900 |
| C11A—C12A | 1.5262 (17) | C11B—C12B | 1.5278 (18) |
| C11A—H11A | 0.9900 | C11B—H11C | 0.9900 |
| C11A—H11B | 0.9900 | C11B—H11D | 0.9900 |
| C12A—H12A | 0.9900 | C12B—H12C | 0.9900 |
| C12A—H12B | 0.9900 | C12B—H12D | 0.9900 |
| C13A—C14A | 1.5256 (19) | C13B—C14B | 1.5249 (19) |
| C13A—H13A | 0.9900 | C13B—H13C | 0.9900 |
| C13A—H13B | 0.9900 | C13B—H13D | 0.9900 |
| C14A—H14A | 0.9800 | C14B—H14D | 0.9800 |
| C14A—H14B | 0.9800 | C14B—H14E | 0.9800 |
| C14A—H14C | 0.9800 | C14B—H14F | 0.9800 |
| | | | |
| C13A—S1A—C1A | 104.36 (6) | C13B—S1B—C1B | 104.78 (6) |
| C2A—C1A—C6A | 109.67 (10) | C2B—C1B—C6B | 109.61 (10) |
| C2A—C1A—C7A | 111.20 (10) | C2B—C1B—C7B | 111.00 (10) |
| C6A—C1A—C7A | 112.51 (10) | C6B—C1B—C7B | 112.67 (10) |
| C2A—C1A—S1A | 110.62 (8) | C2B—C1B—S1B | 110.81 (8) |
| C6A—C1A—S1A | 104.83 (8) | C6B—C1B—S1B | 104.68 (8) |
| C7A—C1A—S1A | 107.81 (8) | C7B—C1B—S1B | 107.89 (8) |
| C3A—C2A—C1A | 113.55 (10) | C3B—C2B—C1B | 113.36 (10) |

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| C3A—C2A—H2A1 | 108.9 | C3B—C2B—H2B1 | 108.9 |
| C1A—C2A—H2A1 | 108.9 | C1B—C2B—H2B1 | 108.9 |
| C3A—C2A—H2A2 | 108.9 | C3B—C2B—H2B2 | 108.9 |
| C1A—C2A—H2A2 | 108.9 | C1B—C2B—H2B2 | 108.9 |
| H2A1—C2A—H2A2 | 107.7 | H2B1—C2B—H2B2 | 107.7 |
| C2A—C3A—C4A | 110.79 (11) | C2B—C3B—C4B | 110.77 (11) |
| C2A—C3A—H3A1 | 109.5 | C2B—C3B—H3B1 | 109.5 |
| C4A—C3A—H3A1 | 109.5 | C4B—C3B—H3B1 | 109.5 |
| C2A—C3A—H3A2 | 109.5 | C2B—C3B—H3B2 | 109.5 |
| C4A—C3A—H3A2 | 109.5 | C4B—C3B—H3B2 | 109.5 |
| H3A1—C3A—H3A2 | 108.1 | H3B1—C3B—H3B2 | 108.1 |
| C5A—C4A—C3A | 110.49 (11) | C5B—C4B—C3B | 110.68 (11) |
| C5A—C4A—H4A1 | 109.6 | C5B—C4B—H4B1 | 109.5 |
| C3A—C4A—H4A1 | 109.6 | C3B—C4B—H4B1 | 109.5 |
| C5A—C4A—H4A2 | 109.6 | C5B—C4B—H4B2 | 109.5 |
| C3A—C4A—H4A2 | 109.6 | C3B—C4B—H4B2 | 109.5 |
| H4A1—C4A—H4A2 | 108.1 | H4B1—C4B—H4B2 | 108.1 |
| C4A—C5A—C6A | 111.50 (11) | C4B—C5B—C6B | 111.64 (11) |
| C4A—C5A—H5A1 | 109.3 | C4B—C5B—H5B1 | 109.3 |
| C6A—C5A—H5A1 | 109.3 | C6B—C5B—H5B1 | 109.3 |
| C4A—C5A—H5A2 | 109.3 | C4B—C5B—H5B2 | 109.3 |
| C6A—C5A—H5A2 | 109.3 | C6B—C5B—H5B2 | 109.3 |
| H5A1—C5A—H5A2 | 108.0 | H5B1—C5B—H5B2 | 108.0 |
| C5A—C6A—C1A | 113.11 (10) | C5B—C6B—C1B | 112.80 (10) |
| C5A—C6A—H6A1 | 109.0 | C5B—C6B—H6B1 | 109.0 |
| C1A—C6A—H6A1 | 109.0 | C1B—C6B—H6B1 | 109.0 |
| C5A—C6A—H6A2 | 109.0 | C5B—C6B—H6B2 | 109.0 |
| C1A—C6A—H6A2 | 109.0 | C1B—C6B—H6B2 | 109.0 |
| H6A1—C6A—H6A2 | 107.8 | H6B1—C6B—H6B2 | 107.8 |
| C8A—C7A—C1A | 118.12 (10) | C8B—C7B—C1B | 117.75 (10) |
| C8A—C7A—C12A | 104.50 (10) | C8B—C7B—C12B | 104.56 (10) |
| C1A—C7A—C12A | 114.47 (10) | C1B—C7B—C12B | 114.68 (10) |
| C8A—C7A—H7A | 106.3 | C8B—C7B—H7B | 106.4 |
| C1A—C7A—H7A | 106.3 | C1B—C7B—H7B | 106.4 |
| C12A—C7A—H7A | 106.3 | C12B—C7B—H7B | 106.4 |
| O1A—C8A—C9A | 121.75 (12) | O1B—C8B—C9B | 121.68 (12) |
| O1A—C8A—C7A | 125.31 (12) | O1B—C8B—C7B | 125.40 (12) |
| C9A—C8A—C7A | 112.83 (11) | C9B—C8B—C7B | 112.78 (11) |
| C8A—C9A—C10A | 110.85 (11) | C8B—C9B—C10B | 110.92 (11) |
| C8A—C9A—H9A1 | 109.5 | C8B—C9B—H9B1 | 109.5 |
| C10A—C9A—H9A1 | 109.5 | C10B—C9B—H9B1 | 109.5 |
| C8A—C9A—H9A2 | 109.5 | C8B—C9B—H9B2 | 109.5 |
| C10A—C9A—H9A2 | 109.5 | C10B—C9B—H9B2 | 109.5 |
| H9A1—C9A—H9A2 | 108.1 | H9B1—C9B—H9B2 | 108.0 |
| C11A—C10A—C9A | 110.79 (11) | C11B—C10B—C9B | 110.61 (11) |
| C11A—C10A—H10A | 109.5 | C11B—C10B—H10C | 109.5 |
| C9A—C10A—H10A | 109.5 | C9B—C10B—H10C | 109.5 |
| C11A—C10A—H10B | 109.5 | C11B—C10B—H10D | 109.5 |

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| C9A—C10A—H10B | 109.5 | C9B—C10B—H10D | 109.5 |
| H10A—C10A—H10B | 108.1 | H10C—C10B—H10D | 108.1 |
| C10A—C11A—C12A | 110.75 (11) | C10B—C11B—C12B | 110.84 (11) |
| C10A—C11A—H11A | 109.5 | C10B—C11B—H11C | 109.5 |
| C12A—C11A—H11A | 109.5 | C12B—C11B—H11C | 109.5 |
| C10A—C11A—H11B | 109.5 | C10B—C11B—H11D | 109.5 |
| C12A—C11A—H11B | 109.5 | C12B—C11B—H11D | 109.5 |
| H11A—C11A—H11B | 108.1 | H11C—C11B—H11D | 108.1 |
| C11A—C12A—C7A | 110.79 (10) | C11B—C12B—C7B | 110.76 (10) |
| C11A—C12A—H12A | 109.5 | C11B—C12B—H12C | 109.5 |
| C7A—C12A—H12A | 109.5 | C7B—C12B—H12C | 109.5 |
| C11A—C12A—H12B | 109.5 | C11B—C12B—H12D | 109.5 |
| C7A—C12A—H12B | 109.5 | C7B—C12B—H12D | 109.5 |
| H12A—C12A—H12B | 108.1 | H12C—C12B—H12D | 108.1 |
| C14A—C13A—S1A | 107.95 (9) | C14B—C13B—S1B | 107.97 (10) |
| C14A—C13A—H13A | 110.1 | C14B—C13B—H13C | 110.1 |
| S1A—C13A—H13A | 110.1 | S1B—C13B—H13C | 110.1 |
| C14A—C13A—H13B | 110.1 | C14B—C13B—H13D | 110.1 |
| S1A—C13A—H13B | 110.1 | S1B—C13B—H13D | 110.1 |
| H13A—C13A—H13B | 108.4 | H13C—C13B—H13D | 108.4 |
| C13A—C14A—H14A | 109.5 | C13B—C14B—H14D | 109.5 |
| C13A—C14A—H14B | 109.5 | C13B—C14B—H14E | 109.5 |
| H14A—C14A—H14B | 109.5 | H14D—C14B—H14E | 109.5 |
| C13A—C14A—H14C | 109.5 | C13B—C14B—H14F | 109.5 |
| H14A—C14A—H14C | 109.5 | H14D—C14B—H14F | 109.5 |
| H14B—C14A—H14C | 109.5 | H14E—C14B—H14F | 109.5 |
| | | | |
| C13A—S1A—C1A—C2A | 49.82 (10) | C13B—S1B—C1B—C2B | -47.41 (10) |
| C13A—S1A—C1A—C6A | 167.97 (8) | C13B—S1B—C1B—C6B | -165.49 (8) |
| C13A—S1A—C1A—C7A | -71.97 (9) | C13B—S1B—C1B—C7B | 74.30 (10) |
| C6A—C1A—C2A—C3A | -52.42 (14) | C6B—C1B—C2B—C3B | 53.18 (14) |
| C7A—C1A—C2A—C3A | -177.51 (11) | C7B—C1B—C2B—C3B | 178.28 (11) |
| S1A—C1A—C2A—C3A | 62.73 (12) | S1B—C1B—C2B—C3B | -61.86 (13) |
| C1A—C2A—C3A—C4A | 55.83 (15) | C1B—C2B—C3B—C4B | -56.10 (14) |
| C2A—C3A—C4A—C5A | -56.44 (15) | C2B—C3B—C4B—C5B | 56.19 (14) |
| C3A—C4A—C5A—C6A | 56.25 (14) | C3B—C4B—C5B—C6B | -55.90 (15) |
| C4A—C5A—C6A—C1A | -54.81 (15) | C4B—C5B—C6B—C1B | 54.72 (15) |
| C2A—C1A—C6A—C5A | 51.57 (14) | C2B—C1B—C6B—C5B | -52.08 (14) |
| C7A—C1A—C6A—C5A | 175.90 (10) | C7B—C1B—C6B—C5B | -176.21 (10) |
| S1A—C1A—C6A—C5A | -67.21 (12) | S1B—C1B—C6B—C5B | 66.82 (12) |
| C2A—C1A—C7A—C8A | 51.95 (15) | C2B—C1B—C7B—C8B | -51.91 (14) |
| C6A—C1A—C7A—C8A | -71.53 (14) | C6B—C1B—C7B—C8B | 71.45 (14) |
| S1A—C1A—C7A—C8A | 173.37 (9) | S1B—C1B—C7B—C8B | -173.49 (9) |
| C2A—C1A—C7A—C12A | 175.66 (10) | C2B—C1B—C7B—C12B | -175.58 (10) |
| C6A—C1A—C7A—C12A | 52.18 (14) | C6B—C1B—C7B—C12B | -52.22 (14) |
| S1A—C1A—C7A—C12A | -62.92 (12) | S1B—C1B—C7B—C12B | 62.84 (12) |
| C1A—C7A—C8A—O1A | 14.12 (19) | C1B—C7B—C8B—O1B | -14.47 (19) |
| C12A—C7A—C8A—O1A | -114.43 (14) | C12B—C7B—C8B—O1B | 114.15 (14) |

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| C1A—C7A—C8A—C9A | −169.72 (11) | C1B—C7B—C8B—C9B | 169.73 (11) |
| C12A—C7A—C8A—C9A | 61.73 (13) | C12B—C7B—C8B—C9B | −61.65 (13) |
| O1A—C8A—C9A—C10A | 118.09 (13) | O1B—C8B—C9B—C10B | −117.53 (14) |
| C7A—C8A—C9A—C10A | −58.22 (14) | C7B—C8B—C9B—C10B | 58.46 (14) |
| C8A—C9A—C10A—C11A | 51.80 (15) | C8B—C9B—C10B—C11B | −52.11 (15) |
| C9A—C10A—C11A—C12A | −53.81 (14) | C9B—C10B—C11B—C12B | 53.95 (14) |
| C10A—C11A—C12A—C7A | 61.17 (14) | C10B—C11B—C12B—C7B | −61.06 (14) |
| C8A—C7A—C12A—C11A | −62.58 (13) | C8B—C7B—C12B—C11B | 62.32 (13) |
| C1A—C7A—C12A—C11A | 166.68 (10) | C1B—C7B—C12B—C11B | −167.23 (10) |
| C1A—S1A—C13A—C14A | 174.74 (9) | C1B—S1B—C13B—C14B | −176.72 (10) |