Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

# Poly[[tetrakis(*µ*-1,2-di-4-pyridylethylene- $\kappa^2 N: N'$ )tetrakis(selenocvanato- $\kappa N$ )dimanganese(II)] 1,2-di-4-pyridylethylene disolvate]

#### Susanne Wöhlert,\* Inke Jess and Christian Näther

Institut für Anorganische Chemie, Christian-Albrechts-Universität Kiel, Max-Eyth-Strasse 2, 24118 Kiel, Germany Correspondence e-mail: swoehlert@ac.uni-kiel.de

Received 26 April 2011; accepted 28 April 2011

Key indicators: single-crystal X-ray study; T = 293 K; mean  $\sigma$ (C–C) = 0.007 Å; R factor = 0.056; wR factor = 0.124; data-to-parameter ratio = 14.5.

The crystal structure of the title compound, {[Mn(NCSe)<sub>2</sub>- $(C_{12}H_{10}N_2)_2$  ·  $C_{12}H_{10}N_2$  · independent Mn cations, four selenocyanate anions, four 1,2di-pyridylethylene (bpe) ligands (two of which are located on centers of inversion) and two bpe solvent molecules. Each manganese(II) cation is coordinated by two terminally Nbonded selenocyanate anions and four bpe ligands within a slightly distorted octahedron. The manganese(II) cations are linked by the bpe ligands into chains that are further connected by these ligands into layers. These layers are stacked, forming cavities in which additional bpe solvent molecules are embedded.

#### **Related literature**

For background to this work, see: Boeckmann & Näther (2010); Wriedt et al. (2009); Wriedt & Näther (2010). For a description of the Cambridge Crystallographic Database, see: Allen (2002).



 $\beta = 91.211 \ (2)^{\circ}$ 

Z = 4

V = 7219.6 (2) Å<sup>3</sup>

Mo  $K\alpha$  radiation

 $0.19 \times 0.13 \times 0.08 \text{ mm}$ 

68086 measured reflections

12826 independent reflections

10467 reflections with  $I > 2\sigma(I)$ 

 $\mu = 2.42 \text{ mm}^{-1}$ 

T = 293 K

 $R_{\rm int} = 0.054$ 

### **Experimental**

#### Crystal data

$[Mn_2(NCSe)_4(C_{12}H_{10}N_2)_4]$
$2C_{12}H_{10}N_2$
$M_r = 1623.12$
Monoclinic, $P2/c$
a = 14.3102 (3) Å
b = 13.9933 (2) Å
c = 36.0614 (6) Å

#### Data collection

```
Stoe IPDS-2 diffractometer
Absorption correction: numerical
  (X-SHAPE and X-RED32; Stoe
  & Cie, 2008)
  T_{\min} = 0.689, T_{\max} = 0.853
```

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.056$	883 parameters
$wR(F^2) = 0.124$	H-atom parameters constrained
S = 1.10	$\Delta \rho_{\rm max} = 0.79 \ {\rm e} \ {\rm \AA}^{-3}$
12826 reflections	$\Delta \rho_{\rm min} = -0.78 \ {\rm e} \ {\rm \AA}^{-3}$

#### Table 1

Selected bond lengths (Å).

Mn2-N16	2.172 (4)	Mn1-N14	2.176 (3)
Mn2-N17	2.185 (3)	Mn1-N13	2.183 (4)
Mn2-N41	2.317 (3)	Mn1-N21	2.324 (3)
Mn2-N121	2.322 (3)	Mn1-N1	2.338 (3)
Mn2-N22	2.325 (3)	Mn1-N81	2.339 (3)

Data collection: X-AREA (Stoe & Cie, 2008); cell refinement: X-AREA; data reduction: X-AREA; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XP in SHELXTL (Sheldrick, 2008); software used to prepare material for publication: XP in SHELXTLand DIAMOND (Brandenburg, 2011).

# metal-organic compounds

We gratefully acknowledge financial support by the DFG (project No. NA 720/3-1) and the State of Schleswig-Holstein. We thank Professor Dr Bensch for access to his experimental facilities.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5535).

#### References

Allen, F. H. (2002). Acta Cryst. B58, 380-388.

Boeckmann, J. & Näther, C. (2010). Dalton Trans. pp. 1119-1126.

Brandenburg, K. (2011). DIAMOND. Crystal Impact GbR, Bonn, Germany.

Sheldrick, G. M. (2008). Acta Cryst. A64, 112–122. Stoe & Cie (2008). X-AREA, X-RED32 and X-SHAPE. Stoe & Cie, Darmstadt, Germany.

Wriedt, M., Jess, I. & Näther, C. (2009). Eur. J. Inorg. Chem., pp. 1406-1413. Wriedt, M. & Näther, C. (2010). Chem. Commun. 46, 4707-4709.

# supporting information

Acta Cryst. (2011). E67, m693-m694 [doi:10.1107/S1600536811016199]

# Poly[[tetrakis( $\mu$ -1,2-di-4-pyridylethylene- $\kappa^2 N$ :N')tetrakis(selenocyanato- $\kappa N$ )dimanganese(II)] 1,2-di-4-pyridylethylene disolvate]

# Susanne Wöhlert, Inke Jess and Christian Näther

## S1. Comment

In our recent work on the synthesis, structures and properties of new coordination polymers based on paramagnetic transition metals, thiocyanato or selenocyanato anions and *N*-donor ligands we have shown that ligand-deficient compounds can be prepared by thermal decomposition reactions (Wriedt *et al.*, 2009, Boeckmann & Näther, 2010 and Wriedt & Näther, 2010). Within this project we tried to prepare new ligand-rich precursor compounds based on manganese(II) selenocyanato and the *N*-donor ligand 1,2-bis(4-pyridyl)-ethylene (bpe) which resulted in the formation of the title compound that were identified by single crystal X-ray diffraction.

In the crystal structure of the title compound each managnese(II) cation is coordinated by two terminally *N*-bonded selenocyanato anions and four bridging bpe ligands (Fig. 1). The MnN<sub>6</sub> octahedra are slightly distorted with distances in the range of 2.172 (4) Å to 2.339 (3) Å. The angles around the metal cations are in the range of 86.94 (12)  $^{\circ}$  to 179.89 (16)  $^{\circ}$ . The manganese(II) cations are linked by the bpe ligands into chains that are further connected into layers by additional coligands (Fig. 2). The layers are stacked in order that cavities are formed, in which non-coordinated bpe molecules are located (Fig. 3). The Mn—Mn distances within the chains are in range of 13.9666 (8) Å to 13.9933 (8) Å.

It must be noted that according to a search in the CCDC database (ConQuest Ver.1.12.2010; Allen, 2002) compounds based on manganese(II) selenocyanate and 1,2-bis(4-pyridyl)-ethylene are unknown.

## S2. Experimental

Manganese chloride tetrahydrate (MnCl<sub>2</sub> × 4H<sub>2</sub>O) and 1,2-bis(4-pyridyl)-ethylene (bpe) were obtained from sigma aldrich. Potassium selenocyanate (KNCSe) and acetonitrile (MeCN) were obtained from alfa aesar. All chemicals were used without further purification. 0.15 mmol (25.4 mg) MnCl<sub>2</sub> x 4H<sub>2</sub>O, 0.3 mmol (43 mg) KNCSe and 0.6 mmol (107.5 mg) bpe were reacted with 1 ml MeCN in a closed test-tube at 120 ° C for three days. On cooling orange block-shaped single crystals of the title compound were obtained.

## S3. Refinement

H atoms were positioned with idealized geometry and were refined isotropically with  $U_{iso}(H) = 1.2 U_{eq}(C)$  and C—H distances of 0.93 Å using a riding model.



# Figure 1

Crystal structure of the title compound showing the Mn coordination (top) and the noncoordinating bpe ligands (bottom) with labelling and displacement ellipsoids drawn at the 50% probability level. [Symmetry codes: (I) x, y+1, z; (ii) -x+2, y, -z+1/2; (iii) -x+1, y, -z-1/2.]



# Figure 2

Crystal structure of the title compound with view onto the layers. The noncoordinated bpe ligands have been omitted for clearity.



### Figure 3

Crystal structure of the title compound with view approximately along the crystallographic c axis.

# Poly[[tetrakis( $\mu$ -1,2-di-4-pyridylethylene- $\kappa^2 N:N'$ )tetrakis(selenocyanato- $\kappa N$ )dimanganese(II)] 1,2-di-4-pyridylethylene disolvate]

#### Crystal data

$[Mn_{2}(NCSe)_{4}(C_{12}H_{10}N_{2})_{4}] \cdot 2C_{12}H_{10}N_{2}$ $M_{r} = 1623.12$ Monoclinic, $P2/c$ Hall symbol: -P 2yc a = 14.3102 (3) Å b = 13.9933 (2) Å c = 36.0614 (6) Å $\beta = 91.211$ (2)° V = 7219.6 (2) Å <sup>3</sup> Z = 4	P(000) = 3256 $D_x = 1.493 \text{ Mg m}^{-3}$ Mo K $\alpha$ radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 68086 reflections $\theta = 1.1-25.2^{\circ}$ $\mu = 2.42 \text{ mm}^{-1}$ T = 293  K Block, orange $0.19 \times 0.13 \times 0.08 \text{ mm}$
Data collection Stoe IPDS-2 diffractometer Radiation source: fine-focus sealed tube Graphite monochromator $\omega$ scans Absorption correction: numerical (X-SHAPE and X-RED32; Stoe & Cie, 2008) Twin = 0.689. Twar = 0.853	68086 measured reflections 12826 independent reflections 10467 reflections with $I > 2\sigma(I)$ $R_{int} = 0.054$ $\theta_{max} = 25.2^{\circ}, \theta_{min} = 1.1^{\circ}$ $h = -17 \rightarrow 17$ $k = -16 \rightarrow 16$ $l = -43 \rightarrow 43$

Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.056$	Hydrogen site location: inferred from
$wR(F^2) = 0.124$	neighbouring sites
S = 1.10	H-atom parameters constrained
12826 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0431P)^2 + 9.7881P]$
883 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.001$
Primary atom site location: structure-invariant	$\Delta  ho_{ m max} = 0.79 \ { m e} \ { m \AA}^{-3}$
direct methods	$\Delta  ho_{ m min} = -0.78 \  m e \  m \AA^{-3}$

#### Special details

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F<sup>2</sup>, conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2$ sigma( $F^2$ ) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F<sup>2</sup> are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(Å^2)$ 

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
Mn2	0.62315 (4)	0.01645 (4)	0.122255 (15)	0.03700 (14)
Mn1	-0.12571 (4)	0.21901 (4)	-0.119724 (14)	0.03640 (14)
N1	-0.1268 (2)	0.0521 (2)	-0.12307 (8)	0.0420 (7)
C1	-0.1725 (3)	0.0065 (3)	-0.15087 (11)	0.0464 (9)
H1	-0.2060	0.0428	-0.1682	0.056*
C2	-0.1722 (3)	-0.0915 (3)	-0.15496 (11)	0.0514 (10)
H2	-0.2047	-0.1195	-0.1748	0.062*
C3	-0.1239 (3)	-0.1482 (3)	-0.12975 (11)	0.0445 (9)
C4	-0.0762 (3)	-0.1013 (3)	-0.10109 (11)	0.0473 (9)
H4	-0.0421	-0.1360	-0.0835	0.057*
C5	-0.0797 (3)	-0.0031 (3)	-0.09895 (11)	0.0472 (9)
Н5	-0.0474	0.0266	-0.0795	0.057*
C6	-0.1261 (3)	-0.2528 (3)	-0.13367 (12)	0.0525 (10)
H6	-0.1363	-0.2774	-0.1574	0.063*
C7	-0.1148 (3)	-0.3143 (3)	-0.10635 (12)	0.0525 (10)
H7	-0.1037	-0.2894	-0.0828	0.063*
C8	-0.1182 (3)	-0.4190 (3)	-0.10983 (11)	0.0467 (9)
C9	-0.1231 (3)	-0.4765 (3)	-0.07876 (11)	0.0540 (11)
H9	-0.1237	-0.4492	-0.0552	0.065*
C10	-0.1271 (3)	-0.5740 (3)	-0.08270 (11)	0.0527 (10)
H10	-0.1311	-0.6113	-0.0614	0.063*
C11	-0.1207 (3)	-0.5623 (3)	-0.14530 (12)	0.0549 (11)
H11	-0.1197	-0.5915	-0.1685	0.066*
C12	-0.1169 (3)	-0.4645 (3)	-0.14405 (11)	0.0537 (11)
H12	-0.1135	-0.4291	-0.1658	0.064*

N2	-0.1256 (2)	-0.6180 (2)	-0.11562 (9)	0.0483 (8)
N21	0.0030(2)	0.2156 (2)	-0.07933 (8)	0.0422 (7)
C21	0.0906 (3)	0.2202 (4)	-0.09136 (12)	0.0589 (11)
H21	0.0994	0.2310	-0.1165	0.071*
C22	0.1680 (3)	0.2100 (4)	-0.06881 (13)	0.0675 (13)
H22	0.2271	0.2160	-0.0788	0.081*
C23	0.1600 (3)	0.1911 (3)	-0.03208(13)	0.0558 (11)
C24	0.0698 (3)	0.1901 (4)	-0.01895 (12)	0.0641 (13)
H24	0.0600	0.1812	0.0062	0.077*
C25	-0.0056(3)	0.2024 (3)	-0.04298(11)	0.0549 (11)
H25	-0.0653	0.2014	-0.0333	0.066*
C26	0.2465 (4)	0.1687 (4)	-0.01021(14)	0.0691 (13)
H26	0.3028	0.1894	-0.0198	0.083*
C27	0.2501 (4)	0.1240 (4)	0.02013 (13)	0.0701 (13)
H27	0.1935	0.1084	0.0308	0.084*
C28	0 3371 (3)	0.0938(3)	0.04045(13)	0.0563 (11)
C29	0.3289(3)	0.0599(5)	0.07585(14)	0.0851 (18)
H29	0.2701	0.0552	0.0862	0.102*
C30	0.2761 0.4062(3)	0.0326(4)	0.09612(13)	0.0737(15)
H30	0.3978	0.0108	0.1202	0.088*
C31	0.5007 (3)	0.0635(3)	0.04844(11)	0.0541 (11)
H31	0.5595	0.0626	0.0380	0.065*
C32	0.4251(3)	0.0941(3)	0.02680(12)	0.002 0.0625(12)
H32	0.4347	0.1151	0.0027	0.0025 (12)
N22	0.4927(2)	0.0356(2)	0.0027 0.08342 (8)	0.073 0.0438(7)
N41	0.6187(2)	-0.1488(2)	0.11906(9)	0.0433(7)
C41	0.5692(3)	-0.1929(3)	0.09229(11)	0.0472(9)
H41	0.5347	-0.1556	0.0756	0.057*
C42	0.5664 (3)	-0.2906(3)	0.08807(11)	0.057
H42	0.5309	-0.3174	0.0688	0.062*
C43	0.6161(3)	-0.3492(3)	0.11222 (11)	0.002
C44	0.6675(3)	-0.3037(3)	0.14042(12)	0.0519(10)
H44	0.7022	-0 3396	0.1575	0.062*
C45	0.6666 (3)	-0.2057(3)	0.14279(12)	0.002
H45	0 7010	-0.1770	0.1619	0.060*
C46	0.6144 (3)	-0.4537(3)	0 10791 (12)	0.0527(10)
H46	0.6020	-0.4780	0.0843	0.063*
C47	0.6291 (3)	-0.5155(3)	0.13501 (12)	0.0527 (10)
H47	0.6416	-0 4907	0.1585	0.063*
C48	0.6277(3)	-0.6200(3)	0 13131 (11)	0.0464 (9)
C49	0.6207(3)	-0.6768(3)	0.16241(12)	0.0559(11)
H49	0.6161	-0.6492	0.1858	0.067*
C50	0.6205 (3)	-0.7749(3)	0.15851 (11)	0.0530(11)
H50	0.6145	-0.8118	0.1798	0.064*
C51	0.6363 (3)	-0.7641(3)	0.09618 (11)	0.0526 (10)
H51	0.6431	-0.7935	0.0733	0.063*
C52	0.6351 (3)	-0.6657 (3)	0.09734 (11)	0.0514 (10)
H52	0.6393	-0.6303	0.0756	0.062*
-	· · · · <del>·</del>	· · · · · · · · ·		

N42	0.6283 (2)	-0.8201 (2)	0.12618 (9)	0.0463 (8)
N61	0.8742 (4)	0.4435 (5)	0.01294 (14)	0.1056 (18)
C61	0.8781 (6)	0.5184 (5)	0.0353 (2)	0.111 (2)
H61	0.8820	0.5785	0.0245	0.133*
C62	0.8768 (5)	0.5137 (5)	0.07329 (16)	0.0887 (17)
H62	0.8790	0.5696	0.0872	0.106*
C63	0.8722 (3)	0.4262 (4)	0.09082 (13)	0.0657 (13)
C64	0.8668 (4)	0.3469 (4)	0.06801 (14)	0.0739 (14)
H64	0.8628	0.2858	0.0780	0.089*
C65	0.8673 (5)	0.3602 (5)	0.02991 (16)	0.0937 (19)
H65	0.8625	0.3060	0.0151	0.112*
C66	0.8729 (3)	0.4229 (4)	0.13149 (14)	0.0674 (13)
H66	0.8765	0.4813	0.1438	0.081*
C67	0.8690 (4)	0.3472 (4)	0.15213 (14)	0.0709 (13)
H67	0.8654	0.2886	0.1400	0.085*
C68	0.8698 (4)	0.3450 (4)	0.19293(13)	0.0708 (14)
C69	0.8767 (5)	0.3130(1) 0.4244(5)	0.21498(16)	0.100(2)
H69	0.8812	0.4847	0 2043	0.120*
C70	0.8771 (6)	0.4151 (6)	0.25317(18)	0.116 (3)
е70 Н70	0.8818	0.4705	0.2673	0.139*
C71	0.8637 (6)	0.2587 (6)	0.2675 0.24944 (19)	0.119(3)
H71	0.8588	0 1994	0.2609	0.143*
C72	0.8522 (6)	0.2603 (5)	0.21140 (16)	0.103(2)
U72 H72	0.8559	0.2005 (5)	0.1982	0.124*
N62	0.8559 0.8713 (4)	0.2030 0.3331 (5)	0.1702 0.27070 (14)	0.124 0.1031 (17)
N81	-0.2528(2)	0.3331(3) 0.2278(2)	-0.16142(9)	0.1031(17) 0.0448(8)
C81	-0.3403(3)	0.2276(2) 0.2265(4)	-0.14929(13)	0.0440(0) 0.0637(13)
H81	-0.3490	0.2250	-0.1238	0.0057 (15)
C82	-0.4178(3)	0.2230 0.2272(4)	-0.17205(14)	0.0758(15)
H82	-0.4768	0.2272 (1)	-0.1617	0.091*
C83	-0.4103(3)	0.2209	-0.20947(14)	0.051 0.0634(12)
C84	-0.3201(4)	0.2204(4) 0.2293(4)	-0.220947(14)	0.0635(12)
H84	-0.3104	0.2295 (4)	-0.22205(12)	0.0055 (12)
C85	-0.2444(3)	0.2295(3)	-0.19821(11)	0.070
H85	-0.1846	0.2299 (3)	-0.2078	0.065*
C86	-0.4971(4)	0.2309 0.2283 (4)	-0.23282(13)	0.0800 (16)
H86	-0 5532	0.2283	-0.2202 (13)	0.096*
N101	-0.3906(4)	-0.4425(5)	0.00920 (14)	0.0989 (16)
C101	-0.3648(5)	-0.5210(6)	-0.00920(14)	0.0000(10)
H101	-0 3497	-0.5746	0.0057	0.120*
C102	-0.3589(4)	-0.5281(5)	-0.04643(17)	0.0925 (18)
H102	-0.3421	-0.5856	-0.0573	0.111*
C103	-0.3780(4)	-0.4503(5)	-0.06783(15)	0.0788 (16)
C104	-0.4056(5)	-0.3687(5)	-0.04989(17)	0.0963 (10)
H104	-0.4208	-0.3142	-0.0635	0.0903 (19)
C105	-0.4100 (5)	-0.3675 (6)	-0.01181.(18)	0.110
H105	-0.4295	-0.3113	-0.0003	0.103 (2)
C106	-0 3716 (A)	-0.4604 (6)	-0.10878(17)	0.123
0100	0.3/10(4)	0.7007(0)	0.100/0(1/)	0.097 (4)

H106	-0.3609	-0.5218	-0.1176	0.116*
C107	-0.3786 (4)	-0.3983 (6)	-0.13192 (19)	0.105 (2)
H107	-0.3873	-0.3365	-0.1232	0.126*
C108	-0.3749 (4)	-0.4100 (6)	-0.17381 (17)	0.096 (2)
C109	-0.3663 (6)	-0.4953 (6)	-0.1922 (2)	0.122 (3)
H109	-0.3608	-0.5520	-0.1788	0.146*
C110	-0.3657 (7)	-0.4969 (6)	-0.2306 (2)	0.135 (3)
H110	-0.3599	-0.5558	-0.2422	0.161*
C111	-0.3801 (5)	-0.3401 (6)	-0.23341 (19)	0.106 (2)
H111	-0.3854	-0.2844	-0.2474	0.127*
N102	-0.3728 (5)	-0.4211 (5)	-0.25138 (15)	0.115 (2)
N121	0.7525 (2)	0.0070 (2)	0.16209 (8)	0.0428 (7)
C121	0.7446 (3)	-0.0039 (3)	0.19864 (11)	0.0541 (11)
H121	0.6849	-0.0061	0.2084	0.065*
C122	0.8207 (3)	-0.0121 (3)	0.22282 (12)	0.0602 (12)
H122	0.8114	-0.0200	0.2481	0.072*
C123	0.9097 (3)	-0.0085 (3)	0.20951 (12)	0.0548 (11)
C124	0.9175 (3)	0.0022 (4)	0.17212 (13)	0.0725 (15)
H124	0.9764	0.0049	0.1617	0.087*
C125	0.8391 (3)	0.0092 (4)	0.14988 (12)	0.0644 (13)
H125	0.8472	0.0159	0.1245	0.077*
C126	0.9966 (4)	-0.0124 (4)	0.23262 (12)	0.0684 (13)
H126	1.0524	-0.0152	0.2199	0.082*
N13	-0.0334 (3)	0.2265 (3)	-0.16706 (9)	0.0530 (9)
C13	0.0280 (3)	0.2525 (3)	-0.18490 (11)	0.0493 (10)
Se13	0.12347 (4)	0.29280 (5)	-0.211172 (17)	0.0860 (2)
N14	-0.2189 (2)	0.2092 (3)	-0.07297 (9)	0.0523 (8)
C14	-0.2774 (3)	0.2041 (3)	-0.05138 (11)	0.0475 (9)
Se14	-0.36807 (4)	0.19698 (5)	-0.018923 (14)	0.07474 (17)
N16	0.5330 (3)	0.0107 (3)	0.16991 (9)	0.0547 (9)
C16	0.4719 (3)	0.0262 (3)	0.18959 (11)	0.0503 (10)
Se16	0.37749 (4)	0.05023 (6)	0.219058 (15)	0.0860 (2)
N17	0.7137 (2)	0.0225 (3)	0.07427 (9)	0.0511 (8)
C17	0.7777 (3)	0.0433 (3)	0.05675 (11)	0.0459 (9)
Se17	0.87743 (3)	0.07271 (4)	0.030650 (14)	0.06694 (15)
C112	-0.3805 (5)	-0.3312 (6)	-0.19489 (18)	0.106 (2)
H112	-0.3846	-0.2712	-0.1839	0.128*

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Mn2	0.0391 (3)	0.0353 (3)	0.0365 (3)	0.0009 (2)	-0.0034 (2)	0.0015 (2)
Mn1	0.0385 (3)	0.0337 (3)	0.0368 (3)	0.0003 (2)	-0.0042 (2)	0.0031 (2)
N1	0.0491 (18)	0.0320 (16)	0.0449 (17)	0.0005 (14)	0.0006 (14)	0.0011 (14)
C1	0.058 (3)	0.033 (2)	0.048 (2)	0.0017 (17)	-0.0094 (19)	0.0031 (17)
C2	0.061 (3)	0.044 (2)	0.049 (2)	0.000 (2)	-0.0102 (19)	-0.0026 (18)
C3	0.048 (2)	0.034 (2)	0.051 (2)	-0.0010 (17)	0.0011 (18)	0.0024 (17)
C4	0.053 (2)	0.037 (2)	0.052 (2)	0.0006 (18)	-0.0081 (19)	0.0053 (18)

C5	0.057 (2)	0.036(2)	0.048 (2)	-0.0041 (18)	-0.0057 (19)	0.0028 (17)
C6	0.065 (3)	0.039 (2)	0.054 (2)	0.000 (2)	-0.009(2)	-0.0029 (19)
C7	0.068 (3)	0.039 (2)	0.051 (2)	-0.002(2)	-0.003(2)	-0.0034 (19)
C8	0.057 (2)	0.037 (2)	0.046 (2)	-0.0006 (18)	-0.0037 (18)	0.0013 (17)
C9	0.077 (3)	0.044 (2)	0.041 (2)	-0.001 (2)	-0.004 (2)	-0.0001 (18)
C10	0.077 (3)	0.039 (2)	0.042 (2)	-0.001(2)	-0.002(2)	0.0059 (18)
C11	0.081 (3)	0.038 (2)	0.046 (2)	0.000 (2)	-0.007(2)	0.0013 (18)
C12	0.079 (3)	0.035 (2)	0.047 (2)	-0.001 (2)	0.000 (2)	0.0063 (18)
N2	0.060 (2)	0.0342 (17)	0.0508 (19)	-0.0013 (15)	-0.0068 (16)	0.0038 (15)
N21	0.0422 (18)	0.0410 (18)	0.0431 (17)	0.0009 (14)	-0.0068 (14)	0.0045 (14)
C21	0.048 (3)	0.077 (3)	0.052 (2)	0.000 (2)	-0.004 (2)	0.005 (2)
C22	0.045 (3)	0.092 (4)	0.065 (3)	0.005 (2)	-0.008 (2)	0.004 (3)
C23	0.049 (2)	0.048 (3)	0.070 (3)	0.0023 (19)	-0.017 (2)	0.006 (2)
C24	0.071 (3)	0.077 (3)	0.044 (2)	-0.007 (3)	-0.017 (2)	0.014 (2)
C25	0.049 (2)	0.071 (3)	0.044 (2)	-0.001 (2)	-0.0067 (18)	0.007 (2)
C26	0.064 (3)	0.076 (3)	0.067 (3)	-0.011 (3)	-0.012 (2)	0.009 (3)
C27	0.063 (3)	0.086 (4)	0.061 (3)	0.002 (3)	-0.012 (2)	0.006 (3)
C28	0.049 (2)	0.057 (3)	0.062 (3)	0.003 (2)	-0.017 (2)	0.003 (2)
C29	0.045 (3)	0.151 (6)	0.060 (3)	0.017 (3)	-0.003 (2)	0.013 (3)
C30	0.047 (3)	0.124 (5)	0.050 (2)	0.010 (3)	-0.003 (2)	0.015 (3)
C31	0.047 (2)	0.070 (3)	0.046 (2)	-0.004 (2)	-0.0049 (18)	0.012 (2)
C32	0.072 (3)	0.069 (3)	0.045 (2)	-0.009 (2)	-0.017 (2)	0.017 (2)
N22	0.0393 (18)	0.050 (2)	0.0424 (17)	0.0013 (14)	-0.0058 (14)	0.0040 (15)
N41	0.0440 (18)	0.0382 (17)	0.0475 (18)	0.0006 (14)	-0.0004 (14)	0.0017 (14)
C41	0.057 (2)	0.038 (2)	0.047 (2)	0.0035 (18)	-0.0054 (18)	0.0042 (17)
C42	0.060 (3)	0.046 (2)	0.048 (2)	-0.002 (2)	-0.0099 (19)	-0.0002 (19)
C43	0.052 (2)	0.040 (2)	0.048 (2)	0.0006 (18)	0.0020 (18)	-0.0028 (18)
C44	0.058 (3)	0.037 (2)	0.060 (2)	-0.0020 (19)	-0.012 (2)	0.0058 (19)
C45	0.055 (2)	0.040 (2)	0.054 (2)	-0.0036 (19)	-0.0102 (19)	0.0029 (19)
C46	0.063 (3)	0.039 (2)	0.055 (2)	0.003 (2)	-0.008 (2)	-0.0049 (19)
C47	0.061 (3)	0.043 (2)	0.055 (2)	-0.002 (2)	0.000 (2)	-0.004 (2)
C48	0.054 (2)	0.032 (2)	0.053 (2)	0.0029 (17)	-0.0031 (19)	0.0003 (17)
C49	0.079 (3)	0.042 (2)	0.046 (2)	0.002 (2)	0.001 (2)	-0.0072 (19)
C50	0.077 (3)	0.042 (2)	0.040 (2)	-0.001 (2)	0.001 (2)	0.0044 (18)
C51	0.075 (3)	0.042 (2)	0.041 (2)	0.002 (2)	-0.004 (2)	-0.0037 (18)
C52	0.073 (3)	0.037 (2)	0.044 (2)	-0.002 (2)	0.000 (2)	0.0043 (17)
N42	0.059 (2)	0.0347 (17)	0.0454 (18)	-0.0002 (15)	-0.0045 (15)	-0.0010 (14)
N61	0.135 (5)	0.115 (5)	0.066 (3)	0.001 (4)	0.005 (3)	0.020 (3)
C61	0.159 (7)	0.083 (5)	0.090 (5)	-0.016 (5)	-0.005 (5)	0.021 (4)
C62	0.115 (5)	0.080 (4)	0.071 (4)	-0.012 (4)	-0.003 (3)	-0.001 (3)
C63	0.059 (3)	0.079 (4)	0.059 (3)	-0.001 (2)	0.003 (2)	0.004 (3)
C64	0.087 (4)	0.071 (3)	0.064 (3)	0.000 (3)	0.010 (3)	0.000 (3)
C65	0.111 (5)	0.100 (5)	0.071 (4)	0.009 (4)	0.009 (3)	-0.009 (3)
C66	0.066 (3)	0.073 (3)	0.064 (3)	-0.003 (3)	0.000 (2)	-0.005 (3)
C67	0.076 (3)	0.074 (4)	0.063 (3)	-0.002 (3)	-0.002 (2)	-0.005 (3)
C68	0.072 (3)	0.086 (4)	0.054 (3)	0.003 (3)	-0.002 (2)	-0.005 (3)
C69	0.149 (6)	0.085 (4)	0.065 (3)	0.003 (4)	-0.005 (4)	-0.002 (3)
C70	0.173 (8)	0.103 (6)	0.071 (4)	0.007 (5)	-0.004 (4)	-0.019 (4)

# supporting information

C71	0.191 (9)	0.095 (5)	0.072 (4)	0.010 (5)	0.003 (5)	0.015 (4)
C72	0.175 (7)	0.069 (4)	0.064 (3)	0.000 (4)	-0.007 (4)	0.001 (3)
N62	0.135 (5)	0.111 (5)	0.063 (3)	0.009 (4)	-0.002 (3)	0.002 (3)
N81	0.0441 (19)	0.046 (2)	0.0434 (17)	-0.0011 (15)	-0.0091 (14)	0.0050 (14)
C81	0.047 (3)	0.092 (4)	0.052 (2)	-0.002 (2)	-0.006 (2)	0.012 (2)
C82	0.045 (3)	0.117 (5)	0.065 (3)	-0.005 (3)	-0.011 (2)	0.016 (3)
C83	0.048 (3)	0.073 (3)	0.068 (3)	-0.005 (2)	-0.016 (2)	0.008 (2)
C84	0.072 (3)	0.074 (3)	0.044 (2)	-0.002 (3)	-0.015 (2)	0.005 (2)
C85	0.053 (3)	0.064 (3)	0.047 (2)	0.001 (2)	-0.0084 (19)	0.005 (2)
C86	0.066 (3)	0.110 (5)	0.063 (3)	-0.009 (3)	-0.013 (3)	0.012 (3)
N101	0.109 (4)	0.121 (5)	0.067 (3)	-0.005 (4)	0.000 (3)	-0.011 (3)
C101	0.101 (5)	0.119 (6)	0.081 (4)	0.013 (4)	0.009 (4)	0.025 (4)
C102	0.092 (4)	0.104 (5)	0.083 (4)	0.012 (4)	0.011 (3)	0.000 (4)
C103	0.058 (3)	0.115 (5)	0.063 (3)	-0.009 (3)	0.005 (2)	-0.006 (3)
C104	0.116 (5)	0.092 (5)	0.081 (4)	-0.002 (4)	0.000 (4)	0.011 (4)
C105	0.130 (6)	0.099 (5)	0.078 (4)	-0.002 (4)	0.002 (4)	-0.014 (4)
C106	0.083 (4)	0.131 (6)	0.076 (4)	-0.007 (4)	0.003 (3)	0.021 (4)
C107	0.074 (4)	0.148 (7)	0.093 (5)	-0.020 (4)	-0.004 (3)	0.025 (5)
C108	0.078 (4)	0.143 (7)	0.067 (4)	-0.019 (4)	-0.004 (3)	0.015 (4)
C109	0.164 (8)	0.118 (6)	0.082 (5)	-0.013 (5)	-0.018 (5)	0.050 (5)
C110	0.209 (10)	0.105 (6)	0.089 (5)	-0.010 (6)	-0.014 (6)	0.022 (4)
C111	0.119 (6)	0.111 (6)	0.086 (5)	-0.003 (5)	0.002 (4)	0.036 (4)
N102	0.152 (6)	0.119 (5)	0.073 (3)	0.003 (4)	-0.003 (3)	0.017 (4)
N121	0.0401 (18)	0.0459 (19)	0.0419 (17)	0.0007 (14)	-0.0073 (14)	-0.0015 (14)
C121	0.045 (2)	0.071 (3)	0.046 (2)	-0.001 (2)	-0.0069 (18)	0.003 (2)
C122	0.072 (3)	0.068 (3)	0.040 (2)	-0.002 (2)	-0.014 (2)	0.002 (2)
C123	0.045 (2)	0.061 (3)	0.058 (3)	0.004 (2)	-0.010 (2)	-0.003 (2)
C124	0.045 (3)	0.117 (5)	0.055 (3)	0.004 (3)	-0.006 (2)	-0.003 (3)
C125	0.047 (3)	0.098 (4)	0.048 (2)	0.003 (2)	-0.007 (2)	-0.001 (2)
C126	0.059 (3)	0.093 (4)	0.053 (2)	0.006 (3)	-0.006 (2)	-0.005 (2)
N13	0.051 (2)	0.063 (2)	0.0445 (18)	0.0014 (18)	0.0033 (16)	0.0029 (17)
C13	0.055 (3)	0.050 (2)	0.043 (2)	0.006 (2)	-0.0079 (19)	-0.0003 (18)
Se13	0.0671 (3)	0.1123 (5)	0.0793 (4)	-0.0190 (3)	0.0135 (3)	0.0173 (3)
N14	0.052 (2)	0.059 (2)	0.0462 (19)	-0.0037 (17)	0.0025 (16)	0.0048 (16)
C14	0.054 (2)	0.047 (2)	0.041 (2)	-0.0002 (19)	-0.0088 (19)	0.0046 (18)
Sel4	0.0579 (3)	0.1092 (5)	0.0576 (3)	0.0026 (3)	0.0133 (2)	0.0142 (3)
N16	0.053 (2)	0.067 (2)	0.0444 (19)	-0.0019 (18)	0.0019 (17)	0.0009 (17)
C16	0.053 (3)	0.054 (3)	0.043 (2)	-0.002 (2)	-0.010 (2)	0.0060 (19)
Se16	0.0584 (3)	0.1393 (6)	0.0605 (3)	0.0185 (3)	0.0091 (2)	-0.0018 (3)
N17	0.052 (2)	0.058 (2)	0.0436 (18)	0.0064 (17)	0.0037 (16)	0.0018 (16)
C17	0.055 (2)	0.041 (2)	0.041 (2)	0.0048 (19)	-0.0081 (19)	0.0016 (17)
Se17	0.0544 (3)	0.0778 (4)	0.0687 (3)	-0.0104 (2)	0.0053 (2)	0.0131 (3)
C112	0.113 (5)	0.129 (6)	0.078 (4)	-0.003 (5)	0.007 (4)	0.018 (4)

Geometric parameters (Å, °)

Mn2—N16	2.172 (4)	С52—Н52	0.9300
Mn2—N17	2.185 (3)	N42—Mn2 <sup>ii</sup>	2.293 (3)

$Mn2$ — $N42^{i}$	2.293 (3)	N61—C65	1.320 (8)
Mn2—N41	2.317 (3)	N61—C61	1.323 (9)
Mn2—N121	2.322 (3)	C61—C62	1.371 (9)
Mn2—N22	2.325 (3)	C61—H61	0.9300
Mn1—N14	2.176 (3)	C62—C63	1.380 (8)
Mn1—N13	2.183 (4)	C62—H62	0.9300
Mn1—N2 <sup>i</sup>	2.285 (3)	C63—C64	1.382 (7)
Mn1—N21	2.324(3)	C63—C66	1.467 (7)
Mn1—N1	2.338(3)	C64—C65	1.387(7)
Mn1—N81	2.330(3)	C64—H64	0.9300
N1—C5	1335(5)	C65—H65	0.9300
N1	1.335(5) 1 346(5)	C66—C67	1 297 (7)
$C_1$ $C_2$	1.340 (5)	C66 H66	0.9300
C1 H1	0.0300	C67 C68	1.471(7)
$C_1$	1 381 (6)	C67 H67	0.0300
$C_2 = C_3$	0.0200	C68 C72	1 264 (8)
$C_2 = H_2$	0.9300	C08 - C/2	1.304 (8)
$C_3 - C_4$	1.392 (5)	C68—C69	1.369 (8)
$C_3 - C_6$	1.470 (6)	C69—C70	1.383 (8)
C4—C5	1.377 (5)	C69—H69	0.9300
C4—H4	0.9300	C/0—N62	1.314 (9)
C5—H5	0.9300	C/0_H/0	0.9300
C6—C7	1.315 (6)	C71—N62	1.296 (9)
С6—Н6	0.9300	C71—C72	1.371 (8)
C7—C8	1.472 (6)	C71—H71	0.9300
С7—Н7	0.9300	С72—Н72	0.9300
C8—C9	1.383 (6)	N81—C81	1.335 (5)
C8—C12	1.389 (6)	N81—C85	1.335 (5)
C9—C10	1.373 (6)	C81—C82	1.366 (6)
С9—Н9	0.9300	C81—H81	0.9300
C10—N2	1.338 (5)	C82—C83	1.356 (7)
C10—H10	0.9300	C82—H82	0.9300
C11—N2	1.328 (5)	C83—C84	1.387 (7)
C11—C12	1.369 (6)	C83—C86	1.486 (7)
C11—H11	0.9300	C84—C85	1.386 (6)
C12—H12	0.9300	C84—H84	0.9300
N2—Mn1 <sup>ii</sup>	2.285 (3)	C85—H85	0.9300
N21—C25	1.332 (5)	C86—C86 <sup>iii</sup>	1.240 (10)
N21—C21	1.336 (5)	C86—H86	0.9300
C21—C22	1.368 (6)	N101—C105	1.323 (8)
C21—H21	0.9300	N101—C101	1.326 (9)
C22—C23	1.358 (6)	C101—C102	1.379 (8)
С22—Н22	0.9300	C101 - H101	0.9300
C23—C24	1.384(7)	C102 - C103	1 359 (9)
$C^{23}$ $C^{26}$	1.361 (7)	C102 - H102	0.9300
$C_{24}$ $C_{25}$	1 380 (6)	C102 - 1102	1 375 (0)
C24_H24	0.0300	C103 - C104	1.373(7) 1.488(8)
C25—H25	0.2300	C103 - C100	1 377 (8)
$C_{25}$ $C_{25}$ $C_{27}$	1 260 (7)	C104—H104	0 0300
$\nabla \Delta \nabla = \nabla \Delta \Delta T$	1.200 (/)	UIUT 1110T	0.7500

C26—H26	0.9300	C105—H105	0.9300
C27—C28	1.492 (6)	C106—C107	1.208 (9)
С27—Н27	0.9300	C106—H106	0.9300
C28—C32	1.362 (6)	C107—C108	1.522 (9)
C28—C29	1.370 (7)	С107—Н107	0.9300
C29—C30	1.367 (6)	C108—C112	1.341 (9)
C29—H29	0.9300	C108—C109	1.372 (10)
C30—N22	1.330 (5)	C109—C110	1.385 (10)
C30—H30	0.9300	С109—Н109	0.9300
C31—N22	1.328 (5)	C110—N102	1.302 (9)
C31—C32	1.388 (6)	C110—H110	0.9300
C31—H31	0.9300	C111—N102	1 310 (9)
C32—H32	0.9300	C111-C112	1 395 (9)
N41—C41	1 337 (5)	C111—H111	0.9300
N41—C45	1.345(5)	N121—C125	1 325 (5)
$C_{41}$ $C_{42}$	1 376 (6)	N121—C121	1.325(5) 1.334(5)
$C_{41} = C_{42}$	0.9300	$C_{121} = C_{121}$	1.334 (5)
C42 - C43	1 381 (6)	C121—C122	0.9300
$C_{42} = C_{43}$	0.9300	C122 - C123	1 372 (6)
$C_{42} = 1142$	1 306 (6)	C122 - C123	0.0300
$C_{43}$ $C_{46}$	1.390 (0)	C122 - C122	1 363 (6)
$C_{44}$ $C_{45}$	1.471 (0)	$C_{123} = C_{124}$	1.303 (6)
$C_{44} = C_{43}$	0.0300	C123 - C120	1.465 (6)
$C_{44}$ $H_{45}$	0.9300	C124 - C123	0.0300
$C_{45}$	1 318 (6)	$C_{124} = 11124$ $C_{125} = H_{125}$	0.9300
$C_{46} = U_{46}$	0.0300	$C125 - C126^{iv}$	1.256 (0)
C47 $C48$	1 469 (6)	C126 H126	0.0300
C47 - C48	0.0200	N12 C12	0.9300
$C_{4} = C_{4}$	0.9300	$\begin{array}{c} \text{N15} \\ \text{C13} \\ \text{Sa13} \end{array}$	1.138(3) 1.771(5)
$C_{40} = C_{49}$	1.379(0) 1.288(5)	N14 C14	1.771(3) 1.157(5)
C40 - C50	1.300(3) 1.281(6)	$\frac{14}{14}$	1.137(3)
$C_{49} = C_{50}$	0.0200	N16 C16	1.709 (4)
$C_{49} =$	0.9300	$\begin{array}{c} N10 - C10 \\ C16 - Sol6 \\ \end{array}$	1.138(3)
$C_{50}$ $H_{50}$	1.555 (5)	N17 C17	1.709(3)
C51 N42	0.9300	N1/-C1/	1.101(3)
$C_{51} = 1042$	1.343(3)		1.774 (4)
C51 - C52	1.578 (0)	С112—н112	0.9300
С51—н51	0.9300		
N16—Mn2—N17	179.89 (16)	C48—C49—C50	119.3 (4)
N16—Mn2—N42 <sup>i</sup>	90.39 (13)	C48—C49—H49	120.3
N17—Mn2—N42 <sup>i</sup>	89.53 (13)	С50—С49—Н49	120.3
N16—Mn2—N41	89.22 (13)	N42-C50-C49	124.1 (4)
N17—Mn2—N41	90.87 (12)	N42-C50-H50	117.9
N42 <sup>i</sup> —Mn2—N41	179.27 (12)	C49—C50—H50	117.9
N16—Mn2—N121	89.27 (12)	N42—C51—C52	123.9 (4)
N17—Mn2—N121	90.81 (12)	N42—C51—H51	118.1
N42 <sup>i</sup> —Mn2—N121	89.70 (12)	C52—C51—H51	118.1
N41—Mn2—N121	89 68 (11)	$C_{51}$ $C_{52}$ $C_{48}$	119.2 (4)
.,			

N16—Mn2—N22	89.90 (12)	С51—С52—Н52	120.4
N17—Mn2—N22	90.01 (12)	C48—C52—H52	120.4
N42 <sup>i</sup> —Mn2—N22	86.94 (12)	C50—N42—C51	116.0 (3)
N41—Mn2—N22	93.68 (12)	C50—N42—Mn2 <sup>ii</sup>	121.6 (3)
N121—Mn2—N22	176.53 (12)	C51—N42—Mn2 <sup>ii</sup>	122.4 (3)
N14—Mn1—N13	178.94 (14)	C65—N61—C61	114.7 (6)
$N14$ — $Mn1$ — $N2^{i}$	90.72 (13)	N61—C61—C62	124.8 (6)
N13—Mn1—N2 <sup>i</sup>	90.19 (13)	N61—C61—H61	117.6
N14—Mn1—N21	90.20 (12)	С62—С61—Н61	117.6
N13—Mn1—N21	90.37 (12)	C61—C62—C63	120.1 (6)
N2 <sup>i</sup> —Mn1—N21	88.85 (12)	C61—C62—H62	119.9
N14—Mn1—N1	88.46 (12)	С63—С62—Н62	119.9
N13—Mn1—N1	90.63 (13)	C62—C63—C64	116.2 (5)
$N2^{i}$ — $Mn1$ — $N1$	179.16 (12)	C62—C63—C66	119.1 (5)
N21—Mn1—N1	90.96 (11)	C64—C63—C66	124.6 (5)
N14—Mn1—N81	91.17 (12)	C63—C64—C65	118.7 (6)
N13—Mn1—N81	88.29 (12)	С63—С64—Н64	120.7
N2 <sup>i</sup> —Mn1—N81	89.37 (12)	С65—С64—Н64	120.7
N21—Mn1—N81	177.77 (11)	N61—C65—C64	125.4 (6)
N1—Mn1—N81	90.84 (11)	N61—C65—H65	117.3
C5—N1—C1	116.2 (3)	С64—С65—Н65	117.3
C5—N1—Mn1	122.8 (3)	C67—C66—C63	126.9 (5)
C1—N1—Mn1	121.0 (2)	С67—С66—Н66	116.6
N1—C1—C2	123.3 (4)	С63—С66—Н66	116.6
N1—C1—H1	118.4	C66—C67—C68	126.3 (5)
C2—C1—H1	118.4	С66—С67—Н67	116.9
C1—C2—C3	120.2 (4)	С68—С67—Н67	116.9
C1—C2—H2	119.9	C72—C68—C69	115.3 (5)
С3—С2—Н2	119.9	C72—C68—C67	120.5 (5)
C2—C3—C4	116.7 (4)	C69—C68—C67	124.3 (5)
C2—C3—C6	119.9 (4)	C68—C69—C70	120.1 (6)
C4—C3—C6	123.4 (4)	С68—С69—Н69	120.0
C5—C4—C3	119.6 (4)	С70—С69—Н69	120.0
С5—С4—Н4	120.2	N62—C70—C69	124.2 (7)
C3—C4—H4	120.2	N62—C70—H70	117.9
N1-C5-C4	124.0 (4)	С69—С70—Н70	117.9
N1—C5—H5	118.0	N62—C71—C72	125.3 (7)
С4—С5—Н5	118.0	N62—C71—H71	117.4
C7—C6—C3	125.2 (4)	С72—С71—Н71	117.4
С7—С6—Н6	117.4	C68—C72—C71	120.2 (6)
С3—С6—Н6	117.4	С68—С72—Н72	119.9
C6—C7—C8	125.8 (4)	С71—С72—Н72	119.9
С6—С7—Н7	117.1	C71—N62—C70	115.0 (6)
С8—С7—Н7	117.1	C81—N81—C85	115.6 (4)
C9—C8—C12	117.1 (4)	C81—N81—Mn1	120.7 (3)
C9—C8—C7	120.8 (4)	C85—N81—Mn1	123.6 (3)
C12—C8—C7	122.1 (4)	N81—C81—C82	123.9 (4)
С10—С9—С8	119.8 (4)	N81—C81—H81	118.0

С10—С9—Н9	120.1	C82—C81—H81	118.0
С8—С9—Н9	120.1	C83—C82—C81	121.2 (5)
N2—C10—C9	123.2 (4)	С83—С82—Н82	119.4
N2-C10-H10	118.4	С81—С82—Н82	119.4
С9—С10—Н10	118.4	C82—C83—C84	116.1 (4)
N2-C11-C12	124.3 (4)	C82—C83—C86	118.8 (5)
N2—C11—H11	117.9	C84—C83—C86	125.2 (5)
C12—C11—H11	117.9	C85—C84—C83	119.9 (4)
C11—C12—C8	119.1 (4)	С85—С84—Н84	120.1
C11—C12—H12	120.5	С83—С84—Н84	120.1
C8—C12—H12	120.5	N81—C85—C84	123.4 (4)
C11—N2—C10	116.5 (3)	N81—C85—H85	118.3
C11—N2—Mn1 <sup>ii</sup>	122.3 (3)	С84—С85—Н85	118.3
C10—N2—Mn1 <sup>ii</sup>	121.1 (3)	C86 <sup>iii</sup> —C86—C83	127.2 (7)
C25—N21—C21	115.7 (3)	C86 <sup>iii</sup> —C86—H86	116.4
C25—N21—Mn1	122.1 (3)	С83—С86—Н86	116.4
$C_{21} = N_{21} = M_{11}$	122.1 (3)	C105—N101—C101	116.3 (6)
$N_{21} - C_{21} - C_{22}$	123.7 (4)	N101—C101—C102	124.1 (7)
N21—C21—H21	118.1	N101—C101—H101	118.0
$C_{22} = C_{21} = H_{21}$	118.1	C102 - C101 - H101	118.0
$C_{23}$ $C_{22}$ $C_{21}$ $C$	121.1 (5)	C102 - C101 - C101	119.0
$C_{23}$ $C_{22}$ $C_{21}$ $C_{23}$ $C_{22}$ $H_{22}$	119 5	C103 - C102 - H102	120.3
$C_{23} = C_{22} = H_{22}$	119.5	C101 - C102 - H102	120.3
$C_{21} = C_{22} = 1122$	115.5 115.7(4)	$C_{101} = C_{102} = 11102$	120.5 117.1(5)
$C_{22} = C_{23} = C_{24}$	113.7(4) 118.2(4)	C102 - C103 - C104	117.1(5) 118.1(6)
$C_{22} = C_{23} = C_{20}$	116.2 (4) 126.0 (4)	C102 - C103 - C106	124.8(6)
$C_{24} = C_{23} = C_{20}$	120.0(4)	$C_{104} = C_{103} = C_{100}$	124.0(0)
$C_{25} = C_{24} = C_{25}$	120.4 (4)	$C_{103} = C_{104} = C_{103}$	120.1 (0)
$C_{23} = C_{24} = H_{24}$	119.0	$C_{105} = C_{104} = H_{104}$	120.0
N21 C25 C24	119.0 122.2(4)	103 - 104 - 1104	120.0 122.1(7)
$N_{21} = C_{23} = C_{24}$	123.2 (4)	N101 - C105 - C104	123.1 (7)
$N_{21} = C_{23} = H_{23}$	110.4	N101 - C105 - H105	110.5
$C_{24} = C_{23} = H_{23}$	116.4	C104 - C105 - H103	110.5
$C_{27} = C_{20} = C_{23}$	123.7 (3)	C107 - C106 - C103	127.7 (8)
$C_2/-C_{20}-H_{20}$	117.1	C107 - C106 - H106	116.2
$C_{23}$ $C_{26}$ $C_{126}$ $C_{126$	117.1	C103 - C106 - H106	116.2
$C_{26} = C_{27} = C_{28}$	125.8 (5)	C106 - C107 - C108	127.2 (9)
$C_{26} = C_{27} = H_{27}$	117.1	C106—C107—H107	116.4
C28—C27—H27	117.1	C108—C107—H107	116.4
C32—C28—C29	115.8 (4)	C112—C108—C109	116.6 (6)
C32—C28—C27	126.2 (4)	C112—C108—C107	118.1 (8)
C29—C28—C27	118.0 (4)	C109—C108—C107	125.3 (7)
C30—C29—C28	120.8 (5)	C108—C109—C110	119.9 (7)
C30—C29—H29	119.6	C108—C109—H109	120.0
C28—C29—H29	119.6	С110—С109—Н109	120.0
N22—C30—C29	123.8 (4)	N102—C110—C109	124.1 (8)
N22—C30—H30	118.1	N102—C110—H110	117.9
C29—C30—H30	118.1	C109—C110—H110	117.9
N22—C31—C32	123.0 (4)	N102—C111—C112	124.9 (7)

N22—C31—H31	118.5	N102—C111—H111	117.6
С32—С31—Н31	118.5	C112—C111—H111	117.6
C28—C32—C31	120.8 (4)	C110—N102—C111	115.2 (6)
С28—С32—Н32	119.6	C125—N121—C121	115.5 (3)
С31—С32—Н32	119.6	C125—N121—Mn2	122.2 (3)
C31—N22—C30	115.8 (4)	C121—N121—Mn2	122.3 (3)
C31—N22—Mn2	121.5 (3)	N121—C121—C122	123.3 (4)
C30—N22—Mn2	122.0 (3)	N121—C121—H121	118.3
C41—N41—C45	116.2 (3)	C122—C121—H121	118.3
C41—N41—Mn2	120.7 (3)	C123—C122—C121	120.0 (4)
C45—N41—Mn2	123.1 (3)	C123—C122—H122	120.0
N41—C41—C42	123.5 (4)	C121—C122—H122	120.0
N41—C41—H41	118.2	C124—C123—C122	116.5 (4)
C42—C41—H41	118.2	C124—C123—C126	118.3 (4)
C41—C42—C43	120.4 (4)	C122—C123—C126	125.1 (4)
C41—C42—H42	119.8	C123—C124—C125	120.3 (5)
C43—C42—H42	119.8	C123—C124—H124	119.9
C42—C43—C44	116.5 (4)	C125—C124—H124	119.9
C42—C43—C46	121.1 (4)	N121—C125—C124	124.4 (4)
C44—C43—C46	122.5 (4)	N121—C125—H125	117.8
C45—C44—C43	119.6 (4)	C124—C125—H125	117.8
C45—C44—H44	120.2	C126 <sup>iv</sup> —C126—C123	127.4 (6)
C43—C44—H44	120.2	C126 <sup>iv</sup> —C126—H126	116.3
N41—C45—C44	123.8 (4)	C123—C126—H126	116.3
N41—C45—H45	118.1	C13—N13—Mn1	158.1 (3)
C44—C45—H45	118.1	N13-C13-Se13	178.6 (4)
C47—C46—C43	124.8 (4)	C14—N14—Mn1	171.5 (3)
C47—C46—H46	117.6	N14-C14-Se14	179.1 (4)
C43—C46—H46	117.6	C16—N16—Mn2	161.2 (3)
C46—C47—C48	125.7 (4)	N16-C16-Se16	179.1 (4)
C46—C47—H47	117.1	C17—N17—Mn2	158.2 (3)
C48—C47—H47	117.1	N17—C17—Se17	178.4 (4)
C49—C48—C52	117.4 (4)	C108—C112—C111	119.3 (8)
C49—C48—C47	120.0 (4)	C108—C112—H112	120.4
C52—C48—C47	122.5 (4)	C111—C112—H112	120.4

Symmetry codes: (i) *x*, *y*+1, *z*; (ii) *x*, *y*-1, *z*; (iii) -*x*-1, *y*, -*z*-1/2; (iv) -*x*+2, *y*, -*z*+1/2.