

1-(2-Chlorophenyl)-6-fluoro-2-methyl-1*H*-indole-3-carbonitrile**Kun Yang,^{a*} Pei-Fan Li,^{b*} Yan Liu^b and Zhi-Zhong Fang^a**

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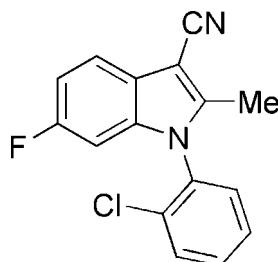
Received 24 March 2011; accepted 25 March 2011

Key indicators: single-crystal X-ray study; $T = 113\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$; R factor = 0.041; wR factor = 0.113; data-to-parameter ratio = 21.4.

In the title compound, $\text{C}_{16}\text{H}_{10}\text{ClFN}_2$, the dihedral angle between the indole ring system and the benzyl ring is $80.91(5)^\circ$. The crystal packing features $\text{C}-\text{H}\cdots\text{Cl}$, $\text{C}-\text{H}\cdots\text{F}$ and $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For the synthesis of the title compound, see: Du *et al.* (2006). For its precursor, see: Jin *et al.* (2009). For related structures, see: Li & Huang (2009); Li *et al.* (2009, 2010a,b).

**Experimental***Crystal data*

$\text{C}_{16}\text{H}_{10}\text{ClFN}_2$
 $M_r = 284.71$

Orthorhombic, $Pbca$
 $a = 7.4581(9)\text{ \AA}$

$b = 16.8480(15)\text{ \AA}$
 $c = 21.356(2)\text{ \AA}$
 $V = 2683.5(5)\text{ \AA}^3$
 $Z = 8$

Mo $K\alpha$ radiation
 $\mu = 0.29\text{ mm}^{-1}$
 $T = 113\text{ K}$
 $0.26 \times 0.22 \times 0.20\text{ mm}$

Data collection

Rigaku Saturn724 CCD
diffractometer
Absorption correction: multi-scan
(*CrystalClear*, Rigaku, 2009)
 $T_{\min} = 0.929$, $T_{\max} = 0.945$

28426 measured reflections
3893 independent reflections
3219 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.113$
 $S = 1.11$
3893 reflections

182 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.39\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.20\text{ e \AA}^{-3}$

Table 1Hydrogen-bond geometry (\AA , $^\circ$). $Cg1$ is the centroid of the C3–C8 benzene ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$C7-\text{H7}\cdots\text{F1}^i$	0.95	2.54	3.1638 (16)	123
$C7-\text{H7}\cdots\text{Cl1}^{ii}$	0.95	2.73	3.5296 (14)	142
$C15-\text{H15}\cdots Cg1^{iii}$	0.95	2.92	3.7246 (14)	143

Symmetry codes: (i) $x + \frac{1}{2}, y, -z + \frac{3}{2}$; (ii) $-x + \frac{3}{2}, y + \frac{1}{2}, z$; (iii) $x - 1, y, z$.

Data collection: *CrystalClear* (Rigaku, 2009); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2009); software used to prepare material for publication: *CrystalStructure*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5823).

References

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supporting information

Acta Cryst. (2011). E67, o1041 [doi:10.1107/S1600536811011214]

1-(2-Chlorophenyl)-6-fluoro-2-methyl-1*H*-indole-3-carbonitrile

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S1. Comment

Indoles are an important compound possessing pharmaceutical properties. Extensive investigation on the crystal structures of indoles helps disclose their structure-activity relationship. For continuing our research, herein, we reported the crystal structure of the title indole derivative.

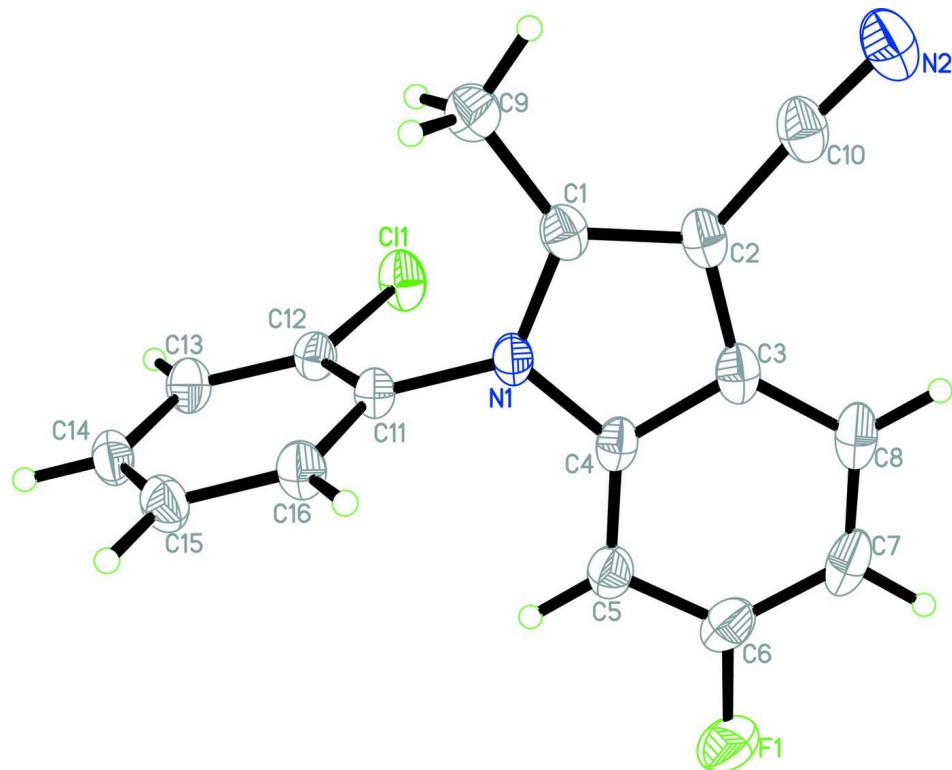
In the molecular structure, (I) (Fig. 1), the indole ring system is almost planar with a dihedral angle of $0.85(6)^\circ$ between its pyrrole ring and fused benzene ring. The indole ring forms an angle of $80.91(5)^\circ$ with the chlorobenzene ring.

S2. Experimental

The title compound was prepared according to the method of the literature (Du, *et al.*, 2006). Colourless prisms were grown from a mixture of ethyl acetate and petroleum ether.

S3. Refinement

All H atoms were positioned geometrically ($C—H = 0.95$ and 0.98 \AA) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{CH})$ or $1.5U_{\text{eq}}(\text{CH}_3)$.

**Figure 1**

The molecular structure of (I) with 50% probability displacement ellipsoids.

1-(2-Chlorophenyl)-6-fluoro-2-methyl-1*H*-indole-3-carbonitrile

Crystal data



$M_r = 284.71$

Orthorhombic, $Pbca$

Hall symbol: -P 2ac 2ab

$a = 7.4581(9)$ Å

$b = 16.8480(15)$ Å

$c = 21.356(2)$ Å

$V = 2683.5(5)$ Å³

$Z = 8$

$F(000) = 1168$

$D_x = 1.409 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71075$ Å

Cell parameters from 10281 reflections

$\theta = 1.5\text{--}31.4^\circ$

$\mu = 0.29 \text{ mm}^{-1}$

$T = 113$ K

Prism, colorless

$0.26 \times 0.22 \times 0.20$ mm

Data collection

Rigaku Saturn724 CCD
diffractometer

Radiation source: rotating anode

Multilayer monochromator

Detector resolution: 14.222 pixels mm⁻¹
 ω scans

Absorption correction: multi-scan
(*CrystalClear*; Rigaku, 2009)

$T_{\min} = 0.929$, $T_{\max} = 0.945$

28426 measured reflections

3893 independent reflections

3219 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.034$

$\theta_{\max} = 30.0^\circ$, $\theta_{\min} = 1.9^\circ$

$h = -10 \rightarrow 9$

$k = -23 \rightarrow 23$

$l = -30 \rightarrow 30$

*Refinement*Refinement on F^2

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.041$$

$$wR(F^2) = 0.113$$

$$S = 1.11$$

3893 reflections

182 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0644P)^2 + 0.1004P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 0.39 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.20 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.50520 (4)	0.220294 (19)	0.641958 (16)	0.03226 (11)
F1	0.61121 (11)	0.54670 (5)	0.76153 (4)	0.0480 (2)
N1	0.39260 (12)	0.37712 (5)	0.59488 (4)	0.0236 (2)
N2	0.79477 (16)	0.43384 (7)	0.42798 (6)	0.0420 (3)
C1	0.45553 (17)	0.36559 (7)	0.53489 (5)	0.0258 (2)
C2	0.60108 (16)	0.41469 (7)	0.52619 (6)	0.0271 (3)
C3	0.63160 (15)	0.45816 (7)	0.58328 (6)	0.0265 (3)
C4	0.49938 (14)	0.43272 (7)	0.62558 (6)	0.0237 (2)
C5	0.48747 (16)	0.46061 (7)	0.68659 (6)	0.0280 (3)
H5	0.3986	0.4427	0.7151	0.034*
C6	0.61408 (17)	0.51607 (7)	0.70243 (6)	0.0337 (3)
C7	0.74735 (17)	0.54359 (8)	0.66235 (7)	0.0361 (3)
H7	0.8315	0.5819	0.6765	0.043*
C8	0.75667 (16)	0.51503 (7)	0.60210 (7)	0.0318 (3)
H8	0.8460	0.5336	0.5740	0.038*
C9	0.36953 (19)	0.30813 (8)	0.49124 (6)	0.0333 (3)
H9A	0.3812	0.2543	0.5081	0.040*
H9B	0.4286	0.3109	0.4503	0.040*
H9C	0.2423	0.3213	0.4866	0.040*
C10	0.70533 (17)	0.42350 (7)	0.47083 (6)	0.0317 (3)
C11	0.25337 (16)	0.33290 (7)	0.62467 (5)	0.0223 (2)
C12	0.28935 (15)	0.25756 (7)	0.64802 (5)	0.0230 (2)
C13	0.15588 (16)	0.21330 (7)	0.67690 (5)	0.0271 (3)
H13	0.1800	0.1612	0.6918	0.032*
C14	-0.01309 (16)	0.24642 (8)	0.68363 (6)	0.0277 (3)

H14	-0.1052	0.2169	0.7037	0.033*
C15	-0.04958 (18)	0.32238 (7)	0.66141 (6)	0.0295 (3)
H15	-0.1658	0.3446	0.6664	0.035*
C16	0.08411 (16)	0.36557 (7)	0.63194 (6)	0.0276 (3)
H16	0.0598	0.4175	0.6167	0.033*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.02051 (19)	0.02941 (18)	0.0468 (2)	0.00492 (11)	0.00001 (12)	0.00824 (12)
F1	0.0450 (5)	0.0454 (5)	0.0535 (5)	-0.0100 (4)	-0.0018 (4)	-0.0201 (4)
N1	0.0198 (5)	0.0213 (4)	0.0298 (5)	-0.0031 (4)	0.0030 (4)	0.0022 (4)
N2	0.0363 (7)	0.0436 (7)	0.0461 (6)	0.0006 (5)	0.0128 (5)	0.0089 (5)
C1	0.0240 (6)	0.0243 (5)	0.0290 (5)	0.0016 (5)	0.0019 (5)	0.0054 (4)
C2	0.0218 (6)	0.0250 (5)	0.0344 (6)	0.0014 (4)	0.0040 (5)	0.0079 (5)
C3	0.0199 (6)	0.0210 (5)	0.0385 (6)	0.0020 (4)	0.0007 (5)	0.0081 (4)
C4	0.0178 (6)	0.0181 (5)	0.0353 (6)	0.0001 (4)	-0.0013 (4)	0.0036 (4)
C5	0.0234 (6)	0.0230 (6)	0.0376 (6)	-0.0001 (4)	0.0013 (5)	0.0006 (5)
C6	0.0311 (7)	0.0253 (6)	0.0446 (7)	0.0007 (5)	-0.0046 (6)	-0.0057 (5)
C7	0.0267 (7)	0.0220 (6)	0.0597 (8)	-0.0050 (5)	-0.0061 (6)	0.0016 (6)
C8	0.0199 (6)	0.0225 (6)	0.0528 (7)	-0.0007 (4)	0.0011 (6)	0.0100 (5)
C9	0.0349 (7)	0.0345 (7)	0.0304 (6)	-0.0037 (6)	0.0011 (5)	0.0007 (5)
C10	0.0259 (7)	0.0291 (6)	0.0402 (6)	0.0026 (5)	0.0047 (5)	0.0088 (5)
C11	0.0194 (6)	0.0216 (5)	0.0259 (5)	-0.0029 (4)	0.0004 (4)	0.0022 (4)
C12	0.0188 (6)	0.0235 (5)	0.0267 (5)	0.0008 (4)	-0.0021 (4)	0.0014 (4)
C13	0.0263 (6)	0.0255 (5)	0.0295 (5)	-0.0022 (5)	-0.0015 (5)	0.0070 (4)
C14	0.0230 (6)	0.0324 (6)	0.0275 (5)	-0.0069 (5)	0.0020 (5)	0.0031 (5)
C15	0.0199 (6)	0.0318 (6)	0.0369 (6)	0.0006 (5)	0.0032 (5)	-0.0002 (5)
C16	0.0229 (6)	0.0241 (6)	0.0358 (6)	0.0015 (5)	0.0009 (5)	0.0041 (5)

Geometric parameters (\AA , ^\circ)

Cl1—C12	1.7328 (12)	C7—C8	1.375 (2)
F1—C6	1.3636 (15)	C7—H7	0.9500
N1—C1	1.3780 (15)	C8—H8	0.9500
N1—C4	1.3934 (14)	C9—H9A	0.9800
N1—C11	1.4276 (14)	C9—H9B	0.9800
N2—C10	1.1457 (16)	C9—H9C	0.9800
C1—C2	1.3775 (16)	C11—C16	1.3857 (16)
C1—C9	1.4892 (17)	C11—C12	1.3900 (15)
C2—C10	1.4227 (17)	C12—C13	1.3883 (16)
C2—C3	1.4404 (18)	C13—C14	1.3857 (17)
C3—C8	1.3963 (17)	C13—H13	0.9500
C3—C4	1.4045 (16)	C14—C15	1.3918 (18)
C4—C5	1.3879 (18)	C14—H14	0.9500
C5—C6	1.3708 (17)	C15—C16	1.3855 (17)
C5—H5	0.9500	C15—H15	0.9500
C6—C7	1.3914 (19)	C16—H16	0.9500

C1—N1—C4	109.73 (9)	C3—C8—H8	120.8
C1—N1—C11	126.05 (10)	C1—C9—H9A	109.5
C4—N1—C11	123.83 (9)	C1—C9—H9B	109.5
C2—C1—N1	108.01 (10)	H9A—C9—H9B	109.5
C2—C1—C9	130.19 (11)	C1—C9—H9C	109.5
N1—C1—C9	121.80 (11)	H9A—C9—H9C	109.5
C1—C2—C10	127.27 (12)	H9B—C9—H9C	109.5
C1—C2—C3	108.40 (10)	N2—C10—C2	176.12 (15)
C10—C2—C3	124.33 (11)	C16—C11—C12	119.89 (10)
C8—C3—C4	119.56 (12)	C16—C11—N1	120.35 (10)
C8—C3—C2	134.28 (11)	C12—C11—N1	119.73 (10)
C4—C3—C2	106.16 (10)	C13—C12—C11	120.76 (11)
C5—C4—N1	129.26 (11)	C13—C12—Cl1	120.31 (9)
C5—C4—C3	123.06 (11)	C11—C12—Cl1	118.91 (9)
N1—C4—C3	107.68 (11)	C14—C13—C12	118.81 (11)
C6—C5—C4	114.73 (11)	C14—C13—H13	120.6
C6—C5—H5	122.6	C12—C13—H13	120.6
C4—C5—H5	122.6	C13—C14—C15	120.85 (11)
F1—C6—C5	118.39 (12)	C13—C14—H14	119.6
F1—C6—C7	117.04 (11)	C15—C14—H14	119.6
C5—C6—C7	124.57 (13)	C16—C15—C14	119.80 (12)
C8—C7—C6	119.68 (12)	C16—C15—H15	120.1
C8—C7—H7	120.2	C14—C15—H15	120.1
C6—C7—H7	120.2	C15—C16—C11	119.86 (11)
C7—C8—C3	118.39 (12)	C15—C16—H16	120.1
C7—C8—H8	120.8	C11—C16—H16	120.1
C4—N1—C1—C2	-1.08 (13)	C4—C5—C6—C7	-0.48 (18)
C11—N1—C1—C2	-174.11 (10)	F1—C6—C7—C8	-179.81 (11)
C4—N1—C1—C9	178.91 (11)	C5—C6—C7—C8	0.4 (2)
C11—N1—C1—C9	5.89 (17)	C6—C7—C8—C3	-0.63 (18)
N1—C1—C2—C10	-178.61 (11)	C4—C3—C8—C7	0.91 (17)
C9—C1—C2—C10	1.4 (2)	C2—C3—C8—C7	-178.87 (13)
N1—C1—C2—C3	0.59 (13)	C1—C2—C10—N2	177 (100)
C9—C1—C2—C3	-179.40 (12)	C3—C2—C10—N2	-2 (2)
C1—C2—C3—C8	179.90 (13)	C1—N1—C11—C16	-104.36 (13)
C10—C2—C3—C8	-0.9 (2)	C4—N1—C11—C16	83.56 (14)
C1—C2—C3—C4	0.10 (13)	C1—N1—C11—C12	77.42 (15)
C10—C2—C3—C4	179.33 (11)	C4—N1—C11—C12	-94.66 (13)
C1—N1—C4—C5	-178.37 (12)	C16—C11—C12—C13	2.09 (17)
C11—N1—C4—C5	-5.16 (18)	N1—C11—C12—C13	-179.69 (10)
C1—N1—C4—C3	1.15 (13)	C16—C11—C12—Cl1	-176.62 (9)
C11—N1—C4—C3	174.36 (10)	N1—C11—C12—Cl1	1.61 (15)
C8—C3—C4—C5	-1.03 (17)	C11—C12—C13—C14	-1.76 (17)
C2—C3—C4—C5	178.81 (11)	Cl1—C12—C13—C14	176.93 (9)
C8—C3—C4—N1	179.41 (10)	C12—C13—C14—C15	0.59 (18)
C2—C3—C4—N1	-0.75 (12)	C13—C14—C15—C16	0.24 (19)

N1—C4—C5—C6	−179.77 (11)	C14—C15—C16—C11	0.07 (19)
C3—C4—C5—C6	0.78 (17)	C12—C11—C16—C15	−1.22 (17)
C4—C5—C6—F1	179.77 (11)	N1—C11—C16—C15	−179.44 (11)

Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C3—C8 benzene ring.

D—H···A	D—H	H···A	D···A	D—H···A
C7—H7···F1 ⁱ	0.95	2.54	3.1638 (16)	123
C7—H7···Cl1 ⁱⁱ	0.95	2.73	3.5296 (14)	142
C15—H15···Cg1 ⁱⁱⁱ	0.95	2.92	3.7246 (14)	143

Symmetry codes: (i) $x+1/2, y, -z+3/2$; (ii) $-x+3/2, y+1/2, z$; (iii) $x-1, y, z$.