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4-(1,3-Benzothiazol-2-yl)-5-methyl-2-phenyl-1-propynyl-1*H*-pyrazol-3(2*H*)-one

 Imane Chakib,^a Abdelfettah Zerzouf,^a Hafid Zouihri,^a El Mokhtar Essassi^a and Seik Weng Ng^{b*}

^aLaboratoire de Chimie Organique Hétérocyclique, Pôle de Compétences Pharmacochimie, Université Mohammed V-Agdal, BP 1014 Avenue Ibn Batout, Rabat, Morocco, and ^bDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

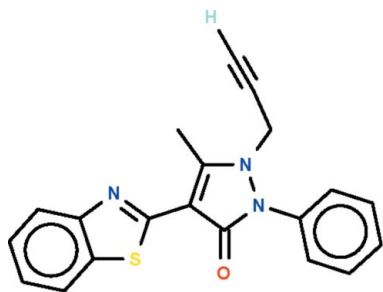
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 Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.042; wR factor = 0.105; data-to-parameter ratio = 16.3.

The title compound, $\text{C}_{20}\text{H}_{15}\text{N}_3\text{OS}$, is a 1*H*-pyrazol-3(2*H*)-one having aromatic 4-(1,3-benzothiazol-2-yl) and 2-phenyl substituents. The five-membered ring and the fused-ring system are close to planar, the r.m.s. deviations being 0.025 and 0.005 Å, respectively. The five-membered ring is aligned at 67.5 (1)° with respect to the phenyl ring and at 4.7 (1)° with respect to the fused-ring system. In the crystal, adjacent molecules are linked through the acetylenic H atom by a C—H···O hydrogen bond into $C(8)$ chains propagating in [010].

Related literature

For the structure of a similar compound, 4-(benzo[*d*]thiazol-2-yl)-2-allyl-3-methyl-1-phenyl-1,2-dihydropyrazol-5-one, see: Chakibe *et al.* (2010). For the structure of a related compound, (*E*)-4-(2,3-dihydro-1,3-benzothiazol-2-ylidene)-3-methyl-1-phenyl-1*H*-pyrazol-5(4*H*)-one, see: Chakib *et al.* (2010).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{15}\text{N}_3\text{OS}$
 $M_r = 345.41$
 Orthorhombic, $P2_12_12_1$
 $a = 4.8221$ (1) Å
 $b = 9.3698$ (2) Å
 $c = 37.6990$ (9) Å
 $V = 1703.32$ (6) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.20$ mm⁻¹
 $T = 293$ K
 $0.40 \times 0.20 \times 0.20$ mm

Data collection

Bruker X8 APEXII diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.923$, $T_{\max} = 0.961$
 10502 measured reflections
 3699 independent reflections
 3008 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.105$
 $S = 0.99$
 3699 reflections
 227 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.19$ e Å⁻³
 Absolute structure: Flack (1983),
 1483 Friedel pairs
 Flack parameter: 0.00 (9)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C14}-\text{H14}\cdots\text{O1}^i$	0.93	2.24	3.174 (3)	179

 Symmetry code: (i) $x, y + 1, z$.

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5048).

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supporting information

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4-(1,3-Benzothiazol-2-yl)-5-methyl-2-phenyl-1-propynyl-1*H*-pyrazol-3(2*H*)-one

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S1. Comment

(*E*)-4-(2,3-Dihydro-1,3-benzothiazol-2-ylidene)-3-methyl-1-phenyl-1*H*-pyrazol-5(4*H*)-one is an amine that can under a nucleophilic substitution with organo bromides to form 2-substituted derivatives if tetra-*n*-butyl ammonium bromide is used as catalyst. In this study, the compound is reacted with propargyl bromide to yield the title compound (Scheme I, Fig. 1).

S2. Experimental

To a solution of (*E*)-4-(2,3-dihydro-1,3-benzothiazol-2-ylidene)-3-methyl-1-phenyl-1*H*-pyrazol-5(4*H*)-one (1 g, 3.25 mmol) in DMF (50 ml), was added sodium carbonate (2.5 g, 23 mmol), tetra-*n*-butyl ammonium bromide (0.15 g, 1 mmol) and propargyl bromide (5.5 g, 46 mmol). The mixture was stirred for 24 h at room temperature. The solid material was removed by filtration and the solution was evaporated under reduced. The residue was washed with dichloromethane and hexane, and the recrystallized from ethanol to afford the title compound as yellow crystals.

S3. Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.93–0.97 Å) and were included in the refinement in the riding model approximation, with $U(\text{H})$ set to 1.2–1.5 $U_{\text{eq}}(\text{C})$.

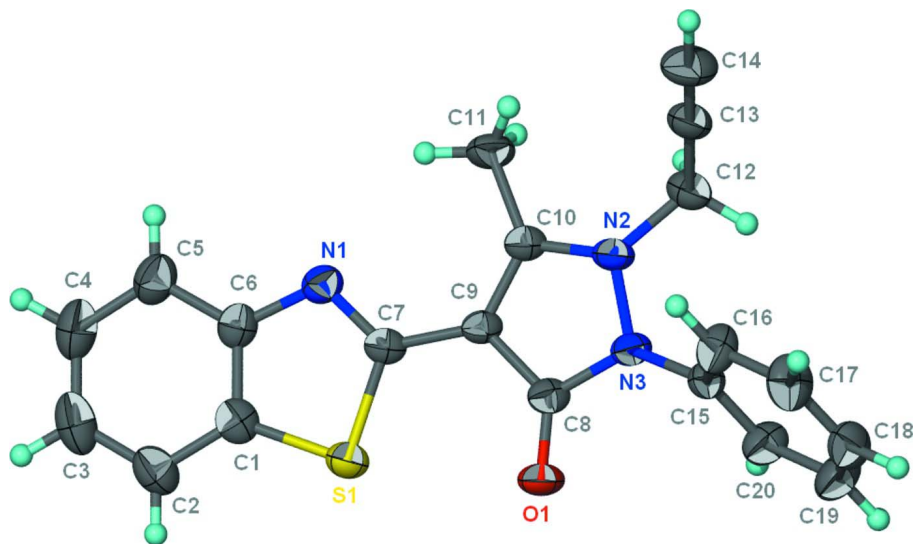


Figure 1

Thermal ellipsoid plot (Barbour, 2001) of $\text{C}_{20}\text{H}_{15}\text{N}_3\text{OS}$ at the 50% probability level; hydrogen atoms are drawn as arbitrary radius.

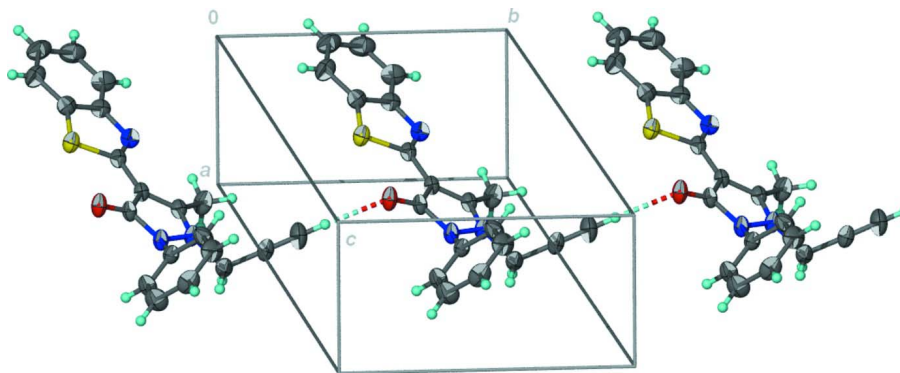


Figure 2

Hydrogen-bonded chain structure.

4-(1,3-Benzothiazol-2-yl)-5-methyl-2-phenyl-1-propynyl-1*H*-pyrazol-3(2*H*)-one*Crystal data* $C_{20}H_{15}N_3OS$ $M_r = 345.41$ Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

 $a = 4.8221$ (1) Å $b = 9.3698$ (2) Å $c = 37.6990$ (9) Å $V = 1703.32$ (6) Å³ $Z = 4$ $F(000) = 720$ $D_x = 1.347$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2503 reflections

 $\theta = 2.4$ – 22.9° $\mu = 0.20$ mm⁻¹ $T = 293$ K

Prism, yellow

 $0.40 \times 0.20 \times 0.20$ mm*Data collection*Bruker X8 APEXII
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 φ and ω scans

Absorption correction: multi-scan

(SADABS; Sheldrick, 1996)

 $T_{\min} = 0.923$, $T_{\max} = 0.961$

10502 measured reflections

3699 independent reflections

3008 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.033$ $\theta_{\max} = 27.1^\circ$, $\theta_{\min} = 3.1^\circ$ $h = -5 \rightarrow 6$ $k = -11 \rightarrow 11$ $l = -47 \rightarrow 46$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.105$ $S = 0.99$

3699 reflections

227 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0613P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.20$ e Å⁻³ $\Delta\rho_{\min} = -0.19$ e Å⁻³Absolute structure: Flack (1983), 1483 Friedel
pairs

Absolute structure parameter: 0.00 (9)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.55911 (13)	0.45298 (6)	0.105599 (16)	0.04655 (17)
O1	0.9129 (4)	0.52382 (15)	0.16533 (4)	0.0487 (4)
N1	0.5947 (4)	0.6723 (2)	0.06358 (5)	0.0445 (4)
N2	1.1641 (4)	0.85264 (18)	0.14034 (5)	0.0422 (4)
N3	1.1570 (4)	0.73580 (18)	0.16370 (5)	0.0408 (4)
C1	0.3603 (5)	0.4536 (3)	0.06743 (6)	0.0469 (5)
C2	0.1735 (6)	0.3514 (3)	0.05524 (7)	0.0599 (7)
H2	0.1422	0.2680	0.0680	0.072*
C3	0.0366 (6)	0.3767 (3)	0.02398 (8)	0.0681 (8)
H3	-0.0894	0.3097	0.0155	0.082*
C4	0.0833 (6)	0.4998 (3)	0.00504 (8)	0.0683 (8)
H4	-0.0121	0.5142	-0.0161	0.082*
C5	0.2662 (5)	0.6015 (3)	0.01643 (7)	0.0593 (7)
H5	0.2961	0.6841	0.0033	0.071*
C6	0.4073 (5)	0.5783 (3)	0.04845 (6)	0.0446 (5)
C7	0.6919 (5)	0.6195 (2)	0.09303 (6)	0.0376 (5)
C8	0.9769 (5)	0.6346 (2)	0.14978 (5)	0.0386 (5)
C9	0.8913 (4)	0.6897 (2)	0.11590 (5)	0.0351 (4)
C10	1.0182 (4)	0.8196 (2)	0.11109 (5)	0.0386 (5)
C11	1.0201 (6)	0.9159 (3)	0.07987 (6)	0.0558 (7)
H11A	0.9830	1.0119	0.0874	0.084*
H11B	0.8800	0.8861	0.0634	0.084*
H11C	1.1985	0.9120	0.0686	0.084*
C12	1.3789 (5)	0.9610 (2)	0.14413 (7)	0.0479 (6)
H12A	1.4965	0.9371	0.1641	0.057*
H12B	1.4934	0.9620	0.1230	0.057*
C13	1.2600 (5)	1.1024 (3)	0.14956 (6)	0.0494 (6)
C14	1.1643 (7)	1.2142 (3)	0.15376 (8)	0.0657 (8)
H14	1.0873	1.3042	0.1571	0.079*
C15	1.2100 (5)	0.7561 (2)	0.20074 (5)	0.0383 (5)
C16	1.0709 (5)	0.8595 (3)	0.21937 (7)	0.0534 (6)
H16	0.9444	0.9189	0.2080	0.064*
C17	1.1218 (6)	0.8741 (3)	0.25519 (7)	0.0625 (7)
H17	1.0293	0.9440	0.2681	0.075*
C18	1.3064 (6)	0.7867 (3)	0.27184 (7)	0.0588 (7)
H18	1.3389	0.7973	0.2960	0.071*
C19	1.4434 (6)	0.6842 (3)	0.25322 (7)	0.0629 (7)
H19	1.5693	0.6248	0.2647	0.075*
C20	1.3951 (5)	0.6682 (3)	0.21707 (6)	0.0505 (6)
H20	1.4880	0.5983	0.2042	0.061*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0594 (3)	0.0375 (3)	0.0427 (3)	-0.0076 (3)	0.0002 (3)	0.0022 (2)

O1	0.0764 (10)	0.0299 (8)	0.0397 (9)	-0.0053 (8)	-0.0008 (9)	0.0053 (6)
N1	0.0509 (11)	0.0449 (11)	0.0376 (10)	-0.0009 (9)	0.0005 (9)	0.0036 (8)
N2	0.0573 (11)	0.0309 (9)	0.0386 (11)	-0.0056 (8)	0.0001 (9)	0.0042 (8)
N3	0.0598 (12)	0.0303 (9)	0.0323 (10)	-0.0002 (8)	-0.0029 (9)	0.0021 (8)
C1	0.0480 (12)	0.0497 (13)	0.0431 (13)	0.0012 (11)	0.0078 (10)	-0.0093 (11)
C2	0.0620 (15)	0.0589 (17)	0.0587 (17)	-0.0136 (13)	0.0045 (13)	-0.0111 (13)
C3	0.0521 (14)	0.086 (2)	0.0662 (19)	-0.0125 (15)	0.0019 (14)	-0.0292 (17)
C4	0.0551 (15)	0.096 (2)	0.0535 (17)	0.0027 (16)	-0.0097 (13)	-0.0122 (15)
C5	0.0586 (15)	0.0776 (19)	0.0418 (15)	-0.0006 (14)	-0.0034 (12)	0.0016 (13)
C6	0.0435 (12)	0.0544 (14)	0.0358 (12)	0.0030 (11)	0.0021 (10)	-0.0053 (10)
C7	0.0442 (11)	0.0346 (11)	0.0340 (12)	0.0014 (9)	0.0085 (9)	-0.0004 (9)
C8	0.0528 (13)	0.0297 (10)	0.0334 (12)	0.0042 (9)	0.0042 (9)	-0.0043 (8)
C9	0.0456 (11)	0.0301 (10)	0.0297 (10)	0.0026 (9)	0.0057 (9)	0.0008 (8)
C10	0.0488 (12)	0.0333 (10)	0.0336 (12)	0.0023 (8)	0.0052 (9)	0.0002 (9)
C11	0.0780 (17)	0.0472 (14)	0.0422 (14)	-0.0119 (12)	0.0012 (12)	0.0132 (10)
C12	0.0470 (12)	0.0391 (12)	0.0577 (15)	-0.0027 (10)	0.0044 (11)	0.0016 (11)
C13	0.0636 (15)	0.0378 (13)	0.0467 (15)	-0.0098 (12)	0.0002 (12)	0.0017 (10)
C14	0.093 (2)	0.0388 (14)	0.0652 (18)	0.0028 (14)	0.0063 (16)	0.0001 (12)
C15	0.0451 (12)	0.0360 (12)	0.0339 (12)	-0.0026 (9)	-0.0016 (9)	-0.0021 (9)
C16	0.0570 (13)	0.0545 (15)	0.0487 (14)	0.0157 (12)	-0.0105 (12)	-0.0128 (11)
C17	0.0676 (16)	0.0679 (18)	0.0519 (16)	0.0054 (14)	-0.0002 (14)	-0.0231 (13)
C18	0.0733 (17)	0.0678 (18)	0.0353 (14)	-0.0125 (15)	-0.0049 (13)	-0.0038 (12)
C19	0.0728 (17)	0.0678 (17)	0.0479 (15)	0.0121 (15)	-0.0133 (14)	0.0063 (13)
C20	0.0628 (15)	0.0459 (13)	0.0428 (13)	0.0117 (12)	-0.0012 (12)	-0.0030 (11)

Geometric parameters (Å, °)

S1—C1	1.729 (2)	C9—C10	1.374 (3)
S1—C7	1.752 (2)	C10—C11	1.483 (3)
O1—C8	1.232 (2)	C11—H11A	0.9600
N1—C7	1.303 (3)	C11—H11B	0.9600
N1—C6	1.385 (3)	C11—H11C	0.9600
N2—C10	1.344 (3)	C12—C13	1.458 (3)
N2—N3	1.405 (2)	C12—H12A	0.9700
N2—C12	1.458 (3)	C12—H12B	0.9700
N3—C8	1.389 (3)	C13—C14	1.156 (4)
N3—C15	1.433 (3)	C14—H14	0.9300
C1—C6	1.389 (3)	C15—C20	1.362 (3)
C1—C2	1.393 (3)	C15—C16	1.372 (3)
C2—C3	1.371 (4)	C16—C17	1.379 (3)
C2—H2	0.9300	C16—H16	0.9300
C3—C4	1.375 (4)	C17—C18	1.362 (4)
C3—H3	0.9300	C17—H17	0.9300
C4—C5	1.368 (4)	C18—C19	1.361 (4)
C4—H4	0.9300	C18—H18	0.9300
C5—C6	1.403 (3)	C19—C20	1.391 (3)
C5—H5	0.9300	C19—H19	0.9300
C7—C9	1.449 (3)	C20—H20	0.9300

C8—C9	1.438 (3)		
C1—S1—C7	88.55 (12)	N2—C10—C9	109.20 (18)
C7—N1—C6	110.14 (19)	N2—C10—C11	120.51 (19)
C10—N2—N3	108.77 (17)	C9—C10—C11	130.3 (2)
C10—N2—C12	127.79 (18)	C10—C11—H11A	109.5
N3—N2—C12	119.92 (19)	C10—C11—H11B	109.5
C8—N3—N2	108.09 (17)	H11A—C11—H11B	109.5
C8—N3—C15	124.80 (18)	C10—C11—H11C	109.5
N2—N3—C15	120.19 (17)	H11A—C11—H11C	109.5
C6—C1—C2	120.9 (2)	H11B—C11—H11C	109.5
C6—C1—S1	109.94 (18)	C13—C12—N2	111.6 (2)
C2—C1—S1	129.1 (2)	C13—C12—H12A	109.3
C3—C2—C1	118.5 (3)	N2—C12—H12A	109.3
C3—C2—H2	120.8	C13—C12—H12B	109.3
C1—C2—H2	120.8	N2—C12—H12B	109.3
C2—C3—C4	120.8 (3)	H12A—C12—H12B	108.0
C2—C3—H3	119.6	C14—C13—C12	179.6 (3)
C4—C3—H3	119.6	C13—C14—H14	180.0
C5—C4—C3	121.9 (3)	C20—C15—C16	121.1 (2)
C5—C4—H4	119.1	C20—C15—N3	118.5 (2)
C3—C4—H4	119.1	C16—C15—N3	120.4 (2)
C4—C5—C6	118.3 (3)	C15—C16—C17	118.9 (2)
C4—C5—H5	120.8	C15—C16—H16	120.5
C6—C5—H5	120.8	C17—C16—H16	120.5
N1—C6—C1	115.4 (2)	C18—C17—C16	120.5 (3)
N1—C6—C5	124.9 (2)	C18—C17—H17	119.7
C1—C6—C5	119.7 (2)	C16—C17—H17	119.7
N1—C7—C9	125.0 (2)	C19—C18—C17	120.2 (2)
N1—C7—S1	115.93 (17)	C19—C18—H18	119.9
C9—C7—S1	119.08 (16)	C17—C18—H18	119.9
O1—C8—N3	123.5 (2)	C18—C19—C20	120.0 (3)
O1—C8—C9	130.8 (2)	C18—C19—H19	120.0
N3—C8—C9	105.66 (18)	C20—C19—H19	120.0
C10—C9—C8	107.89 (19)	C15—C20—C19	119.2 (2)
C10—C9—C7	128.25 (19)	C15—C20—H20	120.4
C8—C9—C7	123.76 (19)	C19—C20—H20	120.4
C10—N2—N3—C8	-6.4 (2)	O1—C8—C9—C7	2.1 (4)
C12—N2—N3—C8	-166.84 (19)	N3—C8—C9—C7	-177.20 (19)
C10—N2—N3—C15	-158.63 (19)	N1—C7—C9—C10	-1.5 (4)
C12—N2—N3—C15	41.0 (3)	S1—C7—C9—C10	-179.60 (17)
C7—S1—C1—C6	-0.38 (17)	N1—C7—C9—C8	174.4 (2)
C7—S1—C1—C2	-179.8 (2)	S1—C7—C9—C8	-3.7 (3)
C6—C1—C2—C3	0.0 (4)	N3—N2—C10—C9	6.0 (2)
S1—C1—C2—C3	179.4 (2)	C12—N2—C10—C9	164.5 (2)
C1—C2—C3—C4	0.2 (4)	N3—N2—C10—C11	-172.1 (2)
C2—C3—C4—C5	-0.1 (4)	C12—N2—C10—C11	-13.7 (3)

C3—C4—C5—C6	-0.3 (4)	C8—C9—C10—N2	-3.4 (2)
C7—N1—C6—C1	1.1 (3)	C7—C9—C10—N2	173.0 (2)
C7—N1—C6—C5	-179.4 (2)	C8—C9—C10—C11	174.5 (2)
C2—C1—C6—N1	179.1 (2)	C7—C9—C10—C11	-9.1 (4)
S1—C1—C6—N1	-0.3 (2)	C10—N2—C12—C13	81.7 (3)
C2—C1—C6—C5	-0.4 (3)	N3—N2—C12—C13	-122.0 (2)
S1—C1—C6—C5	-179.83 (18)	C8—N3—C15—C20	81.6 (3)
C4—C5—C6—N1	-178.9 (2)	N2—N3—C15—C20	-131.0 (2)
C4—C5—C6—C1	0.5 (4)	C8—N3—C15—C16	-96.6 (3)
C6—N1—C7—C9	-179.6 (2)	N2—N3—C15—C16	50.7 (3)
C6—N1—C7—S1	-1.4 (2)	C20—C15—C16—C17	0.2 (4)
C1—S1—C7—N1	1.08 (18)	N3—C15—C16—C17	178.4 (2)
C1—S1—C7—C9	179.36 (17)	C15—C16—C17—C18	-0.3 (4)
N2—N3—C8—O1	-175.2 (2)	C16—C17—C18—C19	0.1 (4)
C15—N3—C8—O1	-24.6 (3)	C17—C18—C19—C20	0.0 (4)
N2—N3—C8—C9	4.2 (2)	C16—C15—C20—C19	-0.1 (4)
C15—N3—C8—C9	154.8 (2)	N3—C15—C20—C19	-178.3 (2)
O1—C8—C9—C10	178.7 (2)	C18—C19—C20—C15	0.0 (4)
N3—C8—C9—C10	-0.6 (2)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C14—H14 \cdots O1 ⁱ	0.93	2.24	3.174 (3)	179

Symmetry code: (i) *x*, *y*+1, *z*.